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Interfacial Properties of Cu/SiO₂ using a Multiscale Modelling Approach in Electronic Packages

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Abstract

Interfacial properties of Cu/SiO₂ in semiconductor devices has continued to be the subject of challenging study for many years because of its difficulties in experimentally quantifying the critical strength of interface. In this paper, a multi-scale modeling approach is built to characterize the interfacial properties between Cu and SiO₂. In this system, the Cu and SiO₂ are bonded together by three types of chemical bonds, Cu-OO, Cu-O, and Cu-Si, which cause three atomistic interfacial structures. For Cu-O and Cu-Si bonded interfaces, the fracture occurs exactly at the interface, however, the fracture for Cu-OO bonded interface occurs at copper layer near the interface, which indicate two different fracture criterions coexist in Cu/SiO₂ system. And, the calculated interfacial strength at macroscale is in agreement with available experimental results.

Introduction

In electronic packaging system, copper in particular, a common metal with high bulk thermal and electrical conductivities, and apparently low electromigration rate [1], is an excellent material for interconnects in integrated circuitry on silicon dioxide substrates. The interface strength and reliability between Cu and SiO₂ has been driving more and more attentions due to the brittle fracture often takes place at the Cu/SiO₂ interface [2] which is one of most easily damage part in electronics products. As we know, the interface is a complicate domain, where two separate materials are bonded together by chemical bonds with various bonding characteristics and adhesion strength. In the meantime, it is tremendous challenging for us to understand and predict the interfacial failure conditions due to the difficulties in experimentally quantifying the fracture energy of interfacial adhesion. And, the inner relationship between interfacial failure mechanisms at nano-level and interfacial strength at macro-level is still not clear for us, which seriously blocks the improvement of reliability for electronic devices.

In this paper, a multi-scale modelling approach was established to characterize the interface properties and simulate the Cu/SiO₂ interface failure under normal and shear loading.

Methodology

The proposed computational procedure of the multi-scale simulation is based on the following ansatz: (1) The micro-scale structure of interface at molecular level is obtained by first principles calculation with density functional theory. This method allows us to elucidate the

effect of local chemistry on the binding nature between copper and silicon dioxide substrate. (2) The obtained interfacial microstructures at first step are used to run MD simulation, the simulation results can indicate many important interfacial mechanical properties, including the traction-separation constitutive relation. (3) The simulation results from the previous step was then applied into the macroscopic scale via the cohesive zone model in Finite Element Method (FEM), the interfacial damage at Cu/SiO₂ interface was then simulated when subjected to shear and normal loading.

Fig. 1: Relaxed structures of the SiO₂ surfaces with different terminated atoms.

On the top surface of SiO₂ substrate, the different oxygen density causes different substrate terminations, which results in three types of SiO₂ surfaces: OO-terminated, O-terminated, and Si-terminated surfaces, as shown in Fig.1. Thus, in first-principles calculations, the oxygen density at the top of SiO₂ slab is changed to mimic different substrate terminations, the SiO₂ slab consists of eight layers in the [001] direction with a 2×2 surface unit cell, and the Cu slab consists of four [001] layers $\sqrt{2}\times\sqrt{2}$ surface unit cell. The first principles calculations are carried out within the local density approximation (LDA) as implemented in the plane-wave code Vienna ab initio Simulation Program (VASP), the ultrasoft pseudopotential is used for treatment of Cu and Si [3], and projector augmented wave potential is applied for O [4].

Fig. 2: Computational structure of the Cu-OO bonded

In MD simulation, the results show that there exist two different fracture mechanisms at Cu/SiO₂ system when subjected to tension loading. As shown in Fig.2.

Fig. 3: The interface stress-displacement curves and atomic structure of interfaces under tension loading, (a) Cu-Si bonded interface, (b) Cu-O bonded interface

For the Cu-O and Cu-Si bonded interfaces, the fracture occurs exactly at interface, as shown in Fig.3 (a) (b), and their stress-displacement relations indicate that the fracture mechanism of Cu-O and Cu-Si bonded interfaces is the breaking of bonds between Cu and O/Si atoms at the interface. However, the fracture of Cu-OO bonded interface does not happen exactly at the interface but Cu layer near the interface, as illustrated in Fig. 4, and the curve shows that the Cu layer goes through four stages: elastic deformation, yield stage, strengthening phase, and accelerated fracture stage. It is a typical stress-displacement curve for metallic materials, which indicates the fracture mechanism of Cu-OO bonded interface is the failure of copper material itself, because the bonding strength at the Cu-OO interface is stronger than that in copper material itself. Furthermore, the fracture energy of Cu-OO bonded interface is 1.32 J/m², which is much larger than the fracture energy of Cu-O and Cu-Si bonded interfaces, just seen in Table.1. Such differences indicate the interfacial oxygen density is very important factor influencing the strength of interfacial adhesion.

Table 1: Calculated maximum traction (N_{max}), initial interface stiffness (K₀), and fracture energy (G_c) from MD simulation.

Type of interface	K ₀ (10 ³ Mpa/nm)	N _{max} (Mpa)	G(J/m ²)
Cu/SiO ₂ :OO	39.1	9620	1.32
Cu/SiO ₂ :O	1.75	1623	0.446
Cu/SiO ₂ :Si	3.73	1552	0.271

Fig. 4: The interface stress-displacement curves and atomic structure of Cu-OO bonded interfaces under tension loading.

In finite element simulation, first, a 3D Finite Element (FE) model in Abaqus was constructed to calculate the overall traction-separation constitutive relation of such multi-interactive interface. Since, we found that the Cu/SiO₂ interface actually is composed by three types of secondary interfaces, as it illustrated in Fig. 5, and approximate 15% and 16% area at the interface is bonded by Cu-O and Cu-Si bonds respectively [6]. Thus, the

Fig. 5: The side and top view of multi-interactive interface structure, the lateral and normal loads are applied on the upper pad, respectively.

sample consists of four Cu-OO interface layer models, four Cu-O layer models, eight Cu-Si layer models and two rigid pads. Those interface layers are modeled by cohesive zone element [7]. After we obtain the overall interface traction-separation constitutive relation, the obtained results can be directly employed to simulate the interfacial delamination test of the Cu/SiO₂ system [8], as is illustrated in Fig.6. According to the geometric and loading configurations of this experiment test [7], the FE model is built, a columnar Cu dot of 50 nm height and 300 nm across on a SiO₂ substrate with a rigid-layer W of 50 nm, and the 4-node axisymmetric elements, CAX4 and COHAX4, are used for the substructures and interfaces respectively.

Fig. 6: Explanation of delamination test for Cu dot on SiO₂ substrate with lateral loading.

Results and discussion

Atomistic interfacial structure

The atomistic structure of Cu-OO, Cu-O and Cu-Si bonded interfaces are shown in Fig.7. For Cu-OO bonded interface, the angle of O-Si-O is changed from 79° to 110°, and the electrons carried by Cu atoms go from 0 to 0.03e, those large rearrangement would indicate bond

formation between Cu and O atoms. The similar changes of atomistic structure and charges transition also can be found in Cu-O and Cu-Si bonded interface, but the magnitude of rearrangements at Cu-O and Cu-Si bonded interface are less than that at the Cu-OO bonded interface. Moreover, the relaxed bond length of Cu-O at Cu-OO and Cu-O bonded interfaces are $\sim 1.962\text{\AA}$ and $\sim 1.963\text{\AA}$, respectively, and bond length of Si-Cu is $\sim 2.269\text{\AA}$.

Fracture mechanisms

The MD simulation results show that there exist two different fracture mechanisms at Cu/SiO₂ system when subjected to tension loading. For the Cu-O and Cu-Si bonded interfaces, the fracture occurs exactly at interface, as shown in Fig.3 (a) (b), and their stress-displacement relations indicate that the fracture mechanism of Cu-O and Cu-Si bonded interfaces is the breaking of bonds between Cu and O/Si atoms at the interface. However, the fracture of Cu-OO bonded interface does not happen exactly at the interface but Cu layer near the interface, as illustrated in Fig. 4 (a), and the curve shows that the Cu layer goes through four stages: elastic deformation, yield stage, strengthening phase, and accelerated fracture stage. It is a typical stress-displacement curve for metallic materials, which indicates the fracture mechanism of Cu-OO bonded interface is the failure of copper material itself, because the bonding strength at the Cu-OO interface is stronger than that in copper material itself. Furthermore, the fracture energy of Cu-OO bonded interface is 1.32 J/m^2 , which is much larger than the fracture energy of Cu-O and Cu-Si bonded interfaces, just seen in Table.2. Such differences indicate the interfacial oxygen density is very important factor influencing the strength of interfacial adhesion.

Macroscopic interfacial strength

The overall traction-separation constitutive relations are shown in Fig.8, the maximum traction $N_{\max}$ is 820Mpa , the initial interface stiffness $K_0=21.5\text{Gpa/nm}$, and fracture energy $G=0.228\text{J/m}^2$. Then, the approximate parameters of stress-displacement relation are applied in the simulation of delamination test.

Fig. 8 (a) The stress contour of delamination test with lateral loading at the displacement of $1.5\text{ nm}/2.5\text{ nm}$, the right graph is the interface stress distribution of Cu dot; (b) the lateral load-displacement curve and the corresponding damage dissipation energy.

Fig. 7: The interfacial strength of the Cu/SiO₂ interfaces with different direction and approximate curve.

In FE simulation, it is observed that some of the cohesive elements have been removed at the lateral displacement of $\sim 2.3\text{ nm}$, which indicates that the onset of the interface damage, as shown in Fig. 8 (a). When the displacement increases to 3.2 nm , the lateral loading decreases to zero, the Cu dot is totally separated from substrate. In this loading process, the maximum lateral force is $\sim 30\mu\text{N}$ at lateral displacement of $\sim 2.8\text{ nm}$, as shown in Fig. 8(b), which is in good consistent with the experimental result [8]. Furthermore, the FE model is used to calculate the normal interface strength of Cu/SiO₂ system, as shown in Fig. 9. The results show that the Cu dot is pulled out from the SiO₂ substrate under $F=52\mu\text{N}$. Thus, the interfacial strengths for Cu/SiO₂ are predicted as $\sim 400\text{Mpa}$ and $\sim 700\text{Mpa}$ in normal and shear directions, respectively.

Conclusion

A multiscale simulation approach is developed to characterize interface properties in Cu/SiO₂ component. The results show the interfacial oxygen density is very important factor influencing the strength of adhesion, the adhesion strength of oxygen-rich interface is larger than that of the low-oxygen density interface. Such interfacial properties cause two different fracture mechanisms in Cu/SiO₂ system. For Cu-O and Cu-Si bonded interface, the fracture happens exactly at the interface, however, the fracture for Cu-OO bonded interface occurs at copper layer near the interface, which, in turns, affects the overall structure of the Cu/SiO₂ system and, ultimately, its macroscopic property.

Fig. 9: (a) The stress contour of delamination test with normal loading at the displacement of 0.4 nm/ 0.7nm, the right graph is the interface stress distribution of Cu dot; (b) the normal load-displacement curve and the corresponding damage dissipation energy graph

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References

- [1]. Murarka S P, Verner I V, Gutmann R J. "Copper-fundamental mechanisms for microelectronic applications. Wiley-Inter science," ISBN: 0471252565 (2000)
- [2]. P. Lin, R.I. Babicheva, M. Xue, H.S. Zhang, H. Xu, B. Liu, K. Zhou, "Effects of temperature and voids on the interface fracture of Si/a-Si₃N₄ bilayer systems," Phys. Status Solid B, 2015, Vol. 252, No. 9, 2013-2019.
- [3]. D. Vanderbilt, "Soft self-consistent pseudo potential in a generalized eigenvalue formalism," Phys. Rev. B, 1990, Vol. 41, No. 11, 7892, doi: 10.1103/PhysRevB.41.7892
- [4]. P.E. Blochl, "Projector augmented-wave method," Phys. Rev. B, 1994, Vol. 50, No. 17, 953.
- [5]. BIOVIA, materials studio modeling and simulation software, version 2017, Dassault Systemes, San Diego, California, USA, 2017.
- [6]. Kazutaka Nagao, J. B. Neaton, and N. W. Ashcroft, "First-principles study of adhesion at Cu/SiO₂ interfaces," Physical Review B, 2003.
- [7]. A. Needleman, "A continuum model for void nucleation by inclusion debonding," J.Appl. Phys. 1987, Vol. 54, No. 3, 525-531

- [8]. Hiroyuki Hirakata, takayuki Kitamura, Yoshitake Yammamoto, "Direct measurement of interface strength between copper submicron-dot and silicon dioxide substrate," JSME international journal, Serial A, 2004, Vol. 47, No. 3, 324-330.