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Subject

Production of Coated Detergent Enzyme Granules

Telephone
015-2855276
015-2573055
015-2782376
015-2783568
015-2784515

### Keywords

Enzyme, protease, fermentation, membrane separation, high shear granulation, fluidized bed drying, fluidized bed coating

Assignment issued	:	1 <sup>st</sup> April 1999
Report issued	:	2 <sup>nd</sup> July 1999
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Faculteit Technische Natuurwetenschappen Scheikundige Technologie en Materiaalkunde

# CPD No. 3237

## Conceptual Process Design Chemical Process Technology

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Authors	Telephone
C. P. Almeida-Rivera	015-2855276
P. Du	015-2573055
R. G. Padios	015-2782376
R. Pitchumani	015-2783568
J. B. Taboada	015-2784515

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## **Conceptual Process Design 3237**

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Kick off :	April, 1 <sup>st</sup> 1999
Report :	July, 2 <sup>nd</sup> 1999

Authors:	C. P. Almeida-Rivera
	P. Du
	R. G. Padios
	R. Pitchumani
	J. B. Taboada

Mentor:	dr.	ir.	G.	M.	H.	Meesters

Coordinators:

ir. C. P. Luteijn prof. ir. J. Grievink





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We hope everybody who reads this report will find it interesting and that it gives new and comprehensible concepts about the topic. The authors have been trying their very best to make this report a useful tool in the future. In this regard, we would like to acknowledge and thank the following people who unselfishly helped us throughout the three months of work:

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Cristhian Almeida-Rivera	Ecuador
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Ricardo G. Padios	Philippines
Ramanan Pitchumani	India
Jerome B. Taboada	Philippines

## **SUMMARY**

In recent times, the use of enzymes has been more and more common. Several industries like the starch industry, the dairy industry, and the detergent industry, are already using enzymes. In the detergent industry, for example, all enzymes customary for detergent and cleaning purposes are processed into enzyme granules.

In this project, the objective is to process enzyme-coated granules. The design of the process has a capacity of 5,000 tons per year of enzyme granules with an activity of  $2.3 \times 10^6$  units per gram of granules, density (bulk) of 1011.32 kg/m<sup>3</sup> and a desired particle size range from 250 µm to 1000 µm.

For the enzyme production, specific microorganisms like *Bacillus licheniformis* and *Bacillus amyloliquefaciens* and the corresponding complex medium which is mainly composed of glucose, ammonium source, trace elements, are charged in a designed batch bioreactor to produce the alkaline serine protease enzyme. After twenty hours (20 hrs) of batch fermentation time, 487.40 tons enzyme is taken out and purified using a membrane separator and concentrated to yield a concentrated enzyme solution with an activity of 500,000 units per gram enzyme. This concentrated enzyme solution containing 85.98% enzyme is used to produce the enzyme-coated granule. Whereas, 482.92 tons per batch of the by-product, mainly composed of biomass and unreacted substances from the reactor is sent to the appropriate waste-treatment plant, which is already outside the battery limit of the plant.

The Builder is sent through a Lödige Mixer granulator and is sprayed with the concentrated enzyme solution together with the Binder consisting of PEG 6000 and PVP 30K (Poly Ethylene Glycol and Polyvinyl Pyrrolidilone respectively). The throughput of the granulator is sent to the fluidized bed dryer, where excess moisture from the granule is removed. To meet the particle size specification, the granules are sieved. The over-sized particles are passed through a roll mill before sending it back to the granulator; while the fine particles are sent directly back to the granulator. Finally, the granules are coated with a Coating consisting of PEG 6000, TiO<sub>2</sub> and Magnesium Silicate, in the fluidized bed coating equipment to produce an enzyme-coated granule with an activity mentioned above

This report includes a detailed definition of the process, as well as the process options. For the process streams, process conditions like pressure, temperature and other stream properties are discussed. Also included are the properties and the value of the mass flow of components in the stream.

In chapter 3, all basic assumptions are stated. This includes the capacity of the plant, the location, the battery limit, and the definition of all in-and-out going streams/substances passing through the battery limits. For the plant capacity, the feedstock, the products, and the by-products are described.

Lastly, the economic potential of the designed project is evaluated.

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## **1 INTRODUCTION**

Gross world sales for enzymes in 1987 was US \$ 450 M. The following enzymes are currently produced commercially:

- 1. Enzymes used in industry such as, amylases, proteases, catalases, etc.
- 2. Enzymes for analytical purposes such as glucose oxidase, cholesterol oxidase, etc.
- 3. Enzymes used in medicine such as asparaginase, lipases, proteases, etc.

These areas of application require varying levels of quality and quantity. On the tonnage basis, the most important are the industrial enzymes. The United States and Western Europe produce about 40-45% of the world market, and nine firms account for 90% of the total market. Table 1 and Figure 1 show the production of industrial enzymes and percentage of industrial enzymes used in various markets

Enzyme	Enzyme preparation Tons/year	Sales, % of total
Bacillus proteases	6200	45
Glucoamylases	4500	13
Bacillus amylase	4200	5
Glucose isomerase	1900	6
Rennin (microbial)	1500	10
Amylases (fungal)	800	4
Pectinases	100	3
Protease (fungal)	100	2
Others	<b>i</b>	12







Percentage of Industrial Enzymes used in various markets

Legend

Detergent enzymes are produced in large amounts (1300-1500 tons/year) and make up about 40% of the selling price (US \$ 300 M) of industrial enzymes. By 1969, 80% of all laundry detergents contained enzymes. Today, 95% of the laundry detergents in Europe contain the enzyme protease although such detergents are much less widely used in the United States.

Other areas in which proteases are used include the pharmaceutical industry, the leather industry, the manufacture of protein hydrolysates, the food industry, the film industry, and waste processing.

The detergent enzymes are the alkaline serine protease Subtilisin Carlsberg formed by Bacillus licheniformis and protease Subtilisin NOVO by Bacillus amyloliquefaciens. They have a pH optimum of 8-11 and a temperature optimum of 80°C. In this design project, only B. licheniformis is mainly dealt with.

Bacillus licheniformis is a facultative aerobic bacterium used in the fermentation industry to produce exocellular enzymes, among which the alkaline serine protease is the most abundant. However, in spite of the industrial importance of the enzyme, published literature on the production using a complex medium is limited.

The media used in the cultivation of microorganisms must contain all elements in a form suitable for synthesis of cell substances and for the production of metabolic products. Although medium preparation seems to be a simple step, it is essential to consider that careless preparation of media may cause complete failure of the process productivity. In industrial fermentations, complex, almost undefinable substances are frequently used for economic reasons. Depending on the particular process, 25% to 75% of the total cost of fermentation may be due to the carbohydrates source. The typical complex media composition is found in Chapter 3.

The process options and selection for the production of enzyme-coated granules is fully described in Chapters 2 and 5. The batch fermentation process is preferred over the fed-batch due to a higher enzyme activity achieved, on the other hand, maintaining a sterile condition is difficult in a continuous process. For the primary separation of the products of the bioreactor, a continuous rotary vacuum drum filter is used for relatively simple operation compared to continuous centrifuge which operation involves too many operating variables. There are many available processes for the concentration of the enzyme product, ultrafiltration using a hollow fiber is the most economical and other equally important reasons are cited in the above-mentioned chapters.

For this design project, the Lödige high shear mixer has been chosen for granulating the enzyme. It is a batch granulation system which has an operating range of 300-2000 kg/h.

The granules from the granulator still contain high moisture due to the mixing of the ultra-filtrate. It is difficult to handle and transport. The fluidized bed dryer is excellent to use for removal of excess moisture. One major advantage of these types of dryers is the close control of conditions so that a predetermined amount of free moisture may be left with the solids to prevent dusting of the product during subsequent material handling operations.

The desired product should have a size range from  $250 \ \mu m$  to  $1000 \ \mu m$ . Industrial screens have to be used to classify the product into the desired size class. The oversize and undersize are re-processed to meet the product size range. This increases the throughput of the granulator and hence the feed has to be reduced depending on its own performance i.e. the amount of over size and under size produced.

Coating of enzyme granules is important in order to protect the enzyme from the environment. In this project a fluid bed coating system will be used to coat polyethylene glycol (PEG 6000) on the enzyme granules. Fluid bed equipment has been found to be useful for forming and coating granules particularly smaller granules which are difficult to handle by other methods. The dust is also controlled by placing a coating around the granules which binds fines to larger particles or prevents attrition of the granule.

Ecological and safety aspects are also important in this design project. Based on literature survey and pathogenity tests, the microorganism used is non-toxic. However, during the cultivation process, contamination tests are regularly performed and contact with the medium is avoided since the enzymes cause allergic reaction to humans. For handling of raw materials, products and wastes, the usual safety and disposal regulations hold true. More details on this matter are presented in Chapters 9 and 10.

The design project is also limited by the availability of data. Since most the processes are proprietary in nature, some data are not available for use in this design. But somehow, this limitation is abated by the use of the Super Pro 2.7 simulation package for the bioreactor design and initial design process of the whole plant, e.g. material and energy balances. Also, scaling-up to industrial scale processes successfully developed at the laboratory scale is not a simple procedure especially when dealing with biochemical reactions, this is not dealt with. In addition, the separation of the specific enzyme from the enzyme mixture is not within the scope of this design project and lastly, optimization of the process parameters (e.g. medium composition) are not dealt with. Detailed discussion of the process structure and description is presented in Chapter 5.

# **2 PROCESS OPTIONS AND SELECTION**

The production of coated detergent enzyme granules is composed of the following operations:

#### 2.1 ENZYME PRODUCTION

Proteases are one of the most important industrial enzymes currently in industrial production. Proteases, having molecular weights ranging from 25000 - 30000, are produced commercially both from bacteria and fungi and are stable up to a temperature of  $65^{\circ}$ C.

Screening of the microorganism should be done to assure its stability under alkaline conditions. Strains of B. licheniformis and B. subtilis showed optimal growth in the pH range 6-7, but the new genetically modified strains can grow at pH 8-11. Protein engineered Bacillus has improved properties such as substrate specificity, pH optimum, and stability to bleaching agents. At their optimum pH range, at least 90% of the maximal growth rate is attained.

The detergent enzymes are alkaline serine protease Subtilisin Carlsberg formed by *Bacillus licheniformis* and protease Subtilisin NOVO<sup>TM</sup> by *Bacillus amyloliquefaciens*. The proteases of this type have many features of value for use as detergent enzymes:

- stability at high temperature (maximum 60°C)
- stability in the alkaline range (pH 9-11)
- stability in association with chelating agents and perborates, however, their stability in the presence of surface-active agent is low, thus limiting their shelf life.

The cultivation with complex medium used in the industry are yielding protease activity of  $0.5 \times 10^9$  U/kg<sub>cells</sub>. The concentration of cells is 30 kg<sub>cells</sub>/m<sup>3</sup><sub>liquid</sub>.

The enzyme production can be carried out in three modalities:

#### 2.1.1 Batch process

A batch fermentation is considered as a closed system. At t=0 sec., the sterilized nutrient solution in the fermenter is inoculated with the microorganisms and incubation proceeds under optimal physiological conditions. During the fermentation, only oxygen (air), an antifoam agent and acid or base to control pH are added. The composition of the culture medium, the biomass concentration, and the metabolite concentration generally change constantly as a result of the metabolism of the cells.

#### 2.1.2 Fed-Batch process

The fed-batch process is the modification of the conventional batch process. In this process, the substrate is added in increments as the fermentation progresses. The formation of many secondary metabolites is subject to catabolic suppression by high concentrations of glucose, other carbohydrates, or nitrogen compounds. So, in the fed-batch system the critical elements of the nutrient solution are added in small concentrations at the start of the fermentation and are continuously added in small doses during the production phase. Experiments showed that the biosynthesis of the alkaline serine protease is dependent on the growth rate of the culture and concentration of nitrogen and carbon sources in the medium. Ammonium was reported to inhibit protease production. The experiments showed that in the fed-batch fermentation a constant low level of ammonium concentration should be maintained for optimum protease production. In a fed-batch procedure, production could be increased 4.6-fold in comparison to an uncontrolled batch process.

#### 2.1.3 Continuous process

Continuous process is considered as an open system. Sterile nutrient solution is added to the bioreactor continuously and an equivalent amount of converted nutrient solution with microorganisms is withdrawn out of the system. The reactor could be a homogenously mixed bioreactor or a plug flow type. A homogenously mixed reactor is either a chemostat where the cell growth is controlled by adjusting the concentration of one substrate (limiting factor) or a turbistat, cell growth is maintained by using turbidity to monitor the biomass concentration and the rate of feed of the nutrient solution is properly adjusted.

The batch and the fed-batch processes are generally used in the fermentation process. The latter is employed to keep down the concentration of ammonium ions and amino acids, since these nitrogenous materials repress protease

production. But on the other hand, the enzyme activity achieved in a fed-batch process is two times lower than in a batch process. Although continuous processes have been described, they are not commercially used. For this design project, the batch process will be used since many published data are available.

The following reactions take place in the bioreactor:

(1) $1.10 C_6 H_{12} O_6 + 1.2 N H_3$	$\rightarrow$	$C_6H_{10.8}N_{1.2}O_{3.0} + 2.4 H_2O + 0.6 CO_2$	enzyme formation
$(2) \ 1.03 \ C_6 H_{12} O_6 \ + \ 1.74 \ NH_3$	$\rightarrow$	$C_6H_{9.48}N_{1.74}O_{2.04} + 0.18CO_2 + 3.78H_2O$	biomass formation
$(3) \qquad C_6 H_{12} O_6 + 6 O_2$	$\rightarrow$	$6 CO_2 + 6 H_2O$	
$(4) \ 3.13 \ C_6 H_{12} O_6 + 2.94 N H_3 + 6 O_2$	$\rightarrow$	$C_6H_{10.8}N_{1.2}O_{3.0} + C_6H_{9.48}N_{1.74}O_{2.04} + 6.18 H_2$	$_{2}O + 6.78 CO_{2}$

Where  $C_6H_{12}O_6$  is the glucose,  $C_6H_{10.8}N_{1.2}O_{3.0}$  the protein and  $C_6H_{9.48}N_{1.74}O_{2.04}$  the biomass.

#### 2.2 SEPARATION

The first downstream step in a biological process is usually the removal of suspended cell mass and other particulate/colloidal debris from the suspending medium. The concentration of extra-cellular enzymes in typical biotechnological processes is 0.5-1% (w/w).

In the past, the preferred methods of cell harvesting were filtration or centrifugation. To be economical on an industrial scale, the method must be a continuous process. Continuous centrifuges, where the cells are removed by the action of high G forces, are expensive to buy, operate, and maintain. The production rate of a centrifuge is a function of many variables, among them the square of the particle diameter, the difference in density between the particles and the suspending medium, the G force and the viscosity of the suspending medium. The production rate and the cell recovery (yield) are strongly dependent on particle size and inversely proportional to each other. In addition, there may be substantial heat generation requiring complicated cooling system; there may be production of aerosols, and the capacity of most tubular centrifuges is limited and may require frequent teardowns and cleaning (Cheryan, 1998).

Filtration, on the other hand, is not limited in throughput by the particle size, and the separation rate does not go down as smaller microorganisms are harvested. The most common method is rotary drum filters, where a vacuum draws the broth to the drum surface. The microbial cells and other insoluble solids accumulate on the filter cloth and are scraped off with a knife.

#### 2.3 CONCENTRATION

Many traditional cell-harvesting processes are being replaced by cross-flow membrane filtration. With proper selection of the membrane, the module, and operating parameters, 100% cell recovery can be obtained, and capacity can be increased by carefully adding more membrane area.

Over 130 materials have been used to manufacture membranes but only a few have achieved commercial status and fewer still have obtained regulatory approval for use in food, pharmaceutical, and kindred applications. The family of polysulfone membranes has been chosen in the concentration step of this process due principally to the following favorable characteristics:

- Wide temperature limits: typically, temperatures up to 75°C
- Wide pH tolerances: these membranes can be continuously exposed to pHs from 1 to 13
- Fairly good chlorine resistance: up to 200 ppm chlorine for cleaning and up to 50 ppm chlorine for short storage of the membrane
- Easy to fabricate membranes in a wide variety of configurations and modules
- Wide range of pore sizes available for ultrafiltration applications, ranging from 10A° (1000 MWCO) to 0.2μm in commercial-size modules.
- Good chemical resistance to aliphatic hydrocarbons, fully halogenated hydrocarbons, alcohols and acids.

Membrane technology presents six basic designs of equipment: (1) tubular, with inner channel diameters>4 mm; (2) hollow fibers, with inner diameters of 0.2-3 mm; (3) plate units; (4) spiral-wound modules; (5) pleated-sheet cartridges; and (6) rotary modules.

A hollow fiber design has been chosen due to the following characteristics:

• It is one of the more economical designs in terms of energy consumption because of combination of pressure drops and flow rates. Operating velocity of 0.5-2.5 m/s (Reynolds numbers of 500-3000) and pressure drops of typically 1.35-2.35 bara, depending on flow rate, fiber diameters and lengths.

- Hollow fibers have the highest surface area-to-volume ratio among the modules. Holdup volume is low.
- A big advantage with hollow fibers is its back-flushing capability, which improves its cleanability. This is possible because the fibers are self-supporting.

Though ultrafiltration is the most economical method for concentrating the filtrate, it cannot achieve very high concentrations (~99%) due to reduction in efficiency as the concentration gradient is very low and the osmotic pressure becomes low. More membrane systems are required in order to achieve such high concentrations.

Another possibility could be freeze concentration, which is done by freezing the ultra filtrate and then the ice formed on the surface of the liquid is scraped off. The refrigeration of the ultra filtrate is energy intensive compared to evaporation.

#### 2.4 GRANULATION

The protease rich ultra filtrate can be spray dried and introduced into the detergent in powdered form (mean particle size 0.3 mm with considerable amount of small particles) (Kadam, 1991). The dust in the environment has high protease concentration and can cause allergenic reactions in workers handling these products. Hence the proteases are an environment hazard in the powdered form. In order to eliminate this hazard, the protease rich ultrafiltrate can be mixed with Builder and Binder and granulated using a granulator. In industry, granulators are bought as there are standard designs. For the present project, Lödige High Shear Mixer/Granulator has been chosen for granulating the enzyme. It is a batch granulation system where the Builder, Binder and Ultra filtrate are mixed under high shear conditions leading to the formation of granules

#### 2.5 FLUIDIZED BED DRYING

The granule formed in the granulator has high moisture content due to the mixing of the ultra-filtrate. The wet granules are difficult during handling and transportation. Drying reduces the cohesive forces between the individual granules enabling ease in handling. The best and most widely used dryer in industry is the fluidized bed dryer. The reason being that heat transfer is very rapid due to the convection currents formed around each individual granule in the bed. Another major advantage of these type of dryers is the close control of conditions so that a predetermined amount of free moisture may be left with the solids to prevent dusting of the product during subsequent material handling operations.

#### 2.6 SCREENING, CRUSHER AND RECYCLE

The product stream has a wide size distribution. The desired product should have a size range from 250  $\mu$ m to 1000  $\mu$ m. Industrial screens have to be used to classify the product into the desired size class. The over size i.e. the size above 1000  $\mu$ m has to be crushed with a crusher and sent back to the granulator and the under size i.e the size class below 250  $\mu$ m has to be sent directly to the granulator. This increases the throughput of the granulator and hence the feed to the granulator has to be reduced depending on its own performance i.e. the amount of over size and under size produced.

#### 2.7 FLUIDIZED BED COATING

Coating of enzyme granules is important in order to protect the enzyme from the environment. The Coating is usually a mixture of polymer (PEG 6000) along with  $TiO_2$  and Magnesium Silicate which effectively isolates the enzyme from the environment. One can also control the dust by placing a coating around the granules which binds fines to larger particles, or prevents attrition of the granule. In this project a fluid bed coating system will be used to coat the Coating on the enzyme granule. Fluid bed equipment has been found to be useful for forming and coating granules particularly smaller granules which are difficult to handle by other methods. They are designed to keep even small particles in the processing chamber where they may become coated.

#### 2.8 SELECTED OPTION

#### **2.8.1 Description of the process**

For this project the components shown in all the process flow schemes and calculations consist of the following subcomponents:

- 1) Ammonium phosphate (Di-ammonium hydrogen phosphate)
- 2) Biomass (inoculum or micro-organism which produces the enzyme)
- 3) Glucose

- 4) Inerts
  - a) Cellulose
  - b) Insoluble salts
  - c) Antifoam Ke 211
- 5) Nutrients
  - a) Magnesium sulfate heptahydrate
  - b) Manganese sulfate monohydrate
  - c) Di-sodium hydrogen phosphate dodecahydrate
  - d) Potassium di-hydrogen phosphate
  - e) Ferrous sulfate heptahydrate
- 6) Builder
  - a) Cellulose powder
  - b) Sodium sulphate (filler)
  - c) Yellow dextrine
  - d) Calcium acetate
  - e)  $TiO_2$
- 7) Binder
  - a) PEG 6000 (Polyethylene Glycol, MW : 6000)
  - b) PVP K30 (Polyvinyl Pyrolidone, MW : 30,000)
- 8) Coating
  - a) PEG 6000 (Polyethylene Glycol, MW : 6000)
  - b) TiO<sub>2</sub>
  - c) Magnesium silicate

The process starts with the 1 hour of filling of substrate and water in the blending tank (T-101). The mixing in the blending tank occurs for 5 hours. The contents of the blending tank are then discharged into the sterilizer (E-101) which takes about 30 minutes. The mixture is sterilized at 121°C for 15 minutes with steam. The sterilized stream is then cooled to cultivation temperature. The cooling process takes about 15 minutes. The mixture from the sterilizer is fed to the fermentor (R-101) along with another stream consisting of the biomass which runs for 20 hours. The end point of the process is when the maximum enzyme activity is reached which is usually after 20 hours of fermentation time, which is shown in Appendix. The oxygen for the fermentation is supplied using a compressor (K-101) which blows ambient air through an Air Filter (S-101) for the removal of impurities and the pure air (devoid of impurities like dust etc.) is sent through the fermenter. After the fermentation, the broth is cooled down to 5°C in the cooler (E-102) using Brine solution as a cooling agent and transferred to the separation and concentration units.

The separation unit consists of a rotary drum vacuum filter (S-102), where a vacuum draws the broth to the drum surface. The microbial cells and other insoluble suspended solids accumulate on the filter cloth and are scraped off with a knife. The filtrate from the rotary drum vacuum filter is sent to the Ultra Filter (S-104) for concentration.

A hollow fiber design has been chosen for the next operation and has the following characteristics

 Table 2. Characteristics of the selected concentration equipment

 PARAMETER
 VALUE

 Membrane design
 Hollow fibers

IMANIDIDA	VALUE
Membrane design	Hollow fibers
Membrane material	Polysulfone
Pore Size [A°]	200
Fiber diameter [mm]	0.5
Fiber length [m]	0.1
Temperature and pressure rating	80°C – 1 atm
pH value	8-11
Manufacturer	A/G Technology

For the present project, Lödige High Shear Mixer (G-101) has been chosen for granulating the enzyme. It is a batch granulation system which consists of a cylindrical vessel having plow-shaped mixing tools mounted on a horizontal shaft rotating at very high speeds of 3000 rpm. At the bottom of the vessel, there are choppers which break the large granules into smaller sizes in order to maintain a narrow size distribution. At first the Builder is fed into the High Shear Mixer and then the Ultrafiltrate from the Ultrafilter along with Binder. The contents are mixed under high shear condition for 10 minutes and discharged to the fluidized bed dryer.

Afterwards, the granules are dried in fluidized bed dryer (D-101), because the heat transfer is very rapid due to the convection currents formed around each individual granule in the bed. Another major advantage of these type of dryers

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is the close control of conditions so that a predetermined amount of free moisture may be left with the solids to prevent dusting of the product during subsequent material handling operations. The drying process takes about 30 minutes. The dried granules have a wide size distribution and industrial screens (S-105) have to be used to classify the product into the desired size class (250  $\mu$ m to 1000  $\mu$ m). The over size (i.e. the size above 1000  $\mu$ m) has to be crushed with a crusher (A-101) and sent back to the high shear mixer and the under size (i.e the size class below 250  $\mu$ m) has to be sent directly to the high shear mixer.

In this project a fluid bed coating system will be used to coat PEG 6000 on the enzyme granule. Fluid bed equipment has been found to be useful for forming and coating granules particularly smaller granules which are difficult to handle by other methods.

#### 2.8.2 Batch versus Continuous

The reactor will be operated in batch mode as the micro-organisms which produce the enzymes tend to change their DNA structure after every batch. So a new culture has to be introduced after every batch. This also involves sterilising the vessel and the substrate. The granulator and all subsequent downstream processing will be batch mode as control of particle size distribution and strength of the granules is extremely difficult in continuous mode.

# **3 BASIS OF DESIGN (BOD)**

#### 3.1 DESCRIPTION OF THE DESIGN

The objective of this project is to produce enzyme-coated granules. The design of the process is such that it has a capacity of 5,000 tons per year of enzyme granules with an activity of 2.34 x  $10^6$  units per gram granules, density (bulk) of 1011.32 kg/m<sup>3</sup> and a desired particle size range from 250 µm to 1000 µm.

For the enzyme production, specific microorganisms (*Bacillus licheniformis* and *Bacillus amyloliquefaciens*) and the complex medium are charged in a designed batch bioreactor to produce the enzyme. After twenty hours of batch fermentation time, the reactor products are sent for downstream processing. The first process is the separation of the fermentation broth from the solid materials using a vacuum drum filter. The broth is then concentrated using ultra-filtration to yield a concentrated enzyme solution with an activity of 8.97 x  $10^6$  units per gram enzyme. This concentrated enzyme solution is used to produce the enzyme-coated granules. 5071.898 tons per batch of wastes, mainly composed of the biomass, is sent to the appropriate waste-treatment plant for treatment, which is already outside the battery limit of the plant.

Sodium Sulfate (Na<sub>2</sub>SO<sub>4</sub>), which is used as a core granule, is sent through a Lodige mixer and is sprayed with the concentrated enzyme solution together with the binder and builder mixture composed of PEG 6000, PVP K30, cellulose powder, calcium acetate, TiO<sub>2</sub>, and yellow dextrin. The discharge of the granulator is sent to the fluidized bed dryer, where excess moisture from the granule is removed. To meet the particle size specification, the granules are sieved. The over-sized particles are passed through a hammer mill before sending it back to the granulator; while the fine particles are sent directly back to the granulator. Finally, the granules are coated with the coating composed of polyethylene glycol (PEG 6000), magnesium silicate and TiO<sub>2</sub> in the fluidized bed coating equipment to produce 24.45 tons per batch of enzyme-coated granule with an activity mentioned above.

For the calculations, the following assumptions have been made:

- 1) conversion of the bio-reactor is 95%, i.e. 95% of the main reactants (corn starch, soy flour and corn steep liquor) are converted into products following the stoichiometric coefficients.
- 2) rotating drum filter separates the enzyme and water from the rest of the components present in the bio-reactor outlet stream.
- 3) efficiency of the hollow fiber membrane ultra-filter is 90%, i.e. 90% of the water present in the inlet stream is removed.
- 4) the activity of the enzyme is not affected by the processing conditions, e.g. pressure, temperature, etc.
- 5) no particle breakage during the coating process
- 6) hydrolyzed starch contains 90% starch, 10% inert
- 7) soy flour contains 74.0% carbohydrates, 12.0% protein, 14.0% inert
- 8) sodium-caseinate contains 95.0% protein (% min.), 5.0% inert
- 9) 80% of the glucose requirement is from hydrolyzed starch and 20% from the soy flour

#### 3.2 **PROCESS DEFINITION**

#### 3.2.1 Process Concepts chosen

In Chapter 2, some of the options of the processes for the enzyme production, downstream processing, granulation and coating of enzyme granules are presented and these will be briefly discussed here. Since there are many possible options, it was narrowed down to the most feasible for industrial application. The choice was based on technological insight, interdependence of the processes, common sense and insight from experts of this field.

#### 3.2.1.1 Enzyme production

The production of the enzyme could be done in three modes, namely; batch, fed-batch, and continuous fermentation processes. Each of the processes has its inherent advantages and disadvantages, which is discussed profoundly in the preceding chapter.

For the batch process, the sterilized nutrient solution in the fermenter is inoculated with the microorganisms and incubation proceeds under optimal physiological conditions. During the fermentation, only oxygen (air), an antifoam agent and acid or base to control pH are added and the broth containing the products is withdrawn only at the end of

each batch run. Fed-batch process is the modification of the conventional batch process. In this process, the critical elements of the nutrient solution are added in increments as the fermentation progress but the culture broth containing the metabolites (products) is taken only after each batch run. In a fed-batch procedure, production could be increased 4.6-fold in comparison to an uncontrolled batch process. Sterile nutrient solution is added to the bioreactor continuously and an equivalent amount of converted nutrient solution with microorganisms is withdrawn out of a continuous system. The continuous operation has been generally regarded as the most efficient one, but maintaining a high activity microorganism in the continuous culture is difficult to due to their genetic variations. Also, long-term continuation of pure culture is hard because of microbial contamination, so in reality, continuous process is hardly employed in industrial scale.

The batch and the fed-batch processes are generally used in the fermentation process. But the enzyme activity achieved in a fed-batch process is two times lower than in a batch process. Although continuous processes have been described, they are not commercially used, since maintaining a sterile condition for a long period of time is difficult. For this design project, the batch reactor will be used since many published data are available.

The fermentation process is distinguished by three characteristic phases; (1) glucose consumption alone (no protein consumption) with pH reduction, (2) consumption of sugar and protein with pH increase, maximum metabolic activity, maximum amino acid concentration and increase of cell mass, DNA, and protease concentration, (3) protease activity attained maximum and then its reduction with increase of pH, amino acids and  $NH_4^+$  concentrations.

The following reactions take place in the bioreactor:

(1)	$1.10 C_6 H_{12} O_6 +$	1.2 NH <sub>3</sub>	$\rightarrow C_6 H_{10.8} N_{1.2} O_{3.0}$	$+ 2.4 H_2 O + CO_2$	enzyme production
(2)	$1.03 C_6 H_{12} O_6 +$	1.74 NH3	$\rightarrow C_6 H_{9.48} N_{1.74} O_{2.04}$	$+ 0.18CO_2 + 3.78 H_2O$	biomass production
(3)	$C_6H_{12}O_6 +$	60,	$\rightarrow 6 CO_2 + 6 H_2O$		

For simplicity the oxidized form of NAD has been omitted. In this design, the stoichiometric reactor model in the Super Pro 2.7 simulation tool is used.

#### 3.2.1.2 Separation

The first downstream step in a biological process is usually the removal of suspended cell mass and other particulate/colloidal debris from the suspending medium. The concentration of extra-cellular enzymes in typical biotechnological processes is 0.5-1% (w/w).

Filtration and centrifugation are the most common methods use in biological separation. To be economical on an industrial scale, the method must be a continuous process. Continuous centrifuges are difficult to operate and maintain due to the many variables, e.g. the production rate and the cell recovery (yield) are strongly dependent on particle size and inversely proportional to each other. In addition, there may be substantial heat generation requiring complicated cooling system.

On the other hand, filtration is not limited in throughput by the particle size, and the separation rate does not go down as smaller microorganisms are harvested. The most common method is rotary drum filters which can efficiently operated as a batch process.

#### 3.2.1.3 Concentration

After the product is recovered and isolated, it needs to be purified or concentrated further. The purification can be done by precipitation, freeze concentration and ultrafiltration. Precipitation causes denaturation of the product while freeze concentration is energy intensive. Therefore, cross-flow membrane filtration is chosen. With proper selection of the membrane, the module, and operating parameters, 100% cell recovery can be obtained, and capacity can be increased by carefully adding more membrane area.

#### 3.2.1.4 Granulation

The proteases are an environment hazard in the powdered form. The dust in the environment with high protease concentration can cause allergenic reactions in workers handling these products. To avoid this problem, it has to be in granular form. In industry, granulators are bought as there are standard designs. For the present project, the Lodige mixer has been chosen for granulating the enzyme. It is a batch granulation system which has an operating range of 300-2000 kg/h. The machines are provided with lid for charging and inspection of the product.

#### Fluidized bed drying 3.2.1.5

The granule formed in the granulator has high moisture content due to the mixing of the ultra-filtrate which is difficult during handling and transportation. The best and most widely used dryer in industry is the fluidized bed dryer. The reason being that heat transfer is very rapid due to the convection currents formed around each individual granule in the bed. Another major advantage of these types of dryers is the close control of conditions so that a predetermined amount of free moisture may be left with the solids to prevent dusting of the product during subsequent material handling operations.

#### 3.2.1.6 Screening, Crusher and Recycle

The product stream has a wide size distribution. The desired product should have a size range from 250 µm to 1000 µm. Industrial screens have to be used to classify the product into the desired size class. The over size i.e. the size above 1000 µm has to be crushed with a hammer mill and sent back to the granulator and the under size i.e. the size class below 250 µm has to be sent directly to the granulator. This increases the throughput of the granulator and hence the feed to the granulator has to be reduced depending on its own performance i.e. the amount of over size and under size produced.

#### Fluidized bed coating 3.2.1.7

Coating of enzyme granules is important in order to protect the enzyme from the environment. The barrier (coating) is usually a polymer (PEG 6000) which effectively isolates the enzyme from the environment. One can also control the dust by placing a coating around the granules which binds fines to larger particles, or prevents attrition of the granule. In this project a fluid bed coating system will be used to coat PEG 6000 on the enzyme granule. Fluid bed equipment has been found to be useful for forming and coating granules particularly smaller granules which are difficult to handle by other methods. They are designed to keep even small particles in the processing chamber where they may become coated.

#### 3.2.2 **Block Schemes**

The block scheme is presented in the Appendix 3-1 at the end of this report.

#### 3.2.3 **Thermodynamics Properties**

The thermodynamics properties of the components are tabulated in Chapter 4, section 4.3

#### 3.2.4 **Pure Components Properties**

The following table shows the properties of the pure components present in the process

	Table 3. List of pl	ure compo	nents			
NAME	STRUCTURAL	MW TEMPER		RATURE [°C]	DENSITY	LD <sub>50</sub>
	FORMULA	[g/mol]	BOILING	MELTING	[kg/m3]	[mg/kg]
		[9]	Dominio	in the started	[]	[
Glucose	$C_6H_{12}O_6$	180.0	83.0	120.0	1544.0	9000
Di-ammonium hydrogen phosphate	$(NH_4)_2HPO_4$	132.07	-	155.0	1619	-
Di-sodium hydrogen phosphate	Na <sub>2</sub> HPO <sub>4</sub> (12H <sub>2</sub> O)	358.14			1679	17000
Dodecahydrate						
Potassium di-hydrogen phosphate	KH <sub>2</sub> PO <sub>4</sub>	136.09	÷.	256	2338	-
Manganese sulfate monohydrate	$MnSO_4(H_2O)$	168.99	850.00	700	3235	332
Ferrous sulfate heptahydrate	FeSO <sub>4</sub> (7H <sub>2</sub> O)	277.91	300.00	64.0	1899	319
Magnesium sulfate heptahydrate	MgSO <sub>4</sub> (7H <sub>2</sub> O)	246.38	-	70.0	2660	5344*
Oxygen	O <sub>2</sub>	32.00	-182.96	-218.0	1.429**	-
*1.01						

LDLo

reported at 0°C

LCLo,ppm/5M

\*\*\*\* TCLo,mg/m<sup>3</sup>/6h

\*\*\*\*\*\* TCLo,mg/kg/19w

Lewis, R., SAX's Dangerous Properties of Industrial Materials, New Yor, 8th ed., 1989 Source: Perry, R., Handbook of Chemical Engineering, New York, McGraw-Hill, 1982

NAME	STRUCTURAL FORMULA	MW	TEMPER	RATURE [°C]	DENSITY	LD <sub>50</sub>
	TORMOLA	[g/mol]	BOILING	MELTING	[kg/m3]	[mg/kg]
Nitrogen	N <sub>2</sub>	28.00				-
Water (liquid)	$H_2O$	18.00	100.00	0.0	1000	-
Biomass	$(C_6H_{10.8}N_{1.2}O_{3.0})$	147.6	-	-	-	-
Protein	(C <sub>6</sub> H <sub>9.48</sub> N <sub>1.74</sub> O <sub>2.04</sub> )	138.48	-	-	-	-
Carbon dioxide	CO <sub>2</sub>	44.00	-78.5	-56.6	1.53	90000
Sodium sulfate	$Na_2SO_4$	142.05		888.0	2671	5989
Calcium acetate	Ca(CH <sub>3</sub> COO) <sub>2</sub>	158.0	-	-	2500	52.0
Titanium dioxide	TiO <sub>2</sub>	79.8	<3000	1640.0	4220	250 <sup>***</sup>
Magnesium silicate	Mg(SiO <sub>4</sub> )	278.91	2230.0	1710.0	2500	100 <sup>***</sup>

LDLo

reported at 0°C

LCLo,ppm/5M

\*\*\*\*\* TCLo,mg/m<sup>3</sup>/6h

\*\*\*\*\* TCLo,mg/kg/19w

Source: Lewis, R., SAX's Dangerous Properties of Industrial Materials, New Yor, 8<sup>th</sup> ed., 1989 Perry, R., Handbook of Chemical Engineering, New York, McGraw-Hill, 1982

#### 3.3 BASIC ASSUMPTIONS

#### **3.3.1** Plant Capacity

The plant operates as batch process with a total output of 5000 tons of PEG 6000 coated enzyme granules per year having an activity of  $2.34 \times 10^6$  U/g. There are 300 batches annually with a cycle time of 35.5 hours.

489.75 tons per batch of medium is mainly composed of glucose, ammonium source, trace elements, water and other growth factors. The waste streams amount to 482.92 tons per batch of liquid and solid liquid wastes and 681.59 tons per batch of gaseous wastes. The complete stream report is found in the appendices.

Based on normal operating hours of 7200 hrs per year, the plant operates at 300 days per year. The remainder number of days is used for maintenance and legal holidays. The plant life is assumed to be 15 years (including 1.5 years for construction) since the product is competitive in the market.

#### 3.3.2 Location

The plant will be located in Pune, India. The country has an average ambient temperature of 25 °C throughout the year which is favorable for the fermenter as compared to Europe where low temperatures is prevalent which is unsuitable for fermentation. Since the road networks are favorable, transport of the raw materials and products is easy. The plant can be located in the outskirts of the city in order to avoid contamination of the surrounding air in case of accident or leaks. Also, the environmental regulations will not be a critical issue at present since no existing policies for this type of industries. But this should not be understood as taking advantage of the present situation.

#### **3.3.3** Battery Limit

The design project includes the following important equipment inside the battery limit. The reference block diagram can be found in the Appendix of this report.

 R-101 Aerated Stirred Tank Bioreactor with cooling jacket. The reactor is also equipped with baffles and mechanical foam separator. There are 2 units of this reactor with a total volume of 290.0 m<sup>3</sup> each and the construction material is SS316.

- 2) RV-101 Rotating Drum Vacuum Filter. The suspension is sucked through a filter cake on a rotating drum and the filter cake is removed with a blade. The filter cake is rinsed during the rotation. Three units of this filter with a filter area of 96.0 m<sup>2</sup> each.
- 3) UF-101 Hollow Fiber Membrane Ultra-filtration system. Two units having a membrane area of 446.0 m<sup>2</sup> per unit are used. The concentration factor (the ratio of feed to retentate) is 100.
- 4) G-101 Lodige mixer/granulator. It is a batch granulation system where the builder, binder and ultra filtrate are mixed under high shear conditions leading to the formation of granules
- 5) D-101 Fluidized Bed Dryer. One unit with a volume of 1.70 m<sup>3</sup>, height is 6.0 m and a diameter of 0.6 m. The drying temperature is 40°C. Steam heated air is used in this equipment.
- S-105 Screens (Industrial Sieves). These are of vibrating type since large capacity and high efficiency are required. The screening system classifies the particle in the range of 1,000 μm to 250μm.
- 7) A-101 Hammer mill. It used for pulverizing and disintegration at high speed. The rotor shaft is horizontal with Tshaped hammers. A cylindrical screen serves as an internal classifier. The size of the product can be controlled by changing the rotor speed, feed rate, or clearance between the hammers and grinding plates.
- 8) D-102 Fluidized Bed Spray Coating System.
- 9) E-101 Coolers. This is a plate and frame type with a heat transfer area of 79.8 m<sup>2</sup>. NaCl brine is used as the cooling medium to cool down the reactor product to 5°C.
- 10) E-102 Coolers. This is also a plate and frame type with a heat transfer area of 0.05 m<sup>2</sup>. NaCl brine is used as the cooling medium to cool down the enzyme granules product to 20°C.
- 11) S-101 & S-102 Air filters. The max throughput of the filters is  $4.0 \text{ m}^3/\text{s}$ .
- 12) T-101 Blending tank. It is used to homogenized the medium prior to fermentation. Two units which have a volume of 283.0 m<sup>3</sup> per tank. The agitation rate is 1.0 kW/m<sup>3</sup> for 6 hours. Chilled water is used as cooling agent.
- 13) P-101, 102, 103 & 104 Centrifugal Pumps. These are widely used in chemical industries for pumping liquids with very wide ranging properties and suspensions with high solids contents.
- K-101, 102, & 103 Compressors. The compressors can carry a pressure change of 2 bars and have an efficiency of 70%.
- 15) E-101 Sterilizer. Three units of sterilizer are used with a diameter of 70cm and the holding tube length is 196.8 m. The heating medium used is steam at a temperature of 152.0 °C and a built cooler which uses a cooling water.

Other facilities available outside the battery limit:

- 1) Ion-exchanger for the process water used in the medium preparation and cleaning of interior and internals of bioreactor
- 2) Vessels for the cooling water, washwater, and other water requirement of the plant
- 3) Air receivers for the oxygen supply into the bio-reactor and other oxygen requirement.
- 4) Boilers for the low pressure steam requirement of the plant.
- 5) Tanks and other vessels for the storage of the raw materials.
- 6) Seed tank for the inoculum preparation.
- 7) Vessel for the NaCl brine container

#### **3.3.4** Definition In- and Out-going streams

The values are valid at the battery limit.

Listed below are the compositions of the typical technical (complex) media for the production of the enzyme.

COMPONENT	COMPOSITION [kg/m <sup>3</sup> ]
Hydrolysed starch	100.00
Sodium caseinate	27.00
Soy-flour	23.00
Corn steep	7.00
Di-ammonium hydrogen phosphate	0.50
Di-sodium hydrogen phosphate dodecahydrate	0.50
Potassium di-hydrogen phosphate	0.30
Manganese sulfate monohydrate	0.02
Ferrous sulfate heptahydrate	0.05
Magnesium sulfate heptahydrate	0.05
Antifoam agent Ke 211	4.00

Tablad	Tachnical	madium	anumonition	110	tan wat	0.14
I UDIE 4.	rechnical	теашт	composition	in	iup waie	21

Since this process is not available from literature or previous published works, it is not possible to mention commercial specifications for the streams. The design stream values are shown in tabulated form.

#### 3.3.4.1 Feedstocks

#### Substrate (IN)

This stream is composed of complex or undefined substances listed above and the biomass. For simplicity, only the necessary constituents are found in the table below. These substances (for the medium) are taken from the storage and weighed to desired proportion. The inoculum is from the seed tank which is outside the BL.

Stream Name	: 1			Subst	rate
Comp.	Units	Specifi	cations		Additional information
	%wt	Available	Design	Notes	( also ref. note numbers )
Glucose			54.27	(1)	(1) hydrolyzed starch and soy flour are the
Biomass			1.49		source of the glucose, although some other
Ammonium p	phosphate		3.12	(2)	hydrates are present, they are assumed to be
Nutrients			0.92	(3)	reduced to glucose.
Inert			40.20	(4)	(2) source of nitrogen, also, amino acids from
					steep and casein are significant, they are
					not participating in the reaction based on stoich
Total			100.00		(3) composed of hydrated salts of Mn, Mg, Fe,
	00		05.00		V
Temp.	C		25.00		agents
Press.	bara		1.01		(4) other components such as antifoam agent,
Phase	V/L/S		L		insoluble cellulose substances
Price	US\$/kg		1.0596		

#### Water (IN)

Process water can be defined as the water used in the bioreactor. It also includes the medium make-up, in association with direct liquid additions, and for cleaning of the vessel interior and internals. The process water is classified under the low-endotoxin water. It is from the water storage, which requires precautions against microbial proliferation. It is fed through a distribution system (e.g. hoses, pipes) which is regularly sterilized.

Stream Name	2: 2	Water					
Comp.	Units	Specifi	cations		Additional information		
	%wt	Available	Design	Notes	( also ref. Note numbers )		
Water			100				
		8					
Total			100.00				
P	Process Condi	tions and Price	;				
Temp.	°C		25				
Press.	bara		1				
Phase	V/L/S		L				
Price	US\$/kg		0.01				

#### Discharge T101

This stream is the homogenized mixture of the medium, inoculum, nutrients and water. This is done to ensure that the solids in suspension is adequately agitated when it is transferred to the fermentor.

Stream Name:	3	Discharge T101				
Comp.	Units	Specifi	cations		Additional information	
	%wt	Available	Design	Notes	(also ref. Note numbers)	
Ammonium ph	osphate		0.32			
Biomass			0.15			
Glucose			5.51			
Inert			4.08			
Nutrients			0.09			
Water			89.84			
Total			100.00			
Process Conditions and Price			;			
Temp.	°C		25			
Press.	bar <sub>a</sub>		1			
Phase	V/L/S		S+L			
Price	US\$/kg		-			

#### **Discharge E101**

Sufficient steam sterilization is necessary to kill and remove included microbial contaminants. It is done using a platetype sterilizer. The process conditions, e.g. time and temperature, should be monitored to eliminate the bacteria and ensure no severe alterations of the medium (due to chemical reaction).

Stream Name:	4	Discharge E101				
Comp.	Units	Specifi	cations		Additional information	
	%wt	Available	Design	Notes	( also ref. Note numbers )	
Ammonium ph	osphate		0.32	(1)		
Biomass			0.15			
Glucose			5.51			
Inert			4.08			
Nutrients			0.09			
Water			89.84			
Total			100.00			
Process Conditions and Price			;			
Temp.	°C		37			
Press.	Bar <sub>a</sub>		1.01			
Phase	V/L/S		S+L			
Price	US\$/kg		-			

#### Air for R101 (IN)

Process air is supplied to the bioreactor for aerating the culture medium. It is also used, in smaller quantity, for performing pressure transfers and breaking vacuum after steam sterilization. As a material that enters the process, it has the potential to contaminate the process. It should be free from atmospheric contaminants and contamination by the compression system itself. It is from the centralized site and distributed through pipelines. Air receivers are installed in the system to cope with peak loads and short-term drops in supply due to compressor change-over or start-up of additional units.

Stream Name:	5	Process Air				
Comp.	Units	Specifi	cations		Additional information	
	%mol	Available	Design	Notes	( also ref. Note numbers )	
Nitrogen			79.0		т. Т	
Oxygen			21.0			
Total			100.00			
Pro	ocess Condi	tions and Price	;			
Temp.	°C		25.00			
Press.	Bar <sub>a</sub>		1.01			
Phase	V/L/S		V			
Price	US\$/kg		-		5	

#### **Discharge K101**

The sterile condition of the process air is maintained after compression.

Stream Name:	6	Process Air				
Comp.	Units	Specifi	cations		Additional information	
	%mol	Available	Design	Notes	(also ref. Note numbers)	
Nitrogen			79.0	(1)		
Oxygen			21.0			
Total			100.00			
Pro	ocess Condi	tions and Price				
Temp.	°C		40.00			
Press.	Bar <sub>a</sub>		3.01			
Phase	V/L/S		V			
Price	US\$/kg		-			

#### **Discharge S101**

Sterilization of air is done by filtration using fibrous or membrane filters since sterilization by heat is economically impractical and ineffective due to the low heat-transfer of air compared to liquids. Airborne particles are also removed from the air.

Stream Name:	7	Discharge S101				
Comp.	Units	Specifi	cations		Additional information	
	%mol	Available	Design	Notes	(also ref. Note numbers)	
Nitrogen			79.0			
Oxygen			21.0			
Total			100.00			
Pr	ocess Condi	tions and Price	100.00			
	Contraction of the second second	nons and i nee	40.00	-		
Temp.	°C		40.00			
Press.	Bar <sub>a</sub>		3.01			
Phase	V/L/S		V			
Price	US\$/kg		-			

#### Bottom R101

Stream Name:	10			Botto	om R101
Comp.	Units	Specifi	cations		Additional information
	%wt	Available	Design	Notes	(also ref. Note numbers)
Ammonium pho	osphate		0.02		(1) may also contain several other enzymes and
Biomass			1.04		Inactive protein which has synergistic effects on
Enzyme			0.84	(1), (2)	Activity of enzymes
Glucose			2.13		(2) activity is $8.44 \times 10^4$ U/ml
Inert			4.10		
Nutrients			0.09		
Water			91.77		
Total			100.00		
Pro	cess Condi	tions and Price	;	1	
Temp.	°C		37		
Press.	bar <sub>a</sub>		1.01		
Phase	V/L/S		L		
Price	US\$/kg		-		

### Discharge E102

Stream Name:	11	Discharge E102			
Comp.	Units	Specifi	cations		Additional information
	%wt	Available	Design	Notes	(also ref. Note numbers)
Ammonium ph	osphate		0.02		(1) The temperature of the reactor products is
Biomass			1.04		Reduced to 5°C to avoid self-digestion of
Enzyme			0.84		
Glucose			2.13		
Inert			4.10		
Nutrients			0.094		
Water			91.77		
Total			100.00		
Pro	ocess Condi	tions and Price	;		
Temp.	°C		5.00	(1)	
Press.	bar <sub>a</sub>		1.01		
Phase	V/L/S		L		
Price	US\$/kg		-		

#### **Discharge S103**

Stream Name:	14		Discharge S103				
Comp.	Units	Specifi	cations		Additional information		
	%wt	Available	Design	Notes	( also ref. Note numbers )		
Ammonium ph	osphate		0.02		(1) activity is $8.97 \times 10^4$ U/ml		
Enzyme			0.89	(1)			
Glucose			2.25				
Water			96.85				
Total			100.00				
Pro	ocess Condi	tions and Price	:				
Temp.	°C		5.00		×		
Press.	bar <sub>a</sub>		1.01				
Phase	V/L/S		L				
Price	US\$/kg		-				

### Discharge S104

Stream Name:	16	Discharge S104				
Comp.	Units	Specifi	cations		Additional information	
	%wt	Available	Design	Notes	( also ref. Note numbers )	
Ammonium ph	osphate		0.01		(1) activity is 8.97 x 10 <sup>6</sup> U/ml	
Enzyme			31.79	(1)		
Glucose			38.66			
Water			29.54			
Total			100.00			
Pro	ocess Condi	tions and Price	•			
Temp.	°C		7.00			
Press.	bara		1.01			
Phase	V/L/S		L			
Price	US\$/kg		-			

#### Builder + Binder (IN)

This stream composed the main core of the granules. For ecological reasons, these raw materials should be as natural as possible. These agents improve the activity stability, dust forming behavior, and handling. For example, the addition of the cellulose powder prevents the adherence of the wet granulate mass on the walls of the granulator.

Stream Name:	17			Bind	er + Builder
Comp.	Units	Specifi	cations		Additional information
	%wt	Available	Design	Notes	( also ref. Note numbers )
Builder			88.24	(1)	(1) composed of 79.13% Na <sub>2</sub> SO <sub>4</sub> , 8.70%
Binder			11.76	(2)	dextrin, 6.96% Ca(OAc) <sub>2</sub> , 3.48% TiO <sub>2</sub> ,
					and 1.74% cellulose powder
					(2) composed of 85.71% PEG 6000, 14.29%
					PVP K30
Total			100.00		
Pro	ocess Condi	tions and Price	:		
Temp.	°C		25.00		
Press.	bar <sub>a</sub>		1.01		
Phase	V/L/S		S		
Price	US\$/kg		1.588		

#### Discharge G101

This is the product of the granulating equipment. Based on the performance of the equipment, 98.% of the product is within the desired particle size range ( $1,000 \,\mu\text{m} - 250 \,\mu\text{m}$ ) and 2.0% goes to the oversize and undersize streams.

Stream Name:	18	Discharge G101					
Comp.	Units	Specifi	cations		Additional information		
	%wt	Available	Design	Notes	( also ref. Note numbers )		
Enzyme			19.99				
Glucose			24.32				
Builder			32.74				
Binder			4.36				
Water			18.59				
Ammonium ph	osphate		Trace				
Total			100.00				
Pro	ocess Condi	tions and Price	>				
Temp.	°C		18.10				
Press.	bar <sub>a</sub>		1.01				
Phase	V/L/S		S+L				
Price	US\$/kg		-				

#### **Discharge D101**

Stream Name:	19	Discharge D101				
Comp.	Units	Specifi	cations		Additional information	
	%wt	Available	Design	Notes	( also ref. Note numbers )	
Enzyme			20.02			
Glucose			24.36			
Builder			32.79			
Binder			4.37			
Water			18.45			
Total			100.00			
Pro	ocess Condi	tions and Price	;			
Temp.	°C		18.1			
Press.	bara		1.01			
Phase	V/L/S		S			
Price	US\$/kg		-			

#### **Oversize S105**

This stream is sent to the hammer mill for crushing before recycling it to the granulator. This stream is 1.0% of the output of the fluid bed dryer.

Stream Name:	20	Oversize S105				
Comp.	Units	Specifi	cations		Additional information	
	%wt	Available	Design	Notes	(also ref. Note numbers)	
Enzyme			20.02			
Glucose			24.36			
Builder			32.79			
Binder			4.37			
Water			18.45			
Total			100.00			
Pre	ocess Condi	tions and Price	;			
Temp.	°C		18.10			
Press.	bar <sub>a</sub>		1.01			
Phase	V/L/S		S			
Price	US\$/kg		-			

### Discharge A101

Stream Name:	21			Discl	Discharge A101				
Comp.	Units	Specifi	cations		Additional information				
	%wt	Available	Design	Notes	( also ref. Note numbers )				
Enzyme			20.02						
Glucose			24.36						
Builder			32.79						
Binder			4.37						
Water			18.45						
Total			100.00						
Pro	ocess Condi	tions and Price							
Temp.	°C		18.10						
Press.	bar <sub>a</sub>		1.01						
Phase	V/L/S		S						
Price	US\$/kg		-						

#### **Undersize S105**

This stream is directly sent back to the granulating equipment. This stream is 1.0% of the fluid bed dryer output.

Stream Name:	22	Undersize S105				
Comp.	Units	Specifi	cations		Additional information	
	%wt	Available	Design	Notes	(also ref. Note numbers)	
Enzyme			20.02			
Glucose			24.36			
Builder			32.79			
Binder			4.37			
Water			18.45			
Total			100.00			
Pr	ocess Condi	tions and Price				
Temp.	°C		18.10			
Press.	bara		1.01			
Phase	V/L/S		S			
Price	US\$/kg		-			

#### **Discharge S105**

Stream Name:	26		Discharge S105				
Comp.	Units	Specifi	cations		Additional information		
	%wt	Available	Design	Notes	( also ref. Note numbers )		
Enzyme			20.02				
Glucose			24.36	2			
Builder			32.80				
Binder			4.37				
Water			18.45				
Total			100.00				
Pro	ocess Condi	tions and Price					
Temp.	°C		40.00	1			
Press.	bar <sub>a</sub>		1.01				
Phase	V/L/S		S+L				
Price	US\$/kg		-				

#### PEG 6000 (IN)

After drying the granules, it can be coated with a protective coating. The coating can serve to color the granulate or for the protection of the enzyme, or can cause retardation of the release of the enzyme or enzyme mixture. The binder selected from water soluble polymers, especially PEG with molecular weights in the range of 1,500 to 10,000.

Stream Name:	27	PEG 6000				
Comp.	Units	Specific	cations		Additional information	
	%wt	Available	Design	Notes	( also ref. Note numbers )	
Coating			32.00	(1)	(1) composed of 60.0% PEG 6000, 26.67%	
Builder			48.00		13.33% MgSiO <sub>2</sub>	
Water			20.00			
Total			100.00			
Dr	Condi	tions and Price	100.00			
II II	ocess Collul	tions and Frice				
Temp.	°C		60.00			
Press.	bar <sub>a</sub>		1.01			
Phase	V/L/S		S+L			
Price	US\$/kg		1.122			

#### **Mixture for D101**

Stream Name:	28		Mixture for D101				
Comp.	Units	Specifi	cations		Additional information		
	%wt	Available	Design	Notes	( also ref. Note numbers )		
Enzyme			15.93				
Glucose			19.38				
Coating			6.54				
Builder			35.90				
Binder			3.48				
Water			18.77				
Ammonium pho	osphate		Trace				
Total			100.00	1			
Pro	ocess Condi	tions and Price	•	1			
Temp.	°C		45.97				
Press.	bar <sub>a</sub>		1.01				
Phase	V/L/S		S+L				
Price	US\$/kg		-				

### 3.3.4.2 Product

#### **Enzyme Granules**

This is the final product, coated enzyme granules which has an of activity of  $2.34 \times 10^{6}$  U/ml.

Stream Name:	33	Enzyme Granules				
Comp.	Units	Specifi	cations		Additional information	
	%wt	Available	Design	Notes	( also ref. Note numbers )	
Enzyme			15.93		(1) as dry substance, amount with respect to the	
Glucose			19.38		particles of the uncoated enzyme granulate	
Coating		5.0 - 26.0*	6.54	(1)*		
Builder			35.90			
Binder			3.48			
Water			18.77			
Ammonium ph	osphate		Trace			
Total			100.00			
Pro	ocess Condi	tions and Price	;			
Temp.	°C		20.00			
Press.	bara		1.01			
Phase	V/L/S		S			
Price	US\$/kg		15.00			

### 3.3.4.3 Utilities

#### Air for D101

This stream is used for the fluidized bed dryer for the drying of the granules.

Stream Name:	23		Air for D101					
Comp.	Units	Specifi	cations		Additional information			
	%mol	Available	Design	Notes	( also ref. Note numbers )			
Nitrogen			79.00					
Oxygen			21.00					
2								
Total			100.00					
Pro	ocess Condi	tions and Price		1				
Temp.	°C		25.00	1				
Press.	bar <sub>a</sub>		1.01					
Phase	V/L/S		V					
Price	US\$/kg		-					

#### Discharge K102

Stream Name:	24	Discharge K102				
Comp.	Units	Specifi	cations		Additional information	
	%mol	Available	Design	Notes	( also ref. Note numbers )	
Nitrogen			79.00			
Oxygen			21.00			
Total	an Act offered		100.00			
Pro	ocess Condi	tions and Price	;	1		
Temp.	°C		40.00			
Press.	bar <sub>a</sub>		3.01			
Phase	V/L/S		V			
Price	US\$/kg		-			

#### Air for D102

This is used for the fluidized bed for the coating of the granules.

Stream Name:	29			Air	for D102
Comp.	Units	Specifi	Specifications		Additional information
	%mol	Available	Design	Notes	(also ref. Note numbers)
Nitrogen			79.00		
Oxygen			21.00		
Total			100.00		
Pro	ocess Condi	tions and Price	;		
Temp.	°C		25.00		
Press.	Bar <sub>a</sub>		1.01		
Phase	V/L/S		V		
			-		

#### Discharge K103

Stream Name:	30		Discharge K103				
Comp.	Units	Specifi	cations		Additional information		
	%mol	Available	Design	Notes	(also ref. Note numbers)		
Nitrogen			79.00				
Oxygen			21.00				
Total			100.00				
Pr	ocess Condi	tions and Price	;	1			
Temp.	°C		40.00				
Press.	Bar <sub>a</sub>		3.01				
Phase	V/L/S		G				
Price	US\$/kg		-				

### Discharge E103

Stream Name:	31		Discharge E103				
Comp.	Units	Specifi	cations		Additional information		
-	%mol	Available	Design	Notes	( also ref. Note numbers )		
Nitrogen			79.00				
Oxygen			21.00				
Total			100.00				
Pro	ocess Condi	tions and Price	3	1			
Temp.	°C		20.00	1			
Press.	Bar <sub>a</sub>		1.03				
Phase	V/L/S		G				
Price	US\$/kg		=				

#### NaCl Brine

This utility is used as a cooling agent of the coolers and ultrafilter.

Stream Name	tream Name:				ine
Comp.	Units	Speci	Specifications		Additional information
_	%wt	Available	Design	Notes	( also ref. Note numbers )
NaCl Brine			100.00		
	and the second second second second			-	
Total					
	Process Con	ditions and Pri	ce		
Temp.	°C		-9.00		
Press.	$Bar_a$		1.01		
Phase	V/L/S		L		
Price	US\$/kcal		28.00/10 <sup>6</sup> kcal		

#### **Cooling water**

Cooling water for the sterilizer and fermentor.

Stream Nan	Stream Name:				Vater
Comp.	Units	Specif	ications		Additional information
	%wt	Available	Design	Notes	( also ref. Note numbers )
Water			100.00		
Total			100.00		
	Process Cond	itions and Pric	e		
Temp.	°C		5.00		
Press.	Bara		1.01		
Phase	V/L/S		L		
Price	US\$/kcal		1.00/10 <sup>6</sup> kcal		

#### Low Pressure Steam

Heating agents for Sterilizer and Fluidized bed dryer

Stream Nam	ne:				
Comp.	Units	Speci	fications		Additional information
	%wt	Available	Design	Notes	(also ref. Note numbers)
Water			100.00		
Total			100.00	1	
	Process Con	ditions and Pri	ce	1	
Temp.	°C		152.0	]	
Press.	Bar <sub>a</sub>				
Phase	V/L/S		V		
Price	US\$/kcal		10.00/10 <sup>6</sup> kcal		

#### Wash water

This is used for washing of the cake of the vacuum drum filter.

Stream Name:	12		Wash Water					
Comp.	Units	Specifi	cations		Additional information			
	%wt	Available	Design	Notes	(also ref. Note numbers)			
Water			100.00					
Total			100.00					
Pr	ocess Condi	tions and Price	;	]				
Temp.	°C		25.00	]				
Press.	Bar <sub>a</sub>		1.01					
Phase	V/L/S		L					
Price	US\$/kg		0.005					

### 3.3.4.4 Wastes

#### Gas R101

This is the exhaust gas of the fermentor which is mainly composed of air and  $CO_2$  and possibly traces of fermentation product.

Stream Name:	8	Gas R101				
Comp.	Units	Specifications			Additional information	
	%mol	Available	Design	Notes	( also ref. Note numbers )	
Nitrogen			78.99			
Oxygen			20.23			
Carbon dioxide			0.77			
Total			100.00			
Process Conditions and Price			;			
Temp.	°C		37.00			
Press.	Bar <sub>a</sub>		1.01			
Phase	V/L/S		V			
Price	US\$/kg		-			

### Discharge S102

Stream Name:	9	Discharge S102					
Comp.	Units	Specifications			Additional information		
	%mol	Available	Design	Notes	( also ref. Note numbers )		
Nitrogen			78.99				
Oxygen			20.23				
Carbon dioxide			0.77				
Total			100.00				
Pro	ocess Condi	tions and Price	;				
Temp.	°C		37.00				
Press.	Bar <sub>a</sub>		1.01				
Phase	V/L/S		V				
Price	US\$/kg		-				

#### Solid Waste

This stream is sent to the appropriate treatment plant.

Stream Name:	13	Solid waste					
Comp.	Units	Specifications			Additional information		
	%wt	Available	Design	Notes	( also ref. Note numbers )		
Ammonium phosphate			0.01				
Biomass			10.96				
Glucose			1.01				
Inert	5		43.07				
Nutrients			0.99				
Enzymes			0.40				
Water			43.56				
Total			100.00				
Process Conditions and Price			2				
Temp.	°C		5.00				
Press.	Bar <sub>a</sub>		1.01				
Phase	V/L/S		S/L				
Price	US\$/kg		-				
# Liquid Waste S104

This waste stream is sent to the wastewater treatment facility.

Stream Name:	15			Liqu	id waste
Comp.	Units	Specifi	cations		Additional information
	%wt	Available	Design	Notes	( also ref. Note numbers )
Glucose			1.21		
Water			98.77		
Ammonium ph	osphate		0.02		
-	-				
Total			100.00		
Pro	ocess Condi	tions and Price	;	1	
Temp.	°C		7.00	1	
Press.	Bar <sub>a</sub>		1.01		
Phase	V/L/S		L		
Price	Nfl/kg		-		

# Top D101

The exhaust gas of the fluidized bed dryer. This could contain particulates.

	Stre	am Name:	25	Top D101			
Comp.	Units	Specifi	cations		Additional information		
%m0	ol	Available	Design	Notes	(also ref. Note numbers)		
Wate	er		69.44				
Nitrog	gen		24.15				
Oxyg	en		6.41				
Tota	1		100.00				
Pr	ocess Condi	tions and Price	100.00				
	00003 COllar	tions and Trice					
Temp.	°C		40.00				
Press.	bara		1.01				
Phase	V/L/S		V				
Price	US\$/kg		-				

# **Top D102**

This is the exhaust gas from the fluidized bed coating equipment. Particulates are possible contaminants of this stream.

Stream Name:	32	Top D102				
Comp.	Units	Specifi	cations		Additional information	
	%mol	Available	Design	Notes	( also ref. Note numbers )	
Nitrogen			79.00			
Oxygen			21.00			
Water			Trace			
Total			100.00			
Pro	ocess Condi	tions and Price				
Temp.	°C		20.00			
Press.	Bar <sub>a</sub>		1.01			
Phase	V/L/S		V			
Price	US\$/kg		-			

# 3.4 ECONOMIC MARGIN

Below is the tabulated data for the evaluation of economic margin (see Appendix 3-2):

#### Feed Stream Costs

TOTAL VALUE OF FEEDSTOCKS, PROCESS CHEMICALS IN

Stream>			1	2	5	12	17	23	27	29	Total Price
	Prices [US\$/kg]	Units									US\$
Ammonium phosphate	1.00	kg/day	1,550.00		-			•			1,550.00
Biomass	5.00	kg/day	740.00					•	•		3,700.00
Glucose	1.00	kg/day	27,000.00			-	-				27,000.00
Inerts	1.00	kg/day	20,000.00								20,000.00
Nutrients	1.00	kg/day	460.00		-			-			460.00
Coating	2.00	kg/day	•	•					1,600.00		3,200.00
Builder	1.00	kg/day	•		•		6,379.99	-	2,400.00	•	8,779.99
Binder	6.00	kg/day					850.00			-	5,100.00
Water	0.01	kg/day	-	440,000.00	-	24.00		-	1,000.00		3,307.68
Nitrogen	-	kg/day			521,053.82		-	16.84		22,135.80	•
Oxygen		kg/day			158,181.84	•		5.11		6,720.00	
Carbon dioxide		kg/day	-							-	
Total Mass rate		kg/day	49,750.00	440,000.00	679,235.66	24.00	7,229.99	21.95	5,000.00	28,855.80	
										Total US\$	73,097.67

TOTAL VALUE OF PRODUCTS OUT

Stream>			33	Total Price
	Prices [US\$/kg]	Units		US\$
Enzyme granules	12.00	kg/day	24,454.38	293,452.56
Total Mass rate		kg/day	24,454.38	
			Total US\$	293,452.56

# MARGIN = OUT-IN

MARGIN [US\$/year] 66,106,467.00

Margin = (293,452.56)-(73,097.67) = US\$ 220,354.89 per day = US\$ 66 million per year

- 1) The above prices are approximation of the prices available on the internet.
- 2) The above cost analysis has been made considering that the biomass produced is worthless, when actually, it could be dried and incinerated to produce electricity.

Based on the table above, the design project is economically feasible. The estimated cost of feedstocks and other chemicals is US\$ 73,097.67 while the product value is US\$ 293,452.56 on a per day basis. This means a economic margin of US\$ 220,354.89 per day or US\$ 66 million annually. The cost of the raw materials could still be reduced if the medium composition is optimized.

# **4 THERMODYNAMIC PROPERTIES**

	ieuium compositio			
Component	Cp [kJ/kg-K]	ρ[kg/m³]	State	Source
Ammonium phosphate	0.90	1619	Solid	<sup>1</sup> Perry
Biomass	0.7	1050	Solid	<sup>2</sup> SPD
Glucose	1.22	1562	Solid	<sup>2</sup> SPD
Inerts	1.33	1200	Liquid	Calculated
Cellulose	1.31	1350	Solid	<sup>1</sup> Perry
Insoluble salts	4.184	2300	Solid	Assumed
Antifoam Ke 211	24.02	1000	Liquid	Assumed
Nutrients	0.73	1000	Liquid	Assumed
Builder	1.22	1800	Solid	Calculated
Cellulose powder	1.34	1350	Solid	<sup>1</sup> Perry
Sodium sulphate (filler)	137.24	2698	Solid	<sup>1</sup> Perry
Yellow dextrine	1.34	1038	Solid	Assumed
Calcium acetate	0.59	2500	Solid	Assumed
TiO <sub>2</sub>	0.63	4220	Solid	<sup>1</sup> Perry
Binder	1.34	1010	Liquid	Calculated
PEG 6000 (Polyethylene Glycol, MW : 6000)	1.13	1100	Liquid	Assumed
PVP K30 (Polyvinyl Pyrolidone, MW : 30,000)	1.47	1000	Liquid	Assumed
Water	4.184	1000	Liquid	<sup>1</sup> Perry
Coating	1.40	1100	Liquid	Calculated
PEG 6000 (Polyethylene Glycol, MW : 6000)	1.53	1100	Liquid	Assumed
TiO <sub>2</sub>	0.63	4220	Solid	<sup>1</sup> Perry
Magnesium silica	107.11	2500	Solid	Assumed
Water	4.184	1000	Liquid	<sup>1</sup> Perry

Table 5. Technical medium composition in tap water

Source:

<sup>1</sup>Perry, R.H. and Green, D., Perry's Chemical Engineers' Handbook, 6 Ed. McGraw-Hill, 1984 <sup>2</sup> Super Pro Designer, version 2.7 – Simulation Software

#### 4.1 **OPERATING WINDOW**

Temperature range: 5°C [278 K] – 60°C [313 K] Pressure rande: 1 bara – 3.01 bara

Densities and specific heats of all components have been assumed constant for the purpose of calculation throughout the rnge of the operating window.

# 4.2 DATA VALIDATION

4.3 DATA ACCURACY

# **5 PROCESS STRUCTURE AND DESCRIPTION**

# 5.1 CRITERIA AND SELECTIONS

# 5.1.1 Enzyme production

Proteases are one of the most important industrial enzymes currently in industrial production. Proteases, having molecular weights ranging from 25000 - 30000, are produced commercially both from bacteria and fungi and are stable up to a temperature of  $65^{\circ}$ C.

The detergent enzymes are alkaline serine protease Subtilisin Carlsberg formed by *Bacillus licheniformis* and protease Subtilisin NOVO<sup>™</sup> by *Bacillus amyloliquefaciens*.

The cultivation with complex medium used in the industry are yielding protease activity of  $0.5 \times 10^9$  U/kg<sub>cells</sub>. The concentration of cells is 30 kg<sub>cells</sub>/m<sup>3</sup><sub>liq</sub>.

The proteases of this type have many features of value for use as detergent enzymes:

- stability at high temperature
- stability in the alkaline range (pH 9-11)
- stability in association with chelating agents and perborates, however, their stability in the presence of surfaceactive agent is low, thus limiting their shelf life.

Screening of proteases should be done to assure its stability under alkaline conditions. Strains of *B. licheniformis* and *B. subtilis* showed optimal growth in the pH range 6-7, but the new genetically modified strains can grow at pH 8-11. Protein engineered Bacillus has improved properties such as substrate specificity, pH optimum, and stability to bleaching agents. At their optimum pH range, at least 90% of the maximal growth rate is attained. The enzyme production can be carried out in three modalities:

*Batch process:* A batch fermentation is considered as a closed system. At t=0 sec., the sterilized nutrient solution in the fermenter is inoculated with the microorganisms and incubation proceeds under optimal physiological conditions. During the fermentation, only oxygen (air), an antifoam agent and acid or base to control pH are added. The composition of the culture medium, the biomass concentration, and the metabolite concentration generally change constantly as a result of the metabolism of the cells.

*Fed-Batch process*: The fed-batch process is the modification of the conventional batch process. In this process, the substrate is added in increments as the fermentation progresses. The formation of many secondary metabolites is subject to catabolite suppression by high concentrations of glucose, other carbohydrates, or nitrogen compounds. So, in the fed-batch system the critical elements of the nutrient solution are added in small concentrations at the start of the fermentation and are continuously added in small doses during the production phase. Experiments showed that the biosynthesis of the alkaline serine protease is dependent on the growth rate of the culture and concentration of nitrogen and carbon sources in the medium. Ammonium was reported to inhibit protease production. The experiments showed that in the fed-batch fermentation a constant low level of ammonium concentration should be maintained for optimum protease production. In a fed-batch procedure, production could be increased 4.6-fold in comparison to an uncontrolled batch process.

*Continuous process*: Continuous process is considered as an open system. Sterile nutrient solution is added to the bioreactor continuously and an equivalent amount of converted nutrient solution with microorganisms is withdrawn out of the system. The reactor could be a homogenously mixed bioreactor or a plug flow type. A homogenously mixed reactor is either a chemostat where the cell growth is controlled by adjusting the concentration of one substrate (limiting factor) or a turbistat, cell growth is maintained by using turbidity to monitor the biomass concentration and the rate of feed of the nutrient solution is properly adjusted.

The batch and the fed-batch processes are generally used in the fermentation process. The latter is employed to keep down the concentration of ammonium ions and amino acids, since these nitrogenous materials repress protease production. Although continuous processes have been described, they are not commercially used. For this design project, the <u>batch reactor</u> will be used since many published data are available.

The process starts with the sterilization of the seed and fermentation (with medium) tank at  $121^{\circ}$ C for 30 minutes with steam. The sterilized medium is then cooled to cultivation temperature. The inoculum will be fed to the seed tank (1-20 m<sup>3</sup>) which runs from 20-70 hours depending on the volume and the microorganisms used. At this sterile conditions, the inoculant is transferred to the fermenter, which is typically between 40-100 m<sup>3</sup>, for the fermentation process. The end point of the process is when the maximum enzyme activity is reached which is usually after 24-28 hours of fermentation time. After the fermentation, the broth is cooled down and transferred to the drop tank for further

separation and concentration. The batch process will usually take a time range of 45 to 130 hours depending on the microorganisms, the type, amount and quality of the raw materials.

The following reactions take place in the bioreactor

### 5.1.2 Separation

The first downstream step in a biological process is usually the removal of suspended cell mass and other particulate/colloidal debris from the suspending medium. The concentration of extra-cellular enzymes in typical biotechnological processes is 0.5-1% (w/w).

In the past, the preferred methods of cell harvesting were filtration or centrifugation. To be economical on an industrial scale, the method must be a continuous process. Continuous centrifuges, where the cells are removed by the action of high G forces, are expensive to buy, operate, and maintain. The production rate of a centrifuge is a function of many variables, among them the square of the particle diameter, the difference in density between the particles and the suspending medium, the G force and the viscosity of the suspending medium. The production rate and the cell recovery (yield) are strongly dependent on particle size and inversely proportional to each other. In addition, there may be substantial heat generation requiring complicated cooling system; there may be production of aerosols, and the capacity of most tubular centrifuges is limited and may require frequent teardowns and cleaning (Cheryan, 1998).

Filtration, on the other hand, is not limited in throughput by the particle size, and the separation rate does not go down as smaller microorganisms are harvested. The most common method is rotary drum filters, where a vacuum draws the broth to the drum surface.

The microbial cells and other insoluble solids accumulate on the filter cloth and are scraped off with a knife.

## 5.1.3 Concentration

Many traditional cell-harvesting processes are being replaced by cross-flow membrane filtration. With proper selection of the membrane, the module, and operating parameters, 100% cell recovery can be obtained, and capacity can be increased by merely adding more membrane area.

Over 130 materials have been used to manufacture membranes but only a few have achieved commercial status and fewer still have obtained regulatory approval for use in food, pharmaceutical, and kindred applications. The family of polysulfone membranes has been chosen in the concentration step of this process due principally to the following favorable characteristics:

- Wide temperature limits: typically, temperatures up to 75°C
- Wide pH tolerances: these membranes can be continuously exposed to pHs from 1 to 13
- Fairly good chlorine resistance: up to 200 ppm chlorine for cleaning and up to 50 ppm chlorine for short storage of the membrane
- Easy to fabricate membranes in a wide variety of configurations and modules
- Wide range of pore sizes available for ultrafiltration applications, ranging from 10A° (1000 MWCO) to 0.2µm in commercial-size modules.
- Good chemical resistance to aliphatic hydrocarbons, fully halogenated hydrocarbons, alcohols and acids.

Membrane technology presents six basic designs of equipment: (1) tubular, with inner channel diameters>4 mm; (2) hollow fibers, with inner diameters of 0.2-3 mm; (3) plate units; (4) spiral-wound modules; (5) pleated-sheet cartridges; and (6) rotary modules.

A hollow fiber design has been chosen due to the following characteristics:

- It is one of the more economical designs in terms of energy consumption because of combination of pressure drops and flow rates. Operating velocity of 0.5-2.5 m/s (Reynolds numbers of 500-3000) and pressure drops of typically 1.35-2.35 bara, depending on flow rate, fiber diameters and lengths.
- Hollow fibers have the highest surface area-to-volume ratio among the modules. Holdup volume is low.
- A big advantage with hollow fibers is its back-flushing capability, which improves its cleanability. This is possible because the fibers are self-supporting.

The hollow fiber design that suits the process requirement has the following characteristics:

Table	6. C	haracter	ristics	of	the	selected	concenti	ration	equipment
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	11
PARAMETER	VALUE
Membrane design	Hollow fibers
Membrane material	Polysulfone
Pore Size [A°]	200
Fiber diameter [mm]	0.5
Fiber length [m]	0.1
Temperature and pressure rating	80°C − 1 atm
PH value	8-11
Manufacturer	A/G Technology

Though ultrafiltration is the most economical method for concentrating the filtrate, it cannot achieve very high concentrations (~99%) due to reduction in efficiency as the concentration gradient is very low and the osmotic pressure becomes low. More membrane systems are required in order to achieve such high concentrations. In order to avoid this, an evaporator is used to concentrate the ultra-filtrate further. The enzymes begin to degenerate at temperatures beyond 330 K in water. Hence, the boiling point of the ultra filtrate has to be reduced by vacuum. Thus, the evaporator has to function under vacuum conditions.

Another possibility could be freeze concentration, which is done by freezing the ultra filtrate and then the ice formed on the surface of the liquid is scraped off. The refrigeration of the ultra filtrate is energy intensive compared to evaporation.

# 5.1.4 Granulation

The protease rich ultra filtrate can be spray dried and introduced into the detergent in powdered form (mean particle size 0.3 mm with considerable amount of small particles) (Kadam, 1991). The dust in the environment has high protease concentration and can cause allergenic reactions in workers handling these products. Hence the proteases are an environment hazard in the powdered form. In order to eliminate this hazard, the protease rich ultrafiltrate can be mixed with Builder and Binder and granulated using a granulator. In industry, granulators are bought as there are standard designs. For the present project, Lödige High Shear Mixer/Granulator has been chosen for granulating the enzyme. It is a batch granulation system where the Builder, Binder and Ultra filtrate are mixed under high shear conditions leading to the formation of granules.

# 5.1.5 Fluidized bed drying

The granule formed in the granulator has high moisture content due to the mixing of the ultra-filtrate and Binder, both of which are solutions. The wet granules are difficult during handling and transportation. Drying reduces the cohesive forces between the individual granules enabling ease in handling. The best and most widely used dryer in industry is the fluidized bed dryer. The reason being that heat transfer is very rapid due to the convection currents formed around each individual granule in the bed. Another major advantage of these type of dryers is the close control of conditions so that a predetermined amount of free moisture may be left with the solids to prevent dusting of the product during subsequent material handling operations.

# 5.1.6 Screening, Crusher and Recycle

The product stream has a wide size distribution. The desired product should have a size range from 250  $\mu$ m to 1000  $\mu$ m. Industrial screens have to be used to classify the product into the desired size class. The over size i.e. the size above 1000  $\mu$ m has to be crushed with a crusher and sent back to the granulator and the under size i.e the size class below 250  $\mu$ m has to be sent directly to the granulator. This increases the throughput of the granulator and hence the feed to the granulator has to be reduced depending on its own performance i.e. the amount of over size and under size produced.

# 5.1.7 Fluidized bed coating

Coating of enzyme granules is important in order to protect the enzyme from the environment. The Coating is usually a mixture of polymer (PEG 4000) along with  $TiO_2$  and Magnesium Silicate which effectively isolates the enzyme from the environment. One can also control the dust by placing a coating around the granules which binds fines to larger particles, or prevents attrition of the granule. In this project a fluid bed coating system will be used to coat the Coating on the enzyme granule. Fluid bed equipment has been found to be useful for forming and coating granules particularly smaller granules which are difficult to handle by other methods. They are designed to keep even small particles in the processing chamber where they may become coated.

# 5.1.8 Batch/Continuous

The reactor will be operated in batch mode as the micro-organisms which produce the enzymes tend to change their DNA structure after every batch. So a new culture has to be introduced after every batch. This also involves sterilising the vessel and the substrate. The granulator and all subsequent downstream processing will be batch mode as control of particle size distribution and strength of the granules is extremely difficult in continuous mode.

### 5.1.9 Heat Integration

Energy saving plays an important role in the design process. Heat integration reduces the energy requirement of a certain process, thus making it more economically attractive. However, the inclusion of heat integration in this design project is not practically useful. The temperatures of the streams are not large enough because they are restricted by the allowable limit of the enzymes. This is usually the case of bioprocesses, they are working at a relatively low temperature. For this design project, the working temperature is 5°C and the highest is 121°C.

# 5.2 PROCESS FLOW SCHEME

The Process Flow Scheme diagram is shown in Appendix 5-1.

#### 5.3 BATCH CYCLE DIAGRAM

The Batch Cycle diagram is shown in Appendix 5-2.

#### 5.4 PROCESS STREAM SUMMARY

The Process Stream Summary is shown in Appendix 5-3.

# 5.5 UTILITIES

The details on the utilities is shown in Appendix 5-4.

#### 5.6 PROCESS YIELDS

The process yields have been shown in Appendix 5-5.

# 6 PROCESS CONTROL

# 6.1 INTRODUCTION

The overall objective in any process -continuous or batch- is to convert certain raw materials into desired products in the most economical way. During production, the process must meet a number of requirements, such as safety, production specifications, environmental regulations, operating constraints and economics. All these requirements indicate the need for continuous monitoring and control of the operation of a process plant. This is accomplished through a rational arrangement of equipment (measuring devices, valves, controllers, computers) and human intervention (design and operating personnel) which together constitute the control system. The control loops of the detergent enzyme granules manufacturing process have been shown in the process flow scheme. The following sections will discuss in detail the control system of two major units, namely the fermenter and the fluid bed dryer/coater.

#### 6.2 PHYSICAL CONTROL OF THE FERMENTATION

The growth of microorganisms is a process which is regulated by a complex interaction between the external physical, chemical and biological conditions of the liquid environment of the fermentation and the internal biochemical processes within the cells. The control objective therefore is to regulate the environment so that the correct conditions are obtained for cell growth and production. In this process the aerated stirred tank fermenter is supplyed with means of measuring temperature, pressure, pH, dissolved oxygen concentration and the agitator speed. Foaming is not depressed by the controlled addition of antifoam agent since excessive amount may have inhibiting effects on the growth of the microorganism. A sophisticated device "Fundafoam system" (Checap) mounted on to the agitator shaft is used to destroy the foam. The nutrient solution held in the foam flows back into the bioreactor, and the air released from the foam leaves the vessel.

# 6.2.1 Temperature

The temperature measuring instrument for the fermenter is the platinum resistance thermometer with standard Pt 100 probe. Heating is not necessary as the power input from the impeller is sufficient to heat up the vessel. All tanks are fitted with baffles which prevent a large central vortex being formed and improve mixing, and coolant passes through the baffle. As the fermentation is a batch process, the temperature control must be time-dependent. Therefore, full PID control using the control valve of straight-through cooling water is used.

# 6.2.2 Pressure

The fermenter is fitted with a mechanical pressure gauge with local read-out for safety. A manual needle valve can be used on the fermenter outlet to keep a back-pressure in the fermenter, but this will require much fiddling to keep the pressure constant. A pressure regulator gives a more constant back-pressure, but regulators on fermenters tend to block and are not of hygienic design. The best arrangement is a control loop consisting of a pressure transmitter, PID controller, and air-operated air flow control valve on the vessel exit line.

# 6.2.3 Agitation

Reliable and precise agitator speed control is not easily obtained. A precise method consists of a control loop with a frequency sensor reading the fermenter shaft speed and feedback control to the motor drive. Modern A.C. speed controllers and motors are now available. Adjustment of agitator speed is the easiest way to control dissolved oxygen tension in the fermenter. For this reason the motor speed controller has set-point input, sometimes called external or remote set-point control. This input can then be set by the output of the dissolved oxygen controller.

# 6.2.4 pH control

Measurement of pH is based on the absolute standard of the electrochemical properties of the standard hydrogen electrode. Control of pH is actually achieved by the controlled pulse modulated feeding of either acid or base. This operates such that when the pH measurement exceeds the set-point value, acid is fed to the reactor, and when the measured pH is less than the set-point, feeding of base occurs. The provision of either acid or base side of the set-point results in a two-sided discontinuous corrective feeding of either acid or base. A variable dead-band PI controller is used because it can be set up to allow the fermenter pH to drift natually within a defined band, with acid or base control being applied only outside the upper and lower limits of the dead-band.

# 6.2.5 Dissolved oxygen

The dissolved oxygen content of the fermentation broth is a very important fermentation parameter, affecting both cell growth and production formation. Three main methods, namely: (1) the tubing method, (2) the use of mass

spectrometer probes, and (3) electrochemical detectors, may be used for the determination of dissolved oxygen concentration. All require a prior calibration procedure since they use membranes to separate the point of measurement from the broth. DOT control can be achieved via four controlled variables on the fermenter: air flow, pressure agitator speed, and oxygen/inert gas mixing. In this project a control combination of air flow and agitator speed is used, which is most common in practice. This works generally by the controller first regulating agitator speed and varying this up to a fixed maximum value. At this point, the controller the begins to increase gas flow rate. The combined control scheme is cascaded, such that the controller output signal is used to adjust the set-point of the agitator speed control, or the set-point of the gas flow rate controller.

#### 6.3 PHYSICAL CONTROL OF THE FLUID BED DRYING AND COATING

The aim of any industrial drying processes is to produce a solid product of desired quality at minimum cost and maximum throughput and to maintain these consistently. The control of dryers is probably one of the least studied areas of process control because of the highly nonlinear dynamics of drying process. The most important manipulated inputs to a fluid bed dryer are inlet air temperature and air flow rate. The most desirable drying process output variable to control is the product moisture content, but this is generally difficult to measure directly. Often, the moisture content of the dried product can be inferred from the temperature and humidity of the exhaust gas, although care must be taken in applying appropriate mass and heat balances. Moreover, measuring the temperature of the exit gas is simple, accurate, reliable, cheap and has significant effects on the drying rate. Therefore, Among available control strategies, the control system based on exhaust air temperature as the control variable and inlet air temperature as the manipulated variable is found to be adequate for the fluid bed dryer. This control strategy performs quite satisfactory under any load changes because of the fluid bed capacity. Compared to the fluid bed dryer, one more manipulated input is available: the flow rate of the liquid. Therefore, we have two alternative basic types of control systems. Control system A features two quick-response control loops:

- 1. Control of exhaust air temperature by feed rate regulation.
- 2. Control of inlet air temperature by air heater regulate.
- 3. Control system B uses air heater regulation to compensate for any deviation from the desired outlet air temperature. This system also features manual regulation of the feed rate. Both system are commonly used in the industrial dryers. In this project, the control system B is preferred since wide variations in feed rate cannot be handled.
- 4 These control schemes are provided with safety systems that prevent any failure in the feed system in order to prevent the outlet air temperature rising above a specified safety level.

# 7 MASS AND HEAT BALANCES

# 7.1 PRACTICAL ASPECTS

The mass balance for the present process has been solved accurately with a difference of zero. However, the heat balance has not been solved to the same accuracy, but with a difference of 8 kW. The reason could be due to some discrepancies in the specific heats of the materials. The specific heats have been taken constant in the range of the operating window and this has led to the discrepancy. But since 8 kW is very small compared to the total power input and output, it can be neglected.

# 7.2 BALANCE FOR TOTAL STREAMS

The Balance for the total stream is shown in Appendix 7-1.

#### 7.3 BALANCE FOR STREAM COMPONENTS

The Balance for the Stream Components is shown in Appendix 7-2.

# **8 PROCESS AND EQUIPMENT DESIGN**

# 8.1 INTEGRATION BY PROCESS SIMULATION

For the production of enzyme-coated granules, the process can be divided into two parts: 1) the production of the enzymes; and 2) the granulation. To simulate the process, the group used the aide of the SuperPro Designer Version 2.7(hereafter referred to as SPD2.7). The SPD2.7 is one of the many simulation software for process simulation. However, what is distinctive of SPD2.7 is its capability of simulating a bioprocess.

The main problem encountered when simulating the whole process is the gap created by the granulation part of the process. SPD2.7 is not equipped with the simulation of granulation. So from the production of enzymes, the group treated the granulation part as a simple unit operation of mixing of the product enzyme and the binders used in granulation, and then proceeds to the rest of the unit operations (i.e. fluid bed drying, and fluid bed coating). By treating the granulation part as purely mixing, does not at all affects the mass balance assuming that nothing is lost during granulation.

The simulation modeling is given in Figure 2.



Figure 2. Process Simulation Model

#### 8.2 EQUIPMENT SELECTION AND DESIGN

#### 8.2.1 Blending tank

The volume of the tank is calculated as follows (for batch Operation)

(Liquid Volume) = (Total Liquid Volume Processed per Batch) / (Number of Cycles).

#### 8.2.2 Sterilizer

The purpose of this model is to rigorously simulate the holding tube of a continuous sterilizer and to approximately estimate the cost of the entire system, including the holding tube along with its insulation, the heat exchangers for energy conservation, the pumps, etc. Continuous heat sterilization is simulated assuming logarithmic death rate of microorganisms and spores (Wang et al, 1979):

$$-\frac{dN}{dt} = k * N ,$$

where N is the concentration of viable organisms in number/ $m^3$ , k is the specific death rate constant in sec-1, and t is time in minutes. At t = to, N = No.

The specific death rate (k) is related to sterilization temperature by an Arrhenius type equation:

$$k = A * e^{\left(-\frac{\Delta E}{RT}\right)},$$

where A is the frequency factor in sec-1,  $\Delta E$  is the activation energy of death in J/mole, R is the gas constant in J/mole-K, and T is the absolute temperature in K.

Since actual plug flow through the holding tube of a continuous sterilizer is never achieved, an axial dispersion model is assumed to account for residence time distribution. Solving the material balance equation, we get:

$$\frac{N}{No} = \frac{4*\delta*e^{\left(\frac{N_{pe}}{2}\right)}}{\left(1+\delta^2\right)*e^{\left(\frac{N_{pe}\delta}{2}\right)} - \left(1-\delta^2\right)*e^{\left(\frac{N_{pe}\delta}{2}\right)}},$$

where  $\delta=\sqrt{1+\frac{4N_{R}}{N_{pe}}}$  ,  $N_{R}=\frac{kL}{U},~N_{pe}=\frac{UL}{D_{z}}$ 

U is the average medium velocity (m/sec), L is the length of the holding tube (m), Dz is the axial dispersion coefficient ( $m^2$ /sec). Dz is estimated as a function of the Reynolds number from the following equation:

$$\frac{D_z}{\mu/\rho} = 6.0936 * 10^5 - 1.2342 * 10^5 \ln(\text{Re}) + 6279.2[\ln(\text{Re})]^2,$$

where  $\mu$  is the liquid viscosity (kg/m-s) and  $\rho$  is the liquid density (kg/m<sup>3</sup>). This equation was derived by curve fitting experimental data (Figure 40, Levenspiel, 1972).

The tube diameter (dt) and the sterility level  $(N/N_0)$  are specified and the equations are solved iteratively to estimate the required tube length (L).

The size estimation of the heat exchangers that are used for energy conservation as well as the estimation of utilities are based on intermediate and final temperatures of various streams that are specified by the user.

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#### 8.2.3 Fermentor

The fermentor is modeled as a stoichiometric reactor. It can be used to simulate bio-transformations when the reaction kinetics are unknown or unimportant but the mass stoichiometry is known and the extent of reaction can be specified or calculated based on the concentration of a reference component. The extent of reaction is defined as the fractional conversion of the limiting component.

#### 8.2.3.1 Reaction Initialization and Material Balances

The user provides the mass stoichiometric coefficients (Ai) of the various components for the overall reaction. Negative coefficients are used for reactants and positive for products. The sum of all stoichiometric coefficients must be equal to zero to guarantee mass conservation. The extent of reaction (x) is based on the limiting component and it is either specified by the user or calculated based on the specified concentration of a reference component. If the specified reference component concentration is not feasible, the user is warned and the algorithm assumes a reaction extent of 100% or 0% depending on whether the reference component is a product or reactant, respectively. The material balances are based on the following simple algorithm:

#### Case where the reaction extent is specified

First, the limiting component (k) is identified, based on the mass stoichiometry and the composition of the feed streams. Its outlet mass flowrate (Fout,k) as a function of its inlet mass flowrate (Fin,k) and the extent of reaction (x) is:

 $Fout_k = Fin_k(1-x),$ 

where,

Fout<sub>k</sub> is the mass flowrate of the limiting component after the reaction,

Fin<sub>k</sub> is the mass flowrate of the limiting component before the reaction, and

x is the reaction extent.

Second, the outlet mass flowrate (Fouti) of component (i) as a function of its inlet mass flowrate (Fini), the extent of reaction (x), and the mass stoichiometric coefficients  $(A_i)$ , is given by the following equation:

 $Fout_i = Fin_i - Fin_k * x * (A_i/A_k),$ 

where  $A_k$  is the coefficient of the limiting component.

#### Case where the concentration of a reference component is specified

The algorithm starts again with the identification of the limiting component (k). Next it checks whether the specified reference component concentration is feasible (it corresponds to a reaction extent between 0 and 100%). If the reference component concentration is feasible, the algorithm calculates the reaction extent that corresponds to the specified concentration using an iterative computational procedure. If the specified reference component concentration is not feasible, the algorithm assumes a reaction extent of 100% or 0% depending on whether the reference component is a product or reactant, respectively. After the calculation / selection of the extent of reaction, the reaction material balances are completed as in the previous case.

#### 8.2.3.2 Vapor Effluent

Vapor-liquid equilibrium calculations in the fermentor are not carried out. To account for gas components exiting in the gas outlet stream, the user must specify the removal fraction (Percent in Gas Outlet) for each component. The calculation of the gas outlet stream composition is based on materials that are available in the vessel after the completion of the fermentation.

# 8.2.3.3 Intra/Extracellular Component Flowrate

You may also specify the extracellular percent (100% by default) of each of the reactions product components. This feature is useful when tracking of intracellular water is desired because it affects the performance of centrifugation, filtration, etc. further downstream. If the Extra-Cell % of certain reaction product components is less than 100 (or in other words a fraction of the component is intracellular) and the Primary Biomass and Water components are identified (through the component initialization dialog window), then the model automatically associates intracellular water with the intracellular reaction product components. Then, if a separator is used to remove intracellular components (i.e., removal of biomass by a centrifuge), the separation (removal %) of intracellular water will be the same as that of the intracellular component(s).

#### 8.2.3.4 Vessel Sizing

The bioreactor volume (V) for continuous operation is estimated based on the overall feed volumetric flowrate (F) and the average dilution rate (D):

V = F/D

For batch operation, when the reactor volume is not specified by the user, it is set equal to the volume of the material that is processed during each cycle.

### 8.2.3.5 Aeration Rate

When the aeration rate is set by the user (in Volume of air under standard conditions per Volume of fermentation broth per Minute - VVM), the flowrate of the aeration stream is adjusted by the fermentor model. If the aeration stream has a source unit operation (e.g., a compressor), then the adjustment of its flowrate is recursively back - propagated till flowsheet feed streams are reached. At least one of the flowsheet feed streams that feed into the aeration stream must have non-zero flowrate. Only units with a single output stream can be part of the sequence of units that feed into an aeration stream. An exception to this rule is the Custom Mixer which is not allowed to be part of the sequence even though it has a single output stream.

# 8.2.3.6 Utility Calculation

The average agitation power is estimated by multiplying the unit power requirement (kW/m3 of broth) by the liquid volume of the fermentor. To estimate the overall heating or cooling requirement, the model considers the sensible heat of the inlet and outlet streams along with the heat of reaction. The calculation of the heat of reaction for aerobic systems is based on the oxygen uptake rate (Cooney et al., 1968):

#### $Q_{\rm f} = 0.12 \ Q_{\rm O2},$

where  $Q_f$  is heat release in J/m<sup>3</sup>-s and  $Q_{02}$  is the oxygen uptake rate in mol/m<sup>3</sup>-s. A default value of -15690 kJ/(kg of oxygen utilized) is used based on the above model.

#### 8.2.3.7 Equipment Purchase Cost

The equipment purchase cost includes the cost of the vessel along with the agitator cost. If the vessel is checked as an ASME Vessel (i.e., constucted according to standards published by the American Society of Mechanical Engineers) then it assumed to withstand vacuum to 35 psig and its purchase cost is penalized by 20% over the base vessel cost. If the operating pressure of the vessel is set to a pressure higher than 3 atm, then the vessels purchase cost is penalized by an 80% increase over the base cost.

# 8.2.4 Compressor

Compressors operate in pressure ranges and with compression ratios that often require external cooling to prevent damage to sensitive seals and metal surfaces. This physical situation falls between the isentropic and isothermal extremes and is called polytropic compression. To calculate the power requirement for polytropic compression, the equations developed by Shultz (1962) are used:

$$Power = \frac{\gamma}{\gamma - 1} p_1 v_1 \left[ \left( \frac{p_2}{p_1} \right)^{(\gamma - 1)/\gamma} - 1 \right]$$

$$p_2 = p_1 \left(\frac{T_2}{T_1}\right)^{(\gamma-1)/\gamma}$$

where  $\gamma$  = ratio of specific heat of gas at constant pressure to specific heat of gas at constant volume, v<sub>1</sub> = specific volume of gas at intake conditions, T<sub>1</sub> = absolute temperature of gas at intake conditions, and T<sub>2</sub> = absolute temperature of gas at final delivery conditions.

# 8.2.5 Air Filter

This unit operation model calculates the size and number of units based on the actual throughput.

#### 8.2.6 Rotary Drum

In biochemical processes, centrifugal filtration is mainly used for dewatering and washing of crystalline and fibrous solids. This method uses filtration driven by centrifugal force. The equipment consists of a rotor with a perforated cylindrical shell, lined with an appropriate filter medium, mounted on a bowl bottom, and surmounted by an annular lip ring to retain its contents.

The material balances around a basket centrifuge are determined by the Retention Coefficient of the various components specified by the user. The fractions of solvent and solutes that are retained on the cake are estimated based on the cake porosity.

In design mode, the user specifies the average filtrate flux and the model calculates the filter area, the drum diameter and the number of units. In rating mode, the user specifies the drum diameter and the number of units and the model calculates the average filtrate flux.

Averaged data from vendors were used to estimate the purchase cost of basket centrifuges.

# 8.2.7 Ultrafilter

Tangential flow ultrafiltration is used to concentrate protein solutions and separate proteins from low molecular weight solutes. Ultrafilter membranes are rated based on molecular weight cut-off of spherical proteins that cannot pass through the membrane. Membranes are available with MW cut-off ranging from 2,000 Daltons up to 1,000,000 Daltons.

The average life of ultrafiltration membranes strongly depends on the degree of utilization and the operating conditions and ranges approximately from 6 to 12 months (or 1000 to 2000 hours of operation time).

The modeling equations of the membrane ultrafilter are identical to those of the membrane microfilter.

#### 8.2.7.1 Microfilter

The purpose of this model is to simulate a tangential flow membrane microfilter used for solid-liquid separation and to estimate its capital and operating cost.

In bioprocessing, tangential flow microfiltration is frequently used for cell harvesting, cell debris removal, and sterilization of cell culture media. In tangential microfiltration, cross flow parallel to the filter surface is used to enhance filtration rate. The pore sizes of microfilters usually range from 0.1 to 0.45 microns.

Two modes of filtration are modeled by the membrane microfilter:

Batch Concentration (Figure 3):



Figure 3. Ultrafiltration - Batch Concentration

A schematic of batch concentration is shown above. The retentate is returned to the feed tank for recycling through the module. This is the fastest method of concentrating a given amount of material and it requires the minimum membrane area. In batch concentration, the fraction ( $F_i$ ) of a component (i) remaining in the retentate at the end of concentration is estimated by the following equation (McGregor, 1986):

 $F_i = (CF)R_i-1$ 

Eq. 1

where CF is the concentration factor and is given by

(CF) = (Feed Volumetric Flowrate)/(Concentrate Volumetric Flowrate) and R<sub>i</sub> is the rejection coefficient of component (i) and is defined by

 $R_i = 1 - (C_{pri} / C_{ci})$ 

 $C_{pri}$  and  $C_{ci}$  are the concentrations of component (i) at the end of filtration in the permeate and concentrate streams, respectively.

For batch concentration, a simple algorithm is used to perform the material balances and estimate the size of equipment. The user always specifies the average permeate flux (J) and the rejection coefficients of the various particulate components. In large-scale operation, the filtrate flux of microfilters usually ranges between 10 and 80 liter/m2-h. In Rating Mode, the user also specifies the area (A) of the membrane and the number of identical units operating in paraller while in Design Mode the user specifies the concentration factor. A simple equation relates the average filtrate flux with the membrane area and the amount of material processed by the filter per cycle:

#### $J = V_{feed} (1 - 1/CF) / A / (Process Time) / (Number of microfilters)$

 $V_{\text{feed}}$  is the volume of material fed to the microfilter per cycle. The process time is always specified by the user as part of the process scheduling initialization. In design mode (where the user specifies the concentration factor), the above equation is solved for A (the membrane area). If the calculated membrane area exceeds the maximum allowable membrane area per unit, the system assumes multiple identical units operating in parallel with a total membrane area equal to the calculated. In rating

mode (where the user specifies the area and the number of units), the above equation is solved for CF.

Having CF and R<sub>i</sub>, Eq. 1 is used to calculate the fraction of each component retained in the concentrate and perform the material balances.

#### 8.2.8 Cooler

#### 8.2.8.1 Technical Description

The purpose of this model is to simulate a shell and tube surface condenser. Condensation is the process of converting a gas or vapor to liquid. Any gas can be reduced to a liquid by sufficiently lowering its temperature and/or increasing its

pressure.

The coolant runs in the tube side, while condensation takes place in the shell side. A constant overall heat transfer coefficient is assumed for this model (provided by the user).

The heat transfer for a surface condenser is governed by the following relationship:

 $Q = U A_c \Delta T$ ,

Q is the total heat load, J /s.

U is the overall heat transfer coefficient,  $J/s - K - m^2$ .

 $\Delta T$  is the mean temperature difference driving force, K.

 $A_c$  is the surface area of the condenser, m<sup>2</sup>.

Computational procedures to estimate the condensation efficiency are available for three different sets of specifications for the condenser model:

Operating pressure and temperature (isothermal condensation) Operating pressure and percent of total condensation Operating pressure and percent of condensation of a specific component

For each component the enters the condenser, the following material balance equation holds:

 $F_m * z_i = V_m * y_i + L_m * x_i$ 

where  $F_m$ ,  $V_m$ ,  $L_m$  are the molar flowrates of the feed, vapor and liquid phase respectively and  $z_i$ ,  $y_i$ ,  $x_i$  are the mole fractions of component i in the three phases.

The mole fractions of component i in the vapor  $(y_i)$  and liquid  $(x_i)$  phase are related through the partition coefficient K:

 $K = y_i/x_i$ 

The partition coefficients (K) are assumed to be independent of composition i.e. the model assumes ideal vapor and liquid phases (Antoine's equation is used to estimate the K value).

Since the design equation is a nonlinear equation an iterative numerical algorithm is used for its solution.

# 8.2.9 Granulation circuit

#### 8.2.9.1 Technical Description

The granulator which is the heart of the granulation circuit is the Lödige Ploughshare®-Mixer Type FKM for Batch Operation. In this report it will be called the High Shear Mixer and shown in Fig. 1. It has a total volume of 30,000 l and is equipped with three ploughshare shovels attached to a rotating horizontal shaft and a side-mounted spinning chopper. The rotational speed of the ploughshares is variable and is set usually at 200 rpm. A hammer type chopper is used with fixed speed of 3000 rpm. The action of the chopper and the ploughshare is different as can be expected from their size and particularly from their tip speeds of about 5 m/s for the ploughshares and 25 m/s for the chopper. The ploughshares ensure mixing while the chopper provides impact forces on the granules. The residence time of the particles in the High shear mixer is around 10 minutes.



Figure 4. Lödige Ploughshare®-Mixer Type FKM

The second part of the granulation circuit is the Screens or Industrial sieves. They are placed after the Fluid Bed dryer in order to classify the granules within the size range of  $250 \,\mu m$  to  $1000 \,\mu m$ . There are two sieves/screens with  $250 \,\mu m$  and  $1000 \,\mu m$  sieve opening respectively.

The particles collected below the 250  $\mu$ m sieve are either recycled or discarded as waste and the particles which are collected above the 1000  $\mu$ m sieve are sent to the Crusher which the third part of the granulation circuit. The Crusher is a Hammer Mill which will break the particles in sizes smaller than 1000  $\mu$ m and sent back into the High Shear Mixer for agglomeration. In this project 10% of fines (i.e. particle size <250  $\mu$ m) and coarse (i.e. particle size > 1000  $\mu$ m) are produced which will be sent back into the next batch for agglomeration.

# 8.2.10 Fluid Bed Dryer

# 8.2.10.1 Technical Description

In the fluid-bed dryer, the wet solids are fluidized by the drying gas. Mixing and heat transfer are very rapid. Wet feed is admitted to the top of the bed and dry product is taken out from the side near the bottom. The average time a particle stays in the dryer is typically 30 to 120 s.

A short-cut model is used to simulate the fluid-bed dryer. The user always specifies the drying temperature, the evaporation fraction of the solvent component(s), the specific heating requirement, the specific power requirement, the drying gas requirement, the height to diameter ratio and the average solids velocity.

The material balances are done based on the evaporation fraction of the solvent component(s). The average solids velocity is used to estimate the required height. In design mode, the user in addition specifies the solids residence time which along with

the average solids velocity is used to estimate the height of the vessel. Then, the height to diameter ratio is used to calculate the diameter and the volume of the dryer.

In rating mode, the user specifies the total dryer volume and the number of units and the program calculates the solids residence rate.

# 8.2.11 Granulation

There are some special issue with regards to the calculation of the size distribution obtained from the High Shear Mixer. There are batch data\*\* available on downscaled High Shear Mixers and from these a correlation was obtained between the operating conditions and the size distribution.

The particle size distribution of the granules should lie in the range of 250  $\mu$ m and 1000  $\mu$ m. Now according to batch data, the change in the size distribution is mainly caused due to change in amount of Builder, Binder and Ultra Filtrate. Using Rosin-Rammler Size distribution equation, the parameters were obtained from batch experimental data and correlated to the ratio of (Builder + Binder) and Ultra Filtrate. This correlation was obtained using non-linear regression. From the process design the ratio which is suitable to obtained a most optimum size distribution i.e. 10% above 1000  $\mu$ m and 10% below 250  $\mu$ m is 0.6.

<sup>&</sup>lt;sup>•</sup> The references and the batch experimental data cannot be given as these are confidential.

### 8.3 EQUIPMENT DATA SHEETS

# 8.3.1 Equipment Summary Sheets

The Equipment Summary Sheets are shown in Appendix 8.1 to 8.6.

# 8.3.2 Equipment Specification Sheets

The Equipment Specification Sheets are shown in Appendix 8.7 to 8.22.

# 9 WASTES

This chapter is intended to make an in-depth review of all wastes produced during the production of enzyme granules and the methods of handling them. The management of wastes can be approached at the very start of the design process, e.g. during the selection of the processes and equipment. The emphasis of the new approach is prevention through basic design. The details of the waste treatment process will not be discussed in this design project.

# 9.1 INDIRECT WASTES

These wastes include all pollution occurring during the manufacturing of feedstocks and plant equipment or as a result of product usage. This category of wastes is not included in this design project.

# 9.2 DIRECT WASTES

The waste stream is characterized by organic solid and liquid wastes, contaminated exhaust gases, and particulates. These wastes do not contain toxic components which require specialized treatment and disposal systems. The available waste treatment technologies are applicable to reduce its impact to the environment at accepted level. The highly organic liquid and solid waste could be treated in an anaerobic wastewater system. Air filters are used to treat gaseous wastes. Below, these wastes are categorized and their effects are discussed.

Type (Quantity)	Strea m #	Composition	Source	Effects
Exhaust gases (2.04x10 <sup>5</sup> ton per year)	8	Mainly $CO_2$ , air, traces of gases (e.g. $H_2S$ , CO, SO <sub>x</sub> , NO <sub>x</sub> which are products of possible side reactions)	Bioreactor	<ul> <li>SO<sub>x</sub> causes sensory and respiratory irritation, vegetation damage, corrosion</li> <li>CO reduces the oxygen-carrying capacity of blood</li> </ul>
(8.67x10 <sup>3</sup> ton per year)	24,25	Water , air , particulates	granulator, fluidized bed dryer, coating equipment	<ul> <li>causes allergic reactions to humans</li> <li>particulates containing enzymes could be inhaled which will be</li> </ul>
Solid wastes (1.63x10 <sup>4</sup> ton per year	13	Biomass, glucose, inert, nutrients, enzymes, (with water)	Wastes from the rotary drum vacuum filter	<ul> <li>high BOD reduces the dissolved oxygen of the lakes or streams to a certain level which is harmful to the marine life</li> </ul>
Liquid wastes (1.29x10 <sup>5</sup> ton per year)	15	di-ammonium hydrogen phosphate, glucose, water	Ultrafilter	- same as solid waste
Others (cannot be quantified)			Packaging materials,	
(cannot be quantified)		Contains the solid and liquid wastes mentioned above	<ul> <li>Cleaning water of the bioreactor (interiors, internals)</li> <li>Sanitary water, condensate</li> </ul>	- Same as solid waste

#### Table 7. Characterization of wastes

#### **Population Equivalent**

The volume of organically contaminated industrial wastewater is often compared to that of domestic sewage in terms of the population equivalent (PE). The population equivalent is based on the amount of oxygen necessary to oxidize the domestic wastewater discharged by one person per day. Since the amount and concentration of organic pollutant vary per country, the quantitative definition of the PE varies nationally. The PE concept is used to calculate how much money industries must pay to the government in order to be allowed to discharge their wastewater in the public waters.

In the computation of the PE of the aqueous waste for this design project, we will use the American definition of a PE, that is, 76 BOD<sub>5</sub> /person/day and a BOD<sub>5</sub> of effluent water equal to 1,200 mg/liter. This value is typical in a process industry producing food products. It would be a good approximation since the nature of the wastewater is similar, the organic load is very large. The population equivalent of the wastewater produced is 6,780 people. This value is could affect the economic feasibility of the plant.

# **10 PROCESS SAFETY**

In this chapter all the aspects of process safety of the designed process are discussed. A summary of the hazardous components is given and also a preliminary HAZOP-analysis of fluid bed spray coating is presented. Finally the DOW F&EI for the designed process is evaluated.

# **10.1** SAFETY

The fermentation process is internally not a very dangerous reaction. The energy produced by the dissimilation of glucose (2843kJ/mole glucose) is coupled with the anabolism of micro-organism. Power input of the agitator is the major heat source of the fermenter. Concerning the whole designed process, the most vulnerable part must be drying and coating. The statistics of industrial accidents also show that drying should be regarded as a potentially hazardous operation that has brought a number of reported incidents with serious results for personnel and equipment. Drying is particularly prone to fire and explosion hazards because it is a process in which External heat sources are applied. Most of the materials are combustible. All possible ignition sources may be present in dryers. In this designed process, the major hazard is the dust explosion. Combustible materials which can form an explosible dust cloud in a confined space will explode if: enough oxygen is present; the dust concentration falls within the explosive limits; and an ignition source of sufficient energy, either thermal or electrical, is present.

The minimum explosible concentration (MEC) for most dust clouds is in the range 0.01 to 0.06 kg/m3 and is in most cases above 0.04 kg/m3. The minimum oxygen concentration required to support a dust explosion varies from 3% to 15% v/v, depending on the chemical nature of the dust, its particle size, its moisture content and its temperature. The explosibility characteristics of some dusts are listed in the summary of hazardous components below. There are several methods available in dust explosion protection. The venting system will be applied to the designed process.

# 10.2 HEALTH

Enzymes are proteins an are no different from other proteins, such as pollen or animal dander, in their potential to be respiratory allergens. Inhalation of the enzyme can stimulate the body's immune system to produce antibodies. Antibodies can be generated at very low levels of exposure to an allergen. The exact level of exposure that stimulates allergen production varies from person to person and is allergen-dependent. A second potential health effect of enzymes on workers is skin irritation. Protease catalyzes breakdown of protein strains and is potential skin irritant. Another risk to the health of workers is the presence of the fine powders in the workplace. There are some standards on the concentration of dusts in air. In the designed process, the TWA(time-weighted average) limit of cellulose powder in workplace air is 10 mg/m3 (USA), and that of starch is 1 mg/m3(CSK).

#### **10.3** Summary of hazardous components

Ammonium phosphate Not flammable. Irritating to eyes, nose and throat. May be dangerous if it enters water intakes. Threshold limit value(toxicity): 25 ppm as ammonia. LD50: not available. MAC: not available. Calcium acatate Flammable. Explosion danger. Irritating to eyes. MAC: not available.

Cellulose limits in workplace air(TWA): 10 mg/m3(USA). Limits in ambient air(annual): 10 mg/m3 Corn starch MEC: 0.045 kg/m3, MIT 400 C(cloud), MIE: 0.040 J, MOC: 11% Magnesium silicate (TALC) Not flammable. Irritating to skin and eyes. Harmful if swallowed. MAC: 2 mg/m3

#### PEG6000

No fire hazards at low temperature. Cleveland Open Cup flash point: 271 C. No appreciable irritation to eyes. LD50(g/kg): >50 (male rats)

Sodium sulfate not flammable. Irritating to eyes. MAC: not available.

Soya flour MEC: 0.045 kg/m3, MIT 530 C(cloud), 460 C(layer), MIE: 0.060 J, MOC: 15%

Titanium dioxide Not flammable. Irritating to eyes. MAC: 10 mg/m3

More data can be available in the Appendix.

# **10.1 HAZOP**

Table 8.1 Risk Analysis of Fluid Bed Spray Coating D102

Guide Word	Deviation	Possible Causes	Consequences	Action Required
Not, No	No flow s27	1) P106 fails (motor	Loss of feed of	a)Install low level alarm on FC
		fault, loss of drive,	coating agent.	b)Ensure good communication
		impeller corroded	Mechanical strength	with intermediate dryer operator
		away, etc.)	of final product	c)Install kickback on P106
			decreased.	Covered by a)
			No. 40 102	d)Institute regular patrolling and
		2) Line blockage,	As for 1)	inspection of transfer line
		valve closed in error		e) Install low level alarm on FC
		2) 7		for air
		3) Line fracture	As for 1), Polymer	Covered by e),
			solution discharged	I) Install relief valve on K103
			nito area aujacent to	
			pipenne	
	No flow s31	4)K103 fails (motor	Loss of pressure	
		fault, impeller	overhead, No	
		corroded away, etc.)	fluidization	
		5) line blockage,	As for 4),	
		control valve closed	overpressure on	
Maria	Mara flam 27	in error	compressor K103	a) Install bigh laws along an EC
WIDE	WIDE HOW S27	FC hypass pen in	of coating granules	b) Institute locking off procedure
		error	agglomerate	for FC when not in use
			466101101410	Covered by $g$ ) h) for FC on s31
				i) Install wire mesh on air outlet
	More flow s31			,
		7) FC for air fails	Coating overdried,	
		open or bypass pen in	flooding, blow-off of	
		error	granules	
	More pressure	8) Gas outlet	Transfer line	covered by f)
	D102	blockage	subjected to full pump	j) Install relief venting on D102
			delivery or surge	
	M	0 $OV = f = -1$	pressure	
	More	9) CV of cooling	Coating overdried,	k) Check CV of cooling water
	D102	water E105 closed III	octivity	regularly
Less	Less flow s27	10) Leaking flange or	Material loss adjacent	Covered by d)
Less	LC33 110 W 327	valve stub not	to pipeline	DCheck line and flange ratings
		blanked and leaking	to pipeline	ryeneek mie and nange ratings
	Less flow s31	11) Line fracture or	Inadequate mixing of	Covered by l)
		flange leak	granules with coating	m) adjust the place of atomizer
			material. wide PSD	
	Less	12) CV of cooling	High moisture content	Covered by k)
	Temperature	water E103 fails open	of coating, granules	

	D102	13) Winter conditions	agglomerate Polymer solution more viscous, smaller droplets, less spray formed by atomizer	n) K103 working continuously to keep D102 warm
As well as	Small size granules present	14) Sieve E105 fails	More dusts. Explosion risk	o) Spray of polymers before fluidization begins
Part of	High polymer concentration in s31	15) High polymer diversity of PEG6000	Polymer solution more viscous, smaller droplets, less spray formed by atomizer	Covered by n)

# 10.2 FEI

The calculation is based on the special form shown in Fig 9.2 of Coulson & Richardson's(1998).

*Units:* consider the total plant, no separate areas, but exclude the main storages. The fluid bed dryer would have the greatst impact on the magnitude of any fire or explosion.

Material factor: sodium sulfate is the most abundant material in the builders. Others are minor and none of them has significant fire or explosion hazards.

Estimated material factor for sodium sulfate is: 2.0

General process hazards:

Base factor = 1.0

A. not applicable

B. not applicable

C. not applicable

D. not applicable

E. penalty = 0.35

F. adequate drainage would be provided, factor = 0.0

Special process hazards:

A. sodium sulfate is not toxic, MAC-value not available, factor = 0.2

B. not applicable

C. not applicable

D. dust explosion hazard, factor = 1.5

E. Operation Presure 3 atm = 3\*14.7-14.7 = 29.4 psig. Set relief valve at 20% above the operating pressure = 35.6 psig.

F. not applicable

G. Not applicable

H. Stainless steel is the construction material, allow minimum factor = 0.1

I. Use minimum factor as full equipment details are not known at the flow-sheet stage, factor = 0.1

J. not applicable

K. not applicable

L. large compressors used, factor = 0.5

F1 = 1.35 F2 = 2.4Unit hazard factor F3 = F1 \* F2 = 3.24

Fire and Explosion Index FEI = MF \*F3 = 6.5

The index works out at 6.5, classified as "mild". Sodium sulfate is normally not a dangerous material and non-toxic. The danger of dust-explosion is the main process hazard.

# **11ECONOMY**

The purpose of this chapter is to present the economic evaluation of the project. From the estimated total capital investment and the total operating costs per year, the economic potential of the project is calculated. The detailed estimation of the total capital and the annual operating cost are shown in Appendix 3.2

## **11.1 INVESTMENT**

Table 8 shows the estimation of the Capital requirements.

Table 8. Estimate of Capital requirements	
1 Manufacturing Capital	Total Cost
Total Process Equipment	11,035,477.08
Total manufacturing capital based on (Lang)factor*	34209978.94
2. Nonmanufacturing Capital	
Proportionate share existing capital estimated at 15% manufacturing Capital"	5,131,496.84
3. Total Fixed Capital	
Sum of 1 and 2	39,341,475.78
4. Working Capital	
Raw Material	2192930.1
Store supplies and all other items at 5% gross sales"	1875000
Total Working Capital	4067930.1
5. Total Fixed and Working Capital	43,409,405.88
*Garrett Donald F : CHEMICAL ENGINEERING ECONOMICS New York Van Nostrand Reinhold 1989	

"Douglas, James M.: Conceptual design of chemical processes. New York McGraw-Hill 1988.

# **11.2 OPERATING COSTS**

Table 9 shows the Operating Cost .

Table 9. Operating Cost Summary							
ESTIMATED PRODUCTION COST AT AN	ENZYME-C	OATED GRANUL	ATION PLANT				
ENZYME-COATED GRANULE OUTPUT =	5,000 TON	NES PER YEAR					
PRODUCTION DELIVERED WITH 500,000	DUNITS PE	R GRAM (ACTIVIT	Y)				
TOTAL MANUFACTURING CAPITAL =		34209978.94					
TOTAL FIXED AND WORKING CAPITAL	-	43,409,405.88					
=							
QUANTITY UNIT PRICE COST PER YEA							
	UNIT	PER YEAR	(\$/Unit)	(\$/Year)			
RAW MATERIALS							
Ammonium phosphate	kg	465000.00	1.00	465000.00			
Biomass	kg	222000.00	5.00	1110000.00			
Glucose	kg	8100000.00	1.00	8100000.00			
Inerts	kg	6000000.00	1.00	600000.00			
Nutrients	kg	138000.00	1.00	138000.00			
Coating	kg	480000.00	2.00	960000.00			
Builder	kg	2633997.00	1.00	2633997.00			
Binder	kg	255000.00	6.00	1530000.00			
Water	ka	132307200.00	0.01	992304.00			
			Subtotal :	21929301.00			
DIRECT EXPENSE			oublotal !	21020001100			
Operating Labor	operators	18	100 000 00	1800000			
4.5 operators/shift: 4 shifts/day "			100,000.00	1000000			
Supervision							
10% of Operating labor "				180000			
Plant Overbead				100000			
50% of Operating Jabor"				00000			
Stoom	10AG kool	1000	10.00	900000			
Electricity	IU"O KCal	1022	10.00				
	<b>K</b> WN	20503136	0.10	2260923.76			
				100000 10			
Repaires@4%M.Cap.*	10101	00005		1368399.16			
water-cooling	10^6 Kcal	20325	1.00	509000			
Chilled Water	10^6 kcal	736	35.00	26000			
Water-process							
NaCl Brine	10^6 kcal	4365	28.00	124000			
Factory Supplies &				684199.58			
Laboratory@2% M. Cap."							
			Subtotal :	7862522.497			
INDIRECT EXPENSE							
Depreciation (plant life: 15 years)				2280665.262			
Taxes : 1% of M. Cap "				342099.7894			
Other indirect: 4% of M.Cap "				1368399.157			
			Subtotal :	3991164.209			
			TOTAL :	33782987.71			

"Douglas, James M.: Conceptual design of chemical processes. New York McGraw-Hill 1988.

### **11.3INCOME**

Quantity of Product per year	: 5 MKg of enzyme-coated granule per year
Price of Product	: US\$ 12.00 per Kg
Sales of Product	: M US\$ 60.00 per year

#### 11.4CASH FLOW

Total Operating Cost per year	: M US\$ 34.00 per year
Net Income per year	: M US\$ 26.00 per year

# **11.5ECONOMIC CRITERIA**

Rate of Return	(ROR)	: 59.88%	
Pay Out Time	(POT)	: 1.67 Years	
Discounted Cas	sh Flow	Rate of Return (DCFROR)	: 59.83%

## **11.6COST REVIEW**

In the Total Capital Investment, the total equipment costs and the installation costs tantamount to 79% of the total Investment. Majority of the investment is allotted to the installation cost, which is 3.1 times the equipment cost. This is based on the Lang factor for predominantly solids processing plant. From this assessment, it is suggested to better calculate/look for installation factor of each equipment, rather than using a more general lang factor; thereby placing the estimation of the capital investment in proximity to accuracy.

The price of the equipment is based from the Economic Evaluation Report of Super Pro Designer (software) from the simulation of the enzyme-coated granulation process. The prices are based on the year 1996, where the CE index is given at 381.7. To escalate the cost to the present date, the CE index of 389 as of may1999 is used. To have a more accurate estimation of the equipment costs, a vendor's quotation will give a much better if not best accuracy of the estimate.

# **12CONCLUSIONS AND RECOMMENDATIONS**

# 12.1 CONCLUSIONS

In this chapter, the most important results of this design project will be presented and conclusions will be made. For this design project entitled "Production of Coated Detergent Enzyme Granules", the following conclusions are drawn:

The demand for industrial enzymes for detergent has been increasing. In order to cope up with this, its productivity should be increased through efficient production processes at the same time preserving the enzyme activity. One area that should be looked into is the production of the enzymes. It has been found that the fermentation process to produce this enzyme is very dependent to various parameters like pH, oxygen concentration and the production of metabolites which repress the enzyme production. This exocellular enzyme, alkaline serine protease produce by Bacillus licheniformis, is repressed by high concentration of ammonia in the bioreactor. Several fermentation modes are available to minimize this repression namely, the batch, fed-batch and continuous operations. Fed-batch process is excellent in controlling the inhibitory effects of the critical elements but the activity of the enzyme produced is two times lower than the batch process. On the other hand, continuous process is considered as the most efficient, but maintaining a sterile condition for such a long time is a difficult condition to achieve in an industry. Thus, continuous process is hardly employed in industrial scale. For this design project, the batch process is chosen. Although the production is less compared to the other two, the activity of the enzyme is higher. The selection of the media for the cultivation of microorganism is also an important factor. It must contain all the necessary elements for synthesis of the cell substances and the production of metabolic products. For economic reasons, complex media are frequently used. In spite of its importance, there are only limited published literatures in this topic which has significantly affect the outcome of this design project.

The first downstream process employed is the separation of the solid mass from the fermentation broth. Vacuum drum filtration is chosen because of its simple operation compared to centrifugation which has too many operating variables. Cross-flow membrane filtration is used for the concentration of the enzyme concentrate. With proper selection of the membrane, 100% cell recovery can be obtained. The granulation of the enzyme is done using the Lodige High Shear Mixer/Granulator. It is a batch system where the builder, binder and enzyme concentrate are mixed under high shear conditions leading to the formation of granules. The drying of granules is performed in the fluidized bed dryer. This is preferred because a close control of conditions is possible allowing the predetermined moisture content to be met. Coating of enzymes is very necessary to protect them from the environment. A fluidized bed coating is chosen since it is found useful for forming and coating granules particularly smaller granules which are difficult to handle by other methods.

The above-mentioned processes are preferred for the preliminary plant design. Based on the results, the desired production rate of 5,000 tons of enzyme granules and activity of  $2.34 \times 10^6$  U/ml are achieved. The amounts of solid and liquid wastes produced are  $1.63 \times 10^4$  and  $1.29 \times 10^4$  per year respectively and  $2.13 \times 10^5$  tons per year of exhaust gases.

The whole production process is simulated using the Super Pro Designer 2.7 simulation package for bioprocesses. This software is very useful in simulating the preliminary process structure. Here, the simulation of the important equipment will be discussed briefly.

The fermentor is designed using the stoichiometric model. This is a simple model based on the stoichiometry of the reactions involved. There are more accurate but complicated options available in the software, but the availability of the kinetic data restricted our choice of the fermentor. The granulation and coating parts are simulated as mixing processes.

For the economic margin, the rate of return (ROR) is 59.88% and the pay out time (POT) is 1.67 years with the discounted cash flow rate of return (DCFROR) of 59.83%. The investment cost of the project is US \$ 44 million and the revenues is US \$ 26 million. This means that the project has very high economic potential.

# 12.2 RECOMMENDATIONS

This design project is limited by the availability of the published literature on the kinetics of the fermentor. It is therefore recommended to do more research in this area especially reaction involving a complex medium. These data greatly affect the result of the whole design project and make it more realistic. To the future teams who will be working in this area, more attention should be given in the optimization of the production parameters, e.g, the composition of the media to further reduce the quantity of raw materials which is economically advantageous.

# LIST OF SYMBOLS

Symbol	Meaning	Unit
N		r -31
IN V	Concentration of viable organisms	
Z Z	specific death rate constant	
l NT		[S <sup>-</sup> ]
IN <sub>0</sub>	Initial Concentration of viable organisms	
A	frequency factor	[s <sup>1</sup> ]
ΔE	Activation energy	[J mol <sup>-1</sup> ]
R	Gas constant	[J mol <sup>-</sup> 1 K <sup>-</sup> ']
T	Temperature	[K]
U	Average medium velocity	[m s <sup>-1</sup> ]
L	Length of holding tube	[m]
Dz	Axial dispersion coefficient	[m <sup>2</sup> s <sup>-1</sup> ]
Re	Reynold's Number	[-]
μ	viscosity	[kg m <sup>-1</sup> s <sup>-1</sup> ]
ρ	density	[kg m <sup>-3</sup> ]
dt	tube diameter	[m]
Fout <sub>k</sub>	is the mass flowrate of the limiting component after the reaction	[kg s <sup>-1</sup> ]
Fink	is the mass flowrate of the limiting component before the reaction	[kg s <sup>-1</sup> ]
х	is the reaction extent	[-]
Ai	stoichiometric coefficients	[-]
V	volume	[m <sup>2</sup> ]
F	volumetric flow rate	$[m^{3} s^{-1}]$
D	dilution rate	[s <sup>-1</sup> ]
Q <sub>f</sub>	heat released	[J m <sup>-3</sup> s <sup>-1</sup> ]
Q02	Oxygen uptake rate	$[mol m^{-3} s^{-1}]$
р	gas pressure	[Pa]
γ	ratio of specific heat of gas at constant pressure to specific heat of	
	gas at constant volume	[-]
ν	specific volume of gas	$[m^{3} kg^{-1}]$
Fi	fraction of component 'i'	[-]
CF	concentration factor	[-]
R <sub>i</sub>	rejection coefficient of component 'i'	[-]
C <sub>pri</sub>	concentration of component 'i' in permeate stream	[mol m <sup>-3</sup> ]
C <sub>ci</sub>	concentration of component 'i' in concentrate stream	[mol m <sup>-3</sup> ]
J	permeate flux	$[mol m^{-2} s^{-1}]$
А	area of membrane	$[m^2]$
V <sub>feed</sub>	volume of material fed to the ultrafilter per cycle	$[m^3]$
Q	Q is the total heat load	[J s <sup>-1</sup> ]
U	Overall heat transfer coefficient	$[J m^{-2} s^{-1} K^{-1}]$
$\Delta T$	mean temperature difference driving force	[K]
A <sub>c</sub>	surface area of the condenser	$[m^2]$
K	partition coefficient	[-]
xi	mole fraction of component 'i' in liquid phase	[-]
yi	mole fraction of component 'i' in vapour phase	[-]
zi	mole fraction of component 'i' in feed	[-]
F <sub>m</sub>	molar flow rate of feed	$[mol s^{-1}]$
Vm	molar flow rate of vapour phase	$[mol s^{-1}]$
L <sub>m</sub>	molar flow rate of liquid phase	$[mol s^{-1}]$

# LITERATURE

#### Chapter 1

- PU 1. Asenjo, J.A. and Merchuk, J.C. (1995). Bioreactor System Design, Marcel Dekker, Inc.
- 2. Schugerl, K. (1997). Bioreaction Engineering, v.3, John Wiley & Sons.
- HB463. Boyce, C.O.L. (1986). Novo's Handbook of Practical Biotechnology, Novo Industri A/S.
- 4. Cruguer, W. and Cruguer, A. (1989). Biotechnology: A Textbook of Industrial Microbiology, 2<sup>nd</sup> ed., Sinauer Associates, Inc.
- 5. Lee, J.M. (1992). Biochemical Engineering, Prentice Hall.
- PO36. Demain, A.L. and Solomon, N.A. (1985). Biology of Industrial Microorganisms, The Benjamin/Cummings Publishing Company, Inc.

#### Chapter 2

- PD4 1. Wang, D. I. C., C. L.Cooney, A. L.Demain, P. Dunnil, A. E. Humphrey, and M. D. Lilly (1979). Fermentation and Enzyme Technology, John Wiley & Sons.
- D<sup>G</sup> 2. Levenspiel, O. (1972). Chemical Reaction Engineering, 2nd edition, John Wiley & Sons.

#### Chapter 3

- Asenjo, J.A. and Merchuk, J.C. (1995). Bioreactor System Design, Marcel Dekker, Inc.
  - Z. Schugerl, K. (1997). Bioreaction Engineering, v.3, John Wiley & Sons.
  - 3. Boyce, C.O.L. (1986). Novo's Handbook of Practical Biotechnology, Novo Industri A/S.
  - A. Cruguer, W. and Cruguer, A. (1989). Biotechnology: A Textbook of Industrial Microbiology, 2<sup>nd</sup> ed., Sinauer Associates, Inc.
  - S. Lee, J.M. (1992). Biochemical Engineering, Prentice Hall.
- 6. Demain, A.L. and Solomon, N.A. (1985). Biology of Industrial Microorganisms, The Benjamin/Cummings Publishing Company, Inc.
- PD4.7. Kadam, K.L. (1991). Granulation Technology for Bioproducts, CRC Press.

#### Chapter 8

- PL 1. Cooney, C. L., D. I. C.Wang, and R. I. Mateles (1968). Measurements of Heat Evolution and Correlation with Oxygen Consumption during Microbial Growth. Biotechnol. Bioeng. 11, 269-281.
- HE 2. Shultz, J.M. (1962). The Polytropic Analysis of Centrifugal Compressors, J. Eng. Power, Trans. Am. Soc. Mech. Eng., pp. 69-82 (January 1962).
- Peters, M.S. and K.D. Timmerhaus, (1991). Plant Design and Economics for Chemical Engineers, 4th edition, McGraw-Hill, pp. 523-525.
- 4. M.C. McGregor (editor) (1986). Membrane Separations in Biotechnology, Marcel Dekker, Inc., New York and Basel.
- POHS. Kern D.Q (1965). Process Heat Transfer, McGraw-Hill.
- 6. Henley E. J. and J. D. Seader (1981). Equilibrium-Stage Separation Operations in Chemical Engineering, John Wiley & Sons.
- Theodore L., Buonicore A.J. (1988). Air Pollution Control Equipment, 2 (Gases), CRC Press.
- McCabe W. L., J. C. Smith, and P. Harriott. (1993). Unit Operations of Chemical Engineering, McGraw-Hill, 5th ed., pp. 798-800.
  - 9. McCabe W. L., J. C. Smith, and P. Harriott. (1993). Unit Operations of Chemical Engineering, McGraw-Hill, 5th ed., pp. 798-800.
- PL 10. Brauer, D., Biosafety in rDNA research and production, Biotechnology, 2<sup>nd</sup> ed., vol.12 pp. 63-113

#### Chapter 9

- *X.* Schugerl, K. (1997). Bioreaction Engineering, v.3, John Wiley & Sons.
- 2. Asenjo, J.A. and Merchuk, J.C. (1995). Bioreactor System Design, Marcel Dekker, Inc.
- 3. Perry's Chemical Engineer's Handbook

#### Chapter 11

- 1. Garrett, Donald E.: CHEMICAL ENGINEERING ECONOMICS. New York Van Nostrand Reinhold 1989.
- Douglas, James M.: Conceptual design of chemical processes. New York McGraw-Hill 1988

# **APPENDICES**

BLOCK SCHEMES (3.1) ECONOMIC EVALUATION (3.2) PROCESS FLOW SCHEME (5.1) BATCH CYCLE DIAGRAM (5.2) BATCH PROCESS STREAM SUMMARY (5.3) UTILITIES SUMMARY (5.4) PROCESS YIELDS (5.5) MASS AND HEAT BALANCES PROCESS STREAMS (7.1) PROCESS STREAM SUMMARY (7.2) EQUIPMENT SUMMARY AND SPECIFICATIONS (8) PROCESS SAFETY (10)

# **APPENDIX 3.1**

# BLOCK SCHEME OF THE PROCESS "DETERGENT ENZYME GRANULE PRODUCTION PLANT"



Conceptual Process Design Group 3237

Appendix 3.1

# APPENDIX 3.2 EQUIPMENT SCHEDULE

				Number of	Unit Cost (US\$)	Total Cost (US\$)	CE index	CE index	Present
lt	em No.	Equipment name	Size	Units	In SPD (1996)	In SPD (1996)	of 1996^	of 1999^	Price (US\$)
1	R101	Bioreactor (Stirred Tank)	289.16 m3	2	1,329,000.00	2,658,000.00	381.7	389	2708834.16
2	S103	Rotating Drum Vacuum Filter	97.16 m2	6	148,000.00	888,000.00	381.7	389	904982.97
3	S104	Hallow Fiber Membrane Filter	450.89 m2	4	161,000.00	644,000.00	381.7	389	656316.48
4	E102	Cooler	95.75 m2	5	16,000.00	80,000.00	381.7	389	81529.997
5	G101	Granulator (NIRO HEC 630)	30,000.00 L	1	-	-	-	-	800000.00
6	D101	Fluidized Bed Dryer	1.70 m3	1	9,000.00	9,000.00	381.7	389	9172.12
7	D102	Fluidized Bed Coating system	1.70 m3	1	9,000.00	9,000.00	381.7	389	9172.12
8	K101	Compressor	1315.27 kW	1	911,000.00	911,000.00	381.7	389	928422.85
9	K102	Compressor	0.17 kW	1	43,000.00	43,000.00	381.7	389	43822.37
10	K103	Compressor	50.80 kW	1	44,000.00	44,000.00	381.7	389	44841.50
11	S101	Air Filter	2.66 m3/s	2	22,000.00	22,000.00	381.7	389	22420.75
12	S102	Air Filter	2.67 m3/s	2	22,000.00	44,000.00	381.7	389	44841.50
13	T101	Blending Tank	282.35 m3	2	157,000.00	314,000.00	381.7	389	320005.24
14	E101	Heat Sterilizer	196.8 m (L)	3	857,000.00	2,571,000.00	381.7	389	2620170.29
15	E103	Cooler	6.60 m2	1	2,000.00	2,000.00	381.7	389	2038.25
16		Cost of unlisted Equipment at 20	% of Total						1838906.47

Total Price (US\$) : 11035477.08

^ Chemical Engineering Journal , May 1999 SPD - SuperPro Designer

# APPENDIX 3.2 MANUFACTURING CAPITAL

Item no.	<b>Delivered Cost</b>	Instalation Factor*	Total
R101	2,708,834.16	3.10	8397385.905
S103	904,982.97	3.10	2805447.21
S104	656,316.48	3.10	2034581.085
E102	81,530.00	3.10	252742.9919
G101	800000.00	3.10	2480000
D101	9172.12	3.10	28433.58659
D102	9172.12	3.10	28433.58659
K101	928,422.85	3.10	2878110.82
K102	43,822.37	3.10	135849.3581
K103	44,841.50	3.10	139008.6455
S101	22,420.75	3.10	69504.32277
S102	44,841.50	3.10	139008.6455
T101	320005.24	3.10	992016.2431
E101	2620170.29	3.10	8122527.901
E103	2038.25	3.10	6318.574797
Unlisted Equip	1838906.47	3.10	5700610.06
Total :	11,035,477.08	Total :	34209978.94

\*Coulson's and Richardson's chemical engineering. Vol. 6. An Introduction to chemical engineering design (SI units) / by R.K. Sinnott

# APPENDIX 3.2 WORKING CAPITAL

1. Raw Material 1 month supply @ 10% of Annual R.M.	Cost 2192930.1
2. Other, at 5% Gross Sales <i>5,000,000*7.5*0.05</i>	1875000
total :	4067930.1
# APPENDIX 3.2 ESTIMATE OF CAPITAL REQUIREMENTS

	Total Cost
1. Manufacturing Capital Total Process Equipment Total manufacturing capital based on (Lang)factor*	11,035,477.08 34209978.94
2. Nonmanufacturing Capital Proportionate share existing capital estimated at 15% manufacturing Capital"	5,131,496.84
3. Total Fixed Capital Sum of 1 and 2	39,341,475.78
4. Working Capital Raw Material Store supplies and all other items at 5% gross sales" Total Working Capital	2192930.1 1875000 4067930.1
5. Total Fixed and Working Capital	43,409,405.88

\*Garrett, Donald E.: CHEMICAL ENGINEERING ECONOMICS. New York Van Nostrand Reinhold 1989. "Douglas, James M.: Conceptual design of chemical processes. New York McGraw-Hill 1988.

### APPENDIX 3.2 UTILITIES SUMMARY

			Usage		Unit Price	Cost (\$/yr)
Utility	Item No.	Equipment name	Unit	Annual	]	
1.Cooling Water			Duty(kcal/h)	10^6 kcal	25US\$/10^6 kcal	
	R101	Bioreactor (Stirred Tank)	26211755.1	17125		428000
	K101	Compressor	373395.6	2661		67000
	E101	Heat Sterilizer	2935746.9	436		11000
	K102	Compressor	42	0		0
	K103	Compressor	14403.4	103		3000
					Subtotal (\$/yr):	509000
2 Electricity			Power (kW)	КМЛЬ	0.1115\$/KW/b	
2. Electricity	B101	Bioreactor (Stirred Tank)	1440	9405666	0.1035/1011	941000
	S104	Hallow Eiber Membrane Eilter	260.7	3403000		941000
	0104		300.7	321203		32000
	Dioi	Granulator (NIRO REC 630)	200.05	182982		18598.5
	DIOI	Fluidized Bed Dryer	0	15		1.5
	D102	Fluidized Bed Coating system	0	15		1.5
	K101	Compressor	1315.3	9371933		937000
	K102	Compressor	0.2	1228		123
	K103	Compressor	50.8	361950		36000
	1101	Blending Tank	480	855061		86000
		Unlisted Equipment		5% of Total		102536.225
		General Load		5% of Total		107663.036
					Subtotal (US\$/yr)	2260923.76
3.Steam			Duty (Kcal/h)	10^6 kcal	10US\$/10^6 kcal	
	D101	Fluidized Bed Dryer	11590.2	5		0
	E101	Heat Sterilizer	6850076	1017		10000
					Subtotal (US\$/yr)	10000
4. Chilled Water			Duty (Kcal/h)	10^6 kcal	35US\$/10^6 kcal	
	T101	Blending Tank	413010.1	736		26000
					Subtotal (US\$/yr)	26000
5. NaCl Brine			Duty (Kcal/h)	10^6 kcal	28US\$/10^6 kcal	
	S104	Hallow Fiber Membrane Filter	24111.2	21		1000
	E102	Cooler	14630664.3	4344		122000
	E103	Cooler	291362.3	43		1000
					Subtotal (US\$/yr):	124000

Total (US\$/yr) 2929923.76

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## APPENDIX 3.2 OPERATING COST SUMMARY

ESTIMATED PRODUCTION COST AT AN ENZYME-COATED GRANULATION PLANT									
PRODUCTION DELIVERED WITH 500 000 LINITS PER GRAM (ACTIVITY)									
TOTAL MANUFACTURING CAPITAL = 34209978.94									
TOTAL FIXED AND WORKING CAPITAL =									
	UNIT PRICE	COST PER YEAR							
	UNIT	PER YEAR	(\$/Unit)	(\$/Year)					
RAW MATERIALS									
Ammonium phosphate	kg	465000.00	1.00	465000.00					
Biomass	kg	222000.00	5.00	1110000.00					
Glucose	kg	8100000.00	1.00	8100000.00					
Inerts Nutriopto	kg	128000.00	1.00	6000000.00					
Coating	kg	136000.00	1.00	136000.00					
Builder	ka	2633997.00	1.00	2633997.00					
Binder	ka	255000.00	6.00	1530000.00					
Water	ka	132307200.00	0.00	992304 00					
	Ng	102007200.00	Subtotal :	21929301.00					
DIRECT EXPENSE			ouplottin.	21020001100					
Operating Labor 4.5 operators/shift; 4 shifts/day "	operators	18	100,000.00	1800000					
10% of Operating labor "				180000					
50% of Operating labor "				900000					
Steam	10^6 kcal	1022	10.00	10000					
Electricity	KWh	20503136	0.10	2260923.76					
Compressed Air Repaires@4%M.Cap."				1368399.16					
Water-cooling	10^6 kcal	20325	1.00	509000					
Chilled Water	10^6 kcal	736	35.00	26000					
water-process									
NaCl Brine	10^6 kcal	4365	28.00	124000					
Factory Supplies &				684199.58					
Laboratory@2% M. Cap."				7000500 407					
INDIRECT EXPENSE			Subtotal :	/862522.497					
Depreciation (plant life: 15 years) Taxes : <i>1% of M. Cap</i> " Other indirect: <i>4% of M.Cap</i> "			Subtotal :	2280665.262 342099.7894 1368399.157 <b>3991164.209</b>					
	I		TOTAL :	33782987.71					

"Douglas, James M.: Conceptual design of chemical processes. New York McGraw-Hill 1988.

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## APPENDIX 3.2 WORKING TABLE FOR DISCOUNTED CASH FLOW (DFC) CALCULATIONS

						% Interest = 59.83				
	Cash Flo	w (US\$)	Cash Flow (US\$)		Net Future					
Year	Cap. Cost(US\$)	Accumulation	Cash Flow (US\$) Accumulation		Value	<b>Discount Factor</b>	Present Value (US\$)			
0	43409405.88	43409405.88			-43409405.88	1.0000	-43409405.88			
1			26,217,012.29	26,217,012.29	-17192393.58	0.6257	16403060.94			
2			26,217,012.29	52434024.59	35241631	0.3915	10262817.33			
3			26,217,012.29	78651036.88	113892667.9	0.2449	6421083.23			
4			26,217,012.29	104868049.2	218760717.1	0.1532	4017445.56			
5			26,217,012.29	131085061.5	349845778.5	0.0959	2513574.14			
6			26,217,012.29	157302073.8	507147852.3	0.0600	1572654.79			
7			26,217,012.29	183519086.1	690666938.3	0.0375	983954.69			
8			26,217,012.29	209736098.3	900403036.7	0.0235	615625.79			
9			26,217,012.29	235953110.6	1136356147	0.0147	385175.36			
10			26,217,012.29	262170122.9	1398526270	0.0092	240990.66			
						total net PV=	6976.60			
Total Inves	tment			US \$44 million						
Total Opera	ating cost per year			US \$ 34 million						
Gross inco	me ber vear	=	total amount of pro	duct per vear*pric	e of product					

Total Operating cost per	year	05 \$ 34 millior	US \$ 34 million				
Gross income per year	=	total amount of product per year*p	total amount of product per year*price of product				
	=	5,000,000kg/year * 12 US\$/kg	=	US \$ 60 million			
Net Income	=	Gross Income - Operating Cost	=	US \$ 26 million			

Payout Period (years): Total Investment/ Net Income = 1.655772419

DCFROR (%) : 59.83

Rate of Return (ROR) : 1/(POT) = 0.5988

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Appendix 3.2-7



Appendix 5-1 Process Flow Sheet

Production of coated detergent enzyme granule



Production of coated detergent enzyme granul

#### Production of coated detergent enzyme granule

Production of coated detergent enzyme gra

#### Production of coated detergent enzyme granu

CPD group 323Delft University of Technology

Appendix 5.2-5

Production of coated detergent enzyme granules

#### **APPENDIX 5.3**

#### BATCH REMARKS AND CALCULTATION

	Ср
COMPONENT	[kJ/kgK]
Ammonium phosphate	0.900
Biomass	0.700
Enzyme	0.540
Glucose	1.220
Inerts	1.330
Nutrients	0.730
Coating	1.400
Builder	1.220
Binder	1.340
Water	4.184
Nitrogen	1.050
Oxygen	0.920
Carbon dioxide	0.940

Operation		Time [h]		
	Name	Cycle	Turnaround	Remarks
Mixing T101	tx1	6.00	1.00	
Sterilization E101	ts	0.50	0.00	
Fermentation R101	tf	22.00	2.00	includes 1 hr filling, 20 hr reaction and 1 hr discharging
Blowing air K101	tca1	20.00	0.00	simultaneously with reaction time
Filtration air S101	tfa1	20.00	0.00	simultaneously with reaction time
Filtration gas S102	tfa2	20.00	0.00	simultaneously with reaction time
Cooling E101	tc1	1.00	0.00	
Filtration discharge E102	trd	3.00	1.00	
Ultrafilt. Discharge S103	tuf	3.00	1.00	simultaneously with trd
Granulation G101	tgr	2.67	1.00	
Screening S105	tsc	0.17	1.00	simultaneously with tgr
Crushing A101	tcr	0.17	1.00	simultaneously with tgr
Blowing air K102	tca2	0.50	0.00	simultaneously with tfb1
Drying D101	tfb1	1.50	1.00	includes 0.5 hr filling, 0.5 hr coating, 0.5 hr discharging
Compression K103	tca3	0.50	0.00	simultaneously with tfb2
Cooling E103	tc2	0.50	0.00	simultaneously with tfb2
Coating D102	tfb2	1.50	1.00	includes 0.5 hr filling, 0.5 hr coating, 0.5 hr discharging
Total	tt	35.50		

#### APPENDIX 5.3 BATCH PROCESS STREAM SUMMARY

STREAM Nr.		1,2-3		3-4			
Name:		Mixin	g substrate+w	ater		Sterilization	
COMPONENT	MW	Cycle Cycle Process		Cycle	Cycle	Process	
		kg	kg/h	kg/h avg	kg	kg/h	kg/h avg
Ammonium phosphate	132.07	1550.00	258.33	43.66	1550.00	3100.00	43.66
Biomass	147.00	740.00	123.33	20.84	740.00	1480.00	20.84
Enzyme	138.48						
Glucose	180.00	27000.00	4500.00	760.49	27000.00	54000.00	760.49
Inerts	5000.00	20000.00	3333.33	563.33	20000.00	40000.00	563.33
Nutrients	229.03	460.00	76.67	12.96	460.00	920.00	12.96
Coating	6000.00						
Builder	142.05						
Binder	7000.00						
Water	18.00	440000.00	73333.33	12393.20	440000.00	880000.00	12393.20
Nitrogen	28.00						
Oxygen	32.00						
Carbon dioxide	44.00						
TOTAL		489750.00	81625.00	13794.48	489750.00	979500.00	13794.48
Enthalpy	MJ/MW/MW	0.000	0.000	0.000	22832.986	12.685	0.179
Phase			L			L	
Pressure	Bara		1.01			1.01	
Temperature	°C		25.00			37.00	
Cycle times	[h]						
Cycle & Process			6.00	35.50		0.50	35.50

STREAM Nr.			4,7-8,10			10-11	
Name:			Fermentation			Cooling	
COMPONENT	MW	Cycle	Cycle	Process	Cycle	Cycle	Process
		kg	kg/h	kg/h avg	kg	kg/h	kg/h avg
Ammonium phosphate	132.07	1550.00	70.45	43.66	77.50	77.50	2.18
Biomass	147.00	740.00	33.64	20.84	5088.56	5088.56	143.33
Enzyme	138.48				4079.87	4079.87	114.92
Glucose	180.00	27000.00	1227.27	760.49	10401.23	10401.23	292.96
Inerts	5000.00	20000.00	909.09	563.33	20000.00	20000.00	563.33
Nutrients	229.03	460.00	20.91	12.96	460.00	460.00	12.96
Coating	6000.00					0.00	0.00
Builder	142.05					0.00	0.00
Binder	7000.00					0.00	0.00
Water	18.00	440000.00	20000.00	12393.20	447288.26	447288.26	12598.49
Nitrogen	28.00	521053.82	23684.26	14676.19		0.00	0.00
Oxygen	32.00	158181.84	7190.08	4455.41		0.00	0.00
Carbon dioxide	44.00					0.00	0.00
TOTAL		1168985.66	53135.71	32926.08	487395.41	487395.41	13728.16
Enthalpy	MJ/MW/MW	31144.591	0.393	0.244	-38338.285	-10.650	-0.300
Phase			L+S+G			L	
Pressure	Bara		1.01/6.01			1.01	
Temperature	°C		37.00			5.00	
Cycle times	[h]						
Cycle & Process			22.00	35.50		1.00	35.50

STREAM Nr.		1	6,17,21,22-19			18,24-19,25		
Name:		Gra	anulation Circu	uit	Drying			
COMPONENT	MW	Cycle Cycle Process		Cycle	Cycle	Process		
		kg	kg/h	kg/h avg	kg	kg/h	kg/h avg	
Ammonium phosphate	132.07	0.64	0.24	0.02	0.64	0.43	0.02	
Biomass	147.00							
Enzyme	138.48	3975.11	1488.80	111.96	3975.11	2650.07	111.96	
Glucose	180.00	4835.20	1810.94	136.19	4835.20	3223.47	136.19	
Inerts	5000.00							
Nutrients	229.03							
Coating	6000.00							
Builder	142.05	6510.19	2438.27	183.37	6510.19	4340.13	183.37	
Binder	7000.00	867.35	324.85	24.43	867.35	578.23	24.43	
Water	18.00	3694.05	1383.54	104.05	3694.05	2462.70	104.05	
Nitrogen	28.00				16.84	11.23	0.47	
Oxygen	32.00				5.11	3.41	0.14	
Carbon dioxide	44.00							
TOTAL		19882.54	7446.64	560.02	19904.49	13269.66	560.64	
Enthalpy	MJ/MW/MW	-224.986	-0.023	-0.002	489.436	0.091	0.999	
Phase			L			L		
Pressure	Bara		1.01			1.01		
Temperature	°C		18.10			40.00		
Cycle times	[h]							
Cycle & Process			2.67	35.50		1.50	35.50	

#### APPENDIX 5.3 BATCH PROCESS STREAM SUMMARY

STREAM Nr.		26,27-28 28,31-32-33						
Name:		Mi	xing for Coatin	g	Coating			
COMPONENT	MW	Cycle Cycle Process		Cycle	Cycle	Process		
		kg	kg/h	kg/h avg	kg	kg/h	kg/h avg	
Ammonium phosphate	132.07	0.63	1.26	0.02	0.63	1.26	0.02	
Biomass	147.00							
Enzyme	138.48	3895.60	7791.20	109.72	3895.60	7791.20	109.72	
Glucose	180.00	4738.50	9477.00	133.47	4738.50	9477.00	133.47	
Inerts	5000.00							
Nutrients	229.03							
Coating	6000.00	1600.00	3200.00	45.07	1600.00	3200.00	45.07	
Builder	142.05	8779.99	17559.98	247.30	8779.99	17559.98	247.30	
Binder	7000.00	850.00	1700.00	23.94	850.00	1700.00	23.94	
Water	18.00	4589.66	9179.32	129.27	4589.66	9179.32	129.27	
Nitrogen	28.00				22135.80	44271.60	623.49	
Oxygen	32.00				6720.00	13440.00	189.28	
Carbon dioxide	44.00							
TOTAL		24454.38	48908.76	688.79	53310.18	106620.36	1501.55	
Enthalpy	MJ/MW/MW	863.521	0.480	0.007	-353.019	-0.196	-0.003	
Phase			L+S			L+S		
Pressure	Bara		1.01			1.01		
Temperature	°C		45.97			20.00		
Cycle times	[h]							
Cycle & Process			0.50	35.50		0.50	35.50	

### APPENDIX 5.4 SUMMARY OF UTILITIES

	EQUIPMENT	HEA	TING		COO	LING		PO	WER	
		Load	Consumption	Load	Co	nsumption [t	/h]			
Nr.	Name	[kW]	LP steam	[kW]	Cooling	NaCl brine	Chilled	Actual load	Electr.	REMARKS
			[t/h]		Water		Water	[kW]	[kWh/h]	
T101	Blending tank	480.00					82342.41			Chilled water: Tin = 5°C; Tout = 10°C; Cp = 4.2 J/kg/K
E101	Sterilizer		235416.19		155480.00					Pressure = 1 bara, T superh. = 190°C; T condens. = 133.5°C; Cp = 2 J/kg/K
R101	Fermentor	1440.00			135253.33					Cooling water: Tin = 20°C; Tout = 40°C; Cp = 4.2 J/kg/K
K101	Compressor	1315.30			18607.52					Tin = 5°C; Tout = 10°C; Cp = 4.2 J/kg/K
S101	Air filter									
S102	Air filter									
E102	Cooler					270035.11				
S103	Rotary Vacuum Filter									
S104	Ultrafilter	177.90				1258.82				NaCl brine : Tin = -9°C; Tout = 0°C; Cp = 4.2 J/kg/K
G101	Granulator	149.20								
S105	Screens	20.00								
A101	Crusher	37.30								
K102	Compressor	0.20			2.09					
D101	Fluid Bed Dryer		429.43							
K103	Compressor	50.80			717.77					
E103	Cooler					32265.68				
D102	Fluid Bed Dryer									
	105									
TOTAL		3670.70	235845.63	0.00	310060.71	303559.60	82342.41	0.00	0.00	

#### APPENDIX 5.5 PROCESS YIELDS

			Process Strea	ms			
Name	Ref. Name	kg/da	ау	t/h		t/t granul	
		IN	OUT	IN	OUT	IN	OUT
Substrate	<1>	49750.000		2.073		2.034	
Water	<2>	440000.000		18.333		17.993	
Air for R101	<5>	679235.662		28.301		27.776	
Wash water	<12>	24.000		0.001		0.001	
Builder+Binder	<17>	7229.990		0.301		0.296	
Air for D101	<23>	21.954		0.001		0.001	
Coating	<27>	5000.000		0.208		0.204	
Air for D102	<29>	28855.800		1.202		1.180	
Discharge S102	<9>		681590.248		28.400		27.872
Solid waste	<13>		46431.660		1.935		1.899
Liquid waste S104	<15>		428732.232		17.864		17.532
Top D101	<25>		53.090		0.002		0.002
Top D102	<32>		28855.800		1.202		1.180
Enzyme granules	<33>		24454.380		1.019		1.000
TOTAL		1210117.405	1210117.410	50.422	50.422	49.485	49.485

Utilities											
Name	Ref. Name	t/h	kW	t/t granul.	kWhh/						
					t granul.						
LP steam		235845.627		231463.445							
NaCI Brine		303559.603		297919.247							
CW		310060.708		304299.557							
Chilled water		82342.414		80812.432							
Electricity			3670.700		3602.496						



Figures in italics: Vh Figures in bold font: Vton granules

		IN						OUT		
Plant		EC	UIPMENT	0	EQUIPM.	0	EQUIPMENT	11	Plan	nt
Mass	Heat	Mass	Heat	Stream	IDENTIF.	Stream	Mass	Heat	Mass ko/day	Heat
49750 00	0.00	49750.00	0.00	<1> 1	T101	(3)	489750 00	0.00	ky/uay	NVV
440000 00	0.00	440000 00	0.00	<2>		102	400700.00	0.00		1
110000.00	480.00		480.00		Operation					
			0.000		Duty			480.33		480.33
		489750.00	480.00		Total		489750.00	480.33		
		489750.00	0.00	<3>	E101	<4>	489750.00	264.27		
			7000 04		Operation			0005 77		0005 77
	7268.84	490750.00	7268.84		Duty		480750.00	3995.77		3995.77
679235 66	0.00	679235.66	0.00	(5)	K101	(6)	679235.66	120 25		
079233.00	1315.30	079255.00	1315.30	<b>N</b>	Operation	102	073233.00	120.20		
					Duty			434.26		434.26
		679235.66	1315.30		Total		679235.66			
		679235.66	120.25	<6>	S101	<7>	679235.66	120.25		
					Operation					
		070005 00	100.05		Duty		070005 00	100.05		
		6/9235.66	120.25	-12	I Otal	192	6/9235.66	120.25		
		679235.66	120 25	<7>	RIVI	<10>	487395 41	266.24		
	1440.00	073255.00	1440.00		Operation		407000.41	200.24		
					Duty			3155.91		3155.91
		1168985.66	1824.52		Total		1168985.66	3518.67		
		681590.25	96.52	<8>	S102	<9>	681590.25	96.52	681590.25	96.52
					Operation					
					Duty					
		681590.25	96.52	.10	Total		681590.25	96.52		
		487395.41	266.24	<10>	E102	<11>	487395.41	-443.73		
					Duty			2835 01		2835 91
		487395 41	266 24		Total		487395.41	2392.18		2000.01
		487395.41	-443.73	<11>	S103	<13>	46431.66	-26.80	46431.66	-26.80
24.00	0.00	24.00	0.00	<12>		<14>	440987.75	-416.95		
					Operation					
					Duty					
		487419.41	-443.73		Total		487419.41	-443.75		
		440987.75	-416.95	<14>	S104	<15>	428732.23	-370.45	428732.23	-370.45
	177.00		177.00		0	<16>	12255.52	-4.80		
	177.90		177.90		Operation			12.22		13.22
		440987 75	-239.05		Total		440987 75	-362 04		13.22
		12255 52	-4.80	<16>	G101	<18>	19882 54	-2 60		
7229.99	0.00	7229.99	0.00	<17>			10002.04	2.00		
		198.51	0.06	<21>						
		198.51	0.06	<22>						
	149.20		149.20		Operation					
					Duty					
		19882.54	144.51	-105	l otal	-265	19882.54	-2.60		
		19031.41	5.04	(15)	3105	~20>	19454.55	0.06		
						<22>	198.51	0.06		
	20.00		20.00		Operation	Sec.	100.01	0.00		
					Duty		1			
		19851.41	25.64		Total		19851.41	5.64		
		198.51	0.06	<20>	A101	<21>	198.51	0.06		
	37.30		37.30		Operation					
		100.54	07.00		Duty		100.54	0.00		
01.05	0.00	198.51	37.36	-22-	l otal	-04-	198.51	0.06		
21.95	0.00	21.90	0.00	(23)	Operation	\$242	21.95	0.00		
	0.20		0.20		Duty			0.05		0.05
		21.95	0.20		Total		21.95	0.05		0.00
		19882.54	-2.60	<18>	D101	<25>	53.09	0.03	53.09	0.03
		21.95	0.00	<24>		<19>	19851.41	5.64		
			2000 00000		Operation					
	13.48	10051	13.48		Duty		10001100			
00055.00		19904.49	10.88	-00	Total		19904.50	5.66		
28855.80	0.00	20055.80	0.00	<23>	Operation	<30>	20055.80			
	50.80		50.80		Duty			16 75		16 75
		28855 80	50.80		Total		28855 80	16.75		10.75
		28855.80	5.11	<30>	E103	<31>	28855.80	-1.70		
			0.71		Operation					
					Duty			338.85		338.85
		28855.80	5.11		Total		28855.80	337.15		
		19851.41	5.64	<26>	D102	<32>	28855.80	-1.70	28855.80	-1.70
5000.00	3.79	5000.00	3.79	<27>		<33>	24454.38	-2.38	24454.38	-2.38
		28855.80	-1.70	<31>	Onerting					
					Operation					
		53707 21	7 72		Total		53310.18	-4 09		
1210117.41	10956.81	00/07.21	1.12		, orai		1 00010.10	-4.03	1210117.41	10966.26
• • • • • • • • • • • • • • • • • • •						•				
OUTIN									0.00	0.45

#### APPENDIX 7.1 MASS & HEAT BALANCE TOTAL STREAMS

#### APPENDIX 7.2 PROCESS STREAM SUMMARY

STREAM Nr.		1	IN	2	IN	3 :	=1+2	4 :	=3
Name:		Substrate	+Biomass	Wa	ter	Dischar	ge T101	Dischar	ge E101
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	1550.00	11.74	0.00	0.00	1550.00	11.74	1550.00	11.74
Biomass	147.60	740.00	5.01	0.00	0.00	740.00	5.01	740.00	5.01
Enzyme	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Glucose	180.00	27000.00	150.00	0.00	0.00	27000.00	150.00	27000.00	150.00
Inerts	600.00	20000.00	33.33	0.00	0.00	20000.00	33.33	20000.00	33.33
Nutrients	229.03	460.00	2.01	0.00	0.00	460.00	2.01	460.00	2.01
Coating	2434.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Builder	1080.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Binder	9428.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Water	18.00	0.00	0.00	440000.00	24444.44	440000.00	24444.44	440000.00	24444.44
Nitrogen	28.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Oxygen	32.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TOTAL		49750.00	202.09	440000.00	24444.44	489750.00	24646.54	489750.00	24646.54
Enthalpy	kW	0.0	000	0.0	00	0.0	000	264	.271
Phase		L	-	L	-		L		L
Pressure	Bara	1.0	01	1.0	01	1.	01	1.	01
Temperature	°C	25.	.00	25.	00	25	.00	37	.00
Activity	U/ml	(	)	C	)		0		0

STREAM Nr.		5	IN	6 =	=5	7 =	=6	8	Calc
Name:		Air for	R101	Discharg	ge K101	Dischar	ge S101	Gas	R101
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Biomass	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Enzyme	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Glucose	180.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Inerts	600.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nutrients	229.03	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Coating	2434.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Builder	1080.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Binder	9428.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Water	18.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nitrogen	28.00	521053.82	18609.06	521053.82	18609.06	521053.82	18609.06	521053.82	18609.06
Oxygen	32.00	158181.84	4943.18	158181.84	4943.18	158181.84	4943.18	152525.18	4766.41
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	0.00	0.00	8011.25	182.07
TOTAL		679235.66	23552.25	679235.66	23552.25	679235.66	23552.25	681590.25	23557.55
Enthalpy	kW	0.0	000	120.	249	120	249	96.	522
Phase		(	G	G	à	(	3	(	3
Pressure	Bara	1.01		3.01		3.01		1.01	
Temperature	°C	25.00		40.00		40.00		37.00	
Activity	U/ml	0		0		0		0	

STREAM Nr		9.	-8	10	Calc	11 -	-10	12	IN
Name:		Dischar	00 \$102	Bottor	P101	Dischar	TO E102	Wash	water
COMPONIENT	A AVA/	bischar	ye STU2	bollon ka/day	kmal/day	Discitat	ye LTUZ	ka/day	water kmol/dov
COMPONENT	171.6.4	kg/uay	kmol/day	kg/uay	kmovday	kg/uay	kmol/day	kg/uay	KINOVUAY
Ammonium phosphate	132.07	0.00	0.00	77.50	0.59	77.50	0.59	0.00	0.00
Biomass	147.60	0.00	0.00	5088.56	34.48	5088.56	34.48	0.00	0.00
Enzyme	147.60	0.00	0.00	4079.87	27.64	4079.87	27.64	0.00	0.00
Glucose	180.00	0.00	0.00	10401.23	57.78	10401.23	57.78	0.00	0.00
Inerts	600.00	0.00	0.00	20000.00	33.33	20000.00	33.33	0.00	0.00
Nutrients	229.03	0.00	0.00	460.00	2.01	460.00	2.01	0.00	0.00
Coating	2434.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Builder	1080.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Binder	9428.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Water	18.00	0.00	0.00	447288.26	24849.35	447288.26	24849.35	24.00	1.33
Nitrogen	28.00	521053.82	18609.06	0.00	0.00	0.00	0.00	0.00	0.00
Oxygen	32.00	152525.18	4766.41	0.00	0.00	0.00	0.00	0.00	0.00
Carbon dioxide	44.00	8011.25	182.07	0.00	0.00	0.00	0.00	0.00	0.00
TOTAL		681590.25	23557.55	487395.41	25005.18	487395.41	25005.18	24.00	1.33
Enthalpy	kW	96.	522	266.	.238	-443	.730	0.0	00
Phase			G	ι	-	1	_	L	-
Pressure	Bara	1.	01	1.0	01	1.	01	1.0	01
Temperature	°C	37	.00	37.	.00	5.	00	25.	.00
Activity	U/ml		0	844:	30.7	844	30.7	(	)

#### APPENDIX 7.2 PROCESS STREAM SUMMARY

STREAM Nr.		13	OUT	14 :	=11+12-13	15	OUT	16 =	=14-15
Name:		Solid	waste	Discharg	ge S103	Liquid wa	ste S104	Discharg	ge S104
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	3.50	0.03	74.00	0.56	73.37	0.56	0.63	0.00
Biomass	147.60	5088.56	34.48	0.00	0.00	0.00	0.00	0.00	0.00
Enzyme	147.60	184.26	1.25	3895.60	26.39	0.00	0.00	3895.60	26.39
Glucose	180.00	469.75	2.61	9931.48	55.17	5192.98	28.85	4738.50	26.32
Inerts	600.00	20000.00	33.33	0.00	0.00	0.00	0.00	0.00	0.00
Nutrients	229.03	460.00	2.01	0.00	0.00	0.00	0.00	0.00	0.00
Coating	2434.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Builder	1080.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Binder	9428.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Water	18.00	20225.58	1123.64	427086.67	23727.04	423465.88	23525.88	3620.79	201.16
Nitrogen	28.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Oxygen	32.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TOTAL		46431.66	1197.35	440987.75	23809.17	428732.23	23555.29	12255.52	253.88
Enthalpy	kW	-26.	.805	-416	.948	-370	.455	-4.	799
Phase		L		L	_			1	_
Pressure	Bara	1.01		1.0	01	1.01		1.01	
Temperature	°C	5.00		5.00		7.00		7.00	
Activity	U/ml	40643.5		89713.3		0		8971333.5	

									the second se
STREAM Nr.		17	IN	18	=16+17+21+22	19 =	=20+22+26	20	Calc
Name:		Builder-	+Binder	Discharg	ge G101	Botton	n D101	Oversiz	e S105
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	0.00	0.00	0.64	0.00	0.64	0.00	0.01	0.00
Biomass	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Enzyme	147.60	0.00	0.00	3975.11	26.93	3975.10	26.93	39.75	0.27
Glucose	180.00	0.00	0.00	4835.20	26.86	4835.20	26.86	48.35	0.27
Inerts	600.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nutrients	229.03	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Coating	2434.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Builder	1080.00	6379.99	5.91	6510.19	6.03	6510.19	6.03	65.10	0.06
Binder	9428.00	850.00	0.09	867.35	0.09	867.35	0.09	8.67	0.00
Water	18.00	0.00	0.00	3694.05	205.22	3662.92	203.50	36.63	2.03
Nitrogen	28.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Oxygen	32.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TOTAL		7229.99	6.00	19882.54	265.14	19851.41	263.41	198.51	2.63
Enthalpy	kW	0.0	000	-2.6	504	5.6	538	0.0	056
Phase			S	S	+L	S	+L	S	+L
Pressure	Bara	1.	01	1.	01	1.	01	1.	01
Temperature	°C	25.00		18.10		40.00		40.00	
Activity	U/ml		0	335	1524	335	1524	335	1524

STREAM Nr	1	21	Calc	22	Calc	23	IN	24 =	23
Name:		Dischar	ae A101	Undersiz	e S105	Air for	D101	Discharg	ge K102
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	0.01	0.00	0.01	0.00	0.00	0.00	0.00	0.00
Biomass	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Enzyme	147.60	39.75	0.27	39.75	0.27	0.00	0.00	0.00	0.00
Glucose	180.00	48.35	0.27	48.35	0.27	0.00	0.00	0.00	0.00
Inerts	600.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nutrients	229.03	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Coating	2434.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Builder	1080.00	65.10	0.06	65.10	0.06	0.00	0.00	0.00	0.00
Binder	9428.00	8.67	0.00	8.67	0.00	0.00	0.00	0.00	0.00
Water	18.00	36.63	2.03	36.63	2.03	0.00	0.00	0.00	0.00
Nitrogen	28.00	0.00	0.00	0.00	0.00	16.84	0.60	16.84	0.60
Oxygen	32.00	0.00	0.00	0.00	0.00	5.11	0.16	5.11	0.16
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TOTAL		198.51	2.63	198.51	2.63	21.95	0.76	21.95	0.76
Enthalpy	kW	0.0	056	0.0	56	0.0	000	0.0	04
Phase		S	+L	S-	۰L	(	G	(	3
Pressure	Bara	1.	01	1.0	01	1.	01	3.	01
Temperature	°C	40.00		40.00		25.00		40	.00
Activity	U/ml	335	1524	3351	524		0	(	)

#### APPENDIX 7.2 PROCESS STREAM SUMMARY

STREAM Nr.		25	OUT	26 .	16+17+23-26	27	IN	28 -	=26+27
Name:		Top	D101	Dischard	ge S105	Coa	ting	Mixture	for D102
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	0.00	0.00	0.63	0.00	0.00	0.00	0.63	0.00
Biomass	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Enzyme	147.60	0.00	0.00	3895.60	26.39	0.00	0.00	3895.60	26.39
Glucose	180.00	0.00	0.00	4738.50	26.33	0.00	0.00	4738.50	26.33
Inerts	600.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nutrients	229.03	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Coating	2434.00	0.00	0.00	0.00	0.00	1600.00	0.66	1600.00	0.66
Builder	1080.00	0.00	0.00	6379.99	5.91	2400.00	2.22	8779.99	8.13
Binder	9428.00	0.00	0.00	850.00	0.09	0.00	0.00	850.00	0.09
Water	18.00	31.14	1.73	3589.66	199.43	1000.00	55.56	4589.66	254.98
Nitrogen	28.00	16.84	0.60	0.00	0.00	0.00	0.00	0.00	0.00
Oxygen	32.00	5.11	0.16	0.00	0.00	0.00	0.00	0.00	0.00
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TOTAL		53.09	2.49	19454.38	258.15	5000.00	58.44	24454.38	316.58
Enthalpy	kW	0.0	027	5.5	26	3.7	'88	9.9	994
Phase		(	G	S-	۴L	L-	⊦S	L	+S
Pressure	Bara	1.01		1.01		1.01		1.01	
Temperature	°C	40.00		40.00		60.00		45.97	
Activity	U/mi		0	3360	590.1	(	0	2342	886.4

STREAM Nr.		29	IN	30 =	=29	31 =	=30	32	OUT
Name:		Air for	D102	Discharg	je K103	Dischar	ge E103	Top I	D102
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Biomass	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Enzyme	147.60	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Glucose	180.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Inerts	600.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nutrients	229.03	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Coating	2434.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Builder	1080.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Binder	9428.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Water	18.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nitrogen	28.00	22135.80	790.56	22135.80	790.56	22135.80	790.56	22135.80	790.56
Oxygen	32.00	6720.00	210.00	6720.00	210.00	6720.00	210.00	6720.00	210.00
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TOTAL		28855.80	1000.56	28855.80	1000.56	28855.80	1000.56	28855.80	1000.56
Enthalpy	kW	0.0	000	5.1	09	-1.3	703	-1.3	703
Phase		(	G	G	à	(	G	(	3
Pressure	Bara	1.	01	3.0	01	3.	01	1.	01
Temperature	°C	25.00		40.00		20.00		20.00	
Activity	U/ml	(	)	C	)	(	0	(	)

					OVERA	ENT MASS BA	LANCE		
STREAM Nr.		33	OUT	1+2+5+12+1	7+23+27+29	9+13+15-	25+32+33	OU	T-IN
Name:		Enzyme	granules	TOTA	AL IN	TOTAL OUT		Total Plant	
COMPONENT	MW	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day	kg/day	kmol/day
Ammonium phosphate	132.07	0.63	0.00	1550.00	11.74	77.50	0.59	-1472.50	-11.15
Biomass	147.60	0.00	0.00	740.00	5.01	5088.56	34.48	4348.56	29.46
Enzyme	147.60	3895.60	26.39	0.00	0.00	4079.86	27.64	4079.86	27.64
Glucose	180.00	4738.50	26.33	27000.00	150.00	10401.23	57.78	-16598.77	-92.22
Inerts	600.00	0.00	0.00	20000.00	33.33	20000.00	33.33	0.00	0.00
Nutrients	229.03	0.00	0.00	460.00	2.01	460.00	2.01	0.00	0.00
Coating	2434.00	1600.00	0.66	1600.00	0.66	1600.00	0.66	0.00	0.00
Builder	1080.00	8779.99	8.13	8779.99	8.13	8779.99	8.13	0.00	0.00
Binder	9428.00	850.00	0.09	850.00	0.09	850.00	0.09	0.00	0.00
Water	18.00	4589.66	254.98	441024.00	24501.33	448312.26	24906.24	7288.26	404.90
Nitrogen	28.00	0.00	0.00	543206.46	19400.23	543206.46	19400.23	0.00	0.00
Oxygen	32.00	0.00	0.00	164906.96	5153.34	159250.29	4976.57	-5656.67	-176.77
Carbon dioxide	44.00	0.00	0.00	0.00	0.00	8011.25	182.07	8011.25	182.07
TOTAL		24454.38	316.58	1210117.41	49265.88	1210117.41	49629.82	0.00	363.94
Enthalpy	kW	-2.3	383						
Phase		L	-						
Pressure	Bara	1.0	01						
Temperature	°C	20.	00						
Activity	U/ml	23428	886.4						

### **APPENDIX 8**

### EQUIPMENT SUMMARY AND SPECIFICATIONS

		1	1		
EQUIPMENT NR :	R101	T101			
NAME :	Bioreactor	Blending Tank			
Number of Units :	2	2			
Pressure [bara] :	1.50	1.50			
Temperature [°C] :	37.0	25.0			
Volume [m <sup>3</sup> ] : Diameter [m] : H [m] :	289.15 5.28 13.20	282.35 5.24 13.10			
Materials of Construction (2) :	SS316	SS304			
Other :					
Remarks: SS – Stainless St	eel				
Designers: C.A	lmeida P. Du		Project ID-Number	: CPD 3237	
K.P	autos K.Pito	inumani .	Date	. July 1999	

### **REACTORS AND VESSELS – SUMMARY**

J.Taboada

EQUIPMENT NR : NAME :	S101 Air Filter	S102 Air Filter	S103 Rotary Vacuum Filter	S104 Ultrafilter		
Number of Units :	3	2	6	4		
Throughput [m <sup>3</sup> /s] :	4.00	4.00				
Average Filter Flux [L/m <sup>2</sup> .h] : Filter Area [m <sup>2</sup> ] : Cake Porosity [v/v]:			250.00 97.16 0.45	80.00		
Membrane Area [m <sup>2</sup> ] : Power [kW] :				450.89 360.7		
Materials of Construction (2) :	SS316	SS316	CS	SS316		
Other :						
Remarks: SS – Stainless Steel CS – Carbon Steel						

# FILTERS - SUMMARY

Designers:	C.Almeida R.Padios J.Taboada	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999	
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# FLUIDIZED BED – SUMMARY

EQUIPMENT NR : NAME :	D101 Fluid Bed Dryer	D102 Fluid Bed Coater		
Number of Units :	1	1		
Drying Temperature [°C] :	40.00	20.00		
Volume [m <sup>3</sup> ] : Diameter [m] : H [m] :	1.70 0.60 6.00	1.70 0.60 6.00		
Power [kW] :	0.00	0.00		
Materials of Construction (2) :	SS 316	SS 316		
Other :				
Remarks:	5-10-10-1			×
SS – Stainless Steel				

Designers:	C.Almeida R.Padios	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999	
	J. Laboada				

EQUIPMENT NR :	G101	S105	A101			
NAME :	Granulator	Screens	Crusner			
Number of Units :	1	1	1			
Pressure [bara] :	1	1	1			
Temperature [°C] :	20	20	20			
Volume [m <sup>3</sup> ] : Diameter [m] : H [m] :	30	2	20			
<u>Number</u> - Series : - Parallel :						
Materials of Construction (2) :	SS316	SS316	SS316			
Other :						
Remarks: SS – Stainless Steel						

# **GRANULATION – SUMMARY**

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			

EQUIPMENT NR : NAME :	E101 Sterilizer	E102 Cooler	E103 Cooler			
Number of Units :	3	5	1			
Sterilization Criterion ln(No/N) :	41.00					
Temperature [°C] Sterilization : Preheat outlet : Final Exit :	121.00 100.00 37.00	5.00	20.00			
Volume [m <sup>3</sup> ] : Diameter [m] : H [m] :						
Heat Transfer Area [m <sup>2</sup> ] :		95.75	6.60			
Holding tube length [m] : Holding tube Diameter [m] : Throughput [m <sup>3</sup> /h] :	196.80 70.00 320.00					
Materials of Construction (2) :	SS 316	CS	CS			
Other :						
Remarks SS – Stainless Steel CS – Carbon Steel						
Designers: C.Al R.Pa	meida P. Du dios R.Pitc	chumani D	roject ID-Number ate	: CPD 3237 : July 1999		

# **COOLERS and STERILIZER – SUMMARY**

J.Taboada

EQUIPMENT NR :	G101	G102	G103		
NAME :	Compressor	Compressor	Compressor		
Number of Units :					
Overall Efficiency					
% :	70	70	70		
Power [kW] :	1315.3	0.20	50.8		
Pressure Change [bara] :	2.00	1.00	2.00		
<u>Number</u> - Series : - Parallel :					
Materials of Construction (2) :	CS	CS	CS	5	
Other :					
Remarks:					
CS – Carbon Steel					

## **COMPRESSORS – SUMMARY**

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			

# **BLENDING TANK – SPECIFICATION SHEET**

EQUIPMENT NUMBER NAME	: T101 : Blending Tank
Service : Type : Number :	
	Operating Conditions
Operating Pressure [bara] Exit Temperature [°C] Agitation Rate [kW/m <sup>3</sup> ]	: 1.00 : 25.00 : 1.00
	Equipment Design
Height [m] Diameter [m] Total Volume [m <sup>3</sup> ] Power [kW] Heating [kcal/h] Cooling [kcal/h]	: 13.10 : 5.24 : 282.35 : 480.0 : 0.0 : 413010.0
	Heat Transfer Agent
Medium Inlet Temperature [°C] Outlet Temperature [°C]	: Chilled Water : 5.0 : 10.0
Remarks:	
Designers: C Almeida	P. Du Project ID Number CDD 2227

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			•

## **BLENDING TANK – SPECIFICATION SHEET**

EQUIPMENT NUMBER NAME	: T102 : Blending Tank
Service : Type : Number :	
	Operating Conditions
Operating Pressure [bara] Exit Temperature [°C] Agitation Rate [kW/m <sup>3</sup> ]	: 1.00 : 25.00 : 1.00
	Equipment Design
Height [m] Diameter [m] Total Volume [m <sup>3</sup> ] Power [kW] Heating [kcal/h] Cooling [kcal/h]	: 13.10 : 5.24 : 282.35 : 480.0 : 0.0 : 413010.0
	Heat Transfer Agent
Medium Inlet Temperature [°C] Outlet Temperature [°C]	: Chilled Water : 5.0 : 10.0
Remarks:	

Designers:	C.Almeida R.Padios J.Taboada	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999
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## HEAT STERILIZER – SPECIFICATION SHEET

EQUIPMENT NUMBE NAME	ER : E101 : Heat Sterilizer
Service : Type : Number :	
	Operating Conditions
Sterilization Temperatu Preheat Outlet Tempera Final Exit Temperature	are [°C] : 121.00 ature [°C] : 100.00 [°C] : 37.00
Design Criteria : Sterilization Cri	terion $\{\ln(No/N)\}$ : 41
	Equipment Design
Holding tube Diameter Holding tube Length	[m] : 0.70 [m] : 196.80
	Process Conditions
Reynolds Number Peclet Number Damkohler Number Dispersion Coeff. Specific Death Rate Throughput Linear Velocity Heating Cooling	: $164965.36$ : $1218.54$ : $42.38$ $[m^2/s]$ : $0.0373$ [1/s] : $0.05[m3/h]$ : $320.00[m/s]$ : $0.23[kcal/h]$ : $6850076.0[kcal/h]$ : $2935746.9$
	Heat Transfer Agent
Heating Medium In temp. [°C] Out temp. [°C]	: Steam Cooling Medium : Cooling Water : 152.00 In temp. [°C] : 25.00 Out temp. [°C] : 30.00
Remarks:	

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			-

# FERMENTOR – SPECIFICATION SHEET

EQUIPMENT NUMBER NAME	: R101 : Fermentor			
Service :				
Number :				
	Operating	Conditions		
	1 0			
Operating Pressure [ Fermentation Temperature Agitation Rate [kV	[bara] : 1.00 [°C] : 25.00 V/m <sup>3</sup> ] : 1.00			
	Reaction N	Mass Data		
Component	Mass Stoich.	Extra-cell %	% To Gas Outlet	
Ammonium phosph Binder Biomass	-49.980 0.00 147.60	100.00 100.00 0.00	0.00 0.00	
Builder	0.00	100.00	0.00	
Carb. Dioxide	271.92	0.00	100.00	
Coating	0.00	100.00	0.00	
Glucose	-563.40	20.00	0.00	
Inert	0.00	0.00	100.00	
Nutrients	0.00	100.00	0.00	
Oxygen	-192.00	0.00	100.00	
Proteins	138.48	100.00	0.00	
Water	247.38	100.00	0.00	
	Equipmer	nt Design		
Height [m]	: 13.20			
Diameter [m]	: 5.28			
Total Volume [m <sup>3</sup> ]	: 289.16			
Power [kW]	: 1440.0			
Heating [kcal/h]	: 0.0			
Cooling [Kcal/h]	: 2021755.00	£		
	Heat Irans	ster Agent		
Medium: Cooling WaterInlet Temperature[°C]: 25.0Outlet Temperature[°C]: 30.0				
Remarks:				

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			-

# **COOLER – SPECIFICATION SHEET**

EQUIPMENT NUMBER NAME	: E102 : Cooler
Service : Type : Number :	
	Operating Conditions
Heat Transfer Coeff. [kcal/m Duty [kcal/h] Exit Temperature [°C]	h <sup>2</sup> .h. <sup>o</sup> C] : 1290.66 ] : -14630664.00 : 5.00
	Equipment Design
Heat Transfer Area [m <sup>2</sup> ]	: 95.75
Exchanger type	: Plate and Frame
	Heat Transfer Agent
Medium Inlet Temperature [°C] Outlet Temperature [°C]	: NaCl Brine : -9.01 : -0.01
Remarks:	
Designante	P. Du Project ID Number CDD 2227

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
-	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			

# **COOLER – SPECIFICATION SHEET**

EQUIPMENT NUMB NAME	ER : E103 : Cooler
Service : Type : Number :	
	Operating Conditions
Heat Transfer Coeff. Duty Exit Temperature	[kcal/m <sup>2</sup> .h.°C] : 1290.66 [kcal/h] : -291362.3 [°C] : 20.00
	Equipment Design
Heat Transfer Area	$[m^2]$ : 6.60
Exchanger type	: Plate and Frame
	Heat Transfer Agent
Medium Inlet Temperature Outlet Temperature	: NaCl Brine [°C] : -9.01 [°C] : -0.01
Remarks:	

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			-

# **ROTARY VACUUM FILTER – SPECIFICATION SHEET**

EQUIPMENT NUMBER : S	\$103
NAME : F	Rotary Vacuum Filter
Service :	
Туре :	
Number :	Discuss Caraliziana
	Process Conditions
Retention Coefficients (RC) for So	lids :
Component	RC (%)
Ammonium phosph	0.00
Binder	0.00
Biomass	100.00
Builder	0.00
Carb. Dioxide	0.00
Coating	0.00
Glucose	0.00
Inert	100.00
Oxygen	0.00
Proteins	0.00
Nitrogen	0.00
Nutrients	100.00
Water	0.00
	Operating Condition
Average Filtrate Flux [L/m <sup>2</sup> .	.h] : 250.00
	Equipment Design
	Equipment Dosign
Filter Area $[m^2] \cdot 97.16$	
Cake Porosity $[y/y] : 0.45$	
Remarks:	

Designers:	C.Almeida R.Padios J.Taboada	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999	
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### **ULTRAFILTER – SPECIFICATION SHEET**

EQUIPMENT NUMBER NAME	: S104 : Ultrafilter
Service : Type : Number :	
	Operating Condition
Filtration Mode Exit Temperature Average Filtrate Flux Max. Solids Conc. In Retentate Rejection Coefficient (RC) Rest of components are set at 0 <sup>th</sup>	: Batch Concentration [°C] : 7.00 [L/m <sup>2</sup> .h] : 80.00 [g/L] : 1200.00 : Set Proteins at 100.00% RC % RC
	Other Data
Unit Power Consumption Membrane Repl. Frequency Power Heating Cooling	[W/m <sup>2</sup> ] : 200.00 [oper.h]: 2000.00 [kW] : 360.70 [kcal/h]: 0.00 [kcal/h]: 24111.2
	Equipment Design
Membrane Area[m <sup>2</sup> ] : 450.8	9
	Heat Transfer Agent
Medium Inlet Temperature [°C] Outlet Temperature [°C] Remarks:	: NaCl Brine : -9.01 : -0.00

Designers:	C.Almeida R.Padios	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999	
	J. Laboada				

### FLUID BED DRYER – SPECIFICATION SHEET

EQUIPMENT NUMBE NAME	R : D101 : Fluid Bed Dryer
Service : Type : Number :	
	Operating Condition
Drying Temperature Ave. Solids Residence 7 Ave. Solids velocity Power Heating	$\begin{bmatrix} {}^{\circ}C \end{bmatrix} : 40.00$ Fime $\begin{bmatrix} s \end{bmatrix} : 60.00$ $\begin{bmatrix} m/s \end{bmatrix} : 0.10$ $\begin{bmatrix} kW \end{bmatrix} : 0.00$ $\begin{bmatrix} kcal/h \end{bmatrix} : 11590.2$
	Evaporation Data
Water is set at 5% Evap Rest of the components	orated do not evaporate
	Other Data
Drying Gas Required Specific Power Specific Heating	[kg dry gas/kg evap] : 5.00 [kW/ m <sup>3</sup> dryer vol.] : 0.02 [kJ/ kg evap] : 2230.00
	Equipment Design
Volume [m <sup>3</sup> ] Diameter [m] Height [m]	: 1.70 : 0.60 : 6.00
	Heat Transfer Agent
Medium Inlet Temperature Outlet Temperature	: Steam [°C] : 152.00 [°C] : 152.00
Remarks:	

Designers:	C.Almeida R.Padios I.Taboada	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999
	J. I abbaua			

### FLUID BED DRYER – SPECIFICATION SHEET

EQUIPMENT NUMBE NAME	ER	: D102 : Fluid Bed Dryer
Service :		
Type :		
Number :		Operating Condition
Drying Temperature		[°C] : 20
Ave. Solids Residence 7	Гime	[s] : 60.00
Ave. Solids velocity		[m/s] : 0.10
Power		[kW] : 0.00
neating		
		Other Data
Drying Gas Required	[kg dry	gas/kg evap] : 5.00
Specific Power	[kW/n	o'dryer vol.] : 0.02
Specific Heating	[kj/ kg	evap] : 2230.00
		Equipment Design
Volume [m <sup>3</sup> ]	: 1.70	
Diameter [m]	: 0.60	
Height [m]	: 6.00	
	engrinder solve kalende stare	Heat Transfer Agent
	*	
Medium		: Steam
Inlet Temperature	[°C]	: 152.00
Outlet Temperature	[°C]	: 152.00
Remarks:		

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
R.Padios	R.Pitchumani		Date	: July 1999
J.Taboada				

## **GRANULATOR – SPECIFICATION SHEET**

EQUIPMENT NUMBER NAME	: G101 : Granulator
Service :	
Type : Number :	
ivumber .	Operating Condition
Flow rate: 700 kg/h	
Chopper rpm : 200	
	Other Data
	Equipment Design
XV 1 20 <sup>3</sup>	
Volume : 30 m <sup>-</sup>	
	Heat Transfer Agent
Remarks:	

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
R.Padios	R.Pitchumani		Date	: July 1999
J.Taboada				

### **CRUSHER – SPECIFICATION SHEET**

FOLIPMENT NUMBER	· A101
NAME	: Crusher
Service :	
Type :	
Number :	
	Operating Condition
Flow rate : 700 kg/h	
	Other Data
	Equipment Design
	Equipment Design
Volume : $20m^3$	
volume . 20m	
	Heat Transfer Agent
Remarks:	

Designers:	C.Almeida R.Padios J.Taboada	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999
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# SCREENS - SPECIFICATION SHEET

EQUIPMENT NUMBER	: S105	
NAME	: Screens	
Service :		
Type : Number ·		
		Operating Condition
		Other Data
		Culoi Duu
		Equipment Design
Diameter : 2 m		
Top Sieve opening : 1000 um		
Bottom Sieve opening : 250 µm	n	
		Heat Transfer Agent
		<u> </u>
Remarks:		

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J. Tabbaua			

## **COMPRESSOR – SPECIFICATION SHEET**

EQUIPMENT NUMBER NAME		: K101 : Air Compressor						
Service : Type : Number :								
Operating Condition								
Pressure Change Power Cooling	[bara] [kW] [kcal/h	: 2.00 : 1315.3 ]: 373395						
		Other Data						
Overall Efficiency	[%]	: 70.00						
		Heat Transfer Agent						
Medium Inlet Temperature Outlet Temperature	[°C] [°C]	: Cooling Water : 25.0 : 30.00						
Remarks:								

Designers: C.A R.P J.Tz	lmeida P. Du adios R.Pite	chumani Da	roject ID-Number : Pate :	CPD 3237 July 1999

## **COMPRESSOR – SPECIFICATION SHEET**

EQUIPMENT NUMB NAME	ER : K102 : Air Compressor						
Service : Type : Number :							
Operating Condition							
Pressure Change Power Cooling	[bara] : 1.00 [kW] : 0.20 [kcal/h]: 42.00						
	Other Data						
Overall Efficiency	[%] : 70.00						
	Heat Transfer Agent						
Medium Inlet Temperature Outlet Temperature Remarks:	: Cooling Water [°C] : 25.0 [°C] : 30.00						

Designers:	C.Almeida R.Padios J.Taboada	P. Du R.Pitchumani	Project ID-Number Date	: CPD 3237 : July 1999
------------	------------------------------------	-----------------------	---------------------------	---------------------------
#### **COMPRESSOR – SPECIFICATION SHEET**

EQUIPMENT NUMB	ER	: K103 : Air Compressor
Service : Type : Number :		
		Operating Condition
Pressure Change Power Cooling	[bara] [kW] [kcal/h]	: 2.00 : 50.80 : 14403.4
		Other Data
Overall Efficiency	[%]	: 70.00
		Heat Transfer Agent
Medium Inlet Temperature Outlet Temperature	[°C] [°C]	: Cooling Water : 25.0 : 30.00
Remarks:		

Designers:	C.Almeida	P. Du	Project ID-Number	: CPD 3237
	R.Padios	R.Pitchumani	Date	: July 1999
	J.Taboada			



CONSTRUCTION SPOTLIGHT					
COMPANY	PROJECT	LOCATION	COST, CAPACITY	STATUS	
Enron Corp. (Houston, Tex.)	Gas-fired power plant	New Albany, Miss.	\$70 million	Bidding and construction schedules not yet determined; contract method undetermined; further information within 60-90 days	
Husky Oil Ltd. (Calgary, Alta.)	Heavy-oil production unit	Lloydminster, Sask.	\$50 million	Design drawings to be started soon; target completion set for March 2001	
Constellation Power (Baltimore, Md.)	Power plant	Brevard County, Fla.	\$30 million	Site acquired; further action pending various approvais; construction targeted to begin early 2000, with commercial operation commencing in 2001	
El Paso Energy Corp. (Houston, T <u>e</u> x.)	Cogeneration plant	Windsor Locks, Conn.	\$15-25 million	Construction scheduled to begin October 1999	
Mobil Oil Corp. (St. Louis, Mo.)	Cogeneration plant	Beaumont, Tex.	\$10-15 million	Preliminary plans in progress; contract method undetermined; construction schedules not yet determined; further information within 30–60 days	
Town of Beauceville (Beauceville, Que.)	Filtration plant	Beauceville, Que.	\$5.5 mittion	Architect selected; construction schedul undetermined; further information pending funding	

This partial list of upcoming projects in the chemical process industries is from DataLine, an online-retrieval service of The McGraw-Hill Companies' F.W. Dodge division. Call 800-221-0088 for more information. The jobs selected are in the pre-planning stage, generally pror to finalization of the engineering design or selection of the general contractor

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104.4

for January, February, and March 1994 (i.e., first quarter 1994).

1 Index values have been rounded to the nearest tenth.

Refrigeration systems

thermal oxidizers

Thermal incinerators

December 1995, pp. 88-95.

Regenerative

Watscrubhers

106.1

106.3 107.9 108.4

<sup>2</sup> Each annual average is the arithmetic mean of the quarterly indexes for that year. <sup>3</sup> All fourth quarter 1998 and first quarter 1999 indexes are preliminary.

107.6

108.2 109.4 109.9 110.6 110.9

109.8 109.0 109.4 109.7 109.7

<sup>4</sup> For fabric filters and mechanical collectors, each quarterly value shown is the average of the Pro-

ducer Price Indexes (PPIs) for the three months in question, divided by the average of the PPIs

Note: For a detailed explanation of the development and use of the VAPCCI, see Chem. Eng.,

107.9

108.8

107.8

109.2

107.2

109.1

110.5

109.9

106.1

108.5

109.1

1102

1010

1000

990

1st 2nd 3rd 4th Quarter

Annual Index

Rubber

Related industries

Electrical power

Mining, milling

Refrigerating

Steam Power

1993 = 964.2

1994 = 993.4

1995 = 1.027.5

1,153.0

0637

1,099.0

1,266.2

1,037.6

1,150.9

066 6

1.096.6

1.264.2

1,037.5

1996 = 1,039.2

1997 = 1,056.8

1998 = 1.061.9

1,149.3

966 8

1.097.1

1.262.7

1.036.7

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Chemical Engineering	2
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FOR MORE ECONOMIC INDICATORS, SEE NEXT PAGE

10 Process Safety \_\_\_\_\_ Deta Book Inded. Cr. Wiliss wyes Data Con 1.1 Appendix 10 Process Safety \_ AMMONIUM PHOSPHATE APP Common Synonyms Solid White Weak ammonia odor 6. FIRE HAZARDS 10. HAZARD ASSESSMENT CODE pammonium orthophosphate ponum phosphate, dibasic phoanum hydrogen phosphate phosphate imonium orthophosphate (See Hazard Assess Flash Point: Not flammable Int Han ok) 6.1 Flammable Limits in Air: Not flammable 6.2 SS Sinks and mixes with water. Fire Extinguishing Agents: Not pertinent 6.4 Fire Extinguishing Agents Not to be Used: Not pertine Stop discharge if possible. Keep people away. Avoid contact with solid and dust. Isolate and remove discharged material. Notily local health and pollution control agencies. 6.5 Special Hazards of Combustion 11. HAZARD CLASSIFICATIONS Products: Toxic and irritating fumes of 11.1 Code of Federal Regulations ammonia and oxides of nitrogen may form in fires. Not listed 11.2 NAS Hazard Rating for Bulk Water Transportation: Not listed Behavior in Fire: Data not available 6.6 Ignition Temperature: Not pertinent Electrical Hazard: Not pertinent 6.7 11.3 NFPA Hazard Classification: Not flammable. Irritating gases may be produced when heated. Not listed 6.9 Burning Rate: Not pertinent 6.10 Adiabatic Flame Temperature: Not pertinent 6.11 Stoichiometric Air to Fuel Ratio: Fire Not pertinent 6.12 Flame Temperature: Not pertinent CALL FOR MEDICAL AID. 7. CHEMICAL REACTIVITY DUST Reactivity With Water: No reaction Dus i irritating to eyes, nose and throat. If inhaled will cause coughing or difficult breathing. If in eyes, hold eyelds open and flush with pienty of water. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. 7.2 **Reactivity with Common Materials: Data** not available Stability During Transport: Stable 7.4 Neutralizing Agents for Acids and SOLID Imitating to skin and eyes. Harmful if swallowed. Caustics: Not pertinent 7.5 Polymerization: Not pertinent Exposure narmul if swalowed. Remove containnated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelids open and flush with plenty of water. IF SWALLOWED and victum is CONSCIOUS, have wctim drink water 7.6 Inhibitor of Polymerization: Not pertinent 7.7 Molar Ratio (Reactant to or milk. IF SWALLOWED and victim is UNCONSCIOUS OR HAVING CONVULSIONS, do nothing except keep victim warm. Product): Data not available 7.8 Reactivity Group: Data not available 12. PHYSICAL AND CHEMICAL PROPERTIES 12.1 Physical State at 15°C and 1 atm: Effect of low concentrations on aquatic life is unknown. May be dangerous if it enters water intakes. Solid Water 12.2 Molecular Weight: Monoammonium: 115 Notify local health and wildlife officials. Notify operators of nearby water intakes. nonium: 132 Pollution Diam 12.3 Boiling Point at 1 atm: Not pertinent (begins to decompose at 100°C) Freezing Point: Not pertinent 12.4 1. RESPONSE TO DISCHARGE 2. LABEL 8. WATER POLLUTION (begins to decompose at 100°C) Critical Temperature: Not pertinent 2.1 Category: None 2.2 Class: Not pertinent (See Response Methods Handbook) 8.1 Aquatic Toxicity: 12.5 Disperse and flush 155 ppm/96 hr/fathead minnow/LCse 12.6 Critical Pressure: Not pertinent 8.2 Waterfowl Toxicity: Data not available 12.7 Specific Gravity: 8.3 Biological Oxygen Demand (BOD); Diammonium: 1.8 at 20°C (solid) Monoammonium: 1.6 at 20°C (solid) Data not available 8.4 Food Chain Concentration Potential: 12.8 Liquid Surface Tension: Not pertinent Liquid Water Intertacial Tension: Not pertinent 12.9 None 3. CHEMICAL DESIGNATIONS 4. OBSERVABLE CHARACTERISTICS 12.10 Vapor (Gas) Specific Gravity: 3.1 CG Compatibility Class: Not listed Not pertinent Physical State (as shipped): Solid 3.2 Formula: NH+H2PO+ and 12.11 Ratio of Specific Heats of Vapor (Gas): 4.2 Color: White (NHa)+HPOA 4.3 Odor: Diammonium-faint ammonia Not pertinent 3.3 IMO/UN Designation: Not listed Monoammonium-faint acid 12.12 Latent Heat of Vaporization: 3.4 DOT ID No .: Data not available Not pertinent 12.13 Heat of Combustion: Not pertinent 3.5 CAS Registry No.: 7783-28-0 12.14 Heat of Decomposition: Not pertinent 12.15 Heat of Solution: 42 Btu/ib = 23 cal/g 5. HEALTH HAZARDS 9. SHIPPING INFORMATION = 0.97 X 10<sup>s</sup> J/kg 12.16 Heat of Polymerization: Not pertinent 5.1 Personal Protective Equipment: Dust mask, protective gloves, and goggles. When diammonium 9.1 Grades of Purity: Reagent; Technical phosphate is stored in closed area, self-contained breathing apparatus is required to protect 9.2 Storage Temperature: Ambient 12.25 Heat of Fusion: Data not available against ammonia fumes. 9.3 Inert Atmosphere: Ventilated (forced) 12.26 Limiting Value: Data not available 52 Symptoms Following Exposure: Inhalation of monoammonium form causes irritation of mucous 9.4 Venting: Open 12.27 Reid Vapor Pressure: Data not available membranes; with diammonium form, ammonia vapors in closed area can cause pulmonary edema and asphyxia. Contact with solid or with ammonia gas causes irritation of eyes and skin. 5.3 Treatment of Exposure: INHALATION: if exposed to ammonia fumes from diammonium phosphate, give artificial respiration and oxygen if needed; enforce rest, EYES: flush with water lor at least 15 min.; if irritation persists, get medical attention. SKIN: flush w Threshold Limit Value: 25 ppm as armonia Short Term Inhalation Limits: Data not available 5.5 Toxicity by Ingestion: Data not available 5.7 Late Toxicity: Data not available Vapor (Gas) Irritant Characteristics: Data not available 5.9 Liquid or Solid Irritant Characteristica: Data not available 5.10 Odor Threshold: Odorless 5.11 IDLH Value: 500 ppm as ammonia NOTES **JUNE 1985** 103

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-1 PC/2541

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# NITROGEN

## NXX

			-		
Common Synonym Liquid nitrogen	Ges	Coloriess Odoriess		6. FIRE HAZARDS     6.1 Flash Point: Not pertinent (nonflammable compressed gas)	10. HAZARD ASSESSMENT CODE (See Hazard Assessment Handbook) A-C-D-F-G
Avoid contact v	Avoid contact with liquid.			6.2 Flammable Limits in Air: Not pertinent     6.3 Fire Extinguishing Agents: Not pertinent     6.4 Fire Extinguishing Agents Not to be     Used: Not pertinent	
	Avoid comact with inquid.			6.5 Special Hazards of Combustion Products: Not pertinent     6.6 Behavior in Fire: Containers may explode when heated.     6.7 Ignition Temperature: Not pertinent	11.1 Code of Federal Regulations: Nonflammable gas 11.2 NAS Hazard Rating for Bulk Water Transportation: Not listed
Fire	Not flammable.			Electrical Hazard: Not pertinent     Burning Rate: Not pertinent     Ditabatic Flame Temperature:     Data not available     Data not available     Data not available     Data not available     Electrical and available     Electrical available	I1.3         NFPA Hazard Classification:           Category         Classification           Health Hazard (Ske)         3           Flammability (Red)         0           Reactivity (Yellow)         0
Exposure	CALL FOR MEDICAL AID. VAPOR Not harmful. In high concentrations may of or loss of consciousness. LIQUID Will cause trostbite. Flush affected areas with ple DO NOT RUB AFFECTED A	cause dizziness, difficult breathing. a, inty of water. REAS.		<ol> <li>CHEMICAL REACTIVITY</li> <li>Reactivity With Water: Heat of water will vigorously vaporize liquid nitrogen.</li> <li>Reactivity with Common Materials: No chemical reaction. Low temperature may cause brittleness in nubber and plastics.</li> <li>Stability During Transport: Stable</li> <li>Neutralizing Agents for Acids and Caustics: Not pertinent</li> <li>Polymerization: Not pertinent</li> <li>Inhibitor of Polymerization: Not pertinent</li> <li>Molar Ratio (Reactant to Product): Data not available</li> <li>Reactivity Group: Data not available</li> </ol>	12. PHYSICAL AND CHEMICAL PROPERTIES
Water Pollution	Not hermful to equatic life.				12.1 Physical State at 15°C and 1 atm: Gas     12.2 Molecular Weight: 28.0     12.3 Boiling Point at 1 atm: 320.1°F =195.6°C = 77.6°K     12.4 Freezing Point:     12.4 Freezing Point:
RESPONSE     (Bee Response M     Restrict access	E TO DISCHARGE ethode Handbook)	2. LABEL 2.1 Category: Nonflammable gas 2.2 Class: 2		8. WATER POLLUTION     8.1 Aquatic Toxicity:     None     8.2 Waterfowl Toxicity: None     8.3 Biological Oxygen Demand (BOO):     None     8.4 Food Chein Concentration Potentiat:     None	354'F =215'C = 58'K 12.5 Crttical Temperature: 232.6'F =147.0'C = 126.2'K 12.6 Crttical Pressure: 433 psia = 33.5 atm = 3.40 MN/m <sup>3</sup> 12.7 Specific Gravity: 0.507 at195.5'C (liquid) 12.8 Liquid Surface Tension: 8.3 dynas/gm = 0.083 N/m at193'C
3. CHEMICAL 3.1 CQ Competibility ( 3.2 Formula: Ni 3.3 IMO/UN Designatik 3.4 DOT ID Ho.: 1977 3.5 CAS Registry No.:	. DESIGNATIONS Class: Not listed on: 2/1977 7727-37-9	<ol> <li>OBSERVABLE CHARACTERISTICS</li> <li>Physical State (as shipped): Liquefied gas</li> <li>Color: Coloriess to faint yellow</li> <li>Odor: None</li> </ol>			12.9 Liquid Water Interfactal Tension: Not pertinent 12.10 Vapor (Gas) Specific Gravity: 0.965 12.11 Ratio of Specific Heats of Vapor (Gas): 1.3962 12.12 Latent Heat of Vaportzation: 95 Bb//b = 53 cal/g = 2.2 × 10 <sup>4</sup> J/kg 12.13 Heat of Combustion: Not pertinent
Personal Protecth trousers wom or breathing appars     Symptoms Follow contain oxyger; or eyes causes (     S.3 Treatment of Exp has stopped; cal froetbite; soak in     S.4 Threahold Limit V E.5 Short Term Inheile E.6 Toxicity by Ingest     Late Toxicity; Nor E.4 Uquid or Solid Im     Sed Im	5. HEAL ve Equipment: Safety glass- utaide boots or over high-top stus where insufficient air is a ring Exposure: Inhalation ca diszines; unconsciousness, frostbite burns, sosure: InhALATION: remove i physician. EYES: treat for f hikeverm veter. rake: Non-toxic trans. Not pertinent tor: Not pertinent tor: Characteristics: None teamt Characteristics: Frost	TH HAZARDS is or face shield; insulated gloves; long sleeves; shoes to shed spilled liquid; self-contained present. In cause asphysiation, if atmosphere does not or even death can result. Contact of liquid with skin a to fresh air; apply artificial respiration if breathing roatbite burns caused by liquid. SKIN: treat for		9. SHIPPING INFORMATION 9.1 Grades of Purity: 99.5+% 9.2 Storage Temperature:	12.14 Heat of Decomposition: Not pertinent 12.15 Heat of Solution: Not pertinent 12.16 Heat of Polymerization: Not pertinent 12.25 Limiting Value: Data not available 12.27 Reid Vapor Pressure: Data not available
5.10 Oxfor Threahold: f 5.11 IDLH Value: Deta	Not available			NOT	ES

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OXYGEN

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Common Synom	yme	Gas	Light blue Odorless		6. FIRE HAZARDS	1	0. HAZARD ASSESSMENT CODE
Liquid oxygen				6.1	Flash Point: Not flammable but supports	(5	ee Hazard Assessment Handbook)
2011	1	Sinks and boils in	water.		combustion Flammable Limits in Air: Not flammable	1	A-I
				6.3	Fire Extinguishing Agents: Not pertinent	1	
				- 64	Fire Extinguishing Agents Not to be		
Avoid contac Wear rubber	overclothing	Keep people away. (including gloves).		6.5	Special Hazards of Combustion		11. HAZARD CLASSIFICATIONS
Stop dischar	ge if possible	9.			Products: Not pertinent	11.1 0	Code of Federal Regulations: Nonflammable cas
					fire. Mixtures of liquid oxygen and any	11.2 1	NAS Hazard Rating for Bulk Water
					fuel are highly explosive.	11.3 1	Transportation: Not listed
	Not flamm	nable.	-		Electrical Hazard: Not pertinent		Category Classification
	Cool expo	s may explore in the psed containers with	e. 1 water.	6.9	Burning Rate: Not pertinent		Health Hazard (Blue)
				6.10	Adiabatic Flame Temperature: Data not available		Reactivity (Yellow) 0
Fire				6.11	Stoichiometric Air to Fuel Ratio:		oxy
				6.12	Data not available Flame Temperature: Data not available		
	1				•		
	CALL FO	R MEDICAL AID.			7. CHEMICAL REACTIVITY		
	VAPOR If inhaled	will cause dizziness	s, or difficult breathing.	7.1	Reactivity With Water: Heat of water will		
	LIQUID			7.2	Reactivity with Common Materials: Avoid		
	Flush affe	e frostbite. Icted areas with ple	nty of water.		organic and combustible materials,		
	DO NOT	RUB AFFECTED A	RÉAS.		such as oil, grease, coal dust, etc. It ignited, such mixtures can explode. The		
Exposure					low temperature may cause brittleness		
Exposure				7.3	in some materials. Stability During Transport: Stable		
1				7.4	Neutralizing Agents for Acids and		
				7.5	Polymerization: Not pertinent		
				7.6	Inhibitor of Polymerization:		
				7.7	Not pertinent Molar Ratio (Reactant to		
				7.4	Product): Data not available Reactivity Group: Data not available	12.	Physical AND CHEMICAL PROPERTIES Physical State at 15°C and 1 atm:
Water	Not harm	ful to aquatic life.				12.2	Gas Molecular Weight: 32.0
Pollution	1					12.3	Boiling Point at 1 atm:
						12.1	
1 95590	NEE TO DIS	CHARGE	2 14BEI		8 WATER POLLUTION		-361°F = -218°C = 55°K
(See Response	Methoda H	andbook)	2.1 Category: Oxidzer		Aquatic Toxicity:	12.5	Critical Temperature: 180°E =118°C =155°K
Restrict acco	084		2.2 Class: 5		None	12.6	Critical Pressure:
				8.2	Waterfowl Toxicity: None Biological Oxygen Demand (BOD):	127	738 psia = 50.1 atm = 5.09 MN/m <sup>3</sup> Specific Gravity:
					None		1.14 at
				8.4	Food Chain Concentration Potential: None	12.8	Liquid Surface Tension: 13.47 dvnes/cm = 0.01347 N/m at
2 CHEMI		ATIONS		1			-183°C
S. CHEMIN	the Classe Ma	Allons at Estad	4.1. Bhusical State (as abigment)			12.9	Liquid Water Interfacial Tension: Not pertinent
3.2 Formula: Os	ity canac ivo	A MARIOU	Liquefied gas			12.10	Vapor (Gas) Specific Gravity: 1.1
3.3 IMO/UN Design	nation: 2/107	73	4.2 Color: Light blue			12.11	Ratio of Specific Heats of Vapor (Gas): 1.3962
3.5 CAS Registry M	No: 7782-44-	-7				12.12	Latent Heat of Vaporization:
							91.6 Btu/tb = 50.9 cal/g = 2.13 X 10 <sup>s</sup> J/kg
				-1		12.13	Heat of Combustion: Not pertinent
		5. HEAL	TH HAZARDS		9. SHIPPING INFORMATION	12.14	Heat of Decomposition: Not pertinent Heat of Solution: Not pertinent
5.1 Personal Prot	ective Equip	ment: Safety goggi	es or face shield; insulated gloves; long sleeves;	• 9.1	Grades of Purity: 99.5+%	12.16	Heat of Polymerization: Not pertinent
5.2 Symptoms Fo	n outside bo	ous or over nign-top oeure: inhalation of	100% oxygen can cause nausea, dizziness,	9.2	Storage Temperature:	12.25	Heat of Fusion: Data not available
irritation of k	ungs, pulmon	ary edema, pneumo	onia, and collapse. Liquid may cause frostbite of	9.4	Venting: Safety relief	12.27	Reid Vapor Pressure: Very high
6.3 Treatment of	n. Exposure: If	HALATION: in all t	but the most severe cases (pneumonia), recovery is				
rapid after re	eduction of a	xygen pressure; sup	oportive treatment should include immediate			1	
sedation, and troatbite: and	aconvuisive t	merapy if needed, a m water.	ing rest. EYES: treat frostbite burns. SKIN: treat			1	
5.4 Threshold Lim	ht Value: De	ta not available					
5.5 Short Term In 5.6 Taxiativ by In	heletion Lim	vita: Not pertinent					
5.7 Late Toxicity:	Not pertinen	t					
5.8 Vepor (Ges) in 5.9 Liquid or Solid	ritant Chara d Irritant Chu	eteristics: Data no eracteristics: Data	t available not available				
5.10 Odor Thresho	Het Not pertir	nent				TES	
A 11 IDLH Value: D	eta not availa	8049			NC		
	×						

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#### Na<sub>2</sub>SO

UN-nummer: VRIJ (RC)

## NATRIUMSULFAAT WATERVRIJ

FYSISCHE EIGENSCHAPPEN	BELANGRIJ	BELANGRIJKE GEGEVENS			
Kookpunt, °C > 1700 Smeltpunt, °C 884	WITTE HYGROSCOPISCHE KRISTALLEN OF POEDER Bij sterke verhitting ontstaan prikkelende dampen.				
Relatieve dichtheid (water = 1) 2.7 Oplosbaarheid in water, g/100 ml bij 20°C 16.2	MAC-waarde niet vastgesteld				
Log P octanol/water (berekend) -3.0	2 Wijze van opname/inademingsrisico: De stof kan worden opgenomen in het lict inslikken. Directe gevolgen: De stof werkt irriterend op de ogen. Gevolgen bij langdurige, herhaalde blootstelling: Contact met de huid kan een et tige huidaandoening veroorzaken op basis van een overgevoeligheidsreactie. <sup>11</sup>				
Brutoformule: Na <sub>2</sub> O <sub>4</sub> S Relatieve molecuulmassa 142					
DIRECTE GEVAREN	PREVENTIE	BLUSSTOFFEN			
Brand: niet brandbaar.		bij brand in directe omgeving: aile blusstoffen toegestaan.			
SYMPTOMEN	PREVENTIE	EERSTE HULP			
Inademen:	ventilatie (indien niet in poedervorm), plaatse- lijke afzuiging of ademhalingsbescherming (fil- tertype P2).				
Huid:	handschoenen (butylrubber, PVC).				
		contactlenzen verwijderen), dan naar (oog)ans brengen.			
NOODSITUATIE / OF	PRUIMING / OPSLAG	ETIKETTERING			
Opruimen gemorst produkt: Draag handscho- stofbnl. Gemorst produkt zorgvuldig opzuigen/opscheppe Restant verwijderen met water. Spoelwater atvoe Vaten etiketteren en atvoeren volgens BAGA/KC	enen, laarzen, filtermasker met filtertype P2 en in en eventueel hergebruiken, reen naar riool. A regels.	Afleveringsetiket: vraag leverancier			
		KCA : 05/12			
	OPMERKINGEN				
<sup>1)</sup> Iemand die overgevoeligheidsverschijnselen he stof te vermijden. Natriumsulfaat.10H <sub>2</sub> O is glauberzout.	seft gekregen door blootstelling aan natriumsulfaa	at dient in de toekomst elke blootstelling aan œze			

900

Kaartnummer C-0979

Chemiekaarten veertiende editie 1999



CAS-nummer: [62-54-4]<sup>1)</sup>

Ca(CH<sub>3</sub>COO)<sub>2</sub>

## CALCIUMACETAAT

FYSISCHE EIGENSCHAPPEN	BELANGRIJKE GEGEVENS				
Ontleedt beneden het smeltpunt. °C         160           Zeikontbrandingstemperatuur. °C         680	WITTE, GRIJZE OF BRUINE HYGROSCOPISCHE KRISTALLEN of POEDER     De stof ontleedt bij verhitting boven 160°C of bij contact met sterke zuren onder vorming van bi     dampen (o.a. azijnzuur. zie aldaar). Reageert heftig met oxidatiemiddelen.     S     MAC-waarde niet vastgesteld				
Relatieve dichtheid (water = 1) 1.5 Oplosbaarheid in water, g/100 ml bij 0°C 37					
	Wijze van opname/inademingsrisico: De stof deming van stofdeeltjes en door inslikken. Deze kan echter snel een voor de gezondheid gevaar Directe gevolgen: De stof werkt irriterend op de	kan worden opgenomen in het lichaam door ina i stof verdampt bij 20°C praktisch niet; bij stuiver lijke concentratie in de lucht worden bereikt, a ogen, de huid en de ademhalingsorganen.			
Brutotormule: C4H <sub>6</sub> CaO4 Relatieve molecuulmassa 158.2					
DIRECTE GEVAREN	PREVENTIE	BLUSSTOFFEN			
Franc: brandbaar.		sproeistraal water, poeder.			
ledingsprodukten bij verhitting.					
SYMPTOMEN	PREVENTIE	EERSTE HULP			
nedemen; hoesten.	ventilatie (indien niet in poedervorm), plaatse- lijke afzuiging of ademhalingsbescherming (fil- tertype P2).				
Huid:	handschoenen (butylrubber).				
<b>Ogen: ro</b> odheid en pijn, slecht zien.	stofbril.	minimaal 15 minuten spoeien met water (evt contactlenzen verwijderen), dan naar (oog)arts brengen.			
NOODSITUATIE / OF	PRUIMING / OPSLAG	ETIKETTERING			
Opruimen gemorst produkt: Draag handscho krôbril. Gemorst produkt zorgvuldig opzuigen/opscheppe Restant verwijderen met water. Spoelwater alvoe Vaten etiketteren en alvoeren volgens BAGA/KC Opalag: Gescheiden van oxidatiemiddelen en zu	enen, laarzen, filtermasker met filtertype P2 en en en eventueel hergebruiken. eren naar nool. A regels. gren, droog.	Afleveringsetiket: vraag leverancier BAGA: D.6			
		KCA : 03			
Het CAS-nr. van de gehydrateerde vorm is [57 Calciumacetaat komt ook voor in de gehydrateer	OPMERKINGEN 43-26-0]. de vorm: Ca(C <sub>3</sub> COO) <sub>2</sub> .H <sub>2</sub> O.				
		UN-nummer VRLL (PC)			
	Kaadauma- 0 1007	Gittininer, This (RC)			
	Kaartnummer C-1007 Chemiekaarten veertiende editie 1999				

CAS-nummer: [14807-96-6]

TALK

FYSISCHE EIGENSCHAPPEN	BELANGRIJK	E GEGEVENS	C. Martin
Relatieve dichtheid (water = 1) 2.8 Oplosbaarpeid in water niet	WIT TOT GRUSWIT POEDER		100
	MAC-waarde <sup>1)</sup> (als respirabel stof)	2 mg/m <sup>3</sup>	
Brutoformule: Mq4011Sik.H2O	Hegschulden van een taikbus Deze stof verdampt bij 20°C praktisch niet als poeder verstuiven echter snel een irriterende concentratie in de lucht ontstaan.		
Relatieve molecuulmassa 379,3	DOCVENTIE	PLUSSTOFF	100
Brand: niet brandhaar	FREVENIIC	bij brand in directe omgeving:	alle birector
		toegestaan.	
SYMPTOMEN	PREVENTIE	EERSTE HUL	P 36
nademen: benauwdheid, lichte prikkelhoest.	plaatselijke atzuiging of ademhalingsbescher- ming (filtertype P2).		
Ogen: roodheid.	stofbril.		
NOODSITUATIE / O	PRUIMING / OPSLAG	ETIKETTERIN	G
Opruimen gemorst produkt: Draag handscho stothil. Gemorst produkt afdekken, vervolgens zorgvul fuele iaatste resten verwijderen met water. Spoe Vaten etiketteren en afvoeren volgens BAGA/KC	enen, laarzen, filtermasker met filtertype P2 en dig opzuigen en eventueel hergebruiken, <i>Even- iwater</i> afvoeren naar nool. A regels.	Afleveringsetiket: vraag levera	ncier
		KCA : 05/12	
	OPMERKINGEN		
<sup>11</sup> De MAC-waarde geldt voor stof zonder asbest doeningen aan de longen ten gevolge van inad toegepast. Voor zuivere talk zijn geen bijzondere	vezels. Industněle talksoorten, met name vezelhou eming zijn dan mogelijk. In dat geval moeten de r a voorzorgmaatregelen nodig.	dende, kunnen soms asbestvezel: egelingen aangaande <i>asbest</i> (zie	s bevatten. Aan aldaar) worder
	Kaartnummer C-0836		
1038			
	Chemiekaarten veertiende editie 1999		

0

CAS-nummer: [13463-67-7] anataas rutiel titaanwit

## TITAANDIOXIDE

	[			
FYSISCHE EIGENSCHAPPEN	BELANGRIJKE GEGEVENS			
Kookpunt, °C 2500-3000 Smeltpunt, °C 1840				
Relatieve dichtheid (water = 1) 4.3 Oplosbaarheid in water niet	id (water = 1) water hiet deming en inslikken. Deze stof verdampt bij 20°C praktisch niet; bij verstuiven kai concentratie in de lucht ontstaan. Directe gevolgen: De stof werkt irriterend op de ademhalingsorganen.			
Brutoformule: O <sub>2</sub> Ti Relatieve molecuulmassa 79,9				
DIRECTE GEVAREN	PREVENTIE	BLUSSTOFFEN		
Brand: niet brandbaar.		bij brand in directe omgeving: alle blusstoff toegestaan.		
SYMPTOMEN	PREVENTIE	EERSTE HULP		
nademen: hoesten.	plaatselijke afzuiging of ademhalingsbescher- ming (filtertype P2).	frisse lucht, rust.		
		brengen.		
NOODSITUATIE / OF	PRUIMING / OPSLAG	ETIKETTERING		
uprumen gemorst produkt: Draag tiltermaske afdekken, vervolgens zorgvuldig opzuigen en ev verwijderen met water of stoom. Spoelwater afvo Vaten etiketteren en afvoeren volgens BAGA/KC,	r met nitertype P2 en stofbni. <i>Gemorst produkt</i> ventueel hergebruiken. <i>Eventuele laatste resten</i> eren naar nool. A regels.	Afleveringsetiket: vraag leverancier		
		KCA : 05/12		
	OPMERKINGEN			
<sup>1</sup> De voornaamste typen zijn anataas en rutiel.				
		IN-nummer VPL (84)		
070	Kaartnummer C-0337			

1076

Kaartnummer C-0337

Chemiekaarten veertiende editie 1999



170 World-wide Limits

Limits in Water (ug/l) No Standards Set

Limits in Soil (Permissible Cleanup Levels) (mg/kg or ppm) No Standards Set

## CELLULOSE

CAS 9004-34-6		RTECS	FJ5691460		
(C <sub>6</sub> H <sub>10</sub> O <sub>5</sub> ) <sub>7</sub>	Raw M Esters,	aterial for Cell Paper	ulose	Carcinogen: No Data	
Limits in Workplace A	ir (mg/m <sup>3</sup> )		8		
Location	IWA	51	EL	Notes	Reference
Respirable Dust:					
CHE	6	-	-	_	(1)
FRA	10	-	-	-	(1)
GBR	5	-	_	-	(34)
USA (NIOSH/OSHA)	5	-	-	-	(1)
Total Inhalable Dust: AUS					
BEL	10	-	-	-	(1)
CIS (SUN)	10	-	-	-	(1)
GBR	-	:	2	-	(37)
ISR	10	2	0	-	(34)
USA (ACGIH)	10	-	-	-	(35)
USA (NIOSH/OSHA)	10	-	-	-	(25)
	15	-	-		(1)
Limits in Ambient Air	(ug/m <sup>3</sup> )				
Location	0.5 hr	8.0 hr	24 hr	Annual	References
ND	_	100			(9)
тх	100	-		10	(9,20)

Limits in Water (ug/l) No Standards Set

Limits in Soil (Permissible Cleanup Levels) (mg/kg or ppm) No Standards Set

### **CEMENT (PORTLAND CEMENT)**

CAS 65997-15-1

RTECS VA8770000

 $(CaO)(SiO_2)(Al_2O_3)$ 

Construction Material

Carcinogen: No Data

(9,20)

A Church West

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