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Raman Fingerprint of Pressure-Induced Phase Transitions in TiS_3 Nanoribbons: Implications for Thermal Measurements under Extreme Stress Conditions

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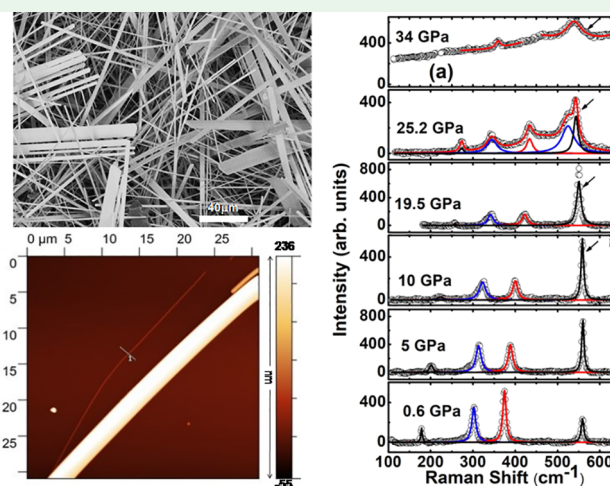
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ABSTRACT: Two-dimensional layered trichalcogenide materials have recently attracted the attention of the scientific community because of their robust mechanical and thermal properties and applications in opto- and nanoelectronics devices. We report the pressure dependence of out-of-plane A_g Raman modes in high quality few-layer titanium trisulfide (TiS_3) nanoribbons grown using a direct solid–gas reaction method and infer their cross-plane thermal expansion coefficient. Both mechanical stability and thermal properties of the TiS_3 nanoribbons are elucidated by using phonon-spectrum analyses. Raman spectroscopic studies at high pressure (up to 34 GPa) using a diamond anvil cell identify four prominent A_g Raman bands; a band at 557 cm^{-1} softens under compression, and others at 175, 300, and 370 cm^{-1} show normal hardening. Anomalies in phonon mode frequencies and excessive broadening in line width of the soft phonon about 13 GPa are attributed to the possible onset of a reversible structural transition. A complete structural phase transition at 43 GPa is inferred from the A_g soft mode frequency (557 cm^{-1}) versus pressure extrapolation curve, consistent with recently reported theoretical predictions. Using the experimental mode Grüneisen parameters γ_i of Raman modes, we estimated the cross-plane thermal expansion coefficient C_v of the TiS_3 nanoribbons at ambient phase to be $1.321 \times 10^{-6}\text{ K}^{-1}$. The observed results are expected to be useful in calibration and performance of next-generation nanoelectronics and optical devices under extreme stress conditions.

KEYWORDS: TiS_3 , high pressure, Raman spectroscopy, phonons, 2D semiconductors



INTRODUCTION

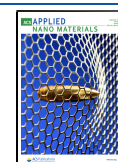
Two-dimensional (2D) van der Waals thin layer structure materials such as MoS_2 , WS_2 , WSe_2 , MoSe_2 , GaS, GaSe, h-BN, SnSe_2 , TiS_3 , GeTe, MoTe_2 , WTe_2 , and black phosphorus have attracted significant research interest because of their extraordinary physical, optical, and electronic properties.^{1–7} These materials are found to be suitable for a wide range of applications such as gas sensors,^{3,8,9} photodetectors,^{10–15} transistors,^{12,16–20} solar cells,^{21,22} energy storage devices,²³ and field emitters.^{13,24–27} The central feature of this 2D class of materials is their finite band gap unlike in graphene, which exhibits zero band gap at the Dirac K-point of the Brillouin zone and larger charge carrier mobility. These 2D materials are considered to be possible alternatives to graphene, and this has invigorated research interest in these materials.⁶ Among these 2D materials, metal trichalcogenide titanium trisulfide (TiS_3), a

direct band gap material ($E_g \sim 1\text{ eV}$) with stable layered structure, exhibits in-plane anisotropic geometry.^{28–34} As a result, it shows remarkable photoresponse, gate tunability, robust thermoelectric performance, and strong quasi-one-dimensional physical properties.^{32,33} *Ab initio* calculations predict anisotropic carrier mobility of a single-layer TiS_3 ^{32,33} with larger carrier mobility along the *b*-axis ($13.87 \times 10^3\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$), and that along the *a*-axis is relatively marginal ($1.01 \times 10^3\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$). The reported carrier mobility in the TiS_3

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nanosheet³² along the “*b*”- and “*a*”-axes are 80 and 40 cm² V^{−1} s^{−1}, respectively. Interestingly, the TiS₃ nanoribbon is found to possess superior optical tunability as compared to that of nanosheets.³² The TiS₃ stabilizes in a monoclinic structure with space group $C_{2h}^2(P2_1/m)$ with two formula units per unit cell ($Z = 2$).³¹ The crystal structure of a monolayer TiS₃ is depicted in Figure 1. The Ti–S bond length along the *b*-axis

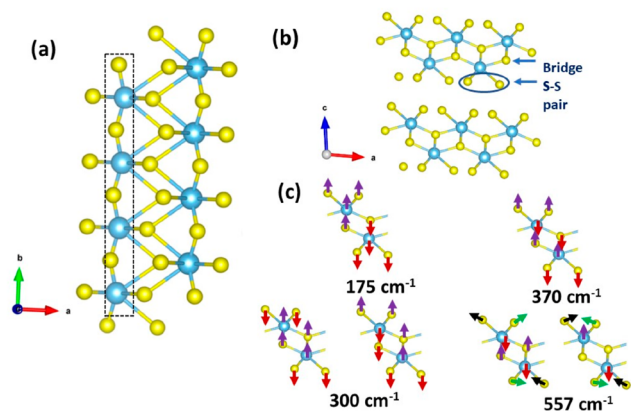


Figure 1. Crystal structure of titanium trisulfide. The central Ti metal atoms (light blue color) are covalently bonded to six S atoms (yellow color), forming a trigonal prism. These prisms share triangular faces to form 1D chains (shown by dashed lines). The black dashed lines mark the Ti chain along crystallographic *b*-axis. (b) Cross-sectional view of the structure of TiS₃, with bridge sulfur and sulfur–sulfur pair atoms. (c) Schematic representation of the motion of atoms for different pressure-sensitive Raman-active phonon modes. The bands at 300 and 557 cm^{−1} are degenerate. Structure models are made by using VESTA.⁵⁵

(0.245 nm) is shorter than the bond along the *a*-axis (0.265 nm)³⁴ which results in its in-plane anisotropic geometry. The metal ions Ti are located at the center of distorted trigonal prisms, and these trigonal faces of prism are connected to form a chain along the *b*-axis (Figure 1a). These covalently bonded parallel chains form a sheet, and such sheets are held together by van der Waals force. Hence, the TiS₃ crystal can be viewed as a set of stacked parallel sheets with each sheet being composed of 1D chains of a triangular TiS₃ unit.³² Two kinds of sulfur atoms are seen in the structure of TiS₃: bridge sulfur and sulfur–sulfur pair atoms (Figure 1b).

Although 2D anisotropic materials such as TiS₃ are of great practical importance,^{35–38} investigations of the effect of pressure on TiS₃ materials are still scarce. Structural study of 2D materials subjected to pressure in a diamond anvil cell is an effective way to tune their structure³⁹ and electronic⁴⁰ properties. Besides, Raman spectroscopic studies on such layer materials can be useful to estimate their thermal transport properties.^{39,40} Interestingly, the mechanical strain in graphene⁴¹ and MoS₂⁴⁰ are reported to alter their electronic structure. TiS₃ exhibits a strong Raman signal^{30,34} as in graphene³⁹ and MoS₂;⁴⁰ hence, Raman spectroscopy is expected to give useful insight about its structural along with mechanical stability and thermal transport properties. Raman studies^{39,41} on seven-layer graphene nanosheets up to 40 GPa indicate a change in G-band slope, and its excessive broadening above 16 GPa is attributed to its deviation from 2D-layer structure. The pressure-induced isostructural electronic transition in multilayered MoS₂ from a semiconducting to a metallic state at 19 GPa has been inferred from high-pressure (HP) Raman spectroscopy.⁴⁰ The out-of-plane A_{1g} Raman mode is an intense band that is sensitive to the electronic transition.⁴⁰ HP Raman spectroscopic studies on a TiS₃ whisker of ~100 μm thickness suggested no phase transition

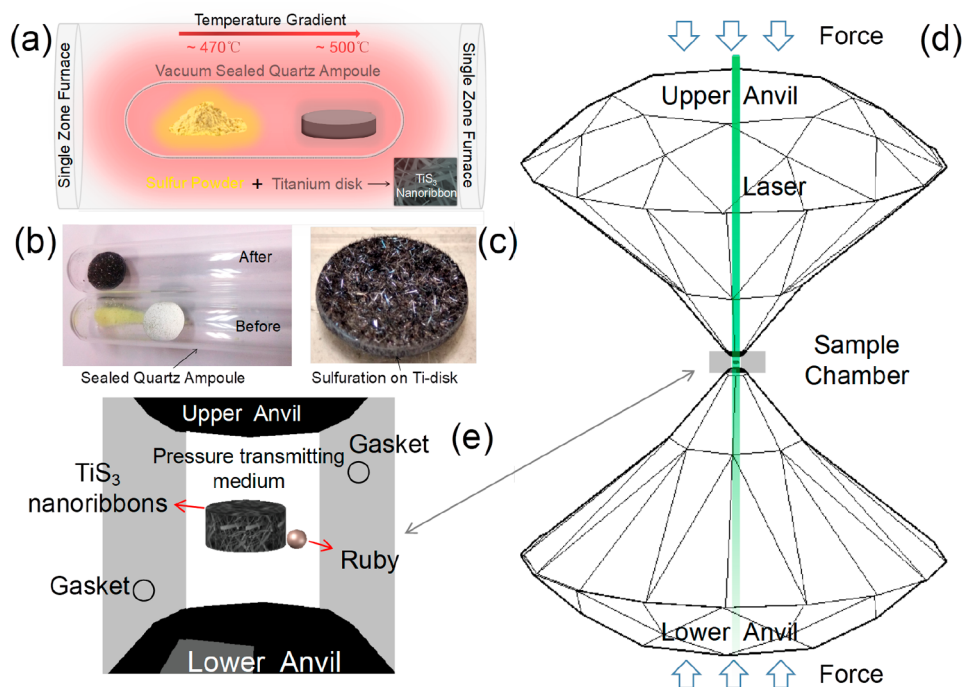


Figure 2. (a) Schematic of solid–gas reaction method involved in fabrication of TiS₃ nanoribbons. For clarity, optical images of (b) sealed ampoules before and after the synthesis process and (c) a zoomed-in view of sulfured titanium disk are shown. Schematics of (d) diamond and (e) gasket used in the diamond anvil cell are also shown. The schematics are made by using AutoCAD software, version 2020.

up to 26 GPa³⁷ within conclusive phonon modes behavior at high pressure. Kang et al. using *ab initio* calculations on the electronic band structure of TiS₃ film suggest its band gap to be independent of the number of layers, stacking order, and vertical strain, and they attribute this result to the absence of charge density redistribution in the interlayer region.⁴² Recently a strain-induced dynamical instability in a TiS₃ monolayer was obtained from phonon dispersion calculations. It was revealed that TiS₃ becomes dynamically unstable for an applied uniaxial tensile strain larger than 6% (8%) along the *a* (*b*)-direction,⁴³ which is yet to be examined experimentally. The anharmonicity of phonon modes is responsible for various thermal properties such as thermal expansion and thermal conductivity of materials. By investigation of Raman spectra as a function of pressure and temperature independently, complete information about the quasi-harmonic and true anharmonic components of the phonon modes and the implications can be analyzed.⁴⁴ In an earlier study,³⁰ we reported our temperature-dependent Raman spectroscopic studies on TiS₃ nanoribbons in the range 88–570 K. Four out-of-plane A_g phonon modes linearly soften (red-shift) with temperature, and the TiS₃ compound was found to be stable in this temperature range. However, the pressure-dependent phonon vibrational properties of few-layer TiS₃ nanoribbons remain unexplored. Therefore, it is of interest to investigate the effect of high pressure on the vibrational properties of TiS₃ to identify the anharmonic phonons and consequently its structural instability. Moreover, the contribution of these phonons to thermal expansion can also be estimated by using high-pressure mode behaviors. To our knowledge, there are no reports of the effect of pressure on vibrational properties of TiS₃ nanoribbons or estimation of its thermal expansion by using the phonon spectroscopic results.

In this work, we report pressure-dependent Raman spectroscopic studies of TiS₃ nanoribbons synthesized by the direct solid–gas reaction method. TEM studies on the as-synthesized materials suggest a highly crystalline nature of the as-prepared sample.³⁰ *In situ* high-pressure Raman spectroscopic studies in a diamond anvil cell were performed up to 34 GPa to study the phonon behavior under compression and any possible phase transition(s) at high pressure. Anomalies in the slope of mode frequency, line width, and their Raman intensity in TiS₃ nanoribbon are attributed to a possible onset of structural transition at 13 GPa, subsequently resulting in a complete structural phase transition at 43 GPa. Pressure coefficients and mode Grüneisen parameters of the A_g modes were obtained from their Raman spectra. The contribution of these phonons to the thermal expansion across the *a*–*b* plane (along the *c*-axis) is estimated by using our experimental mode Grüneisen parameters. To obtain reliable information about the phase transition and Grüneisen parameter, Raman spectra were measured at close intervals of pressure.

RESULTS AND DISCUSSION

The fabrication of TiS₃ nanoribbons was accomplished by using the solid–gas reaction method (Figure 2a–c). The details on the synthesis are reported in the Experimental Section and our earlier studies.^{30,32} A typical schematic of the diamond anvil cell used for high-pressure Raman measurements is depicted in Figure 2d; the close-up view schematic of the sample chamber is shown in Figure 2e to have a better understanding of the sample, ruby, pressure transmitting medium, and gasket position between two anvils.

Morphological Structure. The morphology of as-prepared TiS₃ materials was investigated by using low- and high-magnification SEM (Figure 3a,b). The SEM micrograph

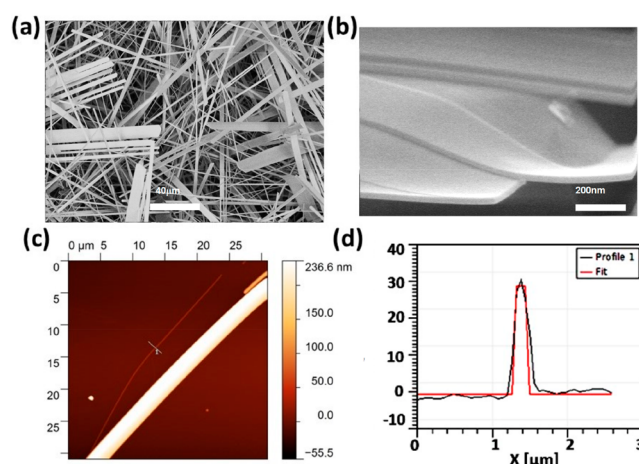


Figure 3. Scanning electron microscopy results on the TiS₃ nanoribbon: (a) low-magnification SEM image recorded from the TiS₃ nanoribbon; (b) highly magnified SEM image of the TiS₃ nanoribbon; (c) AFM image and (d) AFM height profile of the TiS₃ nanoribbon.

shows the nanoribbons of length 30–80 μm and width about 2 μm. The HRTEM image confirms growth of ribbon along a (011) plane with lattice spacing 0.41 nm,³⁰ in accordance with JCPDS data card no. 00-015-0783. The AFM image and its height profile (Figure 3c,d) indicate the thickness of the nanoribbon is about ~30 nm, suggesting that each nanoribbon consists of multilayers of TiS₃ (~43 layers).

High-Pressure Raman Spectroscopy. At ambient temperature and pressure the crystal structure (space group *P*2₁/*m*) of TiS₃ belongs to the *C*_{2h} point group with two formula units per unit cell.⁴⁵ Factor group analysis suggests that the distribution of Raman-active phonons among different irreducible representations⁴⁶ is $\Gamma = 8A_g + 4B_g$. The nonvanishing polarizability tensor corresponding to these Raman modes are⁴⁷

$$A_g = \begin{bmatrix} a & 0 & d \\ 0 & b & 0 \\ d & 0 & c \end{bmatrix} \quad \text{and} \quad B_g = \begin{bmatrix} 0 & e & 0 \\ e & 0 & f \\ 0 & f & 0 \end{bmatrix}$$

The Raman spectrum of the TiS₃ nanoribbon recorded at ambient temperature (Figure 4) shows four prominent A_g Raman bands located at 175, 300, 370, and 557 cm^{−1}, similar to the number of modes reported earlier.³⁰ These band positions are obtained by fitting the Raman spectrum to a Lorentzian function. These observed Raman-active phonon modes vibrate along the *c*-axis, perpendicular to the quasi-1D chain. B_g phonon modes involving atomic displacement along the chain (*b*-axis) are weak;^{30,45} hence, they are not discernible in the Raman spectrum. The Raman band at 175 cm^{−1} arises due to rigid vibrations of the 1D-TiS₃ chain.^{31,45} On the other hand, the bands located at 300 cm^{−1} (degenerate band) and 370 cm^{−1} correspond to the internal vibrations of the polyhedral TiS₆ unit, and the degenerate band at 557 cm^{−1} arises due to a predominant contribution from the nonbridging S–S pair vibration (diatomic vibration) of the trigonal

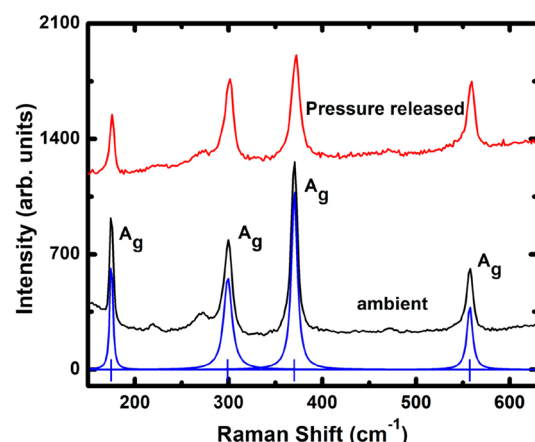


Figure 4. Raman spectra of TiS_3 nanoribbons at ambient and after complete pressure released in a diamond anvil cell. The deconvoluted individual Raman bands (blue color) are also shown. Blue tick marks locate the calculated prominent Raman bands of the ambient phase. For clarity, the spectra are vertically shifted.

prism.^{31,37} For clarity, schematic representations of the motion of atoms for different pressure-sensitive Raman-active phonon modes are represented in Figure 1c. To examine the changes in the Raman spectrum of TiS_3 nanoribbon under the influence of high pressure, the Raman spectra were recorded in the DAC at different pressures (Figure 5). At 0.6 GPa, the spectral features are found to be similar to the ambient one (Figure 4). As the pressure is increased, the intensity of Raman bands except that at 557 cm^{-1} reduces and peaks are broadened. Although the rigid band at 175 cm^{-1} survives at high pressure, its intensity reduces significantly and becomes weak at high pressure. The band intensity of the 557 cm^{-1} mode grows as pressure is increased up to $\sim 13\text{ GPa}$ and reduces thereafter up to 34 GPa , the highest pressure of the present studies. In addition, the line width of this band is found to be insensitive to pressure up to $\sim 13\text{ GPa}$ and softens continuously upon further compression. The band located at 484 cm^{-1} , close to the highest frequency S–S diatomic vibrational band, has been attributed to the nearly degenerate band at 557 cm^{-1} .^{37,45} The appearance of this band at high pressure (20–25 GPa) can be considered to be due to their different pressure dependencies. The line-width behavior of Raman bands with pressure will be discussed later. At high pressure, as can be seen in Figure 5b, the 557 cm^{-1} band shifts to lower frequency. Usually, as can be expected, upon compression, the phonon mode frequencies increase as bond lengths involving atomic vibration are reduced. However, sometimes the mode frequency decreases under the influence of pressure, and such modes are known as soft modes.⁴⁴ The existence of such soft modes often suggests instability of the crystal lattice, and eventually the system undergoes a structural phase transition. In the present case, the decrease of the 557 cm^{-1} mode frequency with pressure is due to the increase of the S–S bond length possibly due to electron–phonon interaction. Often in 2D materials,⁴⁰ the softening of phonon is due to the effect of electron–phonon interaction because of the reduction in interlayer spacing under compression. More evidence of phase transition is noticed from the pressure dependencies of mode frequencies and other spectral parameters such as mode line width and intensity. The spectral parameters are obtained by quantitative analysis of the pressure-dependent Raman spectra by using a Lorentzian

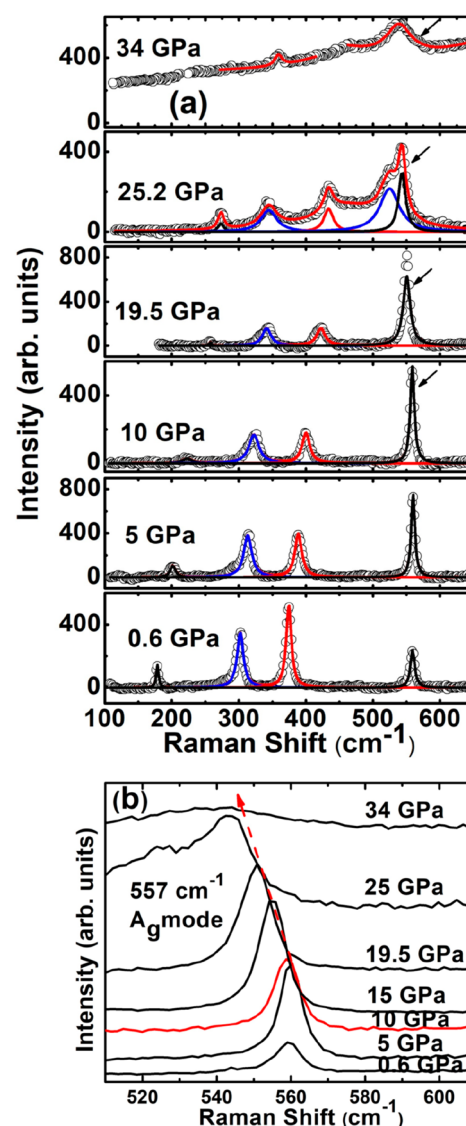


Figure 5. (a) Raman spectra of TiS_3 nanoribbons at different pressures. Open circles are the experimental data, and solid lines are the fits for individual bands and total fit to the data. Softening of the mode is indicated by arrows. (b) For brevity, the A_g mode at 557 cm^{-1} continues to soften upon increase in pressure above 13 GPa .

function. The pressure evolution of Raman mode frequencies (Figure 6) shows anomalous behavior of these modes around $\sim 13\text{ GPa}$ which can be attributed to the onset of structural transition resulting from structural distortion. Upon increasing pressure, a complete structural transition is evident as discussed later. The theory of mean-field approximation relates the soft phonon with structural transition.⁴⁸ The square of the soft phonon mode frequency (ω^2) is related to its transition pressure by the relation $\omega^2 \propto (P - P_c)$, for $P < P_c$ and the soft phonon vanishes as the pressure approaches a transition pressure (P_c). The variation of ω^2 as a function of pressure is depicted in Figure 7a. As can be noticed, ω^2 falls linearly up to 34 GPa , the highest pressure covered in the present investigation. In the absence of experimental data above 34 GPa , the fitted parameters obtained from the linear curve fitting of the experimental data were used to extrapolate the mode frequency with respect to pressure in the high-pressure region ($>34\text{ GPa}$), and we found that the 557 cm^{-1} A_g soft

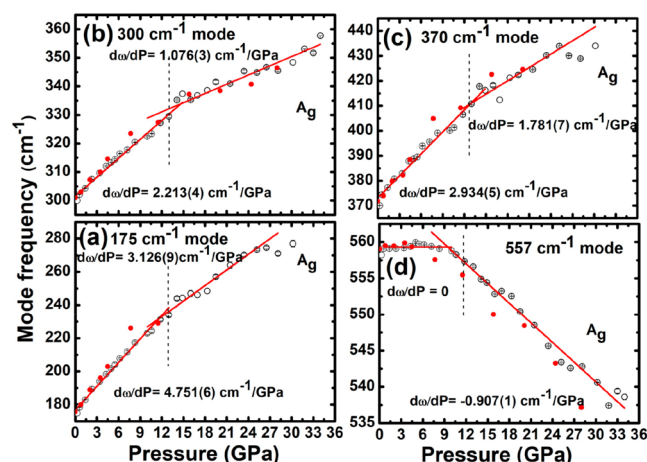


Figure 6. Pressure-dependent Raman mode frequencies of TiS_3 nanoribbons. Open and solid symbols correspond to data during the compression and decompression cycle, respectively. Solid lines are linear fits to the frequencies. The A_g mode at 557 cm^{-1} softens and undergoes a dramatic change in slope $d\omega/dP$ near 13 GPa. All other A_g modes harden with pressure and exhibit a slope change $\sim 13\text{ GPa}$, the onset of transition pressure.

mode vanishes at 43 GPa, the expected complete structural transition pressure. It can be mentioned here that first-principles calculations of the strain dependence of phonon dispersion curves of TiS_3 monolayer predicted a structural instability⁴³ at large strain values of 6%–8%. Upon increasing strain up to 6%–8%, we observed imaginary vibrational frequencies at the Γ -point of the reported phonon dispersion curves pointing toward a structural transition,⁴³ and that theoretically predicted transition is corroborated experimentally in our present HP Raman spectroscopic investigation. The pressure dependencies of the line widths of the observed phonon modes are shown in Figure 8. The line widths of the rigid chain mode at 175 cm^{-1} and internal modes located at 300 and 370 cm^{-1} broaden at high pressure. These internal modes line widths are found to nearly double between 13 and 34 GPa. Upon increasing pressure, the anisotropy of the Ti–S octahedra increases. As a result, the distribution of Ti–S bond lengths is expected to increase. This leads to an increase in the inhomogeneous broadening of these internal vibrations. The broadening of rigid mode and the corresponding decrease in mode intensity with increasing pressure indicate a possibility of

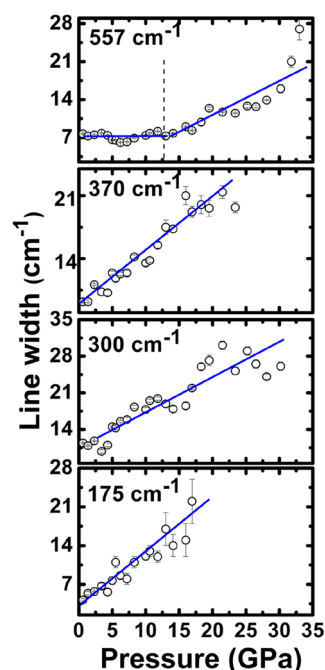


Figure 8. Pressure-dependent line widths of Raman bands of TiS_3 nanoribbons. Open symbols correspond to the evolution of band line widths during compression. Solid lines are a guide to the eye. Vertical dotted lines show change of slope at the onset of the transition pressure.

growth of static positional disorder.^{44,49} On the other hand, the line width of highest frequency band at 557 cm^{-1} is found to be insensitive to pressure up to $\sim 13\text{ GPa}$, followed by an increasing trend, and thus shows an anomaly around 13 GPa as depicted in Figure 8. The line width of this mode increases by a factor of 4 between 13 and 34 GPa. It can be mentioned here that such an anomaly in the line width of phonons was observed when materials undergo either a structural⁵⁰ or an electronic transition (MoS_2 ⁴⁰ and Sb_2Se_3 ⁵¹).

To identify the phase transition in 2D materials, an intensity analysis of Raman mode is useful.⁴⁰ This is due to the fact that light couples directly with the polarization in materials and manifests in its intensity.⁵² The integrated Raman intensity of the band at 557 cm^{-1} was normalized with respect to that of the band at 300 cm^{-1} . The latter band corresponds to the

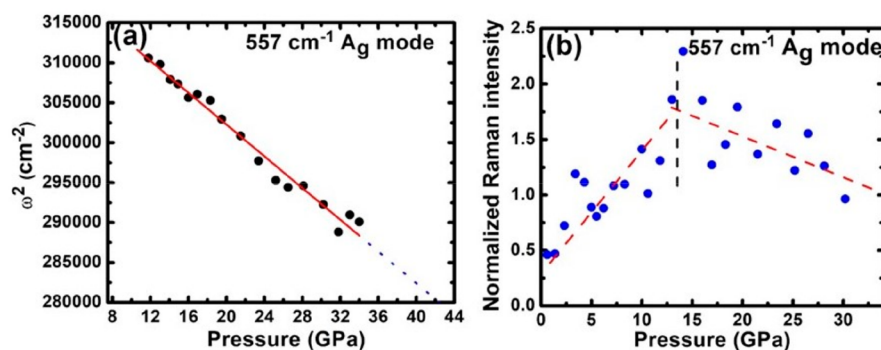


Figure 7. (a) Pressure dependence of the square of the Raman-active A_g soft mode frequency. The solid line is a linear fit to the data, and the dotted line corresponds to extrapolated line into the high-pressure field indicating the complete structural phase transition at 43 GPa. (b) Pressure dependencies of normalized integrated intensity of the 557 cm^{-1} mode with respect to that of mode at 300 cm^{-1} . The curve drawn through the data point is a guide to the eye. The vertical line shows the anomaly in intensity behavior around the onset of transition pressure, $\sim 13\text{ GPa}$.

Table 1. Pressure Coefficients and Grüneisen Parameters of Raman Modes in the Ambient Phase and High-Pressure Phase^a

ambient phase						HP phase	
mode frequency (cm ⁻¹)			P-coefficient (cm ⁻¹ /GPa)	reduced P-coefficient (GPa ⁻¹)	Grüneisen parameter γ_i	P-coefficient (cm ⁻¹ /GPa)	reduced P-coefficient (GPa ⁻¹)
0.6 GPa	300 K	88 K ^b					
178 (A _g)	175	177	4.75(4)	0.027142	1.3571	3.12(8)	0.017828
302 (A _g)	300	302	2.21(9)	0.007366	0.3683	1.07(7)	0.003566
374 (A _g)	370	371	2.93(7)	0.007918	0.3959	1.78(1)	0.004811
559 (A _g)	557	560	0 ^c			-0.91(3)	-0.001634

^aNumbers in the parentheses are the standard errors in the least significant digit. ^bReference 30. ^cError larger than the slope value is considered as zero.

internal vibrations of the TiS₆ octahedron, which persists over the entire pressure range of this study and exhibits a slight change in its intensity upon compression. The pressure dependence of the normalized intensity of the 557 cm⁻¹ mode is shown in Figure 7b. One can notice that the intensity increases up to 13 GPa and subsequently falls beyond this pressure. Hence, the observed anomaly in the mode intensity around 13 GPa is similar to the onset of transition pressure as noticed from Figures 6 and 8 and further corroborates that the TiS₃ nanoribbons undergo a pressure-induced structural transition which begins at ~13 GPa. The Raman spectra were also recorded in the pressure-released cycle, and their peak positions with pressure follow the same trend as observed in the compression cycle is shown in Figure 5. After complete pressure release, the Raman spectrum is found to be reversible and matches with the ambient spectrum obtained from the pristine TiS₃ nanoribbon sample as shown in Figure 4. This suggests that the pressure-induced structural transition of TiS₃ nanoribbons is completely reversible.

The pressure coefficients of Raman bands were obtained by linear fit to ω_i vs P curves in ambient and HP phases (Figure 6). The low-frequency A_g mode which is attributed to the rigid chain vibration has a large pressure dependency, while the 557 cm⁻¹ mode is insensitive to pressure (slope ~0) in the ambient phase. Frequencies of modes at 175, 300, and 370 cm⁻¹ increase upon compression and follow normal hardening behavior, whereas the highest frequency mode located at 557 cm⁻¹ becomes almost constant in the ambient phase below 13 GPa. The pressure dependencies of modes in the HP phase (13–34 GPa) are found to be less compared to those of the ambient phase. The pressure coefficient of the band at 557 cm⁻¹ is found to be negative (Table 1). The reduced slopes $\frac{1}{\omega_i} \frac{\partial \omega_i}{\partial P}$ of phonon modes, which essentially represent the bond strength of phonon vibrational modes⁴⁴ in both phases, were obtained (Table 1). In the ambient phase, the reduced slope of the 175 cm⁻¹ rigid chain mode is larger than the others, indicating that this mode is more sensitive to pressure and gets easily compressed and deformed, whereas a low value of reduced slopes for the 300 and 370 cm⁻¹ internal modes suggests that these vibrations are comparatively less sensitive to pressure. Because the normalized slopes of modes in HP phase are lower than those in the ambient phase (Table 1), the mode vibrations in the HP phase are less influenced by compression, and the bonds involving these atomic vibrations are stiffer. The band at 557 cm⁻¹ has a moderate value of negative reduced slope pointing toward its contribution to negative thermal expansion in the HP phase. Mode Grüneisen parameters of A_g modes which exhibit out-of-plane vibration (along c -axis) are calculated by using the relationship⁴⁴ $\gamma_i = (\omega_i \chi_c)^{-1} \frac{\partial \omega_i}{\partial P}$, where χ_c

is the axial compressibility along the c -axis. Using the reported $\chi_c^{-1} = 50$ GPa,³⁷ γ_i values were calculated by using our experimental pressure dependencies of mode frequencies. Mode Grüneisen parameters γ_i of A_g modes of the ambient phase are listed in Table 1. A larger value of γ_i of the rigid mode 175 cm⁻¹ implies more contribution to the thermal expansion coefficient. These microscopic parameters γ_i are key input for obtaining thermal properties such as the specific heat and thermal expansion coefficient of materials, as described below.

The contribution of the observed A_g phonon modes to thermal expansion coefficient α of the TiS₃ nanoribbon can be estimated by using $\alpha = (\gamma_{av} C_v) / (3V_m \chi_c^{-1})$, where C_v is the molar specific heat at constant volume, γ_{av} is the average Grüneisen parameter, V_m is the molar volume, and χ_c^{-1} is the axial compressibility along the c -axis. Using Einstein's specific heat, $C_i = R[y_i^2 \exp(y_i)] / [\exp(y_i) - 1]^2$, where $y_i = \hbar \omega_i / k_B T$ and R is the gas constant, the total specific heat C_v can be obtained ($i = 1-4$), and γ_{av} is defined as $\gamma_{av} = 1/2(\sum p_i C_i \gamma_i) / C_v$, where p_i is the degeneracy of phonon at Brillouin zone center. Using the unit cell volume, the molar volume is calculated as 4.439×10^{-5} m³/mol; the reported χ_c^{-1} value is taken as 50 GPa,³⁷ and considering our experimental mode Grüneisen parameters, the linear thermal expansion along the c -axis at 300 K is estimated as $\alpha = 1.321 \times 10^{-6}$ K⁻¹, which is found to be of the same order of magnitude as the one reported in MoS₂.⁵³

As mentioned earlier, the change in mode frequency due to the effect of temperature has two contributions. One is due to the phonon–phonon interaction at constant volume known as the intrinsic anharmonic contribution (explicit), and other one is an implicit contribution that solely depends on the lattice volume change. The latter implicit quasi-harmonic contribution is obtained from pressure dependence of mode frequency and enables one to infer the explicit, true anharmonicity of phonon modes. The Raman mode frequencies of TiS₃ observed by compression at 0.6 GPa and cooling to 88 K³⁰ are listed in Table 1. Inspection of these Raman shift values indicates that several of the mode frequencies obtained by way of compression is nearly the same as that obtained by volume contraction achieved by cooling. This suggests that most of the mode frequencies behavior with temperature is mainly due to the quasi-harmonic contribution (caused by only change in lattice volume). However, the mode at 370 cm⁻¹ has a significant difference in its Raman shift with temperature and pressure, indicating the highly anharmonic nature which arises from the phonon–phonon interaction. The present phonon spectral studies at high pressure provide insight into the role of several A_g phonons in the structural phase transition and thermal expansion in TiS₃ nanoribbons. Our experimental

finding is consistent with the theoretically predicted structural phase transition from phonon dispersion calculations.

CONCLUSIONS

In conclusion, we report pressure-dependent Raman spectroscopic studies of out-of-plane A_g phonons in titanium trisulfide nanoribbons prepared by the direct solid–gas reaction method and infer their cross-plane thermal expansion coefficient. Out of four prominent observed A_g modes, the phonon modes at 175, 300, and 370 cm^{-1} show a blue-shift upon compression while that located at 557 cm^{-1} exhibits softening about 13 GPa. Anomalies in phonon mode frequencies and excessive broadening in line width of the soft phonon about 13 GPa are attributed to the onset of a structural transition. The complete structural phase transition at 43 GPa is observed from extrapolation, consistent with recent reported *ab initio* theoretical predictions. The first-order pressure coefficients of A_g modes are found to lie in the range 0.3–1.5 $\text{cm}^{-1}/\text{GPa}$. The experimental mode Grüneisen parameter of the 175 cm^{-1} vibrational A_g mode is large and sensitive to pressure. Using these mode Grüneisen parameters, we estimated cross-plane thermal expansion at 300 K to be $1.32 \times 10^{-6} \text{ K}^{-1}$. The structural change in the TiS_3 nanoribbon under compression is recovered in the pressure release cycle. The observed results can be crucial in calibration and correction of performance of next-generation TiS_3 -based nanoelectronics and optical devices under extreme stress conditions.

EXPERIMENTAL SECTION

Titanium trisulfide layers have been obtained by direct sulfuration of a Ti disc (diameter 10 mm) in a vacuum-sealed ampule at 550 °C for 20 h (see the schematic in Figure 2a–c). The used Ti discs were etched in $\text{HNO}_3\text{:HF}$ (30:4 wt %) for 1 min, followed by washed in distilled water, and subsequently dipped in ethanol. The heating rate was $0.4 \pm 0.05 \text{ }^\circ\text{C}/\text{min}$, and the cooling took place in ambient conditions. The detailed synthesis procedure for TiS_3 nanoribbons is described in our earlier reports.^{30,32} The microstructure of the materials was investigated by using a scanning electron microscope. The thickness of the nanoribbon was characterized by using an atomic force microscope (ICON system, Bruker, Santa Barbara, CA) operated in contact mode. *In situ* high-pressure Raman spectroscopic measurements were performed by using a compact, symmetric diamond anvil cell (DAC) with diamonds of culet diameter 500 μm . Raman spectra were recorded by using a micro-Raman spectrometer (Renishaw inVia) operated in the backscattering configuration. A 514.5 nm line from an Ar-ion laser with an output power of $\sim 0.4 \text{ mW}$ is used for excitation of the sample. A few flakes of TiS_3 sample were loaded into a 200 μm hole of a preindented stainless-steel gasket in the DAC. A 4:1 methanol–ethanol mixture was used as the pressure transmitting medium and ruby was loaded along with the sample as a pressure calibrant (see the schematic in Figure 2d,e). Measurements were performed up to $\sim 34 \text{ GPa}$, covering the wavenumber range 100–650 cm^{-1} . Raman spectra were also recorded during the pressure-reducing cycle. The pressure transmitting medium is quasi-hydrostatic up to 10 GPa, and beyond this pressure the experimental results can be influenced by uniaxial pressure.^{49,54} The spectra were analyzed by Lorentzian line shapes using Peak Fit software (JANDEL).

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Author Contributions

D.J.L. and A.C.-G. conceptualized this project. J.O.I., E.F., J.R.A., C.S., I.J.F., H.S.J.Z.A.P., A.C.-G., and D.J.L. prepared and characterized the materials. K.K.M. and T.R.R. performed the pressure dependence Raman spectroscopy measurements of the TiS_3 nanoribbon and processed the data. A.P. and K.K.M. prepared the all schematic representations. All authors analyzed and discussed the data. K.K.M. and D.J.L. wrote the first draft of the manuscript, with all authors commenting and editing the manuscript.

Notes

The authors declare no competing financial interest.

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