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Low-Impedance PEDOT:PSS MEA Integrated in a Stretchable Organ-on-Chip Device

Affan K. Waafi[®], Nikolas Gaio[®], William F. Quiros-Solano, Paul Dijkstra, Pasqualina M. Sarro, *Fellow, IEEE*, and Ronald Dekker

Abstract—We present the first Organ-on-Chip equipped Poly(3.4-ethylenedioxythiophene) with a low-impedance polystyrene sulfonate (PEDOT:PSS) **MicroElectrode** Array (MEA). The novel device allows simultaneous mechanical stimulation with a stretchable PDMS membrane and electrical monitoring via the PEDOT:PSS MEA of multiple in vitro cell cultures. The surface area enhancement and the morphology of the PEDOT:PSS allows an increase of the charge injection per unit area at the electrode-electrolyte interface, resulting in significantly lower electrochemical impedance of the electrodes. In particular, at 1 kHz the fabricated PEDOT-MEA electrodes show a reduction of the overall impedance up to 99.4 and 93.3 % in comparison with



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benchmark TiN and Pt electrodes. The superior performance of PEDOT:PSS were also confirmed via Cyclic Voltammetry measurement, in which PEDOT:PSS showed a very large capacitive current, compared with the benchmark electrodes both in the forward and the reverse scans. The obtained results confirm the effectiveness of the proposed PEDOT:PSS coating, and introduce this material in the OOC field. Moreover, the quality and morphology of the fabricated PEDOT:PSS based electrodes were assessed via SEM imaging and Raman spectroscopy.

Index Terms—Organ-on-chip, microelectrode array, PEDOT:PSS, PDMS.

I. INTRODUCTION

ORGAN-ON-CHIPS (OOCs) are *in vitro* models, which replicate the minimal functional unit of an organ by combining advanced polymer chip technology with

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biology [1], [2]. They are microfabricated devices consisting of microfluidic components, e.g., channels, chambers, valves, and stretchable or porous membranes, on which cells are cultured [3], [4]. Higher control over phenomena such as cell growth, proliferation, differentiation, maturation and other cell processes are enabled by the dynamically controlled environment in the device [3].

Even though currently available OOCs allowed a deeper understanding of numerous cell processes [5], very frequently they lack the sensing mechanisms necessary to monitor these events, thus requiring the use of bulky conventional imaging techniques to identify and study the phenomena happening in the model [1]. Macroscopic platinum (Pt) wires have been manually inserted in soft-lithography-fabricated channels in order to perform transendothelial electrical resistance (TEER) measurements [6]. The fabrication of these devices and the integration of the electrodes heavily rely on several manual steps [6]. This approach strongly limits the miniaturization of the electrodes, and thus the integration of low-impedance Micro-Electrode Arrays (MEAs), an essential tool to perform in-situ monitoring of electrogenic tissues, such as neurons, muscles and cardiomyocytes, at a subcellular level [7]. Previous work has addressed this issue by combining commercially available MEAs with soft-lithography based microfluidic channels [8]–[10]. Even though this solution guarantees low-impedance MEAs, it does not allow for the integration

1558-1748 © 2019 IEEE. Personal use is permitted, but republication/redistribution requires IEEE permission. See http://www.ieee.org/publications_standards/publications/rights/index.html for more information. of the electrodes in the stretchable membranes, a key element in OOCs.

To overcome these shortcomings, here we present an OOC equipped with a low-impedance MEA. To fabricate this device, the modular OOC platform known as Cytostretch is employed. This platform consists of freestanding and stretchable PDMS membrane fabricated on a silicon chip that can be equipped with different modules directly embedded in the membrane, such as pore arrays, strain gauges, microgrooves and MEAs. These features can be added through dedicated fabrication modules, to yield a chip with specific functionality for a particular OOC model as presented in [11]. The stretchable membrane itself is also an important feature to mechanically stimulate the cell culture [11].

In this work, instead of using only conventional electrode materials such as Pt, Titanium Nitride (TiN) and Gold (Au), known to provide high impedances, the Cytostretch MEA is coated with Poly(3,4-ethylenedioxythiophene) polystyrene sulfonate (PEDOT:PSS). This conductive organic polymer is best known for its low electrochemical impedance, high conductivity, transparency, biocompatibility and long-term stability in the biological environment [12]-[15]. These properties made PEDOT:PSS an attractive material choice in organic solar cells, light emitting devices, stretchable electronics and organic electrodes [7], [12]–[18]. By including PEDOT:PSS in the Cytostretch platform, a low-impedance PEDOT:PSS MEA is realized. This polymer guarantees sufficiently low noiseto-signal ratio during the detection of cellular action potential [7], a crucial requirement for several applications such as cardiotoxicity studies where the extracellular field potential generated by the cells' action potential needs to be studied in details to identify possible QT intervals prolongations often associated with safety-pharmacological risks [19]. Previous work has shown how PEDOT:PSS can be employed to bring the impedance from 330 k Ω down to 20 k Ω (at 1 kHz) on a commercially and non stretchable gold MEA [20]. This work is the first to focus on the integration of PEDOT:PSS MEA into a stretchable OOC device. Moreover, this work compares the resulting performance to more conventional electrode materials.

The use of PEDOT:PSS would also allow the integration of new sensors in the Cytostretch and the OOC field. PEDOT:PSS electrodes and in particular PEDOT:PSS based thin-film transistors have previously used to perform in vitro pH [21], humidity [22] and DNA [23] measurements, and to characterize the electrostatic interaction between the cells and the PEDOT:PSS surface [24].

The use of the Cytostretch platform for the development of this device, not only guarantees a successful integration of these PEDOT:PSS electrodes in an OOC, but also provides a conventional cleanroom-compatible micro-fabrication process, which avoids the often required labor-intensive fabrication steps used during the fabrication of previously presented OOCs.

In order to integrate the PEDOT:PSS electrodes in the Cytostretch, the fabrication process of the platform was improved by covering the TiN electrodes, previously developed by Pakazad *et al.* [25], with a Platinum (Pt) layer to allow the *in situ* electrochemical deposition of PEDOT:PSS [26]. The Pt electrodes were then coated with PEDOT:PSS and fully characterized with Scanning Electron Microscopy (SEM), Raman spectroscopy, Electrochemical Impedance Spectroscopy (EIS) and Cyclic Voltammetry (CV) in order to prove the PEDOT:PSS deposition and quantify the corresponding electrochemical impedance improvement.

II. DEVICE FABRICATION

The fabrication of the novel OOC equipped with lowimpedance PEDOT:PSS MEA, can be divided into two steps.

First, starting from the Cytostretch platform, an OOC with a Pt MEA is fabricated. The OOC consists of a cellculture environment in which a biological tissue grown and can be mechanically stimulated by inflating the stretchable PDMS membrane. The cells cultured in the OOC can be also monitored via the Pt electrodes directly embedded in the PDMS membrane. In order to fabricate this device, the wafer-scale microfabrication process previously presented by Pakazad *et al.* [25], [27] was modified and optimized. Second, the PEDOT:PSS is deposited directly on top of the Pt electrodes by electrochemical polymerization.

A. OOC Equipped With Platinum MEA

The wafer-scale fabrication of the new Cytostretch device with Pt-coated MEA is illustrated in Fig. 1. The process starts with the deposition of 2 and 6 μ m of silicon-dioxide (SiO₂) by plasma-enhanced chemical vapor deposition (PECVD) on the front and back of a 4" wafer, respectively. The SiO₂ layer on the back is patterned by dry-etching to define the membrane area. The process continues by sputtering a 1.5 μ m-thick aluminum (Al) layer on the front side of the wafer. The Al is then patterned by dry-etching to define the contact pads (Fig. 1(a)).

Next, a 100 nm-thick Pt layer, preceded by a 20 nm of Ti as adhesion layer, is evaporated on the wafer and patterned by lift-off to form the electrodes of the MEA (Fig.1(a)). In the next step, the interconnection lines extending from the contact pads to the MEA are fabricated. For this, a 800 nm-thick photosensitive polyimide (PI) (Fujifilm LTC 9305) layer is deposited by spin coating, patterned and cured at 350° C for 1 hour in low-pressure Nitrogen (N_2) atmosphere (Fig. 1(b)). Subsequently, a 200 nm of TiN, preceded by a 10 nm of Ti as adhesion layer, is sputtered on the front side of the wafer, and patterned by dry etching (Fig. 1(c)). A second layer of 800 nm-thick PI is deposited and patterned to provide electrical insulation for the metal lines (Fig. 1(d)). An oxygen plasma treatment (Tepla Plasma 3000 - Power: 600 W -Pressure: 250 ml/min - Time: 1 min) is first performed to promote the adhesion between the two PI layers [28]. Subsequently, a Polydimethylsiloxane (PDMS) layer is deposited by spin coating on the front side of the wafer, at a spin speed of 6000 rpm for 60 seconds and cured for 30 min at 90 °C, resulting in a layer thickness of 10 μ m. The adhesion of the PDMS to the PI was also improved by performing an



Fig. 1. Process flow for the fabrication of OOC with PEDOT:PSS MEA: (a) contact pads and Platinum (Pt) electrodes are fabricated on top of a $2 \mu m$ Silicon Oxide (SiO₂) layer. A 5 μm -thick SiO2 layer is deposited and patterned on the back of the wafer. (b) A 1 μm layer of Polyimide (PI) is deposited and patterned. (c) The 200 nm thick Titanium Nitride (TiN) metal lines are fabricated. (d) The second layer of PI is deposited and patterned to isolate the metal lines. (e) A Polydimethylsiloxane (PDMS) layer is deposited and patterned to access the contact pads. (f,g) The PDMS membrane is released by etching the Si under the PDMS layer making the electrodes accessible. (h) The chips is then loaded, glued and wirebonded on a PCB. The packaging of the device ends by covering the chips with a glass cylinder. (i) PEDOT:PSS is deposited on the Pt electrodes via electrochemical polymerization.

argon plasma treatment (Trikon Sigma 204 - Power 120 W - Pressure: 40 sccm - Time: 15 sec), as previously reported in [29]. The contact pads are then opened by patterning the PDMS layer by means of reactive ion etching using an Al layer as hard mask (Fig. 1(e)). Finally, the membrane is released by removing the Si and the SiO₂ layers underneath the membrane by means of deep reactive ion etching (DRIE) and buffered hydrofluoric acid (BHF), respectively (Fig. 1(g)).

The packaging procedure starts by dicing the 4-inch wafer with an automatic dicing saw. More than 100 Cytostretch chips can be obtained out of each wafer. Four chips are mounted on a PCB with a fully-automatic pick-and-place system. The PCB is designed to fit into a MultiChannel System *in vitro* recording devices (MEA2100-System). The Al contact pads on the chip are subsequently wire-bonded to the PCB. The packaging is then finished by mounting a glass cylinder on top of each dice on the PCB with epoxy glue (EPO-TEK 353ND-T) cured at 75 °C for 90 minutes at atmospheric pressure. This glass cylinder creates an open well structure on top of each chip, where the cells can be seeded. The PDMS membrane can be then stretched by connecting the OOC to an external system that applies pneumatic pressure to the back of the device.

In this work, this well is going to be used as the electrochemical cell for the PEDOT:PSS deposition on top of the Pt electrodes embedded in the stretchable membrane of the Cytostretch. The well can be also directly molded on top of the chips by an automated and monolithic molding technique as previously reported in [30].

B. PEDOT Electrochemical Deposition

The next step is to cover the Pt electrodes of the Cytostretch MEA with a PEDOT:PSS layer (Fig. 1(h)). The in-situ deposition of PEDOT:PSS coating was performed from an aqueous solution containing a mixture of 0.001 M of EDOT monomers, 0.1 M of Sodium Poly(Styrene Sulfonate) (NaPSS,

M_w: 70.000) as the dopant material as well as a surfactant, and 0.005 M of Tetrabutylammonium Tetrafluoroborate (TBABF₄) as secondary doping and to increase the conductivity of the solution. The electrochemical procedures in this study were performed with an Autolab Potentiostat (Metrohm), with a platinum strip and a miniaturized Ag/AgCl (supplied by EDAQ) as the counter and reference electrodes, respectively.

The cyclic voltammetry (CV) deposition method with a three electrodes cell setup [27], was utilized in order to increase the deposition rate by exploiting the capacitive current. In this procedure, the voltage on the working electrode is scanned over a 0.2 - 1.3 V (vs. Ag/AgCl) range with a scan rate of 0.5 V/s. The amount of charge delivered during the deposition is 3 μ C.

After each deposition, the MEA was drained of the remaining solution and rinsed with DI water. Afterwards, vacuum drying was performed at room temperature for 30 minutes to remove all liquid residues.

III. MEA CHARACTERIZATION PROCEDURE

The deposition of the PEDOT:PSS was verified by comparing the SEM images of the electrodes before and after the electrochemical deposition and by performing Raman spectroscopy on the electrodes. This analysis was performed with a Renishaw inVia Raman Spectroscope, with 488 nm wavelength and 300 to 1900 cm⁻¹ spectra coverage.

The performance of the MEA is investigated by means of electrochemical characterization, namely EIS to determine the impedance of the MEA over a certain frequency range and CV measurements to estimate the amount of charge that can be delivered within a certain voltage scan.

The improvement provided by the PEDOT:PSS was measured by characterizing the electrodes before and after the electrochemical deposition of the polymer. The performance of the Pt and PEDOT:PSS electrodes were also compared to



Fig. 2. (a) Optical image of one Cytostretch chips including the Pt MEA embedded in a stretchable PDMS membrane. The chip includes 12 Pt circular electrodes (diameter: $30 \ \mu$ m, pitch $100 \ \mu$ m). (b) SEM image of Pt MEA. (c,d) Optical image of the multi-well plate consisting of four Cytostretch chips mounted on a PCB before (c) and after (d) attaching the glass cylinders on the chips.



Fig. 3. (a) Optical image of one of the electrodes embedded in a stretchable PDMS membrane before (a) and after (b) the electrochemical deposition of PEDOT:PSS. (c) SEM image of the PEDOT:PSS electrode deposited with CV deposition, scanning the voltage on the working electrode between 0.2 - 1.3 V (vs. Ag/AgCl) with a scan rate of 0.5 V/s (PEDOT1). (d) SEM image of the PEDOT:PSS electrode deposited with CV deposition, scanning the voltage on the working electrode between 0.2 - 1.3 V (vs. Ag/AgCl) with a scan rate of 0.5 V/s (PEDOT1). (d) SEM image of the PEDOT:PSS electrode deposited with CV deposition, scanning the voltage on the working electrode between 0.2 - 1 V (vs. Ag/AgCl) with a scan rate of 0.5 V/s (PEDOT2).

standard TiN electrodes included in the previous version of the Cytostretch [11].

All the electrochemical characterization procedures were done with a three electrodes setup [27] in Phosphate Buffered Saline (PBS) medium (by Sigma Aldrich). The electrochemical characterization, i.e. EIS and CV measurements, was performed in a single run for each microelectrode type without changing the PBS medium.

The EIS measurements in this study were performed in the frequency range of 0.01 Hz – 10 kHz, AC signal with an amplitude of 50 mV (RMS). The CV measurements were performed over a voltage range of -0.6 V to 0.8 V, which also corresponds to the electrochemical window of water [31]. The starting potential was 0 V, with the scan rate of 0.5 V/s (forward scan first). Each measurement consisted of 5 repetitive scans.

IV. RESULTS AND DISCUSSION

Fig. 2 shows a successfully fabricated OOC device with Pt electrodes before being mounted on the PCB. The SEM image (top view) of one of the Platinum electrodes (with an area of 700 μ m²) embedded in the stretchable membrane is displayed in Fig.2b. The Ti layer in between the TiN the Pt layers appears to have successfully promote the adhesion, since

none of the fabricated electrodes indicated any Pt detachment after the membrane release, as confirmed by SEM analysis. The chips mounted and wirebonded on the PCB and the glass cylinders glued on the PCB are shown in Fig. 2(c) and (d), respectively.

The deposition of a PEDOT:PSS layer on top of the Pt electrodes was successfully achieved. The material was easily identified on the electrodes by the change of color observed on the Pt electrodes (Fig. 3(a,b)) in agreement with what previously reported [26], [32] and confirmed by SEM imaging. The SEM images of two PEDOT:PSS electrodes are shown in Fig. 3(d,e). The PEDOT:PSS was deposited only on the exposed electrodes, proving that the adhesion between the two layers of PI, was sufficient to prevent any current leakage during the deposition. The glass well and the PEDOT:PSS MEA at the bottom were rinsed 5 times with PBS using a pipette. No PEDOT:PSS peeling off was detected, proving the adhesion of the material to the electrodes.

The results of the Raman spectroscopy measurements on the PEDOT:PSS coated microelectrodes are shown in Fig. 4. The Raman spectrum presents a similar trend with the results observed by Kayinamura *et al.* [26] and Łapkowski and Proń [32]. The maximum peak appears at 1436 cm-1 that corresponds to the symmetric vibration of $C\alpha = C\beta$ bond in the oxidized polymer chain [26]. Several minor peaks at around



Fig. 4. Raman spectra of PEDOT:PSS grown on the Pt electrodes showing a similar trend with the results observed by Kayinamura *et al.* [26] and Łapkowski and Proń [32]. The maximum peak appears at 1436 cm⁻¹ and several minor peaks can be seen around 1249 cm⁻¹.

1249 cm-1 are also visible, which correspond to the $C\alpha$ — $C\alpha$ bond in the doped PEDOT:PSS polymer [32]. These results confirm the successful deposition of the PEDOT:PSS film.

During the CV deposition, charge response (Fig. 5(a)) and current response (Fig. 5(b)) were recorded. Only five of the CV cycles are reported in Fig. 5(b) (Cycle 01,20,40,60 and 80). As can be inferred by the current response (Fig. 5(a) in blue), the deposition rate of the PEDOT:PSS was not constant during the process. In fact, the charge delivered by the electrode during the deposition shows an abrupt increase in the first cycles. This phenomenon is due to the high current peaks caused by the PEDOT:PSS oxidation [33]-[35], appearing between 1 and 1.3 V in the current response graphs of the first few cycles. This can be seen in the first deposition (Cycle 01) reported in Fig. 5.(b). As previously presented by Kayinamura et al., [26] a high deposition rate can cause rapid movement of ions in the solution, which results in a less ordered polymer chain and eventually in a high impedance. In order to avoid these current peaks while minimizing the deposition time, the potential range was narrowed down to 0.2 - 1 V (vs. Ag/AgCl). The current response (Fig. 5(a) in red) and the charge delivery plot (Fig. 5(c)) after this optimization, confirmed that the deposition was in fact occurring at an almost constant deposition rate in the first few cycles, avoiding any current peaks. Moreover, the deposition rate was progressively increasing, due to the increasing current following the decrease of the electrode impedance.

The EIS and CV measurements results are shown in Fig. 6. A significant change was observed in both the impedance and the phase characteristics of the MEA after PEDOT:PSS deposition. The control TiN MEA and Pt MEA exhibit at 1 kHz a 1.3 ± 670 M Ω and 865 ± 35 k Ω impedance, respectively. The impedance decrease is different for each deposition condition. The deposition with potential range between 0.2 and 1.3 V (referred here as PEDOT1) yields a



Fig. 5. (a) Charge delivered through the electrode during of PEDOT1 (in blue) and PEDOT2 (in red) depositions. (b) Current response during cycle 01,20,40,60 and 80 for PEDOT1 deposition. (c) Current response during cycle 01,20,40,60 and 80 for PEDOT2 deposition.

small improvement in term of the electrochemical impedance (705 \pm 21 k Ω 1 kHz,), while the PEDOT:PSS deposited with a potential rate between 0.2 and 1 V (PEDOT2) significantly reduces the impedance, reaching a value of 59 \pm 2 k Ω , which is in the same range of PEDOT:PSS electrodes integrated in non stretchable surfaces [20]. This corresponds to a 95.5 % and a 93% of impedance reduction when compared to TiN and Pt, respectively. The phase of PEDOT2 showed also a more ohmic behavior for frequencies higher than 100 Hz.

When compared to TiN and Pt, the improvement of the electrochemical impedance and the change of phase characteristics obtained with the PEDOT:PSS coating are due to the inherent faradaic activities of the PEDOT:PSS film, attributed to the faradaic doping/undoping reactions, which allow ions to react at a faster rate [36], [37]. This value is even higher than the value reported in earlier studies [38], proving that a slow and ordered deposition of the PEDOT:PSS can improve even further the performances of this material.

The significant difference between PEDOT1 and PEDOT2 characteristics is mainly caused by the different deposition rate and consequently by the orderly structures among the polymer chains.

The CV curves from each of the microelectrodes types are shown in Fig. 6.(c). The CV of PEDOT:PSS coating, particularly PEDOT2, shows very large capacitive current, compared to TiN and Pt (Fig.6.(a)), both in the forward and the reverse scans. This is caused by the large double layer



Fig. 6. (a,b) Bode plot ((a) amplitude and (b) phase) of impedance spectra of a TiN electrode, a Pt electrode and two PEDOT:PSS electrodes (deposited with PEDOT1 and PEDOT2 programs) (30 μ m diameter). EIS at 1 kHz are equal to 9.32 M Ω , 875 k Ω , 720 k Ω and 58 k Ω for TiN, Pt, PEDOT1 and PEDOT2 electrodes, respectively. (c) CV measurement from four types of the microelectrodes combined in one graph.

TABLE I CHARGE DELIVERY CAPACITY (CDC)

Material	CDCa (mC/cm ²)	CDCc (mC/cm ²)
TiN	0.055	0.095
Pt	0.364	0.532
PEDOT:PSS	30.88	33.45
(PEDOT2)		

CDCa: Anodic Charge Delivery Capacity.

CDCc: Cathodic Charge Delivery Capacity

capacitance of the PEDOT:PSS coating. This proves once again that PEDOT:PSS provides a better performance than TiN and Pt in the Cytostretch MEA. Furthermore, the charge delivery capacity (CDC) value for each microelectrode type was derived from this measurement to qualitatively evaluate the microelectrode surface area. The values of CDC are calculated using (1), separately for the cathodic (below 0 A) and anodic scans and listed in Table. 1.

$$CDC = \frac{1}{v \times A} \int_{E_c}^{E_a} [I] \cdot dE \tag{1}$$

where *I* is the measured current (A), *v* is the scan rate (V/s), *A* is the area of the electrode (cm²), E_a and E_c are the peak anodic and cathodic potential, respectively [39]. The CDC values of PEDOT:PSS coated microelectrode are 3 and 2 orders of magnitude higher than those of the TiN and Pt microelectrode, respectively, proving once again the increase of the surface area of the microelectrode. This result confirms the feasibility of integrating PEDOT:PSS coated microelectrodes superior electrochemical performance compared to the more conventional Pt and TiN microelectrodes.

V. CONCLUSION

This work introduces the fabrication, characterization and optimization of the first stretchable OOC equipped with a PEDOT:PSS coated MEA. In order to fabricate this device, the modular OOC platform, known as Cytostretch, was employed as starting point.

The fabrication of the OOC was successfully improved by coating its TiN MEA with a Pt layer in order to guarantee the adhesion between the PEDOT:PSS and the electrodes. The optimized Cytostretch devices were then packaged and assembled on a PCB. After that, the conducting polymer was locally deposited on the electrodes. The deposition was verified by means of optical and SEM imaging. Moreover, the material was analyzed with Raman spectroscopy, and the results compared to previous work so to prove that the deposited material was indeed PEDOT:PSS.

The electrochemical performance of the MEA embedded in the novel device was characterized before and after the PEDOT:PSS deposition, and compared to the previous presented version of the Cytostretch (equipped with a TiN MEA). The PEDOT:PSS electrodes showed an electrochemical impedance reduction of 99.4 and 93.3 % when compared to TiN and Pt electrodes. Moreover, the CV characterization of PEDOT:PSS electrodes confirmed a higher capacitive current, compared to TiN and Pt. In fact, the CDC values for the PEDOT:PSS electrodes were 2 and 3 orders of magnitude higher that the Pt and TiN electrodes, respectively. The impedance might be further optimized by tuning thickness and deposition rate.

This work is a first step toward the integration of lowimpedance PEDOT:PSS MEA in an OOC, that could eventually allows low-noise and on-line monitoring of cells in these *in vitro* models and promote their use in cardiotoxicity tests during the preclinical phase of medicine development.

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