Experimental Study on Electromigration by Using Blech Structure

by

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Abstract

In the advanced semiconductor industry, modern electronic devices are expected to realize complex functions with minimized size, which requires an increase in the density of on-chip interconnects. To meet this demand, the dimension of interconnects is reduced and it requires the narrowing of metal interconnects to carry the increasing current density. With such a developing trend, electromigration is one of the significant reliability challenges for electronic devices. Although lots of works focus on the formulation and simulation for electromigration, they are not complete and consistent with experimental results. Recently, A fully-coupled and self-consistent electromigration theory was developed by Cui et al, but new and extensive experimental data and analysis are needed to further validate their theoretical results.

This research work focuses on the experimental study on electromigration (EM), and the aim is to investigate the different effects on EM. The Blech structure was proposed as our experimental structure. The fabrication of Blech structure is conducted at the EKL cleanroom, and main structures with various dimensions are fabricated by employing sputtering technique for deposition and lithography for defining patterns. The measurement is carried out by accelerated tests with high current stress and at elevated temperature. Experimental results are characterized and analyzed by different tools, such as Keyence 3D laser profilometer and scanning electron microscope

With the current density of $1 \times 10^{10} A/m^2$ and temperature of 250 °C, the result shows that the critical length under certain conditions is 10 µm, and longer stripes have larger drift lengths and a shorter time to form voids in electromigration. Furthermore, it is found that with elevated temperature, the drift length increases and the electromigration lifetime decreases. In addition, the covered SiN passivation layer only for the annealed Al stripes suppresses electromigration and this is because the annealing process improves the coalescence of grain in Al film, reducing the defects at the grain boundary and finally forming a denser microstructure. The influence of the atmosphere on electromigration indicates that the additional oxide on Al interconnects increases that actual current stress and results in a short electromigration lifetime. In general, present experimental results were consistent with existing results in the literature, but several problems are still unsolved, which will be part of our future work.

Keywords: electromigration, aluminum, Blech structure, reliability, interconnects.

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Chapter 1. Introduction

1.1 What's electromigration

Electromigration (EM) is an enhanced mass transport process caused by the momentum transfer between electrons and metal atoms when the conductor is stressed with a high electrical current density. This significant mass transport can cause the depletion of atoms at the cathode side to form voids, and the accumulation of atoms at the anode side to form hillocks, as shown in Figure 1.1. Finally, those voids and hillocks can lead to the formation of open connections or short circuits of interconnects.



Figure 1.1. SEM graph of electromigration in the Al interconnect; (a) voids induced by electromigration at the cathode side; (b) hillocks induced by electromigration at the anode side.

According to the International Technology Roadmap for Semiconductors (ITRS) 2014 Update [1-2], the density of on-chip interconnects increases from generation to generation of modern integrated circuits. This requires the narrowing of metal interconnects which will in turn have to carry an increasing current density, as shown in Figure 1.2. This causes electromigration to become one of the significant reliability challenges for electrical devices. To make matters worse, the decrease in cross-sectional area of the interconnects also means a reduction in allowed current densities for required interconnect reliability. Furthermore, the total length of interconnects per IC will continue to increase. As a consequence, reliability requirements per length unit of the wires need to increase to maintain overall IC reliability. This development trend is contradicted by the future decrease in interconnect reliability due to electromigration. Thus, The ITRS states that not known solutions are available for the EM-related reliability requirements that we will face in the near future (Figure 1.2, red barrier). The limitation of electromigration on interconnects can no longer be overlooked.



Figure 1.2. Evolution of required and maximum current densities in IC interconnect [1-2].

1.2 History of EM research

Before electromigration (EM) became one of the main reliability problems in the modern integrated circuit industry, it has been for a long time only of academic importance. As early as 1861, Gerardin observed the electromigration phenomenon [3]. Then, in 1959 Fiks first published his studies on electromigration [4]. Later that year, Huntington and his group [5] found that electromigration was a process of mass transport, caused by the momentum transformation between moving electrons and metal ions, and they proposed that the electron wind is the driving force for electromigration.

In later studies, people generally realized that electromigration is a complex multiphysics process. In 2003, Arnaud [6] *et al.* found that a large temperature gradient existed in the void-nucleation area. By experimental research, Nguyen *et al.* observed that various temperature gradients of metal interconnects caused the difference of media time to failure [7]. According to previous studies, Cher [8] and his co-workers analyzed the effects of electronic wind, stress gradient and temperature gradient on a system of interconnects, and concluded that the effect of temperature gradient on the formation of voids is as important as the stress gradient. Besides, with the continuous development of experimental technology, people start to investigate the electromigration in multiscale, such as micro-level and nano-level interconnects. Previously, it was thought that only the dimensions and geometries of interconnects were influencing factors. However, the microstructure of metal materials also plays an important role in the effect of electromigration lifetime, which will be discussed in our work.

With the ongoing shrinking of the interconnection line-width of electronic devices, people have found that using pure aluminum is not able to effectively solve this problem. The Al interconnects containing 4% Cu were proved to have a better EM lifetime, approximate 70 times higher than the life-time of pure Al [9]. The longest lifetime of Al alloys was found with 12% Cu [10]. In addition to copper, magnesium, and chromium alloyed with Al have been found to have a better electromigration resistance.

To meet the demand of complex multilayer structures and in order to reduce feature

sizes in integrated circuits, Al-based alloys have been replaced by lower electrical resistivity materials. Copper has come to the attention of researchers, because of its excellent electrical conductivity, which can improve the circuit speed, and provide more reliability in ICs due to higher melting temperature and better current carrying capability. The first company to apply copper as the interconnect material in their products was IBM in 1997. After that, the Cu-based interconnects were widely used by companies to improve chip density and performance. Due to different material properties, the processing of aluminum- and copper-based interconnections are significantly different. The damascene process technology is used for the fabrication of copper-based interconnections[11-13]. And the key steps in this process are electroplating technique and chemical mechanical polishing technique. Since the electroplated copper has a larger grain size, resulting in bamboo or near bamboo microstructure, the diffusion path along grain boundaries for atoms transport is reduced. Thus, the surface or interface is the dominant diffusing path, improving the interface properties of copper became an important way to suppress electromigration.

In addition, EM modeling also is important. V. Sukharev and his group [14-16] simulated the electromigration process of multilayer interconnects and proposed a physical model based on electromigration failure, which can be used to predict the location of void nucleation. Later that year, the EM model was optimized by Cher [17] *et al.* In this model, the vertical diffusion of Joule heat in the interconnect was considered so that it can be used to more accurately predict the location of void formation in narrow interconnects. However, the optimization of EM simulation models never stops. F. Cacho [18] and others found that when considering the effect of the atomic concentration gradient, the predicted EM lifetime corresponds to the real results. In later years, the mathematical physical model of electromigration considering the interaction of electronic wind, stress gradient, temperature gradient, and atomic concentration gradient was proposed [19-20].

1.3 Goal and objectives

In the past few years, there have been many works focusing on the formulation and simulation of electromigration. Although amazing development has been made on the fundamental understanding of electromigration, these models are not able to be consistent with experimental results. Recently, Zhen Cui *et al.* [21] developed a fully-coupled and self-consistent electromigration theory, which includes the effects of the gradients of concentration, stress, temperature, and the electron wind force. In his work, 1-D numerical solutions of a metal line under totally confined configuration and stress-free configuration are provided. Although the predictions are consistent with the original test data presented by Blech, new and extensive experimental data and analysis are needed to further validate their theoretical predictions.

The main goal of the present work is to experimentally study the EM and provide experimental data to validate our theoretical model. The objectives of the present work are summarized as follows,

- 1. Build up a platform to conduct EM testing at the micro-level.
- 2. Investigate the effects of metal length on EM.
- 3. Investigate the effects of the passivation layer on EM.
- 4. Investigate the effects of testing temperature on EM.

1.4 Outline

In this thesis, the experimental study on electromigration by using Blech's structure is introduced, and the thesis consists of six chapters, a brief introduction of every chapter is given as follows:

Chapter 2 is a review of electromigration study in micro-scale interconnection, including the experimental studies on Al-based interconnects and the basic theory of EM.

In Chapter 3, a brief review of the testing structure used for EM is given. Based on Blech's structure, the structure design in the present study is introduced. The fabrication

process is illustrated in detail.

Chapter 4 gives an introduction of measurement equipment and characterization methods used in our work.

In Chapter 5, EM results with different variations of the structure are obtained. The discussion, combining theoretical analysis and experimental results, are given.

After discussion experimental results, a conclusion will be drawn in Chapter 6, and a recommendation for future work is given.

Chapter 2. Review of EM in Micron-Scale Interconnect

This chapter introduces the development of electromigration in pure metal interconnects. In Section 2.1, a review of the theoretical studies on EM is given. The experimental studies on EM are given in Section 2.2. In this section, the different impacts on EM in the micro-level will be discussed and it includes the Al-based metallization and Cu-based interconnects.

2.1 Theory of electromigration

Electromigration is a process of mass transport in the current-carrying metal wire under the driving forces generated by an electric field. On a single metal ion, the net electromigration force (F_{net}) can be divided into two opposing microscopic forces. These forces are called, direct force (F_{direct}) and 'electron wind' force (F_{wind}) . The direct force, also called electric field force, is an electrostatic force on a positive metal ion in the opposite direction to the electron flow under the influence of an applied electric field. The idea of the 'electron wind' was first proposed by F. Skaupy in 1914 [21]. The electrons, driven by the external field, in conductors are scattered by point defects, and the resulting momentum transfer per unit time from scattering electrons to the impurities leads to the so-called 'wind force'.

The total electromigration driving force on a metal ion can be written as [22-25],

$$F_{net} = F_{direct} + F_{wind} = (Z_{direct} + Z_{wind})eE = Z^*eE$$
(2.2)

where $Z^* = Z_{direct} + Z_{wind}$ is referred to as the effective valence (or effective charge number) and represents a parameter that comprises the quantum-mechanical effects of the electromigration phenomenon, Z_{wind} is related to the magnitude and the direction of the momentum exchange between conducting electrons and point defects, Z_{direct} is the direct valence related to the nominal valence of the metallic ion when shielding processes are absent, e is the elementary charge, E is the applied electric field. The sign of the effective valence (Z^*) determines the nature of the transport mechanism. The negative value in the equation means the transport direction is consistent with the electron flow. The following Figure 2.1 shows the two opposing contributions of the electromigration driving force.



Figure 2.1. Two opposing microscopic forces on a metal ion [25].

Finally, the atomic flux induced by the electrical field can be expressed as following equation [25],

$$J = \frac{DCZ^* eE}{k_B T} = \frac{DCZ^* e\rho j}{k_B T}$$
(2.2)

where k_B is the Boltzmann's constant, T is the temperature, C is the atomic concentration, ρ is the resistivity, j is the current density, and D is the diffusion coefficient related to temperature as expressed in following Arrhenius equation,

$$D = D_0 \exp\left(\frac{-E_a}{k_B T}\right) \tag{2.3}$$

where D_0 is the pre-exponential factor and E_a is the activation energy.

If a metal atom moves to the end of the metal line, under the impact of electromigration, a vacancy will be left at the other end of the line [26]. Because of the different volumes between atoms and vacancies, in the lattice site, volumetric relaxations will occur [27], which results in a volume contraction caused by vacancy accumulation at the cathode, on the other side, vacancy depletion produces a volume expansion [28]. Thus, there is mechanical stress gradient along the interconnect length. The stress caused by this gradient is so-called EM induced non-uniform stress [29-31], also known as backflow stress. The direction of this atomic backflow is opposite to the atomic flux induced by electromigration [30]. Therefore, under certain conditions, the opposite flow equals the atomic flux due to electromigration, the damage will stop.

When the metal line is covered by the passivation layer or capping layer, there will be mechanical constraints on the interconnect metal, which further results in the evolution of mechanical stress. It was shown that different passivation materials have different impacts on EM, it depends on the type of stress caused by passivation layers, for example, the interfacial compressive stress will decrease the diffusion rate and suppress the EM [32].

However, the backflow stress is not the only source of stress gradient, another distribution is due to the difference in thermal expansion between the metal material and surrounding materials [33-36].

Here, atomic flux induced by stress gradient can be expressed by the following equation [8],

$$\boldsymbol{J} = \Omega \frac{DC}{k_B T} \frac{\partial \sigma}{\partial x}$$
(2.4)

where Ω is atomic volume, σ is the hydrostatic stress.

In addition, the driving force induced by the temperature gradient is also an important factor that needed to be considered in EM. One of the sources of the temperature gradient is Joule heating in conductors. Joule heating is defined by the process that the current passes through the electrical conductor and produces thermal energy. Another source of the temperature gradient is the geometry of interconnects. For example, line-via structures, the via and the testing line play different roles and have different temperature. Lloyd [37] and Schwarzenberger *et al.* [38] confirmed the correlation between temperature gradient, EM failure. The result indicated that at the point with high temperature gradient, EM failure was typically observed, on the other side, at the site with uniform temperature, the EM failure was randomly found. During the accelerated EM test, it is quite easy to produce the temperature gradient by Joule heating in interconnects.

The atomic flux from temperature gradient induced driving force can be expressed as [8],

$$J = -\frac{DCQ^*}{k_B T^2} \frac{\partial T}{\partial x}$$
(2.5)

where Q^* is the coefficient of heat transfer (eV).

Generally, from the above discussion, three different driving forces on EM are introduced, including the electron wind force, stress gradient, and temperature gradient.

2.2 Experimental studies

To study the electromigration failure behavior, the usage of accelerated tests is necessary, in which failure times are commonly less than 1000 hours. As indicated by Black's equation [39], the lifetime of an interconnect can be reduced by either increasing the current density or by increasing the temperature. Usually, the EM tests are carried out using simple structures. The commonly used structure includes the Blech's structure [29,30] the NIST (National Institute of Standards and Technology) structure [40-42], the SWEAT (Standard Wafer-level Electromigration Acceleration Test) structure [43], and the line-via test structure. The details of those structures will be introduced in Sec. X. In the past decades, many experimental works demonstrated that EM is a multi-physics problem that is controlled by various factors, including mechanical stress, temperature, chemical potential.

2.1.1 The impact of mechanical stress

Blech was one of the first to explain the origin of the stress gradient during electromigration [29,30], and such a force is commonly called "Back Stress". In his experiments [44], Al stripes were patterned on a TiN film, as shown in Figure 2.2. Since Al has better conductivity than TiN, the load current mostly flows through the Al stripes.



Figure 2.2. Blech testing structure

By experimental results, he indicated that the average drift velocity has a linear

correlation with the current density and found that the product of the strip length and critical current density is constant which is called the threshold-length product. Here, the critical current density means when the loading current density is below this value, the drift cannot be detected. In order to figure out the influence factor of threshold current density, he proposed the geometry of a series of Al lines with different lengths (from 10 μ m to 100 μ m) deposited on the TiN layer, as shown in Figure 2.3. After loading $3.7 \times 10^5 A/cm^2$ for 15h at 350 °C, the result showed that the threshold current density has a negative relationship with strip length (for the length range from 30 μ m to 150 μ m). And under the same current density, the longer stripes show lager voids at cathode. Then he carried out the measurement at a temperature between 200 °C to 550 °C and found that the threshold length product depends on temperature. In addition, he also investigated the effect of covering layer on drift velocity of EM, and from Figure 2.4, it showed that the entirely enclosed covering layer on Al stripes can suppress electromigration.



Figure 2.3. SEM image of Al stripes with varying lengths after the passage of $3.7 \times 105 \ A/cm^2$ for 15 h.



Figure 2.4. Average drift velocity of covered and uncovered samples

Generally, Blech's contribution to EM studies can be summarized by three concepts: the "Blech Threshold", the "Blech Length", and the "Blech Condition".

Blech condition: To explain this phenomenon, Blech suggested that the electromigration can be fully stopped when the driving force induced by the difference of mechanical stress at two sides of strip exactly equals to the electron wind force, it shows a balanced state, so-called Blech condition,

$$Z^* e\rho j + \Omega \frac{\Delta\sigma}{\Delta x} = 0 \tag{2.6}$$

Where Ω is the atomic volume, $\Delta \sigma / \Delta x$ is equal to $\Delta \sigma_{xx}$ which is the stress difference, eZ^* is defined as the effective change, and j is the loading current density.

Blech threshold: Integrating the above equation over the length of the metal line and assuming the critical failure condition is the maximum stress that the metal line can withstand, a critical product of metal line length and current density can be obtained as follows,

$$(jL)_c = \frac{(\sigma_{max} - \sigma_{min})\Omega}{Z^* e\rho}$$
(2.6)

below which the electromigration flux can entirely be balanced by the stress-induced counter flux of atoms at the steady-state condition. This critical product is called the Blech product. **Blech length:** For a certain current density, there exists a critical length (L_c) for an Al stripe, below which, the mass transport of electromigration can totally be suppressed. Currently, adding a barrier to divide the entire interconnect into several short segments below "Blech length" is still used as a design rule for interconnects.

Later that year, in order to figure out the reason behind the drift flow behavior, Blech designed a new structure with a large pad covered with Si₃N₄ [45]. After the passage of $4.5 \times 10^5 A/cm^2$ for 19 hours at 350 °C, the aluminum atoms drift into large contact pads, as shown in Figure 2.4 (left side). After placing this stripe at 350 °C without loading current for 762 hours, it was found that the aluminum atoms moved back, as shown in Figure 2.5 (right side). This phenomenon can be explained by the difference in free energy. During the current passage, the large pad serves as a reservoir and elastically stores parts of the aluminum atoms, which causes the free energy difference between the pad and the metal line. When the load is stopped, the uniform stress creates the backflow and the edge of the aluminum film slowly moves to the uncovered metal line.



Figure 2.5. Blech's backflow experiments.

Except for the "back stress", the thermal stress, resulting from the difference in thermal expansion between the passivation and metal upon cooling from high deposition temperatures, is the other source of mechanical stress. Numerous experimental observations have shown [46] that tensile stress induced by thermal mismatch increases the possibility of failure. The increased thickness and rigidity of the passivation layer prevent the relaxation of thermal stress, which results in dielectric cracking and metal extrusion.

2.1.2 The impact of microstructure

With continuous research on EM, people found that various crystal lattice structures of metallization have different effects on the electromigration failure mechanism. The most common microstructures in interconnects are polycrystalline, near-bamboo, and bamboo, as shown in Figure 2.6 [47].



Figure 2.6. Three common microstructures in Al samples

From the above picture, the feature of the polycrystalline microstructure is mainly based on its grain boundaries and the crystal lattice is dominant in Bamboo structures. Generally, the crystal lattice structure of interconnects depends on the grain size or its linewidth, for example, 'bamboo' microstructure is defined as the case that the linewidth of interconnects is less than its grain size.

Numerous studies indicated that there is a certain relationship between grain size and EM lifetime, for example, J. Cho and C. V. Thompson observed that the median time to failure can be affected by the ratio of grain size and linewidth [48]. The bamboo structure of metal interconnects provides a better improvement of EM reliability, which is because in the structure there are fewer grain boundaries and no continuous diffusion paths that are used for atoms transport exist in the direction of the current flow. As a result, the vacancy nucleation time is prolonged and the EM lifetime is increased. The related literature analyzed the influence of the diffusion path on the reliability of EM by studying different microstructures of interconnect, which further verifies the previous conclusion [49].

2.1.2 The impact of the metal interface

The main diffusion paths in metallic interconnect include bulk diffusion, grain boundary diffusion, and interface diffusion. As shown in Table 2.1, the different diffusion paths show different activation energies for atomic diffusion in Cu interconnects.

Diffusion Path	Ea (eV)
Surface	0.8
Grain-boundry	1.2
Bulk	2.3

Table 2.1. The activation energy of different diffusivity paths in Cu interconnects [50].

From Table 2.1, the diffusion path along the Cu surface has the lowest activation energy and the activation energy for diffusion in the bulk is the highest. The interface between copper and the capping layer is known to be the dominant diffusion pathway in damascene interconnects [51]. Hence, improving interface characteristics is an effective way to suppress EM in interconnects. This was demonstrated in many experiments that narrow copper stripes with capping layer and passivation have longer EM lifetime [5]. In general, there are two methods to achieve it, using chemical reaction [52-54] or physical treatment [55] to remove Cu oxides, and adding different capping layers on the copper surface.

Surface treatments can be divided into two categories, physical removal, and chemical reaction. Normally, Ar sputtering process is used for physical cleaning and oxidation-reduction reaction under the plasma process is the principal of chemical treatment.

Adding a capping layer, such as SiN and SiC, etc., on the top of copper has a

positive effect on the adhesion properties of the interconnect surface and they directly influence the rate of EM. It could be simply explained that the transport of atoms along a tightly bonded interface is more difficult and the harder the transport is, the slower diffusion will be. The experimental result from M.W. Lane and his group [56] verified that interfacial diffusion can be suppressed by adhesion and reactive bonds at the interface. Therefore, a series of experiments about the capping layer on copper wires have been carried out.

Hu, C-K, and his coworkers investigated the effect of copper lines covered with Ta/TaN, SiNx, and SiCxNyHz layers [57]. The result showed that a Ta/TaN metal capping layer has a better influence on lifetime and the activation energy increases from 0.87 eV for the bare metal interconnect to 1.4 eV.

Figure 2.7 shows the resistance change of samples with different capping layers such as Pd, CoWP, and CoSnP, and uncoated samples over time [58]. For bare metal lines, the resistance sharply increases in less than 200 hours but increases significantly slower for covered samples; after 2200 hours the resistance only increased by 10%. It is clear that the capping layer significantly prolongs the electromigration lifetime. From Energy-Dispersive X-ray Spectroscopy (EDS) figures, it shows the signal of Cu enters the CoWP layer, in other words, a small amount of Cu atoms is combined with CoWP. It can be considered that the CoWP layer effectively suppresses the diffusion of electromigration along with the interface and improves the electromigration life.



Figure 2.7. The resistance change of samples with or without a passivation layer

Finally, Yan, M.Y. *et al.* [59] studied the effect of Cu surface coated with 20 nm Cu₃Sn metal layer. The result showed that the metal coating layer can effectively suppress the diffusion of Cu atoms along the surface, and increase the electromigration lifetime by an order of magnitude.

2.1.3 Current crowding effect

Normally, when atoms are in a region with higher current density, the atomic transport speed will be faster which results in the acceleration of void formation. Corners are typically a current crowding zone, especially corners with a right angle, where current density can be extremely high.

The first FEM simulation on the current crowding effect with 2D structures was carried out by Kwok and his coworkers [60]. In his study, the current and temperature distribution was given in a multi-level interconnect. In a stud structure, he found that the aspect ratio of interconnects was related to current crowding, the narrower width and the higher aspect ratio result in the increasing of current density. And in a via structure, the current crowding is found to increase with an increasing step angle due to the thinning down of the step cross-section, as shown in Figure 2.8, a cross-section of a via structure with a step angle of 60°.



Figure 2.8. Cross-section view of a via structure

With the demand of complex structures in ICs, the simulation for 2D geometries are not able to meet the requirement, a 3D FEM simulation on current crowding was given by Weide *et al.* [61] In his work, the effect of different via shapes, such as the circular, square and oval like via, on the peak current density was discussed. The results indicated that the circular like via had the largest current density and the oval-like via had the least current density. In addition, the experiment results showed that in a traditional unfilled via, the electromigration lifetime depends on the diameter of the via hole [62]. However, in a tungsten-filled via structure, the diameter no longer plays a role and the reason is not clear. These two structures are shown in the following Figure 2.9.



Figure 2.9. Left: The structure of traditional non-filled via. Right: The structure of tungstenfilled via [62].

Actually, in a current crowding area, since the current and temperature are coupled in the interconnect, the temperature distribution is also inconsistent, which induces the thermo-mechanical stress in corners. Hence, there are several effects in the interconnect in the current crowding region.

2.3 Failure analysis of electromigration

The failure analysis typically is used for the sample after the accelerated EM test. Various analytical methods can be used to realize failure analysis, such as the optical microscope, scanning electron microscope (SEM), Focused ion beam (FIB-SEM), transmission electron microscope (TEM), and electrical resistance measurement.

FIB-SEM is used for cutting the test sample and observe the cross-section of the

structure. The information about the surface topography of the sample is given by SEM. The details about these two tools will be introduced in Chapter 4.

TEM has a better resolution than SEM, so for the sample of line-width below 0.2 mm TEM is the best choice. The working principle of TEM is similar to the light microscope. The major difference is that light microscopes use light rays to focus and produce an image while the TEM uses a beam of electrons to focus on the specimen, to produce an image. In addition, for the very thin sample, the high energy beam from TEM can shine through the sample, and the features such as the crystal structure and features in the structure like dislocations and grain boundaries can be observed.

The electrical resistance measurement is used to determine the EM failure. Since the change of resistance in metal line is found to be strongly dependent on voids nucleation which is equivalent to a slight increase of resistance in experiments [63], we can use the change of resistance to identify the degree of electromigration.

In order to study the electrical properties of metal interconnects the usage of resistivity measurement techniques is necessary, in which the bulk resistivity is commonly concerned. The bulk resistivity ρ of conductors or semiconductors depends on carrier drift. According to Ohm's law, the conductivity is explained as the ratio of current density to electric field intensity. Thus, by measuring the voltage drop along the sample with the sourcing current is the fast method to calculate bulk resistivity of the material. Unfortunately, for metal or other conductors, the magnitude of resistivity is pretty low and this requires accurate measurement. The simplest way to test resistivity can introduce potential errors such as contact resistance, noise and lead resistance. And with the shrinking size of device structures, contact resistance becomes more considerable, and eliminating the effect of contact resistance is crucial. Thus, there are two main test methods still acceptable today, namely four point probe and Van der Pauw technique, as shown in Figure 2.10, which are valid to minimize these errors.



(a)



Figure 2.10. (a) schematic of Four-Point Probe measurement; (b) schematic of Van der Pauw

method.

Chapter 3. Structure Design and Fabrication

In this chapter, a review of test structures used in electromigration will be introduced at first. It includes the Blech test structure, NIST structure, SWEAT structure and Line-via structure. Then a detailed discussion about all experimental structures in our work is given, also the purpose of each designed structure is indicated.

3.1 Test structures review

3.1.1 Blech test structure

There are various types of test structures used in EM testing. One of the simplest structures probably is "Blech's structure". As shown in Figure 3.1, the Al stripe is directly patterned on the TiN film. As Al is a better conductor than TiN, the majority of the current will detour from the TiN and go through Al stripes. By using this simple structure, we can determine the critical length, critical threshold, and drift velocity induced by electromigration.



Figure 3.1. Schematic of Blech structure. [44]

3.1.2 NIST structure

In 1987, considering the effect of the geometry of metal lines and temperature gradients on EM lifetime, Schafft [64] designed a new structure for resistance measurements, that is consist of a straight long line and four terminal structures, as shown in Figure.3.2. In standard NIST guidelines, a metal stripe with a length more than 800 um is suggested, to avoid the short line effect and thermal interferences from two pads. However, in this structure, the inconsistent width of the pad and metal line

(ratio of width is 2 to 1) leads to two different microstructures within the device, i.e. bamboo structure and cluster/ bamboo mixed structure, which results in mostly failures presenting themselves at the end of the pad [65]. To avoid this, various modified test patterns based on NIST structures were proposed later [66, 67], for example, the width of the pads is much larger than the width of the metal lines. The other flaw of the NIST test structure is that there are no significant macroscopic flux divergences and, therefore, experiments always indicate unrealistically long times-to-failure.



Figure 3.2. Schematic of NIST structure. []

3.1.3 SWEAT structure

Another commonly used test structure for electromigration is the SWEAT structure (Standard Wafer-level Electromigration Acceleration Test) proposed by B. J. Root and T. Turner [68]. In this structure, a series of metal stripes with two different widths are connected, as shown in Figure 3.3 [69]. Due to the inconsistency of the geometry, the narrow lines show higher current density and temperature, in which electromigration failure occurs. One of the advantages of such a structure is that it enables one to do multiple EM tests in one case until one of the narrow lines fails. In addition, this structure also can be modified to study the effect of current crowding on EM.



3.3.4 Line-via structure

Above-mentioned Blech, NIST and SWEAT test structures consist of a single level planar metallization. However, in reality, a multilevel interconnection system is required. For example, most of the copper-based metal interconnects use a line–via test structure [70], as shown in Figures 3.4 and 3.5. The EM test on sample of Figure 3.4 (a) is known as "M1 testing" and EM test on sample of Figure 3.4 (b) is known as "M2 testing". The dummy lines are used to conduct electrical current to the testing line. For the M1 testing line, the voids occur at the anode side. For the M2 testing line, the voids may occur at the bottom of the via or the end of the line and via. For three level line-via structure, as shown in Figure 3.5, the different current conditions may cause voids formation at various places.



Figure 3.4. Cross-section view of Line-via structure [70].



Figure 3.5. Cross-section view of Line-via structure with three levels [70].

3.2 Structure design

By comparing those popular structures used in the EM test and considering our fabrication capability, the Blech structure is selected in the present work. The layout of the mask design in this work is shown in Figure 3.6. Various structures are designed to investigate various influential factors for electromigration, such as the effects of mechanical stress, the microstructures of metal materials, the interface/surface of the interconnects, the boundary condition, and current crowding on electromigration.



Figure 3.6. Top view of mask design.

In general, the structure consists of a Si layer, SiO₂ layer, TiN layer, and Al layer, as shown in the following Figure 3.7. The silicon dioxide layer is patterned on the silicon surface because the oxidation layer performs a good adhesion to silicon and provides an excellent electrical isolation layer for the structure. A conducting layer is required at the bottom of the Al film and its resistivity should be higher than the testing wire. Besides, a barrier layer is needed to avoid the atomic diffusion between the dioxide surface and the Al layer. Compared with different materials, the titanium nitride layer is selected as the inter-media layer between Al and SiO₂.



Figure 3.7. Cross-section view of our structure

3.2.1 The structure to measure the drift length

Figure 3.8 shows the structure to measure the drift length. It consists of two parts, metal pads and test lines. To obtain good repeatability and statistics analysis, the long test line is composed of Al stripes repeated five times. In order to avoid forming voids within two big contact pads, the size of the pads is much wider than the metal strip. Besides, a contact pad with a 45-degree bevel is applied to prevent current crowding effect.

In the present study, various dimensions of the structure are designed, as summarized in Table 3.1. First, to investigate the effect of Al length on void length induced by EM, Al stripes with four different lengths, 10 μ m, 30 μ m, 60 μ m and 100 μ m are designed. Second, to determine the critical length of electromigration, Al stripes with 5 μ m, 10 μ m, 15 μ m, 20 μ m and 25 μ m are designed. Thirdly, to further investigate the effect of width on electromigration, four different widths: 3 μ m, 5 μ m, 10 μ m and 15 μ m, are included in our design. In addition to variations of the width and the length of the stripes within a single die, the thickness of the testing line is also varied using different wafers, in order to study its impact on electromigration. As the grain size of aluminum is ~ 300 nm, we choose 200 nm, 400 nm, 600 nm as the thickness of the Al metal layer to obtain different microstructures, i.e. bamboo structure and near bamboo structure.

Table 3.1. Dimensions design of our structure



Figure 3.8. Top view of the structure used for drift length investigation

3.2.2 The structure to investigate the effect of the passivation layer

Apart from the classical Blech's structure, the structure that includes a bare Al stripe and an Al stripe covered by Si₃N₄ in the same line is added in our mask, to investigate the effect of the passivation layer on EM within a single experiment. As shown in Figure 3.10 (a), the right part of the metal line is completely covered by a passivation layer and the other side is exposed. For Figure 3.10 (b), a partly covered aluminum stripe and a bare Al strip are given. In section 5.4, the effect of the passivation layer on EM will be discussed.


Figure 3.10. The structures for investigation passivation layer on EM: (a) completely covered structure (b) partly cover structure.

3.2.3 The structure to investigate the effect of boundary conditions on EM

The diffusion boundary condition is one of the important factors on EM. Figure 3.11 (a) and (b) show structures with two different diffusion boundary conditions, $J_a = 0$ and $J_a = C$ (constant), respectively. The length of the Al stripe is 800 μ m and the width is 5 μ m. For the structure shown in Figure 3.11 (b), two large pads are directly connected to the metal strip, playing a role as a reservoir for atomic diffusion. Thus, the

atomic flux in the metal interconnect is the same as in the metal pad.



Figure 3.11. (a) The structure of metal strip with atomic flux $J_a = 0$; (b) The structure of metal stripe with atomic flux $J_a =$ Constant.

3.2.4 The structure to investigate the effect of current crowding

In our measurements, a relatively large loading current density is applied, which may cause a local temperature increase and can bring unexpected instability to the system. To analyze this effect, a simulation with COMSOL was carried out, where the current density was simulated with three different geometries of the contact pads, pads with a right angle, with 45-degree bevel and with fillets (Figure 3.12 (a), (b) and (c), respectively). The simulation results show that the contact pads with fillets can effectively suppress the current crowding effect. Thus, we also included the following three structures to investigate such effects on electromigration.



Figure 3.12. (a)The structure of contact pads with right angle; (b) The structure of contact pads with 45-degree bevel; (c) The structure of contact pads with fillets.

(c)

3.2.5 The structure to investigate the effect of non-uniform current density

By altering the width of the metal lines, we could obtain a non-uniform current density in the whole stripe, as shown in Figure 3.15.



Figure 3.15. (a) The structure of triangle metal strip; (b) The structure of ladder-shaped metal strip.

3.2.6 The structure to in-situ measure the resistance

All the previous structures only allow for measuring the voltage over the entire line. In our work, the four-point probe structure is used to monitor the resistance change of Al stripes during electromigration. This structure consists of contact pads, test stripes and connecting wires, in which two large pads at the end of the stripe are connected to the ammeter to provide required current, and the voltage drop is measured by the two other pads. Since the input resistance in the voltmeter is large, the voltage drop on the connecting wires can be disregarded. The load current only flows through the pads at the end of the metal stripe, as shown in Figure 3.16. The metal pad is designed to be $300 \times 300 \ \mu m^2$, And the test line is designed with four different lengths, 10 \ m, 30 \ m, 60 \ m and 100 \ m.



Figure 3.16. Four-point resistor test structure

3.2.7 The structure to measure the sheet and contact resistances

For the sheet resistance measurement of different layers, the Van der Pauw structure is employed, as shown in Figure 3.17.



Figure 3.17. Van der Pauw resistor test structure.

Since the titanium nitride also serves as a part of the conductor line, the contact resistance between test stripes and titanium nitride becomes one of the important electrical properties. Due to complex components, the theoretical calculation of contact resistance cannot be accurate. Thus, we chose two main test structures, a Cross Kelvin test structure and transmission line model test structure, to directly achieve this measurement. As shown in Figure 3.18, in the Cross Kelvin structure two different metal layers are stacked together on the isolation layer with a contact area connecting them.



Figure 3.18. Cross Kelvin test structure

The specific contact resistance is given by,

$$\rho_c = R_c A_c \tag{3.1}$$

Where A_c is the area of contact window and R_c is contact resistance measured by Ohm's law.

The structure of the transmission line model is shown in Figure. 3.8 Similarly with Kelvin structure, given the width (W) of metal strip and the distance (L_i) between two adjacent contacts, when contact resistance (R_c) can be expressed by,

$$R_i = \frac{\rho_s L_i}{W} + 2R_c \tag{3.2}$$

Where ρ_s is layer sheet resistivity, R_i is resistance changes between two adjacent contacts.



3.3 Fabrication process

The fabrication steps to prepare the samples are presented here. The Blech's structure is mainly composed of three layers from bottom to top, connecting layer (TiN), conductor layer and the passivation layer. Hence, three different masks are prepared during the fabrication step.

3.3.1 The basic techniques

All fabrication steps are completed in the cleanroom of EKL. The basic fabrication techniques used in our process are,

1. Photolithography

The photolithography technique is based on the principle of optical-chemical reaction. Essentially, the process of the photoresist (photosensitive material) serves as the intermediation to transfer image information from the photomask to wafers. Three main steps are included in this technique: coating, exposure and developing. In our work, we use an EVG120 to automatically complete coating and development steps and the wafer stepper (ASML PAS 5500/80) is used to create 52 dies of the same patterns on one wafer.

2. Etching

The pattern etching is an important step in the interconnection technology of integrated circuits. The purpose of this step is to form the required pattern by selectively removing the material in the unmasked area. The main etching methods are wet etching and dry etching. Wet etching is an isotropic process by using liquid chemicals or etchants to remove material. Due to its property, it is not easy to control the time to accurately pattern without removing material underneath the masked area. However, this process has a high selectivity and etching rate.

On the other side, dry etching is anisotropic, which means etching only occurs in the vertical direction. The advantages of dry etching are accurate control of pattern size and automatic detection of the etching endpoint. Primarily, dry etching can be divided into three types by different reaction methods. Plasma etching and reactive ion etching (RIE) are common techniques used in semiconductor manufacturing. In our work, wet etching is used for aluminum layer and patterns on the titanium nitride layer are transferred by dry etching, the same is used for the contact openings of the passivation layer. Here the reason we use wet etching to pattern Al is that the etching selectivity to TiN is limited.

3. Chemical vapor deposition

Chemical vapor deposition technology is a process that uses gas or vapor mixture to generate chemical reactions and transport reactions on a substrate then deposit thin film. Commonly used CVD methods are plasma chemical vapor deposition (PECVD) and low pressure chemical vapor deposition (LPCVD). In the LPCVD technique, the high temperature is adopted to form high-quality film, on the other side, deposition temperature in the PECVD process is much lower but the deposited film has a poor step coverage compared to LPCVD process. In our work, considering the melting temperature of aluminum we choose the PECVD technique to grow the passivation film.

4. Sputtering

The sputtering deposition is one of physical vapor deposition (PVD) process

techniques, and it is widely used in the semiconductor industry. In this process, the ions generated by gas discharge bombard a metal target by applying an electric field, so that the atoms or molecules of the target are knocked out and are deposited on the substrate surface. Argon is commonly used as the sputtering gas and RF or DC is applied power, during the process. In our work, we use Radio Frequency (RF) sputtering for the growth of pure aluminum And for the deposition of the TiN layer, the reactive sputtering with N₂ and Ar is employed.

3.4.2 Process flow

The process starts from a p-type 525 μ m single side polished wafer that is commonly used as a test wafer in the cleanroom. To provide markers for subsequent exposure steps, the standard zero-layer process is necessary. A SiO₂ layer is deposited on the wafer surface to be served as an isolation layer, as shown in Figure 3.20. Since the quality of the oxidation layer is not important in our process, the wet oxidation technique is selected to reduce processing time, and the desired thickness is 300 nm.





Next, the titanium nitride layer is deposited on the SiO₂ layer at 300 °C by using reactive sputtering, as shown in Figure 3.21. In this process, the reactive gas (N₂) and vapor phase titanium react to form the titanium nitride at the vacuum chamber. This process also forms TiN at the target, which can cause the formation of a non-uniform concentration between the top wafer and the bottom wafer in a batch. In order to avoid this problem, a dummy wafer is added after each processed wafer with a clean recipe e.g. Ti_in_Between_300C. Then the pattern is transferred by using photolithography and dry etching. The etching process is realized by plasma etcher Omega 201 with the

recipe "TINTISVO", the etching time can be calculated by recent etching data from the logbook.









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The deposition of the passivation layer is a required step for some impacts investigation. As shown in Figure 3.23, the Si_3N_4 is deposited by PECVD with a mixture of N₂, SiH₄ and NH₃ at a temperature of 400 °C and the process pressure is 2.8 Torr. Referring to the logbook, the deposition time is calculated. After the Si_3N_4 deposition, the patterning of the Si_3N_4 layer is obtained by using the third mask to form contact openings on the pads.



Due to the limitation of measurement devices, our test process only can be done with small chips. So the dicing process is used and diced chips are $10 \times 10 \ \mu m$ size. In order to protect the aluminum from damage during process, a UV protect foil is used. The following picture (Figure 3.24) shows the diced chip.



Figure 3.24. The diced chip

Chapter 4. Measurement Procedure

In this chapter, the preparation of the EM test is introduced, including the measurement setup, test process, the condition of EM accelerated test and the characterization methods.

4.1 Measurement setup

Under the normal working conditions, the electromigration process in interconnects is very slow, typically the lifetime is from several years to several decades. It is quite difficult to effectively estimate the lifetime and reliability of interconnects in a short time under the conventional loading condition. Thus, an accelerated test is adopted. The accelerated test is a method that shortens the experiment periods by increasing stressing conditions without changing the failure mechanism. In this work, we need a setup that is able to provide the high current density and high temperature in order to accelerate failure. Temperature and current density are typically used as accelerated factors for the EM test and it is confirmed that altering these two factors will not change the failure mechanism. It is known that the bare aluminum is easy to react with oxygen and form the native oxide layer on the surface at room temperature, in addition, at elevated temperatures it is possible to form additional oxidation which is undesired in our work. Thus, a vacuum pump is required to create a vacuum test condition. The measurement setup is consist of three parts:

 A micro probe station (MPS): NEXTRON MPS is used for the in-situ measurement of electrical and optical properties of the devices. Three parts are included in the MPS, a hotplate, a pressure chamber and four micro-probes, as shown in Figure 4.2. The hotplate is connected to a temperature control via a computer and provides a working temperature up to 750 °C. Due to the limited volume of the pressure chamber, only the die-level samples can be tested. The probes used for electrical measurement have spring-assisted clamping, which avoids losing stability in gas flow and scanning experiments. The probes have different materials, here, the tungsten-based probe tip is used.



Figure 4.2. The composition of MPS

2. A vacuum pump: Figure 4.3 shows the PFEIFFER vacuum system, which is used for providing high vacuum pressure (range from 10⁻¹ to 10⁻⁴ Pa)



Figure 4.3. PFEIFFER vacuum system

3. A source meter for electrical measurement: a Keithley 2612B is connected to the probe station by four coaxial connectors and it is used to provide the required high current density and real-time monitoring of the voltage. The measurement can be programmed by using the software ' Keithley Test Script Builder '.

4.2 EM test condition

Since the interconnects are greatly affected by current density and temperature, increasing the current density and test temperature is used for the accelerated test. Selecting an appropriate acceleration condition is important to the experimental results. Based on the results of simulation and literature, in order to study the relationship between the electromigration and applied temperature, the selected temperatures are 200 °C, 250 °C, 300 °C and 350 °C the loading current density is $1 \times 10^{10} A/m^2$. In addition, we could also study the impact of current density at 250 °C and the current

density is $0.5 \times 10^{10} A/m^2$, $0.8 \times 10^{10} A/m^2$ and $1 \times 10^{10} A/m^2$.

4.3 Test process

The electromigration experiment has a long test period, in our work, it requires 20 hours. The test data will be automatically recorded by the program and saved on the computer. The measurement steps are described as follows:

- 1. Before EM accelerated test, an initial characterization test of samples is required by the Keyence 3D laser profilometer and SEM.
- 2. The sample is fixed on the hotplate of MPS by two/ four probes and the vacuum pump is turned on to avoid oxygen in the chamber.
- 3. When the pressure in the chamber is low enough, heat up the sample to the desired temperature.
- 4. Set the required current value and start the program. During the measurement, the sample voltage is monitored and recorded in real-time, the sampling frequency is 20 seconds/ time.

4.4 Characterization methods

SEM is a characterization method using a high power electron beam to observe samples on the micro- and nanometer level. Typically, the energy of the electron beam is range from several *keV* to 30 *keV*. As the electrons interact with the sample and generate secondary electrons. Secondary electrons will be detected and used for the images formation of the surface topography. For our experiments, we mainly study the surface properties of aluminum stripes, so the low energy of the electron beam is used to get accurate images. Figure 4.4 shows the SEM in the cleanroom.



Figure 4.4 Philips FEI XL 50

Keyence 3D Laser Scanning Microscopy, as shown in Figure 4.5, can be used for imaging and measurement. It couples a laser with an optical microscope to create high-resolution images. The 3D image is created by the height information based on reflected light intensity from the sample.



Figure 4.5. Keyence 3D laser scanning microscopy.

Chapter 5. Result on EM Tests

In this chapter, a brief introduction about the design of experiment will be given first, then the impact of different parameters on EM is discussed.

5.1 Design of experiment

Typically, when we study the EM at the micro-scale, there are various input factors that simultaneously affect the results. Design of experiment (DOE) is a technique that can determine the individual and interactive effects of various factors that can influence the output results of measurements. In our work, the impact of different parameters on EM is studied and only a single parameter is varied at a time, as the experiment designs shown in the following Table 5.1 and Table 5.2. According to the Blech experiment, the thickness range of the Al layer is from 200 nm to 1 μ m, the test temperature is from 200 °C to 550 °C and the current density is $4 \times 10^9 A/m^2$. In the initial test, the experiment was carried out at 200 °C and 250 °C and it was found that the electromigration was relatively slow at 200 °C, so the 250 °C is set as the temperature baseline. In addition, We tested 400 nm thickness samples and found that it would exceed our safety voltage, so finally, we chose 200 nm thickness Al for most of our experiments. And the loaded current stress guided by simulation result is $1 \times 10^{10} A/m^2$.

Table 5.1.	Input	factor	in	our	DOE.
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Input factor	
V1	Thickness of conductor layer
V2	Length of conductor
V3	Test ambient environment
V4	Test temperature
V5	Thickness of passivation layer
V6	Thermal treatment

Table 5.2.	The	information	n on test sample	s.

Sample	V1 (nm)	V2 (µm)	V3	V4 (°C)	V5 (nm)	V6 (°C)
#1	200	5/10/15/20/25	air	250	-	-
#2	200	10/30/60/100	air	250	-	-

#3	200	5/10/15/20/25	vacuum	250	-	-
#4	200	10/30/60/100	vacuum	250	-	-
#5	200	10/30/60/100	vacuum	200	-	-
#6	200	10/30/60/100	vacuum	300	-	-
#7	200	5/10/15/20/25	vacuum	300	-	400
#8	200	10/30/60/100	vacuum	300	-	400
#9	200	10/30/60/100	vacuum	250	300	400
#10	200	200/200	vacuum	300	800	400
#11	400	200/200	vacuum	300	900	400

5.2 Experimental study on the impact of conductor length on EM

In Section 2.1.1, we have stated that in the results of Blech experiments, it is found that the drift velocity of EM was inversely proportional to the stripe length and the longer stripe has a larger void at the cathode and higher drift velocity. Besides, based on the equation (2.6) proposed in the "Blech Condition", the critical length of stripes is defined as the length that when the stripe length is below it there is no electromigration failure. Hence, the critical length is an important factor of design structures.

5.2.1 Investigate the critical length under certain condition

As shown in Figure 5.1, in order to find the critical length, Al stripes with five different lengths (5/ 10/ 15/ 20/ 25 μ m) were patterned on a thin TiN layer. The accelerated test is carried out under the current density $J = 1 \times 10^{10} A/m^2$ at 250 °C. As shown in Figure 5.1, after loading the current for 10 hours in a vacuum, the 5 μ m (left end of the stripe) and 10 μ m stripes do not show any signs of electromigration and for the stripes above 10 μ m, obvious voids can be observed at the cathode. Thus, an approximate critical length under a certain condition has been found, the threshold length is 10 μ m.



Figure 5.1. Drift of Al stripes with five different lengths after passage of $1 \times 10^{10} A/m^2$ for 10

hours.

5.2.2 Investigate average drift length

The movement of Al atoms from the cathode to the anode implies an opposite drift of vacancies from the anode to the cathode. As a result, the vacancy accumulates at the cathode of the Al segments and form voids. Thus, the drift velocity can be defined as the ratio of drift length to drift time and the drift length here can be assumed as the average length of voids.

To study the relation between drift velocity and Al stripe length, four different lengths of Al segments are powered in series, as shown in Figure 5.2. The length of each stripe from the left to right are 10 μ m, 30 μ m 60 μ m and 100 μ m. The test condition is current density $J = 1 \times 10^{10} A/m^2$ at 250 °C. To observe the process of electromigration over time, we record the Al stripe in an initial state and different periods after applying the loading current, as shown in Figure 5.2. It can be seen that there is no obvious electromigration at 30 mins, but a small void is observed at 1 hr for the Al stripes with a length of 10, 30, 60 μ m. Also, after 20 hours we find that the void of 100 μ m metal stipe is larger than that of 60 μ m stripe.



Figure 5.2. Drift of Al stripes with varying lengths at different times.

Using the microscope the drift length is measured, which is shown in Figure 5.3.

	[1] 3.046µm	[2] 3	917µm		[8] 3.8559pm
10.00	*	- Cê	*		×
•	• •	0	0	0	•

Figure 5.3. Measurement of drift length for Al stripes with different lengths after 20 hours.

To reduce measurement errors and get more accurate results, five different groups of samples are tested for 20 hours and the test results are given in Table 5.3.

Conductors length(µm)	10	30	60	100
Drift length (µm)				
Sample 1	-	3.078	3.357	4.383
Sample 2	-	3.046	3.917	3.855
Sample 3	-	3.326	3.824	4.290
Sample 4	-	3.575	3.738	3.792
Sample 5	-	3.320	3.873	4.215
Average	-	3.269	3.742	4.107

Table 5.3. the measurement result of drift length for each sample.

The impact of conductor length on the average drift length can be clearly observed in Figure 5.4. In addition, in this Figure, the simulation result from Cui's model is given by a solid blue and red line for 10 hours and 20 hours. The simulation model is based on Blech theory, in the literature [45] the net drift velocity for long stripes can be express by,

$$V = V_{em} - V_{BF} \tag{5.1}$$

Where V_{em} is electromigration drift velocity, $V_{em} = \frac{D}{k_B T} Z^* e \rho j$ and V_{BF} is the backflow induced by the concentration gradient, $V_{BF} = \frac{D}{k_B T} \frac{\Delta C}{L}$. L is the stripe length, C is the vacancy concentration difference between the two ends of the strip. The drift length (L_d) is given by,

$$L_d = \int_o^t V \, dt \tag{5.2}$$

Where t is electromigration time. From the equation, it can be seen that with the increasing of conductor length, the backflow velocity will reduce and the net drift velocity is increasing, which results in the increase of drift length.

Figure 5.4 shows that the longer strip has the increasing drift length and when the length is up to 100 μ m, the drift velocity almost reaches saturation, which matches the simulation predictions and it is consistent with the conclusion in literature.



Figure 5.4. The relation between average drift length and conductor length. The current density was $1 \times 10^{10} A/m^2$ at 250 °C for 10 and 20 hours.

5.3 Experimental study on the impact of temperature on EM

The temperature has a significant effect on EM. On the one hand, higher temperature means the faster diffusion rate of Al atom, which can accelerate the EM process. On the other hand, if the temperature distribution along the Al stripe length is non-uniformed, the atom will migrate driven by the temperature gradient, which will also affect the EM process. The latter effect is not the main topic in the present study. Thus, in this section, four groups of samples are heated up to 200 °C, 250 °C, and 300 °C, 350 °C, respectively, to investigate the effect of uniform temperature on EM The experiment is carried out under the current density $J = 1 \times 10^{10} A/m^2$ for 20 hours. The experiment results for these 4 samples are shown in Figure 5.5, the sample tested at 350 °C shows the largest voids for 30, 60 and 100 µm stripes and in some samples, we could clearly observe the voids at the cathode of 10 µm stripe, which indicates that

the critical length for 350 $^{\circ}$ C is below 10 μ m. In literature, it reported that the threshold length product was found to be a function of temperature and the threshold length product increases with the decreasing temperature [44].



Figure 5.5. Drift of three sets of Al stripes from 200 °C to 350 °C

All the experiments are performed for 4 groups of samples to reduce the measurement error. Based on the test data, the relationship between drift length and the temperature is given in Figure 5.6. It can be seen that for 100 μ m stripes the drift length increases with elevated temperature.



Figure 5.6. The relationship between drift length and temperature.

According to $V_{em} = \frac{D}{k_B T} Z^* e \rho j$, the calculated diffusivity at the different temperatures is given in Table 5.4.

Temperature (K)	Diffusivity (m ² /s)				
	#1	#2	#3	#4	
500	4.91E-16	5.22E-16	6.15E-16	6.99E-16	
550	1.38E-15	1.58E-15	1.67E-15	1.91E-15	
600	3.71E-15	4.46E-15	4.63E-15	5.01E-15	
650	6.90E-15	8.22E-15	8.22E-15	9.18E-15	

Table 5.4. Diffusivity at different temperatures.

From the above result, we could also determine the Arrhenius equation of the diffusion coefficient at different temperatures. The Arrhenius equation is given in equation (5.1),

$$D = D_0 exp(-E_a/kT)$$
(5.3)
$$In(D) = -\frac{E_a}{kT} + In(D_0)$$

Where D is the diffusion coefficient, D_0 is the pre-exponential factor, E_a is the activation energy, k_B is the Boltzmann's constant and T is the temperature.

By calculating, the function of this curve is y = 5.77x + 23.52, which is shown in Figure 5.7. the pre-exponential factor (D_0) and the activation energy (E_a) are obtained through the slope and vertical intercept, $D_0 = 6.1 E - 11 m^2/s$ and $E_a = 0.5 eV$. In literature, Al interconnect has the activation energy of 0.5 eV with grain boundary dominated diffusion as the failure mechanism [70], which is constant with our result. But the pre-exponential factor in our result is two orders of magnitude less than that in literature, this may be because their value is calculated for bulk Al and the Al in our experiment is a thin film.



Figure 5.7. The relation between -In(D) and 1/T.

5.4 Experimental study on the impact of annealing on EM

Annealing is a heat treatment process for metal materials in which the metal is heated up to a certain temperature, kept for the required time, and then slowly cooled. Post annealing is a critical step in film fabrication, it can enhance the properties of thin-film by affecting the phase of the material and altering the microstructure. The annealing process can be divided into three phases, recovery, recrystallization, and the growth of grains. When the annealed temperature is low or at the initial phase of the annealing process, it can cause the removal of dislocations and internal stresses. Figure 5.8 shows an SEM micrograph of a nonannealed stripe (a) and an annealed (annealed 30 mins at 400 °C) stripe (b). From the Figure, a few small hillocks randomly formed at the Al surface can be observed and the boundary between grain and grain becomes blurred. In addition, the size of the grain does not significantly change.



Figure 5.8. SEM micrograph of Al stripes (a) sample before annealed; (b) sample after annealing.

In our experiment, the 200 nm thickness Al film is selected and the samples are annealed at 400 °C for 30 mins. The current density and test temperature as control variables are kept the same as in the previous experiment, $1 \times 10^{10} A/m^2$ and 250 °C. The test results after 10 hours and 20 hours are given in Figure 5.9. In this plot, there are no obvious voids at the cathode of the stripe, however, the hillocks at the anode can be clearly seen. To explain this result, an SEM micrograph of the left edge of the 60 µm strip is given in Figure 5.10. It can be seen that small grooves exist between the grains and they are not able to form larger voids as previously observed.



(a)

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Figure 5.9. Plots of unannealed (a) and annealed (b) samples after 10 hours and 20 hours.



Figure 5.10. SEM micrograph of the cathode of annealed Al stripes after passage of $1 \times 10^{10} A/m^2$ for 20 hours.

To accelerate the experiment, we increase the test temperature to 300 °C and keep the value of the current density. Figure 5.11 shows that after 10 hours the grooves become small voids at the edge of Al stripe.



Figure 5.11. The 3D plot of annealed samples at 300 °C for 10 hours.

In general, the annealing process can be used to prolong the EM lifetime. According to MD simulation performed by Zhen Cui, as shown in Figure 5.12, this is because the annealing process improves the coalescence of grain in Al film, reducing the defects at the grain boundary and forming a denser microstructure, compared to the non-annealed Al film.



Figure 5.12. MD simulation results for the impact of annealing process on microstructure.

5.4 Experimental study on the impact of the passivation layer on EM

In Section 2.2, we discussed the impact of mechanical stress on electromigration,

when the conductor line is covered by the passivation layer, the stripes are subject to mechanical constraints imposed by the surrounding layers. Because of the rigidity of the passivation layer, the accumulations of material at the anode are suppressed and ideally, the volume of the conductor line cannot be changed, which decreases the vacancy diffusion rate and increases the electromigration lifetime.

In our work, an 800 nm SiN passivation layer is deposited on the 200 nm Al film. The test is under 250 °C for 5 hours and 10 hours and the loading current is the same as in previous experiments. It is quite difficult to observe the voids at the cathode for the Al stripe covered with a passivation layer, considering the properties of SiN film, we use the optical microscope to obtain a better observation. As shown in Figure 5.13, the top stripe shows the bare Al after 5 hours and the result of Al covered with the passivation layer is the bottom line. From the Figure, we can observe the voids at the cathode of bare Al stripes and for metal lines covered with SiN, the light region at the edge of the stripe represents the formed voids.

Bare Al at 5h	rs	And in case of the local division of the loc	
0 8 0		0 0	0
Al with passiv	vation at 5hrs		
	· · · · · · · · · · · · · · · · · · ·		
0 0 0	0 0	0 0	0

Figure 5.13. Comparison diagram of 200 nm thickness samples uncovered and covered SiN passivation layer.

By comparing the voids on different lengths of Al stripes, we find that the voids under these two conditions are similar, in other words, the passivation layer appears to not suppress EM. However, this result is not consistent with the conclusion of the literature.

In order to determine the reason behind this, we consider that the defects in the microstructure of materials may simultaneously affect the result., Which was discussed in previous Section 2.1.2 and we decide to anneal the sample at 400 °C for 30 mins.

We have previously shown that the time to form voids for annealed Al is longer than the non-annealed sample. Thus, in order to accelerate the experiments, the test temperature is increased to 300 °C. Figure 5.14 shows the top view of this test structure, two different stripes are powered in series, one is completely confined by the passivation layer, on the other side, the stripe is free. After 20 hours the void on the bare Al stripe is larger than the covered metal line, as shown in Figure 5.15. It indicates the SiN passivation layer takes effect and prolongs the EM lifetime.

For the annealed Al stripes with a thickness of 400 nm, there is no void observed for annealed Al with 900 nm thickness SiN passivation layer, but a big void is observed for the annealed bare Al, as shown in Figure 5.16. That further demonstrated that the passivation layer can reduce the EM in the annealed sample. In other words, for annealed stripes, the passivation layer prolongs the electromigration lifetime as is expected from the literature. However, the SiN passivation layer does not take any effects on reducing electromigration when it covers a non-annealed Al stripe. It could due to the mismatch of thermal expansion and Young's moduli between metallization material and its surrounding materials, several visible voids are induced by thermal stress [71]. These voids may cause that the conductor already has many problems and worse mechanical behaviours than the conductor without depositing SiN. That may explain why the EM development for Al with SiN is almost the same as that for bare Al.



Figure 5.14. Top view of the test structure.





(b)

Figure 5.15. Drift of 200 nm thickness samples after 20 hours at 300 °C, (a) the full covered Al stripe. (b) the bare Al stripe.



Figure 5.16. Drift of 400 nm thickness samples after 3.5 hours at 300 °C, (a) the full covered Al stripe. (b) the bare Al stripe.

5.6 Experimental study on the impact of oxide on EM

It is known that when aluminum is exposed to air, a thin oxide film will form on the surface even at room temperature. Usually, the thickness of the oxidation layer is from 1 to 10 nm, but the development of thickness significantly depends on the temperature and the environment. Considering our test condition is at high temperature and high current density, the oxide on the Al surface could be an important factor on EM. In this experiment, the temperature, load current density, length of the conductor, and test time are as control variables, while the test ambient environment (V3) is the only input variable. After 10 hours the Al stripe which is exposed to air shows much larger voids at the cathode compared to the metal line in a vacuum, as shown in Figure 5.17.

To explain this, the TLM structure (Figure 3.19) is used to monitor the resistance change for the certain Al stripe. From the above discussion, we found that there was no electromigration for 10 μ m stripe in a vacuum, and the resistance change of 10 μ m stripe in a vacuum in Figure 5.18 (green line) showed that the resistance remains unchanged after 16 hours, which indicates that no voids form at the cathode of the stripe. However, the resistance change for the 10 μ m Al stripe in air shows a different tendency, as shown in Figure 5.18 (red line), the resistance keeps increasing and the resistance change exceeds 50% after 16 hours. In Figure 5.18, the 10 μ m stripe in the air has no signs of electromigration. Because of this, we conclude that the resistance increases only as a result of the growth of aluminum oxide on the surface. The growth of the oxidation layer means the reduction of aluminum layer and the decrease of the cross-section for the aluminum layer. Thus, a higher current density actually is applied to Al stripes. As a result, much faster EM failure occurs.



Figure 5.17. Drift of two groups of samples in a vacuum (top) and in the air (bottom) after 10 hours.



Figure 5.18. Resistance change for two sets of 10 μm Al stripes.
Chapter 6. Conclusion and Future work

A brief review of the work that has been done in previous chapters will be given first. Then, several recommendations for future work will be discussed.

6.1 Conclusion

The main goal of this work was to experimentally study EM and provide experimental data to validate our theoretical model. It is known that electromigration is a complex multi-physics process, including electrical, thermal, mechanical, and chemical aspects. However, the reported work on the formulation and modeling is not complete and not consistent with experimental results. A fully-coupled and self-consistent electromigration theory was proposed by Cui *et al*, and to validate their theoretical result, our experiment used the simplest test structure, the so-called "Blech structure", and adopted the accelerated test method with high temperature and current stress.

Electromigration is a multi-scale problem, the structure we designed not only takes into account the effect of dimensions of interconnect at the macro-level, but also the influence of microstructure of metal on electromigration. In addition, different electrical measurement structures were designed using four-point measurement to achieve accurate failure analysis.

The fabrication of our basic structures, together with test structures is a two mask process and masks are used to pattern the TiN and Al layers. The main structure consists of a Si layer, SiO₂ layer, TiN layer, and Al layer. Sputtering deposition, one of the PVD process techniques, was used for the deposition of metal and plasma etching and wet etching were used to etch TiN layer and Al layer, respectively. Besides, the additional covered passivation layer SiN was deposited by PECVD and etched by plasma etcher using a third mask.

The accelerated EM test was carried out under certain current stress at an elevated temperature. When used as control variables, the current density is $1 \times 10^{10} A/m^2$ and

the temperature is 250 °C. First, we studied the impact of conductor length on EM and the measurement results are summarized as follow,

- 1. The critical length under $1 \times 10^{10} A/m^2$ at 250 °C was found to be 10 µm
- 2. Longer stripes have lager drift length and a shorter time failure for electromigration.

Above experimental results match the simulation prediction performed by Zhen Cui and is consistent with the results in Blech experiments.

To investigate the impact of temperature on EM, the setup is heated up from 200 °C to 350°C. With the elevated temperature, the drift length increases and the electromigration lifetime decreases. By using the experiment data, the Arrhenius equation of the diffusion coefficient at different temperatures was obtained. And according to the expression of the fitting curve, the active energy of the Al film was $0.5 \ eV$ and pre-exponential factor was $6.1 \ E - 11 \ m^2/s$. The activation energy is consistent with the value in literature.

In addition, the effect of the SiN passivation layer on EM was studied. Interestingly, it was found that the SiN passivation layer does not take any effects on reducing electromigration when it covers a non-annealed Al stripe. However, for the annealed Al stripe, the SiN passivation layer could obviously decrease the EM process. According to MD simulation performed by Zhen Cui, this is because the annealing process improves the coalescence of grain in Al film, reducing the defects at the grain boundary and forming a denser microstructure, compared to the non-annealed Al film.

Furthermore, we also found that the additional oxidation on the Al surface at elevated temperature can reduce the thickness of Al film and increase the actual current density loaded in the sample, resulting in a shorter EM lifetime for a test performed under ambient pressure instead of vacuum.

In general, present experimental results were consistent with existing results in the literature, and we also discovered that the effects of passivation layer on EM are dependent on the annealing condition of Al stripe. The platform established in the present study can be continually used for our future studies on EM.

6.2 Recommendations and future work

Due to time constraints, several problems and results are not able to be solved and explained in this work, therefore, future work is needed to continue and develop this study. Additional experiments and measurements that would be interesting to be performed are briefly discussed as follow.

1. Based on the reviewed literature, various structures are designed to investigate different impacts on EM, however, we are not able to carry on EM tests for all structures, for example, the structures for studying impact of current crowding effect and interdiffusion on EM should be achieved in the future.

2. According to our results from Section 5.4, samples with 400 nm thickness showed larger voids than the 200 nm thickness Al stripes. While, in order to investigate the impact of conductor thickness, more tests with different thickness samples are necessary to perform.

3. In Section 5.4, we found that the covered samples without thermal treatment are not able to take effect on suppressing electromigration. In literature, due to the mismatch of thermal expansion and Young's moduli between metallization material and its surrounding materials, several visible voids are induced by thermal stress. These voids may cause that the conductor already has many problems and worse mechanical behaviors than the conductor without depositing SiN. That may explain why the EM development for Al with SiN is almost the same as that for bare Al. However, more evidence is necessary to verify this theory and explain our results.

4. As we discussed in Section 5.3, the SEM micrograph shows that after annealing the Al microstructure becomes denser and several hillocks are formed on the surface, but we were not able to use tools to characterize it.

5. In our work, each EM test time takes more than 10 hours, which makes it difficult for us to accomplish a large number of tests in a short time. To be more efficient, the test structure should be improved.

Appendices

A Process flowcharts

Run number: EC2426 Contamination: None

CLEANING : HNO3 99% and 69.5% (OPTIONAL)

Clean	10 minutes in fuming nitric acid at ambient temperature. This will
	dissolve organic materials.
	Use wet bench "HNO ₃ 99% (Si)" and the carrier with the white dot.
Rinse	Rinse in the Quick Dump Rinser with the standard program until the resistivity is 5 M Ω .
Clean	10 minutes in concentrated nitric acid at 110 °C. This will dissolve metal particles.
	Use wet bench "HNO ₃ 69,5% 110C (Si)" and the carrier with the white dot.
Rinse	Rinse in the Quick Dump Rinser with the standard program until the resistivity is 5 M Ω .
Dry	Use the Semitool "rinser/dryer" with the standard program, and the white carrier with a red dot.

COATING

Use the coater station of the EVG120 system to coat the wafers with photoresist. The process consists of:

• a treatment with HMDS (hexamethyldisilazane) vapor, with nitrogen as a carrier gas

- spin coating of Shipley SPR3012 positive resist, dispensed by a pump
- a soft bake at 95 °C for 90 seconds
- an automatic edge bead removal with a solvent

Always check the relative humidity $(48 \pm 2\%)$ in the room before coating and follow the instructions for this equipment.

Use program "Co - 3012 - zero layer". There will be a larger edge bead removal.

EXPOSURE

Processing will be performed on the ASML PAS5500/80 automatic wafer stepper. Follow the operating instructions from the manual when using this machine. Expose masks **COMURK**, with job "**ZEFWAM**" and the correct exposure energy 125 mJ/cm^2

This results in alignment markers for the stepper and contact aligner.

DEVELOPING

Use the developer station of the EVG120 system to develop the wafers. The process consists of:

- a post-exposure bakes at 115 °C for 90 seconds
- developing with Shipley MF322 with a single puddle process
- a hard bake at 100 °C for 90 seconds

Use program "Dev - SP".

INSPECTION: Linewidth and overlay

Visually inspect the wafers through a microscope and check the line width and overlay. No resist residues are allowed.

PLASMA ETCHING: Alignment markers (URK's) in Silicon

Use the Trikon Ω mega 201 plasma etcher.

Follow the operating instructions from the manual when using this machine. It is **not** allowed to change the process conditions and times from the etch recipe! Use sequence **URK_NPD** (with a platen temperature of **20** °C) to etch 120 nm deep ASM URK's into the Si.

Process conditions from chamber recipe URK_ETCH:						
Step	Gasses & flows	Pressure	Platen RF	ICP RF	Platen temp.	Etch time
1. breakthrough	$CF_4/O_2 = 40/20$ sccm	5 mTorr	60 W	500 W	20 °C	0'10"
2. bulk etch	$Cl_2/HBr = 80/40$ sccm	60 mTorr	20 W	500 W	20 °C	0'40"

LAYER STRIPPING: Photoresist

Strip resist Use the Tepla Plasma 300 system to remove the photoresist in an oxygen plasma.

Follow the instructions specified for the Tepla stripper and use the quartz carrier.

Use **program 1**: 1000 watts power and automatic endpoint detection + 2 min. over etching.

CLEANING: HNO3 99% and 69.5%

Clean 10 minutes in fuming nitric acid at ambient temperature. This will dissolve organic materials.
Use wet bench "HNO₃ 99% (Si)" and the carrier with the white dot.
Rinse in the Quick Dump Rinser with the standard program until the

Clean	resistivity is 5 M Ω . 10 minutes in concentrated nitric acid at 110 °C. This will dissolve metal particles.
	Use wet bench "HNO ₃ 69,5% 110 C (Si)" and the carrier with the white dot.
Rinse	Rinse in the Quick Dump Rinser with the standard program until the resistivity is 5 M Ω .
Dry	Use the Semitool "rinser/dryer" with the standard program, and the white carrier with a red dot

WET OXIDATION: 300nm thermal SiO₂

Furnace tube: Total time:	C1	Program	name		WETOX	
PROCESS	TEMP	ERATURE (in °C)	GAS & FLOWS (in liter/min)	TIME (in minutes)	REMARKS	
boat in	800		nitrogen: 3.0 oxygen:0.3	5		
stabilize	800		nitrogen: 3.0 oxygen:0.3	10		
heat up	+10 °C	/min	nitrogen: 3.0 oxygen:0.3	30		
stabilize	1100		nitrogen: 3.0 oxygen:0.3	10		
purge	1100		nitrogen: 3.0 oxygen:2.2 5	1		
oxidation	1100		nitrogen: 2.25 oxygen:3.8 5	16 min 15.8 s		
cool down	-10 °C/	/min	nitrogen: 3.0	55		
boat out	800		nitrogen: 3.0	5		

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MEASUREMENT: OXIDE THICKNESS

Use the Leitz MPV-SP measurement system to measure the oxide thickness: Program: Th. SiO2 on Si, >50nm auto5pts Expected thickness: 300 nm Measured thickness :

TIN PVD DEPOSITION

Use the TRIKON SIGMA sputter coater for the deposition of the TiN metal layer on the process wafers.

Follow the operating instructions from the manual when using this machine.

Use recipe TiN_300nm_300 to sputter a 0.3 µm thick layer of TiN. Add dummy wafers in between (Ti_inbetween_300C) as required. Temperature is 300 °C Perform a target clean

Visual inspection: the metal layer must look shiny with gold-brown color.

COATING AND BAKING

Use the EVG 120 wafertrack to coat the wafers with resist, and follow the instructions specified for this equipment.

The process consists of a treatment with HMDS (hexamethyldisilazane) vapor with nitrogen as a carrier gas,

spin coating with Shipley SPR3012 positive photoresist, and a soft bake at 95degC for 90 seconds.

Always check the temperature of the hotplate and the relative humidity $(48 \pm 2 \%)$ in the room first.

Use coating **Co – 3012 – 1.4µm**.

ALIGNMENT AND EXPOSURE

Processing will be performed on the ASM PAS 5500/80 automatic wafer-stepper. Follow the operating instructions from the manual when using this machine. Energy = TBD mJ/cm² (Perform E-F matrix)

DEVELOPING

Use the developer station of the EVG120 system to develop the wafers. The process consists of:

- a post-exposure bake at 115 °C for 90 seconds
- developing with Shipley MF322 with a single puddle process
- a hard bake at 100 °C for 90 seconds

Always follow the instructions for this equipment.

Use program "1-Dev - SP"

INSPECTION: Linewidth and overlay

Visually inspect the wafers through a microscope, and check the linewidth. No resist residues are allowed.

PLASMA ETCHING OF TITANNIUM NITRIDE

Use the Trikon Ω mega 201 plasma etcher to etch titanium ntride Follow the operating instructions from the manual when using this machine. The process conditions of the etch program may not be changed ! Use sequence **TINTISVO** and set the platen temperature to **25°C**. Check the time is set for 2min 30 sec to etch 300nm TiN (from the logbook)

INSPECTION

Visually inspect the wafers through a microscope, and check if the patterns are indeed existed. No residues are allowed.

LAYER STRIPPING: Photoresist

Strip resist Use the Tepla Plasma 300 system to remove the photoresist in an oxygen plasma.

Follow the instructions specified for the Tepla stripper and use the quartz carrier.

Use program 1: 1000 watts power and automatic endpoint detection + 2 min. over etching.

CLEANING: HNO₃ 99% (Metals)

Clean	10 minutes in concentrated nitric acid.			
	Use wet bench "HNO ₃ 99% (Metals)" and the carrier with the red and			
	yellow dots.			
Rinse	Rinse in the Quick Dump Rinser with the standard program until the resistivity is 5 M Ω .			
Dry	Use the Semitool "rinser/dryer" with the standard program, and the white carrier with a black dot.			

MEASURENMENT : TIN THICKNESS

Use the DEKTAK profilometer for stepheight measurements on the wafer of your batch. Mwasure on the bondflaps.

Following the operating instructions from the manual when using this machine. Stepheight:

ALUMINIUM SPUTTERING

Use the TRIKON SIGMA sputter coater for the deposition of an aluminum metal layer on the wafers.

The target must exist of 99% Al and 1% Si, and deposition must be done at 350 °C. Follow the operating instructions from the manual when using this machine.

Use recipe ""AISi_200_RF_300C" to sputter a 200nm thick aluminum layer at temperature 350 °C.

Note: Check the wafers after aluminum deposition. Metal residuals should be not present on the edge neither on the frontside nor the backside of the wafer.

COATING AND BAKING

Use the coater station of the EVG120 system to coat the wafers with photoresist. The process consists of:

- a treatment with HMDS (hexamethyldisilazane) vapor, with nitrogen as a carrier gas
- spin coating of Shipley SPR3012 positive resist, dispensed by a pump
- a soft bake at 95 °C for 90 seconds
- an automatic edge bead removal with a solvent

Always check the relative humidity $(48 \pm 2 \%)$ in the room before coating, and follow the instructions for this equipment.

Use program "Co - 3012 – 2.1µm".

ALIGNMENT AND EXPOSURE

Alignment and exposure will be done with the ASML PAS5500/80 automatic wafer stepper. Follow the operating instructions from the manual when using this machine.

DEVELOPING

Use the developer station of the EVG120 system to develop the wafers. The process consists of:

- a post-exposure bake at 115 °C for 90 seconds
- developing with Shipley MF322 with a single puddle process
- a hard bake at 100 °C for 90 seconds

Always follow the instructions for this equipment.

Use program "Dev – SP"

INSPECTION: Linewidth and overlay

Visually inspect the wafers through a microscope, and check the line width and overlay. No resist residues are allowed

WET ETCHING OF ALUMINUM

Etch Use wet bench "Al etch (35 C)" and the carrier with the yellow dot. The bath contains PES at temperature 35 C. A typical etch rate of Al is 170 nm/min.

Time 1 minutes and 20 seconds.(first with test wafer)

Rinse Rinse in Quick Dumo Rinser with the standard program until the resistivity is $5 \text{ M}\Omega$.

Etch Use wet bench "Poly-Si etch" and the carrier with the green dot to remove the 1 % silicon.

Time 30 seconds.

Rinse Rinse in Quick Dumo Rinser with the standard program until the resistivity is $5 \text{ M}\Omega$.

Dry Use the "Avenger Ultra-Pure 6" rinser/dryer with standard program, and the white carrier with a black dot.

INSPECTION: No Aluminum residues are allowed on the etched area()

LAYER STRIPPING: Photoresist

Strip resist Use the Tepla Plasma 300 system to remove the photoresist in an oxygen plasma.

Follow the instructions specified for the Tepla stripper, and use the quartz carrier.

Use program 1: 1000 watts power and automatic endpoint detection + 2 min. over etching.

CLEANING: HNO3 99% (Metals)

Clean	10 minutes in concentrated nitric acid.
	Use wet bench "HNO ₃ 99% (Metals)" and the carrier with the red and
	yellow dots.
Rinse	Rinse in the Quick Dump Rinser with the standard program until the
	resistivity is 5 M Ω .
Dry	Use the Semitool "rinser/dryer" with the standard program, and the white
	carrier with a black dot.

PECVD SiN 300 nm

Use the Novellus Concept One PECVD reactor. Check Gases Use group: (recipe **xxx STSiN**) to deposit a 300 nm thick SiN layer. Follow the operating instructions from the manual when using this machine. Change time to get the right thickness according to logbook. (14sec)

MEASUREMENT: OXIDE THICKNESS

Use the Leitz MPV-SP measurement system to measure the oxide thickness: Program: PECVD SiN on Al Expected thickness: 300 nm Measured thickness :

COATING AND BAKING

Use the coater station of the EVG120 system to coat the wafers with photoresist. The process consists of:

• a treatment with HMDS (hexamethyldisilazane) vapor, with nitrogen as a carrier gas

• spin coating of Shipley SPR3012 positive resist, dispensed by a pump

- a soft bake at 95 °C for 90 seconds
- an automatic edge bead removal with a solvent

Always check the relative humidity (48 \pm 2 %) in the room before coating, and follow the instructions for this equipment.

Use program " Co - 3012 – 2.1µm ".

ALIGNMENT AND EXPOSURE

Alignment and exposure will be done with the ASML PAS5500/80 automatic wafer stepper. Follow the operating instructions from the manual when using this machine.

DEVELOPING

Use the developer station of the EVG120 system to develop the wafers. The process consists of:

- a post-exposure bake at 115 °C for 90 seconds
- developing with Shipley MF322 with a single puddle process
- a hard bake at 100 °C for 90 seconds

Use program "Dev – SP"

INSPECTION: Linewidth and overlay

Visually inspect the wafers through a microscope, and check the line width and overlay. No resist residues are allowed

DRY ETCHING SILICION NITRIDE

Use the Drytek Triode 384T plasma etcher.

Follow the operating instructions from the manual when using this machine.

It is **not** allowed to change the process conditions from the etch recipe, except for the etch times!

Use recipe **SCRATCH** to etch the silicon nitride layer with a soft landing on the layer underneath (need new recipe)

Process conditions from recipe SCRATCH:					
Step	Gasses & flows	Pressure	RF power	He pressure	Etch time
1. bulk etch (RIE)	C2F6=65 sccm	130m Torr	250W	8 Torr	
2 landing step	C2F6=65 sccm	130m Torr	100W	8 Torr	

INSPECTION

Visually inspect the wafers through a microscope, and check if the vias are indeed open. No residues are allowed.

MEASUREMENT: SILICION NITRIDE THICKNESS (optional)

Use the Leitz MPV-SP measurement to measure the left TiN layer thickness: Program:

Nitride thickness:

LAYER STRIPPING: Photoresist

Use the Tepla Plasma 300 system to remove the photoresist in an oxygen plasma. Follow the instructions specified for the Tepla stripper, and use the quartz carrier. Use program 1: 1000 watts power and automatic endpoint detection + 2 min.

CLEANING: HNO3 99% (Metals)

Cleaning 10 minutes in concentrated nitric acid. Use wet bench "HNO₃ 99% (Metals)" and the carrier with the red and yellow dots.

Rinse Rinse in the Quick Dump Rinser with the standard program until the

resistivity is 5 MΩ.

Dry Use the Semitool "rinser/dryer" with the standard program, and the white carrier with a black dot.

MEASUREMENT: SIN etching optimization

Use the Keyence microscope for stepheight measurements on the wafers of your batch. Following the operating instructions from the manual when using this machine.

ALLOYING

Furnace no : C4

Program name : ALLOY1 Total time : 50 min

PROCESS	TEMPERATUR	GASSES &	TIME	REMARKS
	Е	FLOWS	(in	
	(in °C)	(in liter/min)	minutes)	
boat in	400	nitrogen: 3.0	5	
stabilize 1	400	nitrogen: 3.0	5	
stabilize2	400	nitrogen: 3.0	10	
alloying	400	nitrogen: 3.0 hydrogen:0.3	20	
purge	400	nitrogen: 3.0	5	
boat out	400	nitrogen: 3.0	5	

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