

## Characterization of bainitic/martensitic structures formed in isothermal treatments below the $M_s$ temperature

Navarro-López, A.; Hidalgo Garcia, J.; Sietsma, J.; Santofimia Navarro, M.J.

**DOI**

[10.1016/j.matchar.2017.04.007](https://doi.org/10.1016/j.matchar.2017.04.007)

**Publication date**

2017

**Document Version**

Final published version

**Published in**

Materials Characterization

**Citation (APA)**

Navarro-López, A., Hidalgo Garcia, J., Sietsma, J., & Santofimia Navarro, M. J. (2017). Characterization of bainitic/martensitic structures formed in isothermal treatments below the  $M_s$  temperature. *Materials Characterization*, 128, 248-256. <https://doi.org/10.1016/j.matchar.2017.04.007>

**Important note**

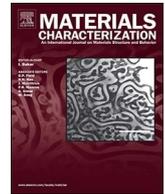
To cite this publication, please use the final published version (if applicable). Please check the document version above.

**Copyright**

Other than for strictly personal use, it is not permitted to download, forward or distribute the text or part of it, without the consent of the author(s) and/or copyright holder(s), unless the work is under an open content license such as Creative Commons.

**Takedown policy**

Please contact us and provide details if you believe this document breaches copyrights. We will remove access to the work immediately and investigate your claim.



# Characterization of bainitic/martensitic structures formed in isothermal treatments below the $M_s$ temperature

A. Navarro-López\*, J. Hidalgo, J. Sietsma, M.J. Santofimia

Department of Materials Science and Engineering, Delft University of Technology, Mekelweg 2, 2628 CD Delft, The Netherlands

## ARTICLE INFO

### Keywords:

Martensite  
Bainitic ferrite  
Austenite decomposition  
Isothermal treatments  
Dilatometry  
EBSD characterization

## ABSTRACT

Advanced Multiphase High Strength Steels are generally obtained by applying isothermal treatments around the martensite start temperature ( $M_s$ ). Previous investigations have shown that bainitic ferrite can form from austenite in isothermal treatments below  $M_s$ , where its formation kinetics is accelerated by the presence of the athermal martensite. That athermal martensite is tempered during the isothermal treatment, and fresh martensite may form during the final cooling to room temperature. The distinction between product phases present after the application of this type of heat treatments is difficult due to morphological similarities between these transformation products. The aim of this study is to characterize the structural and morphological features of the product phases obtained in isothermal treatments below the  $M_s$  temperature in a low-carbon high-silicon steel. Multiphase microstructures, having controlled fractions of product phases, were developed by applying isothermal treatments above and below  $M_s$ , and were further studied by electron back scatter diffraction (EBSD) and scanning electron microscopy (SEM). The bainitic or martensitic nature of these product phases is discussed based on this characterization. Results showed that bainitic ferrite appears in the form of acicular units and irregularly shaped laths. Tempered martensite appears as laths with a sharp tip and as relatively large elongated laths with wavy boundaries containing protrusions.

## 1. Introduction

The development of new Advanced High Strength Steels (AHSS) requires the implementation of improved processing routes in which the resulting microstructures present a high degree of complexity. That is the case of multiphase steels obtained by applying isothermal holdings around the martensite start temperature ( $M_s$ ) after austenitization, leading to microstructures generally formed by a mixture of martensite (fresh and/or tempered), bainite, and retained austenite. The presence of different product phases from austenite decomposition depends on whether the isothermal holding is applied at a temperature above  $M_s$  or below  $M_s$ .

After holding at a temperature above  $M_s$  and further quench, researchers agree that, in hypoeutectoid steels, the matrix is mainly formed by bainitic structures with retained austenite in the form of thin films and/or martensite-austenite (MA) islands [1–7]. Bainite is thus the isothermal product obtained from the decomposition of austenite. Carbides can also be present in bainite, depending on the composition of the steel and the isothermal holding temperature and time. Below  $M_s$ , however, the nature of the products formed during isothermal holding

is unclear due to the coexistence of product phases with morphological similarities such as tempered martensite and bainite.

Cooling from austenitization to a temperature between the martensite start ( $M_s$ ) temperature and room temperature (RT) leads to the formation of a certain volume fraction of athermal martensite. This martensite has a strong accelerating effect on the subsequent transformation kinetics, mainly due to a higher density of potential nucleation sites [8–11]. Although bainite has been reported in hypoeutectoid steels as an isothermal decomposition product from austenite in isothermal treatments below  $M_s$  [4,10–14], there is experimental evidence showing the formation of other types of product phases in similar heat treatments. For example, several authors observed an isothermal product characterized by wide laths with characteristic wavy boundaries with ledges [16,17]. Kim et al. [7,15,16] stated that this isothermal product is neither purely martensitic nor purely bainitic, since it shows similarities with both types of product phases. Somani et al. [17] identified this product as isothermal martensite since it presents clear similarities to athermal martensite. According to the literature, isothermal martensite has mainly been reported in high-carbon steels and high-nickel alloys [18–20], but not in hypoeutectoid

\* Corresponding author.

E-mail addresses: [a.navarrolopez@tudelft.nl](mailto:a.navarrolopez@tudelft.nl) (A. Navarro-López), [j.hidalgogarcia@tudelft.nl](mailto:j.hidalgogarcia@tudelft.nl) (J. Hidalgo), [j.sietsma@tudelft.nl](mailto:j.sietsma@tudelft.nl) (J. Sietsma), [m.j.santofimianavarro@tudelft.nl](mailto:m.j.santofimianavarro@tudelft.nl) (M.J. Santofimia).

<http://dx.doi.org/10.1016/j.matchar.2017.04.007>

Received 31 January 2017; Received in revised form 31 March 2017; Accepted 2 April 2017

Available online 06 April 2017

1044-5803/ © 2017 The Authors. Published by Elsevier Inc. This is an open access article under the CC BY-NC-ND license (<http://creativecommons.org/licenses/by-nc-nd/4.0/>).

steels. Independently of the growth mechanism of these product phases, the dilemma is which microstructural features observed in isothermal treatments below  $M_s$  correspond to each phase formed in those treatments, such as bainitic ferrite, tempered martensite or fresh martensite.

The goals of this study are: to characterize the microstructural features obtained in isothermal treatments performed at temperatures above and below  $M_s$ ; relate those features to the product phases formed at each isothermal temperature; and determine, based on that characterization, the martensitic or bainitic nature of those product phases. For this purpose, isothermal heat treatments were applied at different temperatures around the  $M_s$ -temperature that were selected based on a previous study [11]. Various microstructures formed by combinations of different volume fractions of bainitic ferrite, tempered martensite, and fresh martensite were created. Volume fractions of those product phases were extracted from the dilatometry results presented in earlier work [11]. The identification of phases by dilatometry in combination with the characterization of microstructures formed at different temperatures by electron backscatter diffraction (EBSD) and scanning electron microscopy (SEM) provided a straight and systematic way of phase analysis.

## 2. Experimental procedure

The chemical composition of the investigated steel is 0.2C-3.51Mn-1.52Si-0.25Mo-0.04Al (wt. pct). The as-received material was hot rolled into a 4 mm-thick steel slab. Dilatometry specimens were extracted from hot rolled slabs, parallel to the rolling direction. These were cylindrical of dimensions 10 mm in length and 3.5 mm in diameter. Heat treatments were carried out in a Bähr 805A dilatometer. Specimens were placed between two quartz rods, heated by an induction coil, and cooled using nitrogen gas. A thermocouple was spot-welded in the middle of the specimens to control the temperature.

Different heat treatments were applied above and below the  $M_s$ -temperature, after a full austenitization at 900 °C for 240 s. The applied heat treatments are described as follows (Fig. 1):

- (i) A direct-quench treatment was performed to determine the experimental  $M_s$ -temperature and the kinetics of the martensite formation. The analysis of the dilatometric curve, presented elsewhere [11], determined an experimental  $M_s(1\%) = 320 \text{ °C} \pm 5 \text{ °C}$ , at which temperature the volume fraction of martensite formed is 1%.
- (ii) An isothermal treatment above the  $M_s$ -temperature was carried out at 340 °C to obtain a microstructure formed by a mixture of bainite, retained austenite and fresh martensite.
- (iii) Three isothermal treatments between  $M_s$  and room temperature (RT) were performed at 320 °C, 300 °C and 270 °C to obtain a

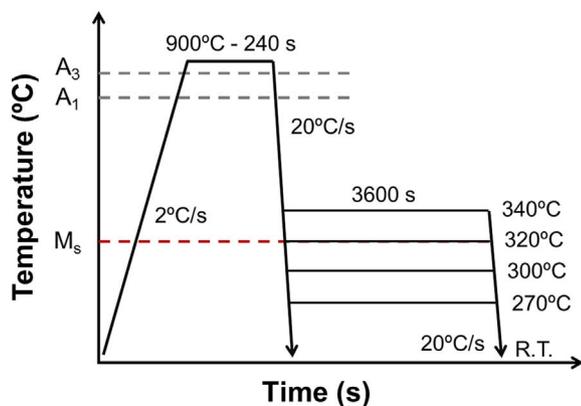


Fig. 1. Schematic representation of the heat treatments applied in the dilatometer to the selected steel.

multiphase microstructure formed by martensite obtained during cooling from austenitization to the isothermal temperature, bainitic ferrite, retained austenite, and fresh martensite obtained in the final cooling to room temperature.

The volume fractions of retained austenite at room temperature were determined by X-ray diffraction (XRD) experiments using a Bruker D8-Advance diffractometer equipped with a Bruker Vantec Position Sensitive Detector. CoK $\alpha$  radiation was used in the  $2\theta$  scan from 40° to 130° with a step size of 0.035°. The fractions of austenite and ferrite were calculated by the integrated area method using the (111), (200), (220), and (311) austenite peaks, and the (110), (200), (211), and (220) ferrite peaks [21]. The heat-treated specimens were metallographically prepared by grinding and polishing. 2% Nital etching was applied to the specimens to reveal the microstructure. A pre-analysis of microstructures by Light Optical Microscopy (LOM) and Scanning Electron Microscopy (SEM) was done and presented elsewhere [11].

Electron Back Scatter Diffraction (EBSD) in combination with SEM was used to perform a detailed characterization of the transformation products obtained in the heat treatments previously described. In order to analyse the same area by EBSD and SEM, various micro-indentations (HV1) were made in the specimens to select representative areas. As a final preparation step, the specimens were mechanically polished with a colloidal silica solution in order to exhaustively clean the specimens and remove surface deformation. Areas selected in the specimens were first analysed by EBSD.

The EBSD patterns were acquired on a FEI Quanta 450 scanning electron microscope equipped with a Field Emission Gun (FEG-SEM) by means of the OIM Data Collection software. The analysis was performed under the following conditions: acceleration voltage of 20 kV, spot size 5, working distance of 16 mm, tilt angle of 70°, and step size of 50 nm in a hexagonal scan grid. The orientation data were post-processed with TSL OIM® Analyses 6.0 software. First, a grain confidence index (CI) standardization was applied to the raw data, selecting a tolerance angle of 5° and a minimum grain size of 6 pixels, and considering that grains are formed by multiple pixel rows. The second step of the post-processing procedure included neighbour-orientation correlation with a tolerance angle of 5° and a minimum confidence index of 0.1. Finally, a down-filtering criterion of confidence index higher than 0.1 was applied to reduce the number of pixels poorly indexed, i.e., not attributed to either fcc or bcc phases.

After a comprehensive EBSD analysis, specimens were carefully polished with a colloidal silica solution for 5 to 10 min in order to remove the carbon deposition layer of the specimen surface. Afterwards, 2% Nital etching was applied to the specimens to reveal the different phases formed during the heat treatments. The EBSD-analysed areas were then studied with a JEOL JSM-6500F Scanning Electron Microscope (SEM) using a 15 kV electron beam and the Secondary Electron Imaging (SEI) detection mode.

## 3. Results and discussion

### 3.1. Volume fraction of phases

There was no evidence of the formation of ferrite, pearlite, or bainite during the initial cooling to the isothermal temperatures above and below  $M_s$  [11]. Only when the temperature decreased below the  $M_s$  temperature, a dilatation was observed due to the formation of a certain fraction of athermal martensite prior to the isothermal holding. Afterwards, dilatometric curves showed a significant dilatation during the isothermal holding, which implies the formation of an isothermal product phase identified as bainitic ferrite in ref. [11]. There was also a small deviation from linearity of the dilatometric curves during the final cooling to room temperature, showing the formation of fresh martensite during that cooling. The determination of the volume fractions of product phases obtained in the applied heat treatments is

**Table 1**  
Phase fractions obtained after the application of the selected heat treatments.  $f^{PM}$ ,  $f^{BF}$ ,  $f^{FM}$ , and  $f^{RA}$  are the experimental fractions of Prior Athermal Martensite, Bainitic Ferrite, Fresh Martensite, and Retained Austenite, respectively [11].

Temperature (°C)	$f^{PM}$	$f^{BF}$	$f^{FM}$	$f^{RA}$
340	–	0.74	0.16	0.10
320	< 0.02	0.84	0.07	0.08
300	0.16	0.71	0.05	0.09
270	0.77	0.12	0.02	0.09

explained in ref. [11] and shown in Table 1.

### 3.2. Microstructural features

Microstructures were characterized by a combined analysis of SEM and EBSD images in order to find patterns regarding the structure and morphology of the different product phases. Distinct microstructural features with specific characteristics were observed and described in detail below. The descriptions will allow a better characterization and understanding of the nature of the product phases obtained in each isothermal treatment. A schematic representation of these features is shown in Fig. 2. Examples of these microstructural features will be shown in further sections.

#### 3.2.1. Thin features ( $S_T$ )

This microstructural feature, defined as ‘ $S_{Thin}$ ’ ( $S_T$ ), appears in the

form of a thin unit with acicular shape.  $S_T$  units are aligned parallel with each other and appear as low-relief microstructures after etching. Their aspect ratio ( $a/c$ ), where ‘ $a$ ’ and ‘ $c$ ’ are the length and width, respectively, can vary in a wide range between 5 and 15. The width of  $S_T$  units is generally less than 1  $\mu\text{m}$ , and the length varies from 3.5  $\mu\text{m}$  to 15  $\mu\text{m}$ . Most of them have a mean length between 5 and 8  $\mu\text{m}$  and a mean width between 0.6 and 0.8  $\mu\text{m}$ . Carbides are not observed within this type of microstructural feature.

#### 3.2.2. Irregular features ( $S_I$ )

This second microstructural feature has an irregular lath-shaped morphology without carbides and is defined as ‘ $S_{Irregular}$ ’ ( $S_I$ ). These lath-shaped features also appear as low-relief microstructures after etching. The length of the  $S_I$  features varies between 4 and 7  $\mu\text{m}$  and the width between 1 and 3  $\mu\text{m}$ , so  $S_I$  features are wider than  $S_T$ , with a lower aspect ratio.

#### 3.2.3. Thick features ( $S_{TK}$ )

The third type of microstructural feature appears in the form of laths with a sharp tip at one of its edges. These lath-shaped features are defined as ‘ $S_{Thick}$ ’ ( $S_{TK}$ ). They are aligned parallel and appear as low-relief microstructures after etching. Carbides are generally visible within these features. The aspect ratio of  $S_{TK}$  varies in the range of 2.5 to 6. These lath-shaped structures have a mean length and width of 5  $\mu\text{m}$  and 1.1  $\mu\text{m}$ , respectively.  $S_{TK}$  features seem to grow from the prior austenite grain boundaries, and are surrounded by  $S_T$  units.

#### 3.2.4. Elongated features ( $S_E$ )

This fourth microstructural feature presents a big elongated, lath-shaped morphology, so it is defined as ‘ $S_{Elongated}$ ’ ( $S_E$ ). This type of lath-shaped features are aligned parallel to adjacent constituent units and appear as low-relief microstructures. Their size is larger than that of the previous microstructural features described.  $S_E$  features have a mean length and width of 12  $\mu\text{m}$  and 2  $\mu\text{m}$ , respectively. In general,  $S_E$  features are longer and wider than  $S_T$ . There is an abundant presence of carbides within most of  $S_E$  features.

The most relevant characteristic of  $S_E$  is the presence of wavy boundaries. These wavy boundaries were first reported by Kim et al. [16], who concluded that the nature of these isothermal products formed below  $M_s$  was neither purely martensitic nor purely bainitic. More recently, Somani et al. [17] identified an isothermal product in the form of laths with wavy boundaries, formed during isothermal treatments below  $M_s$ . Ledge-like protrusions are also observed in these laths. Somani et al. identified this product phase as isothermal martensite.

#### 3.2.5. Blocky features ( $S_B$ ) (also known as MA islands)

This microstructural feature is characterized by a blocky structure which combines small parallel aligned laths within the structure with irregular blocks at its boundaries. This feature, defined as ‘ $S_{Blocky}$ ’ ( $S_B$ ), is generally known as martensite-austenite island (MA), which is formed from partly untransformed austenite in the final cooling to room temperature. SEM shows the MA island as a non-etched structure with a smooth surface. However, EBSD analysis displays that the MA island is formed by an internal structure of very small martensite laths surrounded by small blocks of retained austenite (RA) at the boundaries. These small laths appear aligned parallel within the island. The length of the laths is less than 2  $\mu\text{m}$ .

### 3.3. Microstructures obtained from heat treatments

#### 3.3.1. At 340 °C (above $M_s$ )

According to dilatometry results, bainitic ferrite and fresh martensite coexist in this microstructure in percentages of 74% and 16% volume fractions, respectively (see Table 1), completed by 10% of retained austenite. In this case, there is no formation of athermal

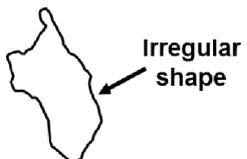
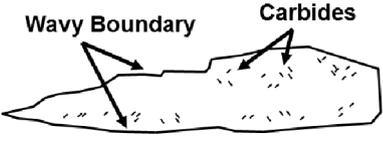
Structural Feature	Schematic Representation
$S_{Thin}$ ( $S_T$ )	 Acicular units
$S_{Irregular}$ ( $S_I$ )	 Irregular shape
$S_{Thick}$ ( $S_{TK}$ )	 Carbides Sharp tip
$S_{Elongated}$ ( $S_E$ )	 Wavy Boundary Carbides
$S_{Blocky}$ ( $S_B$ ) (MA island)	 Small laths RA

Fig. 2. Schematic representations of the different features described, directly derived from actual micrographs.

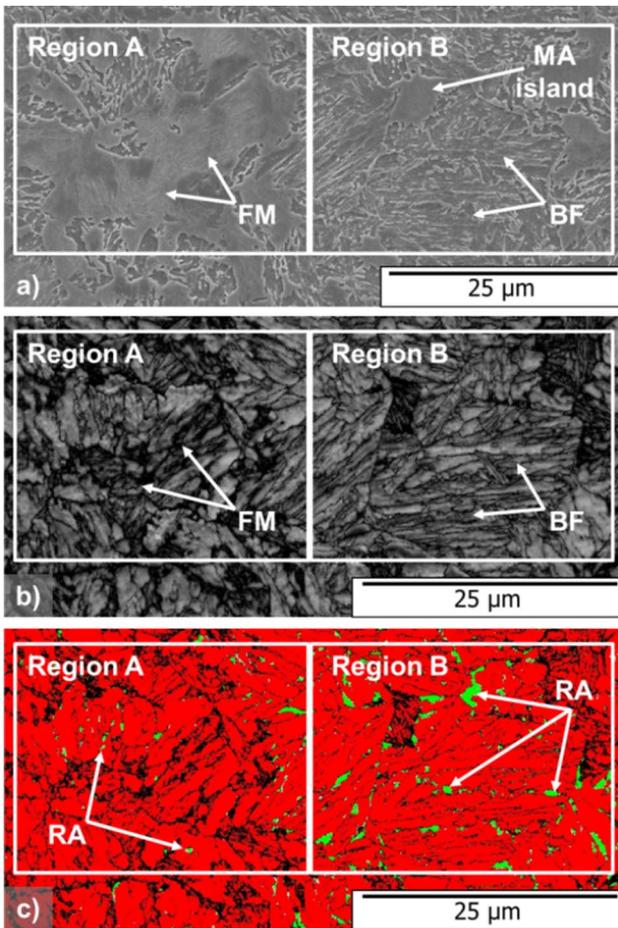


Fig. 3. a) SEM micrograph, b) Image Quality (IQ) map, and c) Phase Distribution (PD) map of a determined area of the specimen isothermally treated at 340 °C (above  $M_s$ ).

martensite prior to the isothermal treatment. The microstructure is thus a mixture of bainitic ferrite, retained austenite and MA islands. Fig. 3 shows the SEM micrograph, the Image Quality (IQ) map, and the Phase Distribution (PD) map of the same area of the specimen. The combined analysis of images allows to distinguish two different sub-areas: Region A, in which different features of fresh martensite appear in combination with small bainitic ones, and Region B, in which bainitic ferrite is the dominant product phase. Bainitic ferrite mainly appears as  $S_T$  features with retained austenite (RA) between them. These bainitic areas are generally better indexed in EBSD than the martensitic ones, so they appear as bright areas in the IQ map due to a higher confidence index (see Fig. 3.b). Conversely, fresh martensite areas have a poorer indexing than the bainitic ones due to internal strains and dislocations. This means that the confidence index will be lower in those areas, and darker regions will appear in the IQ map, as shown in Fig. 3.b.

The distribution and morphology of the retained austenite is also an important factor to distinguish between different product phases. Fig. 3.c shows the Phase Distribution (PD) map of the above-mentioned specimen. As observed, retained austenite (in green) mainly appears within the bainitic ferrite areas (Region B) in the form of small blocks or thin films between bainitic ferrite units. This means that, after the incomplete isothermal transformation (approx. 74% of bainitic ferrite), the remaining austenite in the bainitic areas is more stable than in the areas that become martensitic due to a higher carbon enrichment during the isothermal holding. In the final cooling, while the remaining austenite in bainitic areas will be retained, most of the remaining austenite present in martensitic areas will transform into fresh martensite due to its lower stability. This explains the lower fraction of retained austenite obtained in martensitic areas.

For a detailed description of the different product phases, a prior austenite grain was selected and analysed by a combined study of its SEM micrograph and three different EBSD maps, as shown in Fig. 4. Inverse Pole Figures (IPF) and Kernel Average Misorientation (KAM) maps were used to determine the morphology (size and shape) of the different product phases and the distribution of local grain misorientations, respectively. KAM maps were calculated by considering the 1st

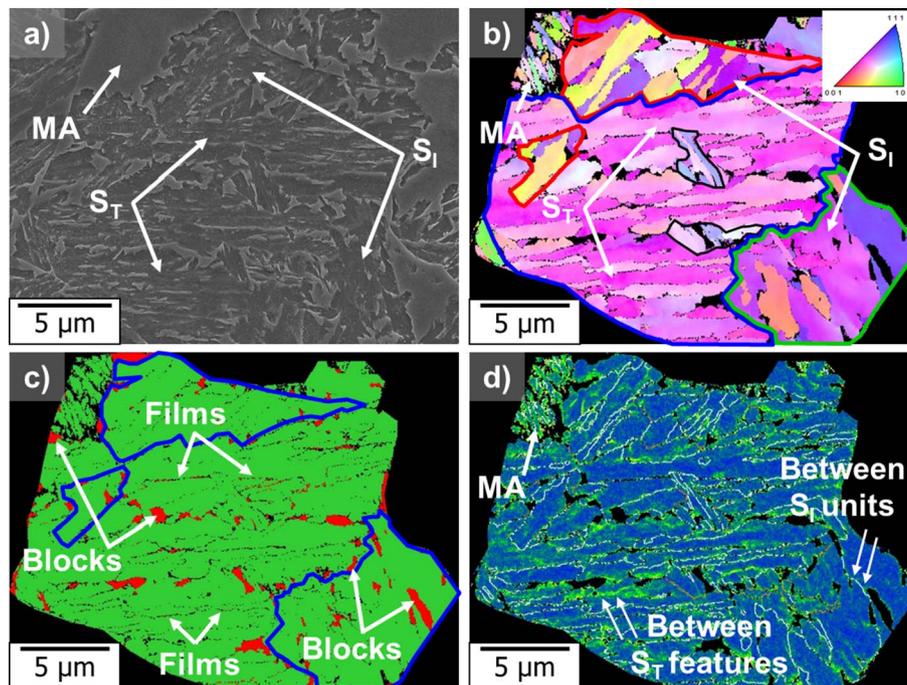


Fig. 4. a) SEM micrograph, b) IPF map, c) PD map, and d) KAM map of a selected prior austenite grain of the specimen isothermally treated at 340 °C. Regions are marked on the IPF map by different colours.

neighbours as the nearest neighbour points of the kernel in order to include information about the surroundings of the measured point.

The selected grain is mainly formed by four regions with different orientation relationships (ORs) (see Fig. 4.b). Analysing the central region of the austenite grain (marked by a blue contour), a typical bainitic structure can be distinguished, which is mainly formed by  $S_T$  features. Retained austenite appears in the form of very thin films between the units (see Fig. 4.c). Elongated blocky structures of retained austenite are also distinguished close to region boundaries and at the boundaries of the MA island, as shown in Fig. 4.c.

$S_I$  features can be observed in the microstructure (see Fig. 4.a, b). Although the morphology of these laths is not acicular, their nature may be bainitic. The reason for this proposition is that there is no dilatometric evidence of the formation of prior athermal martensite, so bainitic ferrite cannot be confused with tempered martensitic structures in this microstructure. Fig. 4 also shows a martensite-austenite (MA) island in which its internal structure of small martensite laths can be observed (Fig. 4.b). The KAM map (Fig. 4.d) shows that these martensitic structures present higher concentrations of local misorientation than the bainitic ones, so the internal structure is more distorted. The KAM map also shows that higher local misorientations are concentrated at the boundaries between  $S_T$  features, and/or between  $S_I$  units, as shown in Fig. 4.d (white arrows).

### 3.3.2. At 320 °C (at $M_s$ )

This isothermal temperature corresponds to the experimental  $M_s$ -temperature at 1%, i.e., when the volume fraction of martensite formed is approximately 1%. This prior athermal martensite formed during cooling from austenitization to the isothermal temperature will be tempered during the isothermal holding, so its appearance will be different from that of the fresh martensite obtained in the final cooling. A complex multiphase microstructure is thus obtained at this temperature (see Table 1), where bainitic ferrite represents more than 80% of the volume fraction.

$S_T$  features containing films of retained austenite appear in the microstructure, as shown in Fig 5.a. and 5.b. According to the previous characterization done above  $M_s$ , these features correspond to bainitic ferrite structures. MA islands can also be distinguished in the microstructure (see Fig. 5.a).  $S_{TK}$  features are also present at this temperature with their characteristic sharp tip at one of the edges. These lath-shaped structures are different from the ones identified at 340 °C, above  $M_s$ . Fig. 6.a shows an  $S_E$  feature, i.e., a large elongated lath containing carbides. This  $S_E$  feature presents wavy boundaries, which can be observed in Fig. 6.b (marked with white arrows) and 6.c (marked with blue dashed line). In addition to this specific characteristic, ledge-like protrusions can be observed in one of the boundaries of this  $S_E$  feature (see Fig. 6.b, c). Since these two types of features are not present in microstructures above  $M_s$ , they should correspond to martensitic structures, which are tempered during the isothermal holding of the heat treatment.

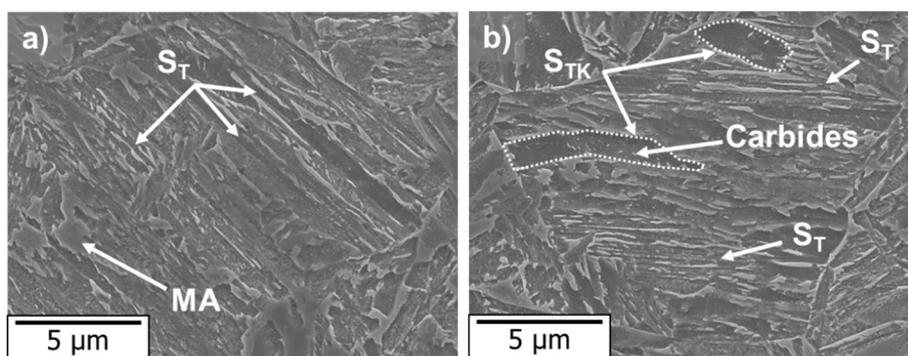


Fig. 5. SEM micrographs of two determined areas of the specimen isothermally treated at 320 °C (at  $M_s$ ), showing the presence of  $S_{Thin}$  and  $S_{Thick}$  features in a selected prior austenite grain.

Retained austenite is present in this microstructure in two different morphologies. Within  $S_T$  features, retained austenite generally appears as thin films. In  $S_E$  features, however, retained austenite appears in the form of elongated blocks along boundaries and/or as irregular blocks between ledge-like protrusions, as shown in Fig. 6.c. The KAM map shows high local misorientations at the boundaries of the  $S_E$  feature, as shown in Fig. 6.d (white arrows). Lower local misorientations are also observed within the  $S_E$  feature, which can correspond to low-angle boundaries of its internal structure.

### 3.3.3. At 300 °C (below $M_s$ )

At this temperature, a volume fraction of approximately 16% of prior athermal martensite is formed during the cooling below  $M_s$  until reaching the isothermal temperature. This prior athermal martensite will be tempered during the isothermal holding. The microstructure is thus a mixture of bainitic ferrite, prior athermal martensite, and MA islands (see Table 1).  $S_T$  features, identified as bainitic ferrite in previously described microstructures, are shown in Fig. 7. This indicates the formation of bainitic ferrite in isothermal treatments below  $M_s$  with similar morphological characteristics as the bainite formed in treatments above and at the  $M_s$  temperature.

Other types of product phases can be distinguished in this microstructure. Fig. 8 shows the SEM micrograph, Inverse Pole Figure (IPF), Phase Distribution (PD), and Kernel Average Misorientation (KAM) maps of a specific prior austenite grain of the specimen treated at 300 °C. An  $S_E$  feature with wavy boundaries can be observed at the top part of the grain (marked with a dashed blue line). This specific feature has also been observed in microstructures formed during isothermal treatments at 320 °C (at  $M_s$ ). As pointed out in the microstructural description of that heat treatment,  $S_E$  features should be characterized as martensite since this type of feature does not appear in treatments above  $M_s$  and its presence in the microstructures becomes more pronounced when the volume fraction of prior athermal martensite increases. Carbides are visible within these  $S_E$  features, and there seem to be multi-directional carbides. Bainitic structures are also characterized at this temperature in the form of  $S_T$  and  $S_I$  features (see Fig. 8.a, b).

Fig. 8.a also shows an interesting feature in the surroundings of  $S_E$ . This feature corresponds to very small acicular units ( $S_T$  features), identified as bainitic ferrite, just next to the wavy boundary of the lath. These  $S_T$  features are indicated by a dashed white or red oval in the SEM figure and EBSD maps. As observed in Fig. 8.a, the  $S_T$  features seem to grow from the  $S_E$  and appear aligned at both sides of the lath-shaped  $S_E$ . However, in the EBSD maps (Fig. 8.b, c, d), there is no clear indication of the existence of these features, apart from a small concentration of high local misorientations (see Fig. 8.d). According to the analysis, bainitic ferrite ( $S_T$ ) may be growing from the prior athermal martensite ( $S_E$ ), maintaining a similar crystallographic orientation relationship. This fact could contribute to the formation of the ledge-like protrusions which can give rise to a wavy appearance of the boundaries of the lath-

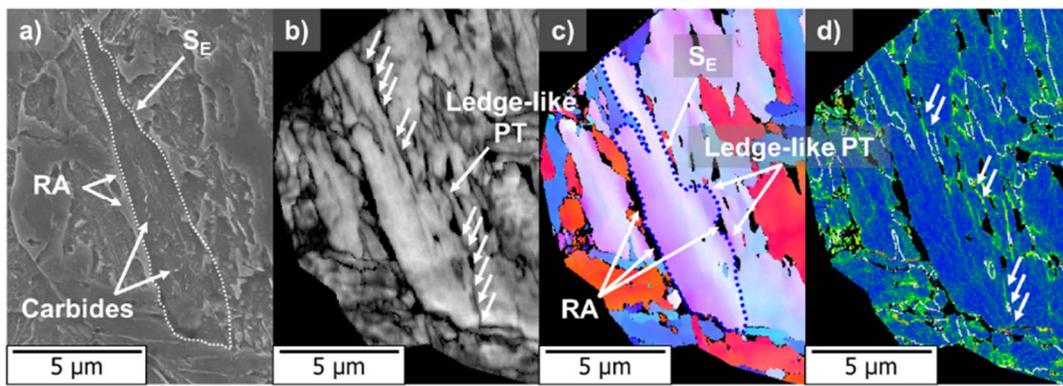


Fig. 6. a) SEM micrograph, b) IQ map, c) IPF map, and d) KAM map of a region within a prior austenite grain of the specimen isothermally treated at 320 °C.

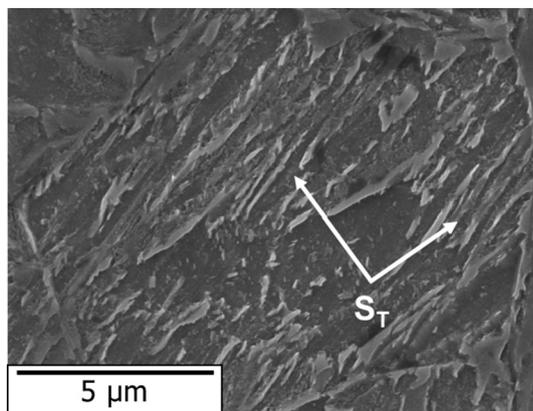


Fig. 7. Presence of  $S_{Tmin}$  microstructural features in the specimen isothermally treated at 300 °C (below  $M_s$ ).

shaped  $S_E$  features.

Retained austenite mainly appears in the form of small elongated blocks, as shown in Fig. 8.c. These blocks could form a continuous film at the boundaries between microstructural constituents, as marked by the dashed white line in Fig. 8.c. A high density of low-angle boundaries with misorientations lower than 5–7° can be observed in Fig. 8.d (marked with white arrows). These boundaries separate the different sub-units present within the same microstructural constituent. The described boundaries coincide with the discontinuous “films” of retained austenite.

### 3.3.4. At 270 °C (below $M_s$ )

At 270 °C, 50 °C below the experimental  $M_s$ -temperature, almost 80% of the microstructure is formed by prior athermal martensite,

which is tempered, with only 12% of bainitic ferrite formed during the isothermal holding (see Table 1) and 9% of retained austenite. The microstructure is mainly formed by lath-shaped structures of different sizes and shapes, containing carbides inside (see Fig. 9.a). In this case, no bainitic structures similar to the ones described to be formed above  $M_s$  are observed. Although the studied steel has a 1.5 wt. percent silicon content, carbide precipitation is not completely suppressed.

Fig. 9.a shows two types of microstructural features:  $S_{TK}$  and  $S_E$ . Wavy boundaries with ledge-like protrusions, a specific microstructural characteristic of  $S_E$ , can be observed in Fig. 9.b, c. Retained austenite is also present in this microstructure in the form of elongated blocks between  $S_E$  features (see Fig. 9.c). Fig. 10 shows another example of  $S_E$  features, which are characteristic for this microstructure. The  $S_E$  feature, indicated by a dashed white contour in Fig. 10.a, is extended along the grain from one boundary to the opposite one. This specific lath-shaped structure is characterized by wavy boundaries at its both sides, marked with dashed blue lines in Fig. 10.b. The KAM map shows high local misorientations at the wavy boundaries of the  $S_E$  feature (see Fig. 10.c).

According to volume fractions obtained and previous microstructural analysis at higher temperatures, both features  $S_{TK}$  and  $S_E$  are definitively characterized as prior athermal martensite, tempered during the isothermal holding. As described in the microstructural characterization at 300 °C, the existence of ledge-like protrusions may produce the characteristic wavy boundaries observed in  $S_E$ . Although there are carbides along the entire  $S_E$  feature, these protrusions do not present any carbides inside, as shown in Fig. 9.a. Based on the EBSD maps (Fig. 9.b, c), the protrusions exhibit a similar orientation as the rest of the  $S_E$  feature. As is pointed out for the microstructure formed at 300 °C, these protrusions can correspond to bainitic ferrite structures growing from the prior athermal martensite.

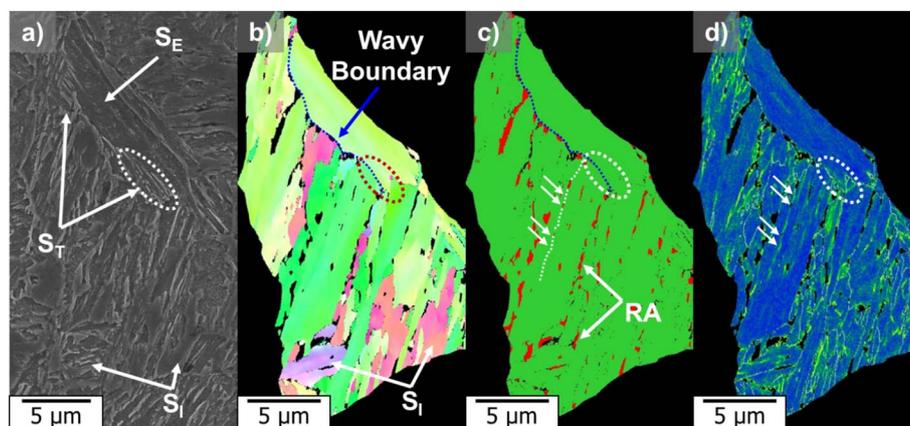


Fig. 8. a) SEM micrograph, b) IPF map, c) PD map, and d) KAM map of a selected prior austenite grain of the specimen isothermally treated at 300 °C.

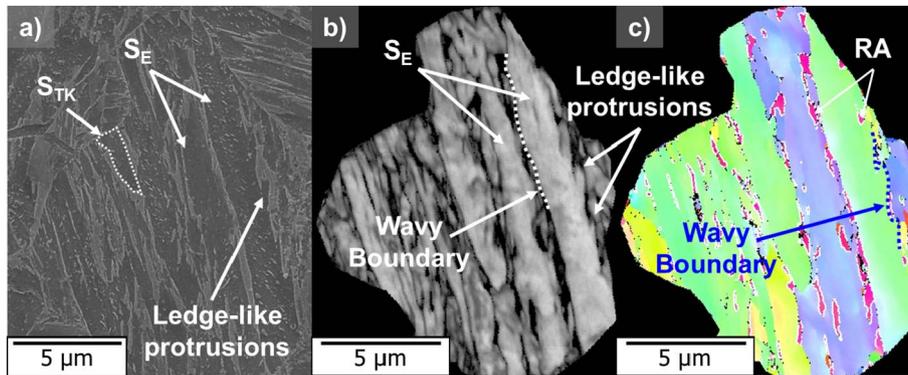


Fig. 9. a) SEM micrograph, b) IQ map, and c) IPF map of a selected prior austenite grain of the specimen isothermally treated at 270 °C. This specimen contains a volume fraction of prior athermal martensite of approximately 80%. This martensite is tempered during the isothermal holding.

### 3.4. Nature of product phases

Table 2 shows a summary of the main characteristics of the five microstructural features observed in the analysis of the four isothermal treatments performed at 340 °C (above  $M_s$ ), and 320 °C (at  $M_s$ ), 300 °C, and 270 °C (below  $M_s$ ), as well as at which temperature that specific feature was observed. The possible nature of the each microstructural feature is also specified.

Considering the thermal cycle followed in these heat treatments, the bainitic structures,  $S_T$  and  $S_b$ , were only formed during the isothermal holdings of one hour performed at temperatures of 300 °C and higher. However, the martensitic structures,  $S_{TK}$  and  $S_E$ , were previously formed during the cooling from austenitization to the selected isothermal temperature. MA islands were finally formed in the cooling to room temperature from the remaining austenite, which was not fully transformed during the isothermal holding.

As shown in Table 2,  $S_T$  and  $S_b$  are considered bainitic structures since they are the main features observed in the isothermal treatment performed above  $M_s$ , where there is no formation of athermal martensite prior to the isothermal holding. Both types of features are also observed in isothermal treatments below  $M_s$ . Besides,  $S_{TK}$  and  $S_E$  are considered martensitic structures. The reason for this conclusion is the majority presence of these two microstructural features after applying the isothermal treatment at 270 °C (below  $M_s$ ), where almost 80% of the volume corresponds to the prior athermal martensite formed during cooling until reaching the isothermal temperature. Furthermore, as the isothermal temperature below  $M_s$  is increased and approaches  $M_s$ , the presence of these microstructural features in the microstructures is reduced. Martensite-austenite islands, formed in the final cooling to room temperature, are present in the microstructures of all heat treatments.

The martensitic nature of  $S_E$  is partially in agreement with previous investigations done by Kim et al. [16] and Somani et al. [17]. Both

authors characterized an isothermal product with wavy boundaries and ledge-like protrusions in isothermal treatments performed below  $M_s$  in low-carbon steels with similar compositions as the one investigated in the present work. In both studies, the authors conclude that the nature of these laths with wavy boundaries is not purely bainitic since they have a stronger similarity with athermal martensite. However, there are some descriptive details which differ between each characterization. These characterization details are presented as follows:

1. Kim et al. [16] compare by TEM observations the product phases obtained in an isothermal treatment above  $M_s$ , below  $M_s$  (with 10% of prior athermal martensite), and in a direct quench. The authors suggest that the presence of wavy boundaries containing ledges in the mentioned laths formed below  $M_s$  is the result of the thickening of those laths during the isothermal transformation. Additionally, multi-variant carbides were also distinguished within those isothermal laths and the athermal martensite, but not in lower bainite. Kim et al. thus state that although there was a stronger similarity with athermal martensite than with lower bainite, the nature of those isothermal laths is neither purely martensitic nor purely bainitic.
2. Somani et al. [17] also compared by TEM observations the nature of the different products phases obtained during two-step quenching and partitioning (Q & P) treatments. Such isothermal laths with wavy boundaries and ledge-like protrusions were also characterized. In this case, these laths are considered the result of the isothermal growth, through the migration of ledges, during the partitioning step of the prior athermal martensite laths already formed. These laths were thus identified as martensite.

Although both identifications are feasible, the thermal conditions of the heat treatments analysed in each investigation are different between them and with respect to the present work. In the investigation

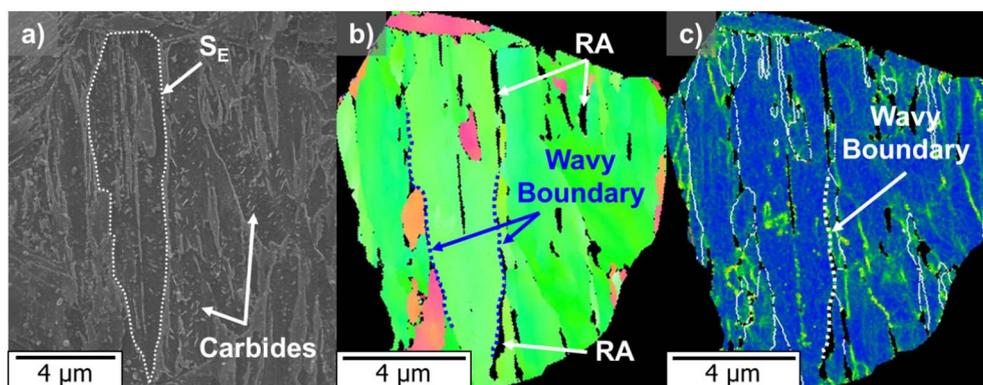


Fig. 10. a) SEM micrograph, b) IPF map, and c) KAM map of a selected prior austenite grain of the specimen isothermally treated at 270 °C.

**Table 2**

Summary of the microstructural features observed in isothermal treatments at 340, 320, 300, and 270 °C. The nature and characteristics of each feature are also detailed.

Structural feature	Characteristics	340 °C	320 °C	300 °C	270 °C	Phase nature
$S_{Thin}$ ( $S_T$ )	<ul style="list-style-type: none"> <li>- Above and below <math>M_s</math>.</li> <li>- Thin units with acicular shape.</li> <li>- Parallel aligned.</li> <li>- Length: 5–8 <math>\mu\text{m}</math>.</li> <li>- No carbides distinguished.</li> </ul>	✓	✓	✓	?	Bainitic
$S_{Irregular}$ ( $S_I$ )	<ul style="list-style-type: none"> <li>- Above and below <math>M_s</math>.</li> <li>- Laths with irregular shape.</li> <li>- Wider than <math>S_T</math>.</li> <li>- No carbides distinguished.</li> </ul>	✓	✓	✓	–	Bainitic
$S_{Thick}$ ( $S_{TK}$ )	<ul style="list-style-type: none"> <li>- Below <math>M_s</math>.</li> <li>- Laths with sharp tip at an edge.</li> <li>- Surrounded by <math>S_T</math>.</li> <li>- Contain carbides.</li> </ul>	–	✓	✓	✓	Martensitic
$S_{Elongated}$ ( $S_E$ )	<ul style="list-style-type: none"> <li>- Below <math>M_s</math>.</li> <li>- Longer and wider than <math>S_T</math>.</li> <li>- Wavy boundaries.</li> <li>- Contain carbides.</li> </ul>	–	✓	✓	✓	Martensitic
$S_{Blocky}$ ( $S_B$ ) (MA island)	<ul style="list-style-type: none"> <li>- Smooth surface (not etched).</li> <li>- Internal structure.</li> <li>- Very small laths.</li> <li>- Surrounded by RA blocks.</li> </ul>	✓	✓	✓	✓	Martensite/austenite

of Kim et al., there is no microstructural characterization of existing tempered martensite formed in the treatment below  $M_s$  (with 10% volume fraction), so those  $S_E$  features could also be considered as tempered martensite. In the investigation of Somani et al., the nature of  $S_E$  is certainly considered martensitic.

However, there is another possible explanation which has not been considered yet. The formation of bainitic ferrite below  $M_s$  could be the reason of the appearance of ledge-like protrusions in the  $S_E$  features, which lead to the formation of wavy boundaries. Several studies have shown that the formation of athermal martensite prior to isothermal treatments leads to the creation of new potential nucleation sites in the form of martensite-austenite interfaces [8–11]. At the same time, as the isothermal temperature decreases and approaches  $M_s$ , there is a reduction in the size of the bainitic units due to an increase of the transformation kinetics. The lower the temperature, the higher the undercooling and driving force for nucleation. So, very small units of bainitic ferrite may grow from the initial martensitic laths ( $S_E$  features) in the form of ledge-like protrusions at the very beginning of the isothermal treatments below  $M_s$ .

This fact can be observed in the microstructure obtained after the isothermal treatment at 300 °C (below  $M_s$ ), shown in Fig. 8.a. In that figure, small units of a few micrometres length (less than 5  $\mu\text{m}$ ) appear just next to the wavy boundary of an  $S_E$  feature. The morphology (shape and size) of these units strongly differs from the one of the  $S_E$ , so they can be considered as different product phases. Their orientation may be the same (or similar) to that of the main  $S_E$  feature (see Fig. 8.b). Only a small concentration of local deformation around the area where those small units appear (see Fig. 8.d) indicates their possible existence. Although these observations can suggest the different nature of these ledge-like protrusions in treatments below  $M_s$ , further investigations are needed for a detailed characterization of these protrusions and a better understanding of the interactions between prior athermal martensite and the subsequently formed bainitic ferrite.

#### 4. Conclusions

The nature of the different microstructural features obtained in isothermal treatments below  $M_s$  is determined as bainitic or martensitic by comparing experimental volume fractions of the product phases,

which were extracted from previous dilatometry and XRD analysis, with the morphology (size and shape) of the specific features characterized at each isothermal temperature by SEM and EBSD analysis. The main conclusions obtained from this methodology are the following:

1. Bainitic ferrite appears in the form of thin acicular units and/or irregularly shaped laths in isothermal treatments performed above and below the  $M_s$ -temperature. Acicular units are aligned parallel to each other. They have a mean length between 5 and 8  $\mu\text{m}$  and a mean width between 0.6 and 0.8  $\mu\text{m}$ . Irregular laths are wider than acicular units. Carbides are not observed within any of these features.
2. Martensite appears as lath-shaped structures in isothermal treatments performed below the  $M_s$ -temperature. Two different martensitic structures are characterized in the form of laths with a characteristic sharp tip at one of the edges and/or laths presenting wavy boundaries with ledge-like protrusions. The latter martensitic structures are longer and wider than acicular units of bainitic ferrite, with a mean length and width of 12  $\mu\text{m}$  and 2  $\mu\text{m}$ , respectively. Both types of lath-shaped structures contain carbides within them. Based on the characterization methodology described above, these microstructures correspond well to tempered martensite.
3. Martensite-austenite islands are characterized in all isothermal treatments performed above and below  $M_s$ . These islands present an internal structure formed by very small martensite laths (smaller than 2  $\mu\text{m}$ ) parallel aligned, surrounded by retained austenite blocks.
4. The appearance of some protrusions in martensitic structures can be related to the formation of bainitic ferrite in isothermal treatments below  $M_s$ , which might lead to the formation of wavy boundaries in such martensitic laths.

#### Acknowledgements

The authors would like to thank Prof. dr. ir. Roumen Petrov for his advice and help on the EBSD analysis. The authors gratefully acknowledge the financial support of the Netherlands Organization for Scientific Research (NWO) and the Dutch Foundation for Applied Sciences (STW)

through the VIDI-Grant 12376.

## References

- [1] K. Sugimoto, T. Iida, J. Sakaguchi, T. Kashima, Retained austenite characteristics and tensile properties in a TRIP type bainitic sheet steel, *ISIJ Int.* 40 (2000) 902–908.
- [2] K. Sugimoto, M. Murata, S.M. Song, Formability of Al-Nb bearing ultra-high strength TRIP-aided sheet steels with bainitic ferrite and/or martensite matrix, *ISIJ Int.* 50 (2010) 162–168.
- [3] Y. Jiang, R. Zhou, R. Zhou, D. Lu, Z. Li, Microstructures and properties of a bainite and martensite dual-phase cast steel fabricated by combination of alloying and controlled cooling heat treatment. *Trans. Tech. Publications, Mater. Sci. Forum* 475–479 (2005) 93–96.
- [4] S.M.C. Van Bohemen, M.J. Santofimia, J. Sietsma, Experimental evidence for bainite formation below  $M_s$  in Fe-0.66C, *Scr. Mater.* 58 (2008) 488–491.
- [5] J.C. Hell, M. Dehmas, S. Allain, J.M. Prado, A. Hazotte, J.P. Chateau, Microstructure-properties relationships in carbide-free bainitic steels, *ISIJ Int.* 51 (2011) 1724–1732.
- [6] I.A. Yakubtsov, G.R. Purdy, Analyses of transformation kinetics of carbide-free bainite above and below the athermal martensite start temperature, *Metall. Mater. Trans. A* 43A (2012) 437–446.
- [7] D. Kim, J.G. Speer, B.C. De Cooman, The isothermal transformation of low-alloy low-carbon CMnSi steels below  $M_s$ , *Trans. Tech. Publications, Mater. Sci. Forum* 654–656 (2010) 98–101.
- [8] H. Kawata, K. Hayashi, N. Sugiura, N. Yoshinaga, M. Takahashi, Effect of martensite in initial structure on bainite transformation, *Mater Sci Forum* 638–642 (2010) 3307–3312.
- [9] M.J. Santofimia, S.M.C. van Bohemen, D.N. Hanlon, L. Zhao, J. Sietsma, Perspectives in high-strength steels: Interactions between non-equilibrium phases, *Inter. Symp. on AHSS, AIST*, 2013, pp. 331–339.
- [10] L. Zhao, L. Qian, J. Meng, Q. Zhou, F. Zhang, Below- $M_s$  austempering to obtain refined bainitic structure and enhanced mechanical properties in low-C high-Si/Al steels, *Scr. Mater.* 112 (2016) 96–100.
- [11] A. Navarro-López, J. Sietsma, M.J. Santofimia, Effect of prior athermal martensite on the isothermal transformation kinetics below  $M_s$  in a low-C high-Si steel, *Metall. Mater. Trans. A* 47A (2016) 1028–1039.
- [12] P. Kolmskog, A. Borgenstam, M. Hillert, P. Hedstrom, S.S. Babu, H. Terasaki, Y.I. Komizo, Direct observation that bainite can grow below  $M_s$ , *Metall Mater Trans A* 43A (2012) 4984–4988.
- [13] E.P. Da Silva, D. De Knijff, W. Xu, C. Föjer, Y. Houbaert, J. Sietsma, R. Petrov, Isothermal transformations in advanced high strength steels below martensite start temperature, *Mater Sci Technol* 31 (2015) 808–816.
- [14] S. Samanta, P. Biswas, S. Giri, S.B. Singh, S. Kundu, Formation of bainite below the  $M_s$  temperature: kinetics and crystallography, *Acta Mater* 105 (2016) 390–403.
- [15] D. Kim, J.G. Speer, B.C. De Cooman, Isothermal transformation of a CMnSi steel below the  $M_s$  temperature, *Metall. Mater. Trans. A* 42A (2011) 1575–1585.
- [16] D. Kim, S.J. Lee, B.C. De Cooman, Microstructure of low carbon steel isothermally transformed in the  $M_s$  to  $M_f$  temperature range, *Metall. Mater. Trans. A* 43A (2012) 4967–4983.
- [17] M.C. Somani, D.A. Porter, L.P. Karjalainen, R.D.K. Misra, On various aspects of decomposition of austenite in a high-Si steel during Q & P, *Metall. Mater. Trans. A* 45A (2014) 1247–1257.
- [18] S.R. Pati, M. Cohen, Nucleation of the isothermal martensitic transformation, *Acta Metall* 17 (1969) 189–199.
- [19] A. Borgenstam, M. Hillert, J. Agren, Critical temperature for growth of martensite, *Acta Metall Mater* 43 (1995) 945–954.
- [20] A. Borgenstam, M. Hillert, Activation energy for isothermal martensite in ferrous alloys, *Acta Mater* 45 (1997) 651–662.
- [21] C.F. Jateczak, J.A. Larson, S.W. Shin, Retained austenite and its measurements by X-ray diffraction, 453 Society of Automotive Engineers, 1980Special Publication.